THIADIAZOLEDIOXIDES AND THIADIAZOLEOXIDES AS CXC- AND CC- CHEMOKINE RECEPTOR LIGANDS

REFERENCE TO RELATED APPLICATION

This Application claims the benefit of US Provisional Application Serial No. 60/417,371 filed October 9, 2002.

FIELD OF THE INVENTION

5

10

15

20

25

30

The present invention relates to novel substituted thiadiazoledioxide and thiadiazolemonooxide compounds, pharmaceutical compositions containing the compounds, and the use of the compounds and formulations in treating CXC and CC-chemokine-mediated diseases.

BACKGROUND OF THE INVENTION

Chemokines are chemotactic cytokines that are released by a wide variety of cells to attract macrophages, T-cells, eosinophils, basophils, neutrophils and endothelial cells to sites of inflammation and tumor growth. There are two main classes of chemokines, the CXC-chemokines and the CC- chemokines. The class depends on whether the first two cysteines are separated by a single amino acid (CXC-chemokines) or are adjacent (CC-chemokines). The CXC-chemokines include, but are not limited to, interleukin-8 (IL-8), neutrophil-activating protein-1 (NAP-1), neutrophil-activating protein-2 (NAP-2), GRO α , GRO β , GRO γ , ENA-78, GCP-2, IP-10, MIG and PF4. CC chemokines include, but are not limited to, RANTES, MIP -1 α , MIP-2 β , monocyte chemotactic protein-1 (MCP-1), MCP-2, MCP-3, CCL19, CCL21 and eotaxin. Individual members of the chemokine families are known to be bound by at least one chemokine receptor, with CXC-chemokines generally bound by members of the CXCR class of receptors, and CC-chemokines by members of the CCR class of receptors. For example, IL-8 is bound by the CXCR-1 and CXCR-2 receptors.

Since CXC-chemokines promote the accumulation and activation of neutrophils, these chemokines have been implicated in a wide range of acute and chronic inflammatory disorders including psoriasis and rheumatoid arthritis. Baggiolini et al., FEBS Lett. 307, 97 (1992); Miller et al., Crit. Rev. Immunol. 12, 17 (1992); Oppenheim et al., Annu. Fev. Immunol. 9, 617 (1991); Seitz et al., J. Clin. Invest. 87,

463 (1991); Miller et al., Am. Rev. Respir. Dis. 146, 427 (1992); Donnely et al., Lancet 341, 643 (1993).

ELRCXC chemokines including IL-8, GROα, GROβ, GROγ, NAP-2, and ENA-78 (Strieter et al. 1995 JBC 270 p. 27348-57) have also been implicated in the induction of tumor angiogenesis (new blood vessel growth). All of these chemokines are believed to exert their actions by binding to the 7 transmembrane G-protein coupled receptor CXCR2 (also known as IL-8RB), while IL-8 also binds CXCR1 (also known as IL-8RA). Thus, their angiogenic activity is due to their binding to and activation of CXCR2, and possible CXCR1 for IL-8, expressed on the surface of vascular endothelial cells (ECs) in surrounding vessels.

5

10

15

20

25

30

Many different types of tumors have been shown to produce ELRCXC chemokines and their production has been correlated with a more aggressive phenotype (Inoue et al. 2000 Clin Cancer Res 6 p. 2104-2119) and poor prognosis (Yoneda et. al. 1998 J Nat Cancer Inst 90 p. 447-454). Chemokines are potent chemotactic factors and the ELRCXC chemokines have been shown to induce EC chemotaxis. Thus, these chemokines probably induce chemotaxis of endothelial cells toward their site of production in the tumor. This may be a critical step in the induction of angiogenesis by the tumor. Inhibitors of CXCR2 or dual inhibitors of CXCR2 and CXCR1 will inhibit the angiogenic activity of the ELRCXC chemokines and therefore block the growth of the tumor. This anti-tumor activity has been demonstrated for antibodies to IL-8 (Arenberg et al. 1996 J Clin Invest 97 p. 2792-2802), ENA-78 (Arenberg et al. 1998 J Clin Invest 102 p. 465-72), and GROα (Haghnegahdar et al. J. Leukoc Biology 2000 67 p. 53-62).

Many tumor cells have also been shown to express CXCR2 and thus tumor cells may also stimulate their own growth when they secrete ELRCXC chemokines. Thus, along with decreasing angiogenesis, inhibitors of CXCR2 may directly inhibit the growth of tumor cells.

Hence, the CXC-chemokine receptors represent promising targets for the development of novel anti-inflammatory and anti-tumor agents.

There remains a need for compounds that are capable of modulating activity at CXC-chemokine receptors. For example, conditions associated with an increase in IL-8 production (which is responsible for chemotaxis of neutrophil and T-cell subsets

into the inflammatory site and growth of tumors) would benefit by compounds that are inhibitors of IL-8 receptor binding.

SUMMARY OF THE INVENTION

5

10

15

20

25

This invention provides novel compounds of formula IA:

and the pharmaceutically acceptable salts (e.g., sodium or calcium) and solvates thereof, wherein A and B are defined below.

This invention also provides a method of treating a chemokine mediated disease in a patient in need of such treatment comprising administering to said patient an effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating a CXCR1 and/or CXCR2 mediated disease in a patient in need of such treatment comprising administering to said patient an effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating a CCR7 mediated disease in a patient in need of such treatment comprising administering to said patient an effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating cancer in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable sait or solvate thereof.

This invention also provides a method of treating Kaposi's sarcoma, melanoma, gastric carcinoma, and non-small cell carcinoma in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating melanoma, gastric carcinoma, and non-small cell carcinoma in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating cancer in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one anticancer agent selected from the group consisting of: (a) microtubule affecting agents, (b) antineoplastic agents, (c) anti-angiogenesis agents, or (d) VEGF receptor kinase inhibitors, (e) antibodies against the VEGF receptor, (f) interferon, and g) radiation. The compound of formula IA can be administered concurrently or sequentially with the anticancer agent.

This invention also provides a method of treating cancer in a patient in need of such treatment comprising administering to said patient at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one (usually 1) antineoplastic agent selected from the group consisting of: gemcitabine, paclitaxel (Taxol®), 5-Fluorourcil (5-FU), cyclophosphamide (Cytoxan®), temozolomide, and Vincristine.

This invention also provides a method of treating cancer in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, concurrently or sequentially with microtubule affecting agent, e.g., paclitaxel.

This invention also provides a method treating cancer in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of: (a) at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, concurrently or sequentially with (b) at least one (usually 1) agent selected from the group consisting of: (1) antineoplastic agents, (2) microtubule affecting agents, and (3) anti-angiogenesis agents.

This invention also provides a method of inhibiting angiogenesis in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

30

5

10

15

20

This invention also provides a method of treating angiogenic ocular disease (e.g., ocular inflammation, retinopathy of prematurity, diabetic retinopathy, macular degeneration with the wet type preferred and corneal neovascularization) in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

5

10

15

20

25

30

This invention also provides a method of treating a chemokine mediated (e.g., CXCR1 and/or CXCR2, or CCR7) disease or condition selected from the group consisting of: acute pain, acute inflammation, chronic inflammation, rheumatoid arthritis, acute inflammatory pain, chronic inflammatory pain, neuropathic pain, psoriasis, atopic dermatitis, asthma, COPD, adult respiratory disease, arthritis, inflammatory bowel disease, Crohn's disease, ulcerative colitis, septic shock, endotoxic shock, gram negative sepsis, toxic shock syndrome, stroke, cardiac reperfusion injury, renal reperfusion injury, glomerulonephritis, thrombosis, Alzheimer's disease, graft vs. host reaction (i.e., graft vs. host disease), allograft rejections (e.g., acute allograft rejection, and chronic allograft rejection), malaria, acute respiratory distress syndrome, delayed type hypersensitivity reaction, atherosclerosis, cerebral ischemia, cardiac ischemia, osteoarthritis, multiple sclerosis, restinosis, angiogenesis, osteoporosis, gingivitis, respiratory viruses, herpes viruses, hepatitis viruses, HIV, Kaposi's sarcoma associated virus (i.e., Kaposi's sarcoma), meningitis, cystic fibrosis, pre-term labor, cough, pruritis, multi-organ dysfunction, trauma, strains, sprains, contusions, psoriatic arthritis, herpes, encephalitis, CNS vasculitis, traumatic brain injury, CNS tumors, subarachnoid hemorrhage, post surgical trauma, interstitial pneumonitis, hypersensitivity, crystal induced arthritis, acute pancreatitis, chronic pancreatitis, acute alcoholic hepatitis, necrotizing enterocolitis, chronic sinusitis, angiogenic ocular disease, ocular inflammation, retinopathy of prematurity, diabetic retinopathy, macular degeneration with the wet type preferred, corneal neovascularization, polymyositis, vasculitis, acne, gastric ulcers, duodenal ulcers, celiac disease, esophagitis, glossitis, airflow obstruction, airway hyperresponsiveness (i.e., airway hyperreactivity), bronchiectasis, bronchiolitis, bronchiolitis obliterans (i.e., bronchiolitis obliterans syndrome), chronic bronchitis, cor pulmonae, dyspnea, emphysema, hypercapnea, hyperinflation, hypoxemia, hyperoxiainduced inflammations, hypoxia, surgical lung volume reduction, pulmonary fibrosis,

10

15

20

25

30

pulmonary hypertension, right ventricular hypertrophy, peritonitis associated with continuous ambulatory peritoneal dialysis (CAPD), granulocytic ehrlichiosis, sarcoidosis, small airway disease, ventilation-perfusion mismatching, wheeze, colds, gout, alcoholic liver disease, lupus, burn therapy (i.e., the treatment of burns), periodontitis, cancer, transplant reperfusion injury, early transplantation rejection (e.g., acute allograft rejection), airway hyperreactivity, allergic contact dermatitis, allergic rhinitis, alopecia areata, antiphospholipid syndromes, aplastic anemia, autoimmune deafness (including, for example, Meniere's disease), autoimmune hemolytic syndromes, autoimmune hepatitis, autoimmune neuropathy, autoimmune ovarian failure, autoimmune orchitis, autoimmune thrombocytopenia, bullous pemphigoid, chronic allograft vasculopathy, chronic inflammatory demyelinating polyneuropathy, cirrhosis, cor pneumoniae, cryoglobulinemia, dermatomyositis, diabetes, drug-induced autoimmunity, epidermolysis bullosa acquisita, endometriosis, fibrotic diseases, gastritis, Goodpasture's syndrome, Graves' disease, Gullain-Barre disease, Hashimoto's thyroiditis, hepatitis-associated autoimmunity, HIV-related autoimmune syndromes and hematologic disorders, hypophytis, idiopathic thrombocytic pupura, interstitial cystitis, juvenile arthritis, Langerhans' cell histiocytitis, lichen planus, metalinduced autoimmunity, myasthenia gravis, myelodysplastic syndromes, myocarditis (including viral myocarditis), myositis, Neuropathies (including, for example, IgA neuropathy, membranous neuropathy and idiopathic neuropathy), nephritic syndrome, optic neuritis, pancreatitis, paroxysmal nocturnal hemoglobulinemia, pemphigus, polymyalgia, post-infectious autoimmunity, primary biliary cirrhosis, reactive arthritis, ankylosing spondylitis, Raynaud's phenomenon, Reiter's syndrome, reperfusion injury, scleritis, scleroderma, secondary hematologic manifestation of autoimmune diseases (such as, for example, anemias), silicone implant associated autoimmune disease, Sjogren's syndrome, systemic lupus erythematosus, thrombocytopenia, transverse myelitis, tubulointerstitial nephritis, uveitis, vasculitis syndromes (such as, for example, giant cell arteritis, Behcet's disease and Wegener's granulomatosis), and Vitiligo in a patient in need of such treatment comprising administering to said patient an effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating a CXCR1 and/or a CXCR2 mediated disease or condition selected from the group consisting of: acute pain, acute

10

15

20

25

30

inflammation, chronic inflammation, rheumatoid arthritis, acute inflammatory pain, chronic inflammatory pain, neuropathic pain, psoriasis, atopic dermatitis, asthma, COPD, adult respiratory disease, arthritis, inflammatory bowel disease, Crohn's disease, ulcerative colitis, septic shock, endotoxic shock, gram negative sepsis, toxic shock syndrome, stroke, cardiac reperfusion injury, renal reperfusion injury, glomerulonephritis, thrombosis, Alzheimer's disease, graft vs. host reaction (i.e., graft vs. host disease), allograft rejections (e.g., acute allograft rejection, and chronic allograft rejection), malaria, acute respiratory distress syndrome, delayed type hypersensitivity reaction, atherosclerosis, cerebral ischemia, cardiac ischemia, osteoarthritis, multiple sclerosis, restinosis, angiogenesis, osteoporosis, gingivitis, respiratory viruses, herpes viruses, hepatitis viruses, HIV, Kaposi's sarcoma associated virus (i.e., Kaposi's sarcoma), meningitis, cystic fibrosis, pre-term labor, cough, pruritis, multi-organ dysfunction, trauma, strains, sprains, contusions, psoriatic arthritis, herpes, encephalitis, CNS vasculitis, traumatic brain injury, CNS tumors, subarachnoid hemorrhage, post surgical trauma, interstitial pneumonitis, hypersensitivity, crystal induced arthritis, acute pancreatitis, chronic pancreatitis, acute alcoholic hepatitis, necrotizing enterocolitis, chronic sinusitis, angiogenic ocular disease, ocular inflammation, retinopathy of prematurity, diabetic retinopathy, macular degeneration with the wet type preferred, corneal neovascularization, polymyositis, vasculitis, acne, gastric ulcers, duodenal ulcers, celiac disease, esophagitis, glossitis, airflow obstruction, airway hyperresponsiveness (i.e., airway hyperreactivity), bronchiectasis, bronchiolitis, bronchiolitis obliterans, chronic bronchitis, cor pulmonae, dyspnea, emphysema, hypercapnea, hyperinflation, hypoxemia, hyperoxia-induced inflammations, hypoxia, surgical lung volume reduction, pulmonary fibrosis, pulmonary hypertension, right ventricular hypertrophy, peritonitis associated with continuous ambulatory peritoneal dialysis (CAPD), granulocytic ehrlichiosis, sarcoidosis, small airway disease, ventilation-perfusion mismatching, wheeze, colds, gout, alcoholic liver disease, lupus, burn therapy (i.e., the treatment of burns), periodontitis, cancer, transplant reperfusion injury, early transplantation rejection (e.g., acute allograft rejection) in a patient in need of such treatment comprising administering to said patient an effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

10

15

20

25

30

This invention also provides a method of treating a CCR7 mediated disease or condition selected from the group consisting of: acute inflammation, chronic inflammation, acute inflammatory pain, acute pain, chronic inflammatory pain, neuropathic pain, acute allograft rejection, acute respiratory distress syndrome, adult respiratory disease, airway hyperreactivity, allergic contact dermatitis, allergic rhinitis, alopecia areata, alzheimer's disease, angiogenic ocular disease, antiphospholipid syndromes, aplastic anemia, asthma, atherosclerosis, atopic dermatitis, autoimmune deafness (including, for example, Meniere's disease), autoimmune hemolytic syndromes, autoimmune hepatitis, autoimmune neuropathy, autoimmune ovarian failure, autoimmune orchitis, autoimmune thrombocytopenia, bronchiolitis, bronchiolitis obliterans syndrome, bullous pemphigoid, burn therapy (i.e., the treatment of burns), cancer, cerebral ischemia, cardiac ischemia, chronic allograft rejection, chronic allograft vasculopathy, chronic bronchitis, chronic inflammatory demyelinating polyneuropathy, chronic sinusitis, cirrhosis, CNS vasculitis, COPD, Cor pneumoniae. Crohn's disease, cryoglobulinemia, crystal-induced arthritis, delayed-type hypersensitivity reactions, dermatomyositis, diabetes, diabetic retinopathy, druginduced autoimmunity, dyspnea, emphysema, epidermolysis bullosa acquisita. endometriosis, fibrotic diseases, gastritis, glomerulonephritis, Goodpasture's syndrome, graft vs host disease, Graves' disease, Gullain-Barre disease, Hashimoto's thyroiditis, hepatitis-associated autoimmunity, HIV-related autoimmune syndromes and hematologic disorders, hyperoxia-induced inflammation, hypercapnea, hyperinflation, hypophytis, hypoxia, idiopathic thrombocytic pupura, inflammatory bowel diseases, interstitial cystitis, interstitial pneumonitis, juvenile arthritis, Langerhans' cell histiocytitis, lichen planus, metal-induced autoimmunity, multiple sclerosis, myasthenia gravis, myelodysplastic syndromes, myocarditis including viral myocarditis, myositis, neuropathies (including, for example, IgA neuropathy, membranous neuropathy and idiopathic neuropathy), nephritic syndrome, ocular inflammation, optic neuritis, osteoarthritis, pancreatitis, paroxysmal nocturnal hemoglobulinemia, pemphigus, polymyalgia, polymyositis, post-infectious autoimmunity, pulmonary fibrosis, primary biliary cirrhosis, psoriasis, pruritis, rheumatoid arthritis, reactive arthritis, ankylosing spondylitis, psoriatic arthritis. Raynaud's phenomenon, Reiter's syndrome, reperfusion injury, restenosis. sarcoidosis, scleritis, scleroderma, secondary hematologic manifestation of

autoimmune diseases (such as, for example, anemias), silicone implant associated autoimmune disease, Sjogren's syndrome, systemic lupus erythematosus, thrombocytopenia, thrombosis, transverse myelitis, tubulointerstitial nephritis, ulcerative colitis, uveitis, vasculitis and vasculitis syndromes (such as, for example, giant cell arteritis, Behcet's disease and Wegener's granulomatosis), and vitiligo in a patient in need of such treatment comprising administering to said patient an effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating a chemokine (e.g., a CXC, or a CC chemokine) mediated disease in a patient in need of such treatment comprising administering to said patient at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one (usually 1) other medicament (e.g., a drug, agent or therapeutic) useful for the treatment of chemokine mediated diseases.

This invention also provides a method of treating a chemokine mediated disease in a patient in need of such treatment comprising administering to said patient at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one (usually 1) other medicament (e.g., a drug, agent or therapeutic) selected from the group consisting of:

20

5

10

15

- a) disease modifying antirheumatic drugs;
- b) nonsteroidal anitinflammatory drugs;
- c) COX-2 selective inhibitors;
- d) COX-1 inhibitors;
- e) immunosuppressives;

25

- f) steroids;
- g) biological response modifiers; and
- h) other anti-inflammatory agents or therapeutics useful for the treatment of chemokine mediated diseases.

This invention also provides a method of treating a pulmonary disease (e.g., COPD, asthma or cystic fibrosis) in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one (usually 1) compound selected from the

group consisting of: glucocorticoids, 5-lipoxygenase inhibitors, β-2 adrenoceptor agonists, muscarinic M1 antagonists, muscarinic M3 antagonists, muscarinic M2 agonists, NK3 antagonists, LTB4 antagonists, cysteinyl leukotriene antagonists, bronchodilators, PDE4 inhibitors, PDE inhibitors, elastase inhibitors, MMP inhibitors, phospholipase A2 inhibitors, phospholipase D inhibitors, histamine H1 antagonists, histamine H3 antagonists, dopamine agonists, adenosine A2 agonists, NK1 and NK2 antagonists, GABA-b agonists, nociceptin agonists, expectorants, mucolytic agents, decongestants, antioxidants, anti-IL-8 anti-bodies, anti-IL-5 antibodies, anti-IgE antibodies, anti-TNF antibodies, IL-10, adhesion molecule inhibitors, and growth hormones.

5

10

15

20

25

30

This invention also provides a method of treating multiple sclerosis in a patient in need of such treatment comprising administering to said patient, a therapeutically effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one compound selected from the group consisting of glatiramer acetate, glucocorticoids, methotrexate, azothioprine, mitoxantrone, chemokine inhibitors, and CB2-selective inhibitors.

This invention also provides a method of treating multiple sclerosis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one compound selected from the group consisting of: methotrexate, cyclosporin, leflunimide, sulfasalazine, β -methasone, β -interferon, glatiramer acetate, prednisone, etonercept, and infliximab.

This invention also provides a method of treating rheumatoid arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating rheumatoid arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one compound selected from the group consisting of COX-2 inhibitors, COX inhibitors,

immunosuppressives (e.g., methotrexate, cyclosporin, leflunimide and sulfasalazine), steroids (e.g., betamethasone, cortisone and dexamethasone), PDE IV inhibitors, anti-TNF-α compounds, MMP inhibitors, glucocorticoids, chemokine inhibitors, CB2-selective inhibitors, and other classes of compounds indicated for the treatment of rheumatoid arthritis.

5

10

15

20

25

30

This invention also provides a method of treating stroke and cardiac reperfusion injury in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one compound (usually 1) of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one compound selected from the group consisting of thrombolitics (e.g., tenecteplase, TPA, alteplase), antiplatelet agents (e.g., gpllb/IIIa), antagonists (e.g., abciximab and eftiifbatide), anticoagulants (e.g., heparin), and other compounds indicated for the treatment of rheumatoid arthritis.

This invention also provides a method of treating stroke and cardiac reperfusion injury in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one compound selected from the group consisting of tenecteplase, TPA, alteplase, abciximab, eftiifbatide, and heparin.

This invention also provides a method of treating psoriasis in a patient in need of such treatment comprising administering to said patient a thereapeutically effective amount of at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, in combination with at least one compound selected from the group consisting of immunosuppressives (e.g., methotrexate, cyclosporin, leflunimide and sulfasalazine), steroids (e.g., β -methasone) and anti-TNF- α compounds (e.g., etonercept and infliximab).

This invention also provides a method of treating COPD in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating acute pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective

amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating acute inflammatory pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating chronic inflammatory pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating neropathic pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a method of treating osteoarthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

This invention also provides a pharmaceutical composition comprising at least one (e.g., 1-3, usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, and a pharmaceutically acceptable carrier.

This invention also provides a pharmaceutical composition comprising at least one (e.g., 1-3, usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, and at least one (e.g., 1-3, usually 1) other agent, medicament, antibody and/or inhibitor disclosed above, and a pharmaceutically acceptable carrier.

DETAILED DESCRIPTION OF THE INVENTION

When any variable occurs more than one time in any moiety, its definition on each occurrence is independent of its definition at every other occurrence. Also,

30

5

10

15

20

combinations of substituents and/or variables are permissible only if such combinations result in stable compounds.

Unless indicated otherwise, the following definitions apply throughout the present specification and claims. These definitions apply regardless of whether a term is used by itself or in combination with other terms. For example, the definition of "alkyl" also applies to the "alkyl" portion of "alkoxy".

"An effective amount" means a therapeutically acceptable amount (i.e., that amount which provides the desired therapeutic effective).

"At least one" means one or more (e.g., 1-3, 1-2, or 1).

"Bu" represents butyl.

5

10

15

20

25

30

"Bn" represents benzyl.

"Composition" includes a product comprising the specified ingredients in the specified amounts, as well as any product that results, directly or indirectly, from combination of the specified ingredients in the specified amounts.

"Et" represents ethyl.

"In combination with" as used to describe the administration of a compound of formula IA with other medicaments in the methods of treatment of this invention, means that the compounds of formula IA and the other medicaments are administered sequentially or concurrently in separate dosage forms, or are administered concurrently in the same dosage form.

"Mammal" includes a human being, and preferably means a human being.

"Patient" includes both human and other mammals, preferably human.

"Ph", as used in the structures herein, represents phenyl.

"Pr" represents propyl.

"Prodrug" represents compounds that are rapidly transformed *in vivo* to the parent compound of the above formula, for example, by hydrolysis in blood. A thorough discussion is provided in T. Higuchi and V. Stella, Pro-drugs as Novel Delivery Systems, Vol. 14 of the A.C.S. Symposium Series, and in Edward B. Roche, ed., Bioreversible Carriers in Drug Design, American Pharmaceutical Association and Pergamon Press, 1987, both of which are incorporated herein by reference.

"Alkyl" means a straight or branched saturated hydrocarbon chain having 1 to 20 carbon atoms, preferably 1 to 12 carbon atoms, more preferably 1 to 6 carbon atoms.

"Alkoxy" means an alkyl-O- group wherein alkyl is as defined above. Nonlimiting examples of alkoxy groups include: methoxy, ethoxy, n-propoxy, iso-propoxy and n-butoxy. The bond to the parent moiety is through the ether oxygen.

"Alkenyl" means a straight or branched aliphatic hydrocarbon group having at least one carbon-carbon double bond, and 2 to 20 carbon atoms, preferably 2 to 12 carbon atoms, and more preferably 2 to 6 carbon atoms. Non-limiting examples of alkenyl groups include: ethenyl, propenyl, n-butenyl, 3-methylbut-2-enyl, n-pentenyl, octenyl and decenyl.

"Alkynyl" means a straight or branched aliphatic hydrocarbon group having at least one carbon-carbon triple bond, and 2 to 15 carbon atoms, preferably 2 to 12 carbon atoms, and more preferably 2 to 4 carbon atoms. Non-limiting examples of alkynyl groups include ethynyl, propynyl, 2-butynyl, 3-methylbutynyl, n-pentynyl, and decynyl.

"Aryl" means an aromatic monocyclic or multicyclic ring system, wherein at least one ring is aromatic, comprising about 6 to about 14 carbon atoms, and preferably about 6 to about 10 carbon atoms. Non-limiting examples of suitable aryl groups include: phenyl, naphthyl, indenyl, tetrahydronaphthyl, indanyl, anthracenyl, and fluorenyl.

"Arylalkyl" means an aryl group, as defined above, bound to an alkyl group, as defined above, wherein the alkyl group is bound to the parent moiety. Non-limiting examples of suitable arylalkyl groups include benzyl, phenethyl and naphthleneylmethyl.

"Bn" represents benzyl.

5

10

15

20

25

30

"Cycloalkyl" means saturated carbocyclic rings having 3 to 10 (e.g., 3 to 7) carbon atoms, preferably 5 to 10 carbon atoms, and more preferably 5 to 7 carbon atoms, and having one to three rings. Non-limiting examples of cycloalkyl groups include: cyclopropyl, cyclopentyl, cyclohexyl, cycloheptyl, norbornyl, and adamantyl.

"Cycloalkylalkyl" means a cycloalkyl group bound to the parent moiety through an alkyl group. Non-limiting examples include: cyclopropylmethyl and cyclohexylmethyl.

"Cycloalkenyl" means a non-aromatic mono or multicyclic ring system comprising 3 to 10 carbon atoms, and preferably 5 to 10 carbon atoms, and having at least one carbon-carbon double bond. Preferred cycloalkenyl rings have 5 to 7

carbon atoms. Non-limiting examples of cycloalkyl groups include cyclopentenyl, cyclohexenyl, cycloheptenyl, and norbornenyl.

"Et" represents ethyl.

5

10

15

20

25

30

"Halo" means fluoro, chloro, bromo, or iodo groups. Preferred are fluoro, chloro or bromo, and more preferred are fluoro and chloro.

"Halogen" means fluorine, chlorine, bromine, or iodine. Preferred are fluorine, chlorine or bromine, and more preferred are fluorine and chlorine.

"Haloalkyl" means an alkyl group as defined above wherein one or more hydrogen atoms on the alkyl is replaced by a halo group defined above.

"Heterocyclyl" or "heterocyclic" or "heterocycloalkyl" means a non-aromatic saturated monocyclic or multicyclic ring system (i.e., a saturated carbocyclic ring or ring system) comprising 3 to 10 ring atoms (e.g., 3 to 7 ring atoms), preferably 5 to 10 ring atoms, in which one or more of the atoms in the ring system is an element other than carbon, for example nitrogen, oxygen or sulfur, alone or in combination. There are no adjacent oxygen and/or sulfur atoms present in the ring system. Preferred heterocyclyls have 5 to 6 ring atoms. The prefix aza, oxa or thia before the heterocyclyl root name means that at least a nitrogen, oxygen or sulfur atom, respectively, is present as a ring atom. The nitrogen or sulfur atom of the heterocyclyl can be optionally oxidized to the corresponding N-oxide, S-oxide or S,S-dioxide. Non-limiting examples of monocyclic heterocyclyl rings include: piperidyl, pyrrolidinyl, piperazinyl, morpholinyl, thiomorpholinyl, thiazolidinyl, 1,3-dioxolanyl, 1,4-dioxanyl, tetrahydrofuranyl, tetrahydrothiophenyl, and tetrahydrothiopyranyl.

The term heterocyclic acidic functional group is intended to include groups such as, pyrrole, imidazole, triazole, tetrazole, and the like.

"Heteroaryl" means an aromatic monocyclic or multicyclic ring system comprising 5 to 14 ring atoms, preferably 5 to 10 ring atoms, in which one or more of the ring atoms is an element other than carbon, for example nitrogen, oxygen or sulfur, alone or in combination. Preferred heteroaryls contain 5 to 6 ring atoms. The prefix aza, oxa or thia before the heteroaryl root name means that at least a nitrogen, oxygen or sulfur atom respectively, is present as a ring atom. A nitrogen atom of a heteroaryl can be optionally oxidized to the corresponding N-oxide. Non-limiting examples of heteroaryls include: pyridyl, pyrazinyl, furanyl, thienyl, pyrimidinyl, isoxazolyl, isothiazolyl, oxazolyl, thiazolyl, pyrazolyl, furazanyl, pyrrolyl, pyrazolyl,

triazolyl, 1,2,4-thiadiazolyl, pyrazinyl, pyridazinyl, quinoxalinyl, phthalazinyl, imidazo[1,2-a]pyridinyl, imidazo[2,1-b]thiazolyl, benzofurazanyl, indolyl, azaindolyl, benzimidazolyl, benzothienyl, quinolinyl, imidazolyl, thienopyridyl, quinazolinyl, thienopyrimidyl, pyrrolopyridyl, imidazopyridyl, isoquinolinyl, benzoazaindolyl, 1,2,4-triazinyl, and benzothiazolyl.

5

10

"Heteroarylalkyl" means a heteroaryl group, as defined above, bound to an alkyl group, as defined above, where the bond to the parent moiety is through the alkyl group.

N-oxides can form on a tertiary nitrogen present in an R substituent, or on =N-in a heteroaryl ring substituent and are included in the compounds of formula IA.

As well known in the art, a bond drawn from a particular atom wherein no moiety is depicted at the terminal end of the bond indicates a methyl group bound through that bond to the atom. For example:

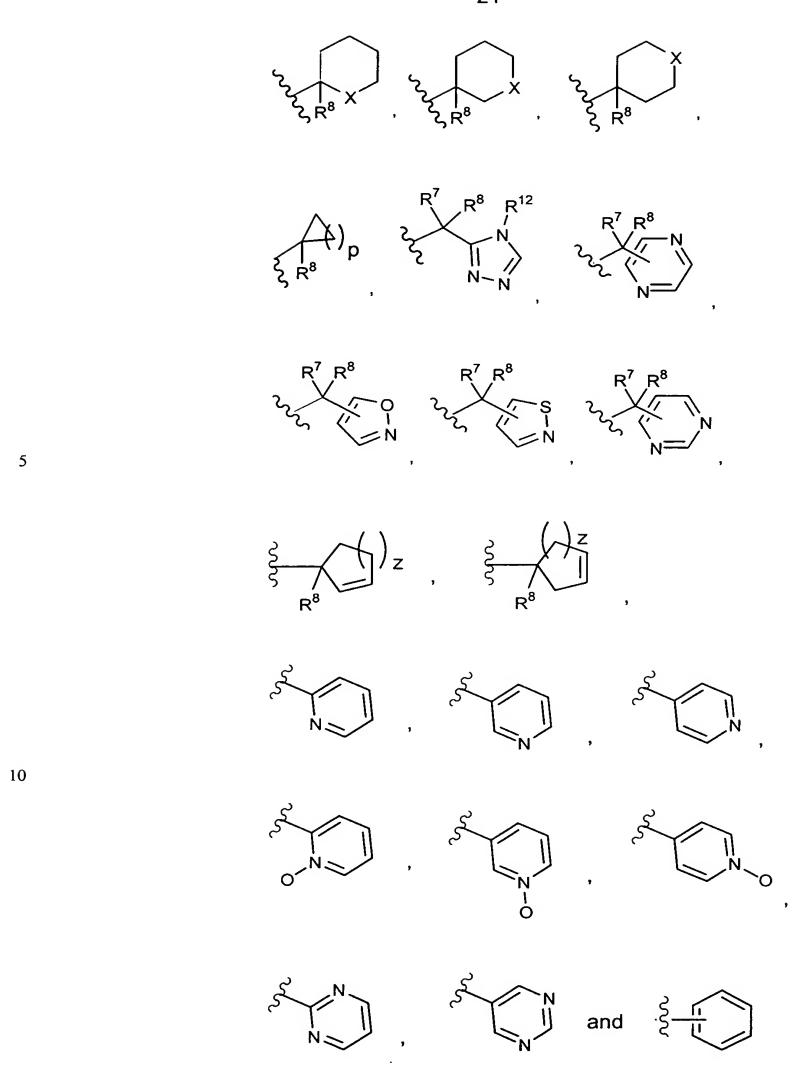
$$\begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array} \end{array}$$
 represents represents
$$\begin{array}{c} \begin{array}{c} CH_3 \\ \\ \\ \\ \end{array} \\ \begin{array}{c} CH_3 \\ \\ \end{array} \\ CH_3 \\ \end{array}$$

The compounds of this invention are represented by formula IA:

and the pharmaceutically acceptable salts (e.g., sodium or calcium salt) and solvates 5 thereof, wherein:

A is selected from the group consisting of:

$$(\mathcal{A}, \mathcal{A})_{p} = (\mathcal{A}, \mathcal{A}$$



wherein the above rings of said A groups are substituted with 1 to 6 substituents each independently selected from the group consisting of: R⁹ groups;

$$\mathbb{R}^7$$
 \mathbb{R}^8 (e.g., \mathbb{R}^7 \mathbb{R}^8), and \mathbb{R}^7 \mathbb{R}^8

wherein one or both of the above rings of said A groups are substituted with 1 to 6 substituents each independently selected from the group consisting of: R⁹ groups;

5

wherein the above phenyl rings of said A groups are substituted with 1 to 3 substituents each independently selected from the group consisting of: R⁹ groups; and

B is selected from the group consisting of

$$R^4$$
 R^5
 R^6
 R^4
 R^6
 R^6

$$R^{12}$$
 R^{3}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{2}

$$R^3$$
 S N R_3 R_2 S and R^{11} S S R^2 ;

n is 0 to 6;

5

10

15

20

p is 1 to 5;

X is O, NR¹⁸, or S;

Z is 1 to 3;

 R^2 is selected from the group consisting of: hydrogen, OH, -C(O)OH, -SH, -SO₂NR¹³R¹⁴, -NHC(O)R¹³, -NHSO₂NR¹³R¹⁴, -NHSO₂R¹³, -NR¹³R¹⁴, -C(O)NR¹³R¹⁴, -C(O)NROR¹³, -C(O)NROR¹³OH, -S(O₂)OH, -OC(O)R¹³, an unsubstituted heterocyclic acidic functional group, and a substituted heterocyclic acidic functional group; wherein there are 1 to 6 substituents on said substituted heterocyclic acidic functional group each substituent being independently selected from the group consisting of: R^9 groups;

each R³ and R⁴ is independently selected from the group consisting of: hydrogen, cyano, halogen, alkyl, cycloalkyl substituted with 1 to 4 alkyl groups (preferably C₁ to C₆ alkyl groups) wherein each alkyl group is independently selected, unsubstituted cycloalkyl, alkoxy, -OH, -CF₃, -OCF₃, -NO₂, -C(O)R¹³, -C(O)OR¹³, -C(O)NHR¹¬, -C(O)NR¹³R¹⁴, -SO(t)NR¹³R¹⁴, -SO(t)R¹³, -C(O)NR¹³OR¹⁴, unsubstituted or substituted aryl, unsubstituted or substituted heteroaryl,

$$\begin{cases} R^{31} & R^{13} \\ P - R^{31} \\ 0 & R^{14} \end{cases}$$
 and
$$\begin{cases} R^{13} \\ N \\ N \\ R^{14} \end{cases}$$

wherein there are 1 to 6 substituents on said substituted aryl group and each substituent is independently selected from the group consisting of: R⁹ groups; and wherein there are 1 to 6 substituents on said substituted heteroaryl group and each substituent is independently selected from the group consisting of: R⁹ groups; or

R³ is and R⁴ taken together with the carbons atoms to which they are bonded to in the the phenyl B substituent

$$R^4$$
 R^5
 R^6
 R^3
 R^2

form a fused ring of the formula:

(preferably Z¹) wherein Z¹ or Z² is an unsubstituted or substituted saturated heterocyclic ring (preferably a 4 to 7 membered heterocyclic ring), said ring Z¹ or Z² optionally containing one additional heteroatom selected from the group consisting of: O, S and NR¹8; wherein there are 1 to 3 substituents on said ring Z¹ or Z², and each substituent is independently selected from the group consisting of: alkyl, aryl, hydroxy, hydroxyalkyl, alkoxy, alkoxyalkyl, arylalkyl, fluoroalkyl, cycloalkyl, cycloalkylalkyl, heteroaryl, heteroarylalkyl, amino, -C(O)OR¹5, -C(O)NR¹5R¹6, -SOtNR¹5R¹6, -C(O)R¹5, -SO₂R¹5 provided that R¹5 is not H, -NHC(O)NR¹5R¹6, -NHC(O)OR¹5, halogen, and a heterocycloalkenyl group (i.e., a heterocyclic group that has at least one, and preferably one, double bond in a ring, e.g.,

examples of the fused ring moiety include, but are not limited to:

20

5

10

$$R^5$$
 R^6
 R^6
 R^6
 R^6
 R^6
 R^6

each R⁵ and R⁶ are the same or different and are independently selected from the group consisting of hydrogen, halogen, alkyl, alkoxy, -CF₃, -OCF₃, -NO₂, -C(O)R¹³, -C(O)OR¹³, -C(O)NR¹³R¹⁴, -SO_(t)NR¹³R¹⁴, -C(O)NR¹³OR¹⁴, cyano, unsubstituted or substituted aryl, and unsubstituted or substituted heteroaryl group; wherein there are 1 to 6 substituents on said substituted aryl group and each substituent is independently selected from the group consisting of: R9 groups; and wherein there are 1 to 6 substituents on said substituted heteroaryl group and each substituent is independently selected from the group consisting of: R9 groups;

each R⁷ and R⁸ is independently selected from the group consisting of: H, unsubstituted or substituted alkyl, unsubstituted or substituted aryl, unsubstituted or substituted heteroaryl, unsubstituted or substituted arylalkyl, unsubstituted or substituted heteroarylalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted cycloalkylalkyl, -CO₂R¹³, -CONR¹³R¹⁴, alkynyl, alkenyl, and cycloalkenyl; and wherein there are one or more (e.g., 1 to 6) substituents on said substituted R⁷ and R⁸ groups, wherein each substitutent is independently selected from the group consisting of:

- a) halogen,
- $-CF_3$, b)
- -COR¹³, C)
- -OR¹³, d)
- $-NR^{13}R^{14}$, e)
- $-NO_2$ f)
- -CN, g)
- -SO₂OR¹³, h) 25
 - -Si(alkyl)₃, wherein each alkyl is independently selected, i)
 - -Si(aryl)₃, wherein each alkyl is independently selected, j)
 - -(R¹³)₂R¹⁴Si, wherein each R¹³ is independently selected,
 - I) $-CO_2R^{13}$,
 - -C(O)NR¹³R¹⁴, m)

20

5

10

15

- n) $-SO_2NR^{13}R^{14}$,
- o) $-SO_2R^{13}$,

10

15

20

25

30

- p) $-OC(O)R^{13}$,
- q) $-OC(O)NR^{13}R^{14}$,
- r) $-NR^{13}C(O)R^{14}$, and
- s) $-NR^{13}CO_2R^{14}$;

(fluoroalkyl is one non-limiting example of an alkyl group that is substituted with halogen);

R^{8a} is selected from the group consisting of: hydrogen, alkyl, cycloalkyl and cycloalkylalkyl;

each R⁹ is independently selected from the group consisting of:

- a) $-R^{13}$,
- b) halogen,
- c) -CF₃,
- d) $-COR^{13}$,
- e) $-OR^{13}$,
- f) $-NR^{13}R^{14}$,
- g) $-NO_2$,
- h) -CN,
- i) $-SO_2R^{13}$,
- j) $-SO_2NR^{13}R^{14}$,
- k) -NR¹³COR¹⁴.
- I) $-CONR^{13}R^{14}$,
- m) $-NR^{13}CO_2R^{14}$,
- n) $-CO_2R^{13}$,
- 0)

- p) alkyl substituted with one or more (e.g., one) –OH groups (e.g., $-(CH_2)_qOH$, wherein q is 1-6, usually 1 to 2, and preferably 1),
- q) alkyl substituted with one or more (e.g., one) $-NR^{13}R^{14}$ group (e.g., -(CH₂)_qNR¹³R¹⁴, wherein q is 1-6, usually 1 to 2, and preferably 1), and

r) -N(R¹³)SO₂R¹⁴ (e.g., R¹³ is H and R¹⁴ is alkyl, such as methyl); each R¹⁰ and R¹¹ is independently selected from the group consisting of R¹³, (e.g., hydrogen and alkyl (e.g., C₁ to C₆ alkyl, such as methyl)), halogen, -CF₃, -OCF₃, -NR¹³R¹⁴, -NR¹³C(O)NR¹³R¹⁴, -OH, -C(O)OR¹³, -SH, -SO_(t)NR¹³R¹⁴, -SO₂R¹³, -NHC(O)R¹³, -NHSO₂NR¹³R¹⁴, -NHSO₂R¹³, -C(O)NR¹³R¹⁴, -C(O)NR¹³OR¹⁴, -OC(O)R¹³ and cyano;

R¹² is selected from the group consisting of: hydrogen, -C(O)OR¹³, unsubstituted or substituted or substituted heteroaryl, unsubstituted or substituted arylalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted alkyl, unsubstituted or substituted cycloalkylalkyl, and unsubstituted or substituted heteroarylalkyl group; wherein there are 1 to 6 substituents on the substituted R¹² groups and each substituent is independently selected from the group consisting of: R⁹ groups;

each R¹³ and R¹⁴ is independently selected from the group consisting of: H, unsubstituted or substituted alkyl, unsubstituted or substituted aryl, unsubstituted or substituted heteroaryl, unsubstituted or substituted arylalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted heterocyclic, unsubstituted or substituted fluoroalkyl, and unsubstituted or substituted heterocycloalkylalkyl (wherein "heterocyloalkyl" means heterocyclic); wherein there are 1 to 6 substituents on said substituted R¹³ and R¹⁴ groups and each substituent is independently selected from the group consisting of: alkyl, -CF₃, -OH, alkoxy, aryl, arylalkyl, fluroalkyl, cycloalkyl, cycloalkylalkyl, heteroaryl, heteroarylalkyl, -N(R⁴⁰)₂, -C(O)OR¹⁵, -C(O)NR¹⁵R¹⁶, -S(O)_tNR¹⁵R¹⁶, -C(O)R¹⁵, -SO₂R¹⁵ provided that R¹⁵ is not H, halogen, and -NHC(O)NR¹⁵R¹⁶; or

R¹³ and R¹⁴ taken together with the nitrogen they are attached to in the groups -C(O)NR¹³R¹⁴ and -SO₂NR¹³R¹⁴ form an unsubstituted or substituted saturated heterocyclic ring (preferably a 3 to 7 membered heterocyclic ring), said ring optionally containing one additional heteroatom selected from the group consisting of: O, S and NR¹⁸; wherein there are 1 to 3 substituents on the substituted cyclized R¹³ and R¹⁴ groups (i.e., there is 1 to 3 substituents on the ring formed when the R¹³ and R¹⁴ groups are taken together with the nitrogen to which they are bound) and each substituent is independently selected from the group consisting of: alkyl, aryl, hydroxy,

hydroxyalkyl, alkoxy, alkoxyalkyl, arylalkyl, fluoroalkyl, cycloalkyl, cycloalkylalkyl, heteroaryl, heteroarylalkyl, amino, -C(O)OR¹⁵, -C(O)NR¹⁵R¹⁶, -SO_tNR¹⁵R¹⁶, -C(O)R¹⁵, -SO₂R¹⁵ provided that R¹⁵ is not H, -NHC(O)NR¹⁵R¹⁶, -NHC(O)OR¹⁵, halogen, and a heterocycloalkenyl group (i.e., a heterocyclic group that has at least one, and preferably one, double bond in a ring, e.g.,

each R¹⁵ and R¹⁶ is independently selected from the group consisting of: H, alkyl, aryl, arylalkyl, cycloalkyl and heteroaryl;

R¹⁷ is selected from the group consisting of: -SO₂alkyl, -SO₂aryl, -SO₂cycloalkyl, and -SO₂heteroaryl;

R¹⁸ is selected from the group consisting of: H, alkyl, aryl, heteroaryl, -C(O)R¹⁹, -SO₂R¹⁹ and -C(O)NR¹⁹R²⁰;

each R¹⁹ and R²⁰ is independently selected from the group consisting of: alkyl, aryl and heteroaryl;

 R^{30} is selected from the group consisting of: alkyl, cycloalkyl, -CN, -NO₂, or -SO₂ R^{15} provided that R^{15} is not H;

each R³¹ is independently selected from the group consisting of: unsubstituted alkyl, unsubstituted or substituted aryl, unsubstituted or substituted heteroaryl and unsubstituted or substituted cycloalkyl; wherein there are 1 to 6 substituents on said substituted R³¹ groups and each substituent is independently selected from the group consisting of: alkyl, halogen and -CF₃;

each R⁴⁰ is independently selected from the group consisting of: H, alkyl and cycloalkyl;

g is 1 or 2 (preferably 1); and t is 0, 1 or 2.

5

10

15

20

25

30

For compounds of formula IA, when R^3 is $-SO_{(t)}NR^{13}R^{14}$ (e.g., $-SO_2NR^{13}R^{14}$), preferably R^{13} and R^{14} are independently selected from the group consisting of: H and alkyl (e.g., methyl, ethyl, isopropyl and t-butyl). Examples include, but are not limited to (1) $-SO_2NH_2$ and (2) $-SO_2NR^{13}R^{14}$ wherein R^{13} and R^{14} are the same or different alkyl group (e.g., methyl, ethyl, isopropyl and t-butyl), e.g., the same alkyl group, such as, for example $-SO_2N(CH_3)_2$.

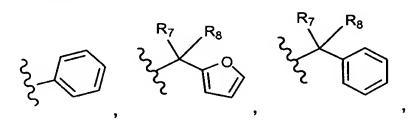
For compounds of formula IA, when R^3 is $-C(O)NR^{13}R^{14}$, preferably R^{13} and R^{14} are independently selected from the group consisting of: H and alkyl (e.g., methyl, ethyl, isopropyl and t-butyl). Examples include, but are not limited to $-C(O)NR^{13}R^{14}$ wherein each R^{13} and R^{14} are the same or different alkyl group, e.g., the same alkyl group, such as, for example $-C(O)N(CH_3)_2$.

For the compounds of formula IA substituent A is preferably selected from the group consisting of:

(1) unsubstituted or substituted:

5

10

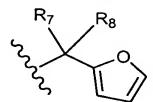


 \mathbb{R}^7 \mathbb{R}^8 \mathbb{R}^7 \mathbb

(2) $\begin{cases} R_7 & R_8 \\ R_{8a} & R_{8a} \end{cases}$

wherein all substitutents are as defined for formula IA.

For the compounds of formula IA substituent A is most preferably:



wherein the furan ring is unsubstituted or substituted with 1 or 2 alkyl groups (e.g., C_1 to C_3 alkyl groups) wherein each alkyl group is independently selected, R^7 is selected from the group consisting of: -CF₃, alkyl (e.g., C_1 to C_4 alkyl) and cycloalkyl (e.g., cyclopropyl), and R^8 is H. More preferably the furan ring is substituted.

For the compounds of formula IA substituent A is even more preferably:

wherein the furan ring is unsubstituted or substituted with 1 or 2 alkyl groups independently selected from the group consisting of methyl, ethyl and isoprpyl, R^7 is selected from the group consisting of: -CF₃, ethyl, isopropyl, t-butyl and cyclopropyl, and R^8 is H. Still more preferably the furan ring is substituted.

5

10

For the compounds of formula IA substituent A is even yet more preferably:

wherein the furan ring is substituted with 1 or 2 alkyl groups independently selected from the group consisting of methyl, ethyl and isopropyl, R⁷ is selected from the group consisting of: ethyl, isopropyl and t-butyl, and R⁸ is H.

Examples of substituent A in formula IA include, but are not limited to:

Substituent A in formula IA is most preferably selected from the group consisting of:

Substituent A in formula IA is more preferably selected from the group consisting of:

10

Substituent A in formula IA is even more preferably selected from the group consisting of:

Substituent B in formula IA is preferably selected from the group consisting of:

wherein all substituents are as defined for formula IA.

5

10

Substituent B in formula IA is most preferably selected from the group consisting of:

Substituent B in Formula IA is more preferably selected from the group consisting of:

and
$$H_2N-S$$
 OH

5

10

15

20

25

Substituent B in Formula IA is even more preferably selected from the group consisting of:

Substituent B in Formula IA is still even more preferably selected from the group consisting of:

An embodiment of the present invention is directed to a method of treating an chemokine mediated disease in a patient in need of such treatment (e.g., a mammal, preferably a human being) comprising administering to said patient a therapeutically effective amount of at least one (e.g., 1-3, and usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

Examples of chemokine mediated (e.g., CXCR1 and/or CXCR2, or CCR7) diseases or conditions include but are not limited to: acute pain, acute inflammation, chronic inflammation, rheumatoid arthritis, acute inflammatory pain, chronic inflammatory pain, neuropathic pain, psoriasis, atopic dermatitis, asthma, COPD, adult respiratory disease, arthritis, inflammatory bowel disease, Crohn's disease, ulcerative colitis, septic shock, endotoxic shock, gram negative sepsis, toxic shock syndrome, stroke, cardiac reperfusion injury, renal reperfusion injury, glomerulonephritis, thrombosis, Alzheimer's disease, graft vs. host reaction (i.e., graft vs. host disease), allograft rejections (e.g., acute allograft rejection, and chronic allograft rejection), malaria, acute respiratory distress syndrome, delayed type hypersensitivity reaction, atherosclerosis, cerebral ischemia, cardiac ischemia, osteoarthritis, multiple sclerosis, restinosis, angiogenesis, osteoporosis, gingivitis, respiratory viruses, herpes viruses, hepatitis viruses, HIV, Kaposi's sarcoma associated virus (i.e., Kaposi's sarcoma),

5

10

15

20

25

30

meningitis, cystic fibrosis, pre-term labor, cough, pruritis, multi-organ dysfunction, trauma, strains, sprains, contusions, psoriatic arthritis, herpes, encephalitis, CNS vasculitis, traumatic brain injury, CNS tumors, subarachnoid hemorrhage, post surgical trauma, interstitial pneumonitis, hypersensitivity, crystal induced arthritis, acute pancreatitis, chronic pancreatitis, acute alcoholic hepatitis, necrotizing enterocolitis, chronic sinusitis, angiogenic ocular disease, ocular inflammation, retinopathy of prematurity, diabetic retinopathy, macular degeneration with the wet type preferred, corneal neovascularization, polymyositis, vasculitis, acne, gastric ulcers, duodenal ulcers, celiac disease, esophagitis, glossitis, airflow obstruction, airway hyperresponsiveness (i.e., airway hyperreactivity), bronchiectasis, bronchiolitis, bronchiolitis obliterans, chronic bronchitis, cor pulmonae, dyspnea, emphysema, hypercapnea, hyperinflation, hypoxemia, hyperoxia-induced inflammations, hypoxia, surgical lung volume reduction, pulmonary fibrosis, pulmonary hypertension, right ventricular hypertrophy, peritonitis associated with continuous ambulatory peritoneal dialysis (CAPD), granulocytic ehrlichiosis, sarcoidosis, small airway disease, ventilation-perfusion mismatching, wheeze, colds, gout, alcoholic liver disease, lupus, burn therapy (i.e., the treatment of burns), periodontitis, cancer, transplant reperfusion injury, early transplantation rejection (e.g., acute allograft rejection), airway hyperreactivity, allergic contact dermatitis, allergic rhinitis, alopecia areata, antiphospholipid syndromes, aplastic anemia, autoimmune deafness (including, for example, Meniere's disease), autoimmune hemolytic syndromes, autoimmune hepatitis, autoimmune neuropathy, autoimmune ovarian failure, autoimmune orchitis, autoimmune thrombocytopenia, bullous pemphigoid, chronic allograft vasculopathy, chronic inflammatory demyelinating polyneuropathy, cirrhosis, cor pneumoniae, cryoglobulinemia, dermatomyositis, diabetes, drug-induced autoimmunity, epidermolysis bullosa acquisita, endometriosis, fibrotic diseases, gastritis, Goodpasture's syndrome, Graves' disease, Gullain-Barre disease, Hashimoto's thyroiditis, hepatitis-associated autoimmunity, HIV-related autoimmune syndromes and hematologic disorders, hypophytis, idiopathic thrombocytic pupura, interstitial cystitis, juvenile arthritis, Langerhans' cell histiocytitis, lichen planus, metal-induced autoimmunity, myasthenia gravis, myelodysplastic syndromes, myocarditis (including viral myocarditis), myositis, Neuropathies (including, for example, IgA neuropathy, membranous neuropathy and idiopathic neuropathy), nephritic syndrome, optic

neuritis, pancreatitis, paroxysmal nocturnal hemoglobulinemia, pemphigus, polymyalgia, post-infectious autoimmunity, primary biliary cirrhosis, reactive arthritis, ankylosing spondylitis, Raynaud's phenomenon, Reiter's syndrome, reperfusion injury, scleritis, scleroderma, secondary hematologic manifestation of autoimmune diseases (such as, for example, anemias), silicone implant associated autoimmune disease, Sjogren's syndrome, systemic lupus erythematosus, thrombocytopenia, transverse myelitis, tubulointerstitial nephritis, uveitis, vasculitis syndromes (such as, for example, giant cell arteritis, Behcet's disease and Wegener's granulomatosis), and Vitiligo.

5

10

15

20

25

30

Examples of CXCR1 and/or CXCR2 mediated diseases or conditions include but are not limited to: acute pain, acute inflammation, chronic inflammation, rheumatoid arthritis, acute inflammatory pain, chronic inflammatory pain, neuropathic pain, psoriasis, atopic dermatitis, asthma, COPD, adult respiratory disease, arthritis, inflammatory bowel disease, Crohn's disease, ulcerative colitis, septic shock, endotoxic shock, gram negative sepsis, toxic shock syndrome, stroke, cardiac reperfusion injury, renal reperfusion injury, glomerulonephritis, thrombosis, Alzheimer's disease, graft vs. host reaction (i.e., graft vs. host disease), allograft rejections (e.g., acute allograft rejection, and chronic allograft rejection), malaria, acute respiratory distress syndrome, delayed type hypersensitivity reaction, atherosclerosis, cerebral ischemia, cardiac ischemia, osteoarthritis, multiple sclerosis, restinosis, angiogenesis, osteoporosis, gingivitis, respiratory viruses, herpes viruses, hepatitis viruses, HIV, Kaposi's sarcoma associated virus (i.e., Kaposi's sarcoma), meningitis, cystic fibrosis, pre-term labor, cough, pruritis, multi-organ dysfunction, trauma, strains, sprains, contusions, psoriatic arthritis, herpes, encephalitis, CNS vasculitis, traumatic brain injury, CNS tumors, subarachnoid hemorrhage, post surgical trauma, interstitial pneumonitis, hypersensitivity, crystal induced arthritis, acute pancreatitis, chronic pancreatitis, acute alcoholic hepatitis, necrotizing enterocolitis, chronic sinusitis, angiogenic ocular disease, ocular inflammation, retinopathy of prematurity, diabetic retinopathy, macular degeneration with the wet type preferred, corneal neovascularization, polymyositis, vasculitis, acne, gastric ulcers, duodenal ulcers, celiac disease, esophagitis, glossitis, airflow obstruction. airway hyperresponsiveness (i.e., airway hyperreactivity), bronchiectasis, bronchiolitis, bronchiolitis obliterans, chronic bronchitis, cor pulmonae, dyspnea, emphysema, hypercapnea, hyperinflation, hypoxemia, hyperoxia-induced inflammations, hypoxia,

surgical lung volume reduction, pulmonary fibrosis, pulmonary hypertension, right ventricular hypertrophy, peritonitis associated with continuous ambulatory peritoneal dialysis (CAPD), granulocytic ehrlichiosis, sarcoidosis, small airway disease, ventilation-perfusion mismatching, wheeze, colds, gout, alcoholic liver disease, lupus, burn therapy (i.e., the treatment of burns), periodontitis, cancer, transplant reperfusion injury, early transplantation rejection (e.g., acute allograft rejection).

5

10

15

20

25

30

Examples of CCR7 mediated diseases or conditions include, but are not limited to: acute inflammation, chronic inflammation, acute inflammatory pain, acute pain, chronic inflammatory pain, neuropathic pain, acute allograft rejection, acute respiratory distress syndrome, adult respiratory disease, airway hyperreactivity, allergic contact dermatitis, allergic rhinitis, alopecia areata, alzheimer's disease, angiogenic ocular disease, antiphospholipid syndromes, aplastic anemia, asthma, atherosclerosis, atopic dermatitis, autoimmune deafness (including, for example, Meniere's disease), autoimmune hemolytic syndromes, autoimmune hepatitis, autoimmune neuropathy, autoimmune ovarian failure, autoimmune orchitis, autoimmune thrombocytopenia, bronchiolitis, bronchiolitis obliterans syndrome, bullous pemphigoid, burn therapy (i.e., the treatment of burns), cancer, cerebral ischemia, cardiac ischemia, chronic allograft rejection, chronic allograft vasculopathy, chronic bronchitis, chronic inflammatory demyelinating polyneuropathy, chronic sinusitis, cirrhosis, CNS vasculitis, COPD, Cor pneumoniae, Crohn's disease, cryoglobulinemia, crystal-induced arthritis, delayed-type hypersensitivity reactions, dermatomyositis, diabetes, diabetic retinopathy, drug-induced autoimmunity, dyspnea, emphysema, epidermolysis bullosa acquisita, endometriosis, fibrotic diseases, gastritis, glomerulonephritis, Goodpasture's syndrome, graft vs host disease, Graves' disease, Gullain-Barre disease, Hashimoto's thyroiditis, hepatitis-associated autoimmunity, HIV-related autoimmune syndromes and hematologic disorders, hyperoxia-induced inflammation, hypercapnea, hyperinflation, hypophytis, hypoxia, idiopathic thrombocytic pupura, inflammatory bowel diseases, interstitial cystitis, interstitial pneumonitis, juvenile arthritis, Langerhans' cell histiocytitis, lichen planus, metal-induced autoimmunity, multiple sclerosis, myasthenia gravis, myelodysplastic syndromes, myocarditis including viral myocarditis, myositis, neuropathies (including, for example, IgA neuropathy, membranous neuropathy and idiopathic neuropathy), nephritic syndrome, ocular inflammation, optic neuritis, osteoarthritis, pancreatitis,

paroxysmal nocturnal hemoglobulinemia, pemphigus, polymyalgia, polymyositis, postinfectious autoimmunity, pulmonary fibrosis, primary biliary cirrhosis, psoriasis,
pruritis, rheumatoid arthritis, reactive arthritis, ankylosing spondylitis, psoriatic arthritis,
Raynaud's phenomenon, Reiter's syndrome, reperfusion injury, restenosis,
sarcoidosis, scleritis, scleroderma, secondary hematologic manifestation of
autoimmune diseases (such as, for example, anemias), silicone implant associated
autoimmune disease, Sjogren's syndrome, systemic lupus erythematosus,
thrombocytopenia, thrombosis, transverse myelitis, tubulointerstitial nephritis,
ulcerative colitis, uveitis, vasculitis and vasculitis syndromes (such as, for example,
giant cell arteritis, Behcet's disease and Wegener's granulomatosis), and vitiligo.

5

10

15

20

25

30

Another embodiment of this invention is directed to a method of treating a CXCR1 and/or CXCR2 mediated disease, as described above, in a patient in need of such treatment comprising administering to said patient an effective amount of a compound selected from the group consisting of the final compounds of Examples 56, 201.1, 201.9, and the pharmaceutically acceptable salts and solvates thereof.

Another embodiment of this invention is directed to a method of treating a CCR7 mediated disease, as described above, in a patient in need of such treatment comprising administering to said patient an effective amount of a compound selected from the group consisting of the final compounds of Examples 2065, 2066, 2105, 2106, and the pharmaceutically acceptable salts and solvates thereof.

Another embodiment of this invention is directed to a method of treating Kaposi's sarcoma, melanoma, gastric carcinoma, and non-small cell carcinoma in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (e.g., 1-3, usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

Another embodiment of this invention is directed to a method of treating melanoma, gastric carcinoma, and non-small cell carcinoma in a patient in need of such treatment comprising administering to said patient an effective amount of at least one (e.g., 1-3, usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof.

Another embodiment of the present invention is directed to a method of treating cancer in a patient (e.g., a mammal, such as a human being) in need of such treatment, comprising administering to said patient, concurrently or sequentially, a

therapeutically effective amount of (a) at least one (e.g., 1-3, and usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, and (b) at least one (e.g., 1, 2 or 3) anticancer agent selected from the group consisting of: (1) microtubule affecting agents, (2) antineoplastic agents, (3) anti-angiogenesis agents, (4) VEGF receptor kinase inhibitors, (5) antibodies against the VEGF receptor, (6) interferon, and (7) radiation.

In further embodiments of this invention that are directed to the treatment of cancer, at least one (e.g., 1-3, and usually one) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, is administered in combination with at least one (e.g., 1 or 2, or 1) antineoplastic agent selected from the group consisting of: gemcitabine, paclitaxel (Taxol®), 5-Fluorouracil (5-FU), cyclophosphamide (Cytoxan®), temozolomide, taxotere and Vincristine.

In another embodiment the present invention provides a method of treating cancer in a patient (e.g., a mammal, such as a human being) in need of such treatment, comprising administering, concurrently or sequentially, an effective amount of (a) at least one (e.g., 1-3, usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, and (b) at least one (e.g., 1-3, usually 1) microtubule affecting agent (e.g., paclitaxel).

In the method of treating a pulmonary disease (e.g., COPD, asthma, or cystic fibrosis), at least one (usually 1) compound of formula IA, or a pharmaceutically acceptable salt or solvate thereof, is administered in combination with at least one compound selected from the group consisting of: glucocorticoids, 5-lipoxygenase inhibitors, β-2 adrenoceptor agonists, muscarinic M1 antagonists, muscarinic M3 antagonists, muscarinic M2 agonists, NK3 antagonists, LTB4 antagonists, cysteinyl leukotriene antagonists, bronchodilators, PDE4 inhibitors, PDE inhibitors, elastase inhibitors, MMP inhibitors, phospholipase A2 inhibitors, phospholipase D inhibitors, histamine H1 antagonists, histamine H3 antagonists, dopamine agonists, adenosine A2 agonists, NK1 and NK2 antagonists, GABA-b agonists, nociceptin agonists, expectorants, mucolytic agents, decongestants, antioxidants, anti-IL-8 anti-bodies, anti-IL-5 antibodies, anti-IgE antibodies, anti-TNF antibodies, IL-10, adhesion molecule inhibitors, and growth hormones. Agents that belong to these classes include, but are not limited to, beclomethasone, mometasone, ciclesonide, budesonide, fluticasone, albuterol, salmeterol, formoterol, loratadine, desloratadine,

tiotropium bromide, MSI-ipratropium bromide, montelukast, theophilline, cilomilast, roflumilast, cromolyn, ZD-4407, talnetant, LTB-019, revatropate, pumafentrine, CP-955, AR-C-89855, BAY-19-8004, GW-328267, QAB-149, DNK-333, YM-40461 and TH-9506 or pharmaceutically acceptable formulations thereof.

Representative embodiments of the novel compounds of this invention are described below. The embodiments have been numbered for purposes of reference thereto.

Embodiment No. 1 is directed to the novel compounds of formula IA wherein B is:

$$R^4$$
 R^5
 R^6
 R^2

10

20

5

and all other substitutents are as defined for of formula IA.

Embodiment No. 2 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 3 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA

Embodiment No. 4 is directed to the novel compounds of formula IA wherein B is:

Embodiment No. 5 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 6 is directed to the novel compounds of formula IA wherein B is:

$$R^4$$
 R^{12}
 R^3
 O
 O
 O

10

5

and all other substitutents are as defined for of formula IA.

Embodiment No. 7 is directed to the novel compounds of formula IA wherein B is:

$$\mathbb{R}^{12}$$

$$\mathbb{R}^{3}$$

$$\mathbb{R}^{2}$$

and all other substitutents are as defined for of formula IA.

Embodiment No. 8 is directed to the novel compounds of formula IA wherein B is:

$$\mathbb{R}^{3}$$

Embodiment No. 9 is directed to the novel compounds of formula IA wherein B is:

$$\mathbb{R}^{3}$$

$$\mathbb{R}^{2}$$

and all other substitutents are as defined for of formula IA.

5

Embodiment No. 10 is directed to the novel compounds of formula IA wherein B is:

$$R^{12}$$
 R^{3}
 R^{2}
 R^{2}

and all other substitutents are as defined for of formula IA.

Embodiment No. 11 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 12 is directed to the novel compounds of formula IA wherein B is:

Embodiment No. 13 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 14 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

10

Embodiment No. 15 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 16 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 17 is directed to the novel compounds of formula IA wherein B is:

$$R_3$$
 R_2
 R_3
 R_2

Embodiment No. 18 is directed to the novel compounds of formula IA wherein B is:

and all other substitutents are as defined for of formula IA.

Embodiment No. 19 is directed to compounds of formula IA wherein B is selected from the group consisting of:

(1)

$$R^4$$
 R^5
 R^6
 R^3
 R^2
 R^6

10

5

and R³ for this B group is selected from the group consisting of: -C(O)NR¹³R¹⁴,

$$\begin{cases} R^{31} & R^{13} \\ R^{31} & R^{14} \\ R^{31} & R^{14} \\ R^{30} & R^{30} \\ R^{30} &$$

and all other substituents are as defined for formula IA.

Embodiment No. 20 is directed to compounds of formula IA wherein B is:

15

and all other substituents are as defined in formula IA.

Embodiment No. 21 is directed to compounds of formula IA wherein B is

$$\begin{array}{c|c}
R^{13} \\
R^{14}
\end{array}$$

$$\begin{array}{c|c}
R^5 \\
R^6 \\
R^{14}
\end{array}$$

R¹³ and R¹⁴ are independently selected from the group consisting of H and alkyl (e.g., methyl, ethyl, isopropyl and t-butyl), and all other substituents are as defined in formula IA.

Embodiment No. 22 is directed to compounds of formula IA wherein B is

$$\begin{array}{c|c}
R^{13} \\
R^{14} \\
R^{14}
\end{array}$$

$$\begin{array}{c|c}
R^{5} \\
C \\
R^{6} \\
R^{2}
\end{array}$$

wherein:

5

10

15

20

(1) R² is –OH and all other substituents are as defined in formula IA, or

(2) R² is–OH, and R¹³ and R¹⁴ are independently selected from the group, consisting of: H and alkyl (e.g., methyl, ethyl, isopropyl and t-butyl), or

(3) R^2 is–OH, and R^{13} and R^{14} are the same or different and alkyl group (e.g., methyl, ethyl, isopropyl and t-butyl), for example the same alkyl group, for example methyl, and

(4) and all other substituents are as defined in formula IA.

Embodiment No. 23 is directed to compounds of formula IA wherein B is

$$R^4$$
 R^5
 R^6
 R^3
 R^2

R³ is selected from the group consisting of:

$$\begin{cases} R^{31} & R^{13} \\ R^{31} & R^{14} \\ R^{31} & R^{14} \\ R^{30} & R^{31} \end{cases} \text{ and } \begin{cases} R^{13} \\ R^{14} \\ R^{14} \\ R^{14} \\ R^{14} \end{cases}$$

and all other substituents are as defined in formula IA.

Embodiment No. 24 is directed to compounds of formula IA wherein B is

$$R^4$$
 R^5
 R^6
 R^3
 R^2

R³ is selected from the group consisting of:

5

15

$$\begin{cases} R^{31} & R^{13} \\ P - R^{31} & R^{14} \\ N & R^{14} \end{cases} \text{ and } \begin{cases} R^{13} \\ N \\ N \\ R^{14} \end{cases}$$

R² is -OH, and all other substituents are as defined in formula IA.

Embodiment No. 25 is directed to compounds of formula IA wherein B is:

$$R^{14}$$
, N Q R^2 S

and all other substituents are as defined in formula IA.

Embodiment No. 26 is directed to compounds of formula IA wherein B is:

$$R^{14}$$
, N Q R^2 S

10 R² is –OH, and all other substituents are as defined in formula IA.

Embodiment No. 27 is directed to compounds of formula IA wherein B is:

$$R^{14}$$
, N Q R^2 S

 R^2 is as defined for compounds of formula IA, R^{13} and R^{14} are independently selected from the group consisting of H and alkyl (e.g., methyl, ethyl, isopropyl and t-butyl), and all other substituents areas defined for compounds of formula IA. For example, R^{13} and R^{14} are the same or different alkyl group. Also, for example, R^{13} and R^{14} are the same alkyl group. Also, for example, R^{13} and R^{14} are methyl.

Embodiment No. 28 is directed to the novel compounds of formula IA wherein B is:

$$R^{14}$$
, N Q R^2 S

R² is –OH, R¹³ and R¹⁴ are independently selected from the group consisting of H and alkyl (e.g., methyl, ethyl, isopropyl and t-butyl), and all other substituents areas defined for compounds of formula IA. For example, R¹³ and R¹⁴ are the same or different alkyl group. Also, for example, R¹³ and R¹⁴ are the same alkyl group. Also, for example, R¹³ and R¹⁴ are methyl.

Embodiment No. 29 is directed to novel compounds of formula IA wherein B is as described in Embodiment No. 23, R⁴ is H, R⁵ is H, R⁶ is H, and all other substituents are as defined for compounds of formula IA.

10

15

20

Embodiment No. 30 is directed to novel compounds of formula IA wherein B is as described in Embodiment No. 24, R⁴ is H, R⁵ is H, R⁶ is H, and all other substituents areas defined for compounds of formula IA.

Embodiment No. 31 is directed to novel compounds of formula IA wherein B is as described in Embodiments Nos. 21, 22, 25 and 26, except that R¹³ and R¹⁴ are each methyl, and all other substituents are as defined in formula IA.

Embodiment No. 32 is directed to compounds of formula IA wherein B is:

$$\mathbb{R}^{3}$$

R¹¹ is H or methyl (preferably H), and all other substituents are as defined in formula IA.

Embodiment No. 33 is directed to compounds of formula IA wherein B is:

$$\mathbb{R}^{3}$$
 \mathbb{R}^{2}

R² is –OH, and all other substituents are as defined in formula IA.

Embodiment No. 34 is directed to compounds of formula IA wherein B is:

R³ is –C(O)NR¹³R¹⁴, and all other substituents are as defined in formula IA.

Embodiment No. 35 is directed to compounds of formula IA wherein B is:

 R^3 is $-S(O)_tNR^{13}R^{14}$ (e.g., t is 2), and all other substituents are as defined in formula IA.

5

Embodiment No. 36 is directed to compounds of formula IA wherein B is:

$$\mathbb{R}^3$$
 \mathbb{R}^2

10 R² is –OH, R³ is –C(O)NR¹³R¹⁴, and all other substituents are as defined in formula IA.

Embodiment No. 37 of this invention is directed to compounds of formula IA wherein B is:

$$\mathbb{R}^3$$
 \mathbb{R}^2

R² is –OH, and R³ is –S(O)_tNR¹³R¹⁴ (e.g., t is 2), and all other substituents are as defined in formula IA.

Embodiment No. 38 is directed to compounds of formula IA wherein B is:

R² is –OH, R³ is –C(O)NR¹³R¹⁴, R¹¹ is H or methyl (preferably H), and all other substituents are as defined in formula IA.

Embodiment No. 39 is directed to compounds of formula IA wherein B is:

$$\mathbb{R}^3$$
 \mathbb{R}^2

R² is –OH, R³ is –S(O)_tNR¹³R¹⁴ (e.g., t is 2), R¹¹ is H or methyl (preferably H), and all other substituents are as defined in formula IA.

Embodiment No. 40 is directed to compounds of formula IA wherein B is:

$$\mathbb{R}^{3}$$

R² is –OH, R³ is –C(O)NR¹³R¹⁴, R¹¹ is H or methyl (preferably H), and R¹³ and R¹⁴ are independently selected from the group consisting of: H, alkyl (e.g., methyl, ethyl, isopropyl and t-butyl), unsubstituted cycloalkyl, substituted cycloalkyl, unsubstituted heteroaryl and substituted heteroaryl, and all other substituents are as defined in formula IA. For example, one of R¹³ or R¹⁴ is alkyl (e.g., methyl). An example of a substituted heteroaryl group is

10

15

20

Embodiment No. 41 is directed to compounds of formula IA wherein B is:

 R^2 is -OH, R^3 is $-S(O)_tNR^{13}R^{14}$ (e.g., t is 2), R^{11} is H or methyl (preferably H), and R^{13} and R^{14} are independently selected from the group consisting of:H, alkyl (e.g., methyl, ethyl, isopropyl, and t-butyl), unsubstituted cycloalkyl, and substituted cycloalkyl, and all other substituents are as defined in formula IA. For example R^3 is (1) $-SO_2NH_2$ or (2) $-SO_2NR^{13}R^{14}$ wherein R^{13} and R^{14} are the same or different alkyl group (e.g.,

methyl, ethyl, isopropyl and t-butyl), e.g., the same alkyl group, such as, for example $-SO_2N(CH_3)_2$.

Embodiment No. 42 is directed to compounds of formula IA wherein B is:

5 R¹¹ is H, and all other substituents are as defined in formula IA.

Embodiment No. 43 is directed to compounds of formula IA wherein B is:

R² is –OH, and all other substituents are as defined in formula IA.

Embodiment No. 44 is directed to compounds of formula IA wherein B is:

10

15

R³ is -C(O)NR¹³R¹⁴, and all other substituents are as defined in formula IA.

Embodiment No. 45 is directed to compounds of formula IA wherein B is:

 R^3 is $-S(O)_tNR^{13}R^{14}$ (e.g., t is 2), and all other substituents are as defined in formula IA.

Embodiment No. 46 is directed to compounds of formula IA wherein B is:

$$R^{11} \xrightarrow{S} \xrightarrow{Z} Z$$

$$R^3 \qquad R^2$$

R² is –OH, R³ is –C(O)NR¹³R¹⁴, and all other substituents are as defined in formula IA.

Embodiment No. 47 of this invention is directed to compounds of formula IA wherein B is:

 R^2 is -OH, and R^3 is $-S(O)_tNR^{13}R^{14}$ (e.g., t is 2), and all other substituents are as defined in formula IA.

Embodiment No. 48 is directed to compounds of formula IA wherein B is:

 R^2 is –OH, R^3 is –C(O)NR¹³R¹⁴, R^{11} is H, and all other substituents are as defined in formula IA.

Embodiment No. 49 is directed to compounds of formula IA wherein B is:

 R^2 is -OH, R^3 is $-S(O)_tNR^{13}R^{14}$ (e.g., t is 2), R^{11} is H, and all other substituents are as defined in formula IA.

Embodiment No. 50 is directed to compounds of formula IA wherein B is:

15

5

10

R² is –OH, R³ is –C(O)NR¹³R¹⁴, R¹¹ is H, and R¹³ and R¹⁴ are independently selected from the group consisting of: alkyl, unsubstituted heteroaryl and substituted heteroaryl, and all other substituents are as defined in formula IA. For example, one of R¹³ or R¹⁴ is alkyl (e.g., methyl). An example of a substituted heteroaryl group is

20

Embodiment No. 51 is directed to compounds of formula IA wherein B is:

 R^2 is -OH, R^3 is $-S(O)_tNR^{13}R^{14}$ (e.g., t is 2), R^{11} is H, R^{13} and R^{14} are independently selected from the group consisting of:H and alkyl (e.g., methyl, ethyl, isopropyl, and t-butyl), and all other substituents are as defined in formula IA. For example R^3 is $(1) -SO_2NH_2$ and $(2) -SO_2NR^{13}R^{14}$ wherein R^{13} and R^{14} are the same or different alkyl group (e.g., methyl, ethyl, isopropyl and t-butyl), e.g., the same alkyl group, such as, for example $-SO_2N(CH_3)_2$.

Embodiment No. 52 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

$$R^{13}$$
 R^{14}
 R^{12}
 R^{12}
 R^{12}
 R^{12}
 R^{12}
 R^{12}
 R^{13}
 R^{14}
 R^{14}
 R^{15}
 R^{12}
 R^{10}
 R^{2}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{2}
 R^{3}
 R^{2}

and

5

10

$$R^3$$
 R^2
 R^2
 R^2

wherein R² to R⁶ and R¹⁰ to R¹⁴ are as defined above for the compounds of formula IA.

Embodiment No. 53 is directed to compounds of formula IA wherein substituent

B in formula is selected from the group consisting of:

15

R⁵

R¹²

R¹²

R¹²

R¹²

$$R^{13} \xrightarrow{R^4} \xrightarrow{R^6} \xrightarrow{R^3} \xrightarrow{R^{12}} \xrightarrow{R^{1$$

wherein

R² is selected from the group consisting of: H, OH, -NHC(O)R¹³ or and -NHSO₂R¹³;

5

10

15

20

25

30

 R^3 is selected from the group consisting of: $-SO_2NR^{13}R^{14}$, $-NO_2$, cyano, $-C(O)NR^{13}R^{14}$, $-SO_2R^{13}$; and $-C(O)OR^{13}$;

R⁴ is selected from the group consisting of: H, -NO₂, cyano, -CH₃, halogen, and -CF₃;

R⁵ is selected from the group consisting of: H, -CF₃, -NO₂, halogen and cyano; R⁶ is selected from the group consisting of: H, alkyl and -CF₃;

each R^{10} and R^{11} is independently selected from the group consisting of: R^{13} , hydrogen, halogen, $-CF_3$, $-NR^{13}R^{14}$, $-NR^{13}C(O)NR^{13}R^{14}$, $-C(O)OR^{13}$, -SH, $-SO_{(1)}NR^{13}R^{14}$, $-SO_2R^{13}$, $-NHC(O)R^{13}$, $-NHSO_2NR^{13}R^{14}$, $-NHSO_2R^{13}$, $-C(O)NR^{13}OR^{14}$, $-OC(O)R^{13}$, $-COR^{13}$, $-OR^{13}$, and cyano;

each R¹³ and R¹⁴ is independently selected from the group consisting of: H, methyl, ethyl and isopropyl; or

R¹³ and R¹⁴ when taken together with the nitrogen they are attached to in the groups -NR¹³R¹⁴, -C(O)NR¹³R¹⁴, -SO₂NR¹³R¹⁴, -OC(O)NR¹³R¹⁴, -CONR¹³R¹⁴. -NR¹³C(O)NR¹³R¹⁴, -SO_tNR¹³R¹⁴, -NHSO₂NR¹³R¹⁴ form an unsubstituted or substituted saturated heterocyclic ring (preferably a 3 to 7 membered ring) optionally having one additional heteroatom selected from the group consisting of: O, S or NR¹⁸; wherein R¹⁸ is selected from the group consisting of: H, alkyl, aryl, heteroaryl, -C(O)R¹⁹, -SO₂R¹⁹ and -C(O)NR¹⁹R²⁰; wherein each R¹⁹ and R²⁰ is independently selected from the group consisting of: alkyl, aryl and heteroaryl; wherein there are 1 to 3 substituents on the substituted cyclized R¹³ and R¹⁴ groups (i.e., the substituents on the ring formed when R¹³ and R¹⁴ are taken together with the nitrogen to which they are bound) and each substituent is independently selected from the group consisting of: alkyl, aryl, hydroxy, hydroxyalkyl, alkoxy, alkoxyalkyl, arylalkyl, fluoroalkyl, cycloalkyl, cycloalkylalkyl, heteroaryl, heteroarylalkyl, amino, -C(O)OR¹⁵, -C(O)NR¹⁵R¹⁶, -SO_tNR¹⁵R¹⁶, -C(O)R¹⁵, -SO₂R¹⁵ provided that R¹⁵ is not H, -NHC(O)NR¹⁵R¹⁶ and halogen; and wherein each R¹⁵ and R¹⁶ is independently selected from the group consisting: of H, alkyl, aryl, arylalkyl, cycloalkyl and heteroaryl.

Embodiment No. 54 is directed to compounds of formula IA wherein substituent B in formula selected from the group consisting of:

$$R^{13}$$
 R^{14}
 R^{14}
 R^{10}
 R^{2}
 R^{2}
 R^{11}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}

wherein:

5

10

15

20

25

R² is selected from the group consisting of: H, OH, -NHC(O)R¹³ and -NHSO₂R¹³;

R³ is selected from the group consisting of: -C(O)NR¹³R¹⁴, -SO₂NR¹³R¹⁴, -NO₂, cyano, -SO₂R¹³; and -C(O)OR¹³;

R⁴ is selected from the group consisting of: H, -NO₂, cyano, -CH₃ or -CF₃;
R⁵ is selected from the group consisting of: H, -CF₃, -NO₂, halogen and cyano; and

R⁶ is selected from the group consisting of: H, alkyl and -CF₃;
R¹¹ is selected from the group consisting of: H, halogen and alkyl; and each R¹³ and R¹⁴ is independently selected from the group consisting of: H, methyl, ethyl and isopropyl; or

R¹³ and R¹⁴ when taken together with the nitrogen they are attached to in the groups -NR¹³R¹⁴, -C(O)NR¹³R¹⁴, -SO₂NR¹³R¹⁴, -OC(O)NR¹³R¹⁴, -CONR¹³R¹⁴, -NR¹³C(O)NR¹³R¹⁴, -SO₁NR¹³R¹⁴, -NHSO₂NR¹³R¹⁴ form an unsubstituted or substituted saturated heterocyclic ring (preferably a 3 to 7 membered ring) optionally having one additional heteroatom selected from O, S or NR¹⁸ wherein R¹⁸ is selected from H, alkyl, aryl, heteroaryl, -C(O)R¹⁹, -SO₂R¹⁹ and -C(O)NR¹⁹R²⁰, wherein each R¹⁹ and R²⁰ is independently selected from alkyl, aryl and heteroaryl, wherein there are 1 to 3 substituents on the substituted cyclized R¹³ and R¹⁴ groups (i.e., on the ring formed when R¹³ and R¹⁴ are taken together with the nitrogen to which they are bound) and each substituent is independently selected from the group consisting of: alkyl, aryl, hydroxy, hydroxyalkyl, alkoxy, alkoxyalkyl, arylalkyl, fluoroalkyl, cycloalkyl, cycloalkyl, heteroaryl, heteroarylalkyl, amino, -C(O)OR¹⁵, -C(O)NR¹⁵R¹⁶, -SO₁NR¹⁵R¹⁶, -C(O)R¹⁵, -SO₂R¹⁵ provided that R¹⁵ is not H, -NHC(O)NR¹⁵R¹⁶ and halogen; and wherein each R¹⁵ and R¹⁶ is independently selected from the group consisting of: H, alkyl, aryl, arylalkyl, cycloalkyl and heteroaryl.

Embodiment No. 55 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

$$R^{13}$$
 R^{14}
 R^{14}
 R^{14}
 R^{15}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{3}
 R^{4}

wherein:

5

10

15

R² is selected from the group consisting of: H, OH, -NHC(O)R¹³ and -NHSO₂R¹³;

 R^3 is selected from the group consisting of: $-C(O)NR^{13}R^{14}$ $-SO_2NR^{13}R^{14}$, $-NO_2$, cyano, and $-SO_2R^{13}$;

R⁴ is selected from the group consisting of: H, -NO₂, cyano, -CH₃ or -CF₃;

R⁵ is selected from the group consisting of: H, -CF₃, -NO₂, halogen and cyano; and

R⁶ is selected from the group consisting of: H, alkyl and -CF₃;

R¹¹ is selected from the group consisting of: H, halogen and alkyl; and each R¹³ and R¹⁴ is independently selected from the group consisting of: H, methyl and ethyl.

Embodiment No. 56 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

$$R^{13}$$
 R^{14}
 R^{14}
 R^{14}
 R^{14}
 R^{14}
 R^{14}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{15}
 R^{2}
 R^{3}
 R^{2}
 R^{2}
 R^{3}

wherein:

 R^2 is -OH;

R³ is selected from the group consisting of: -SO₂NR¹³R¹⁴ and -CONR¹³R¹⁴;

R⁴ is selected form the group consisting of: H, -CH₃ and -CF₃;

R⁵ is selected from the group consisting of: H and cyano;

R⁶ is selected from the group consisting of: H, -CH₃ and -CF₃;

25 R¹¹ is H; and

 R^{13} and R^{14} are independently selected from the group consisting of H and methyl (e.g., for -SO₂NR¹³R¹⁴ both R¹³ and R¹⁴ are H, or both R¹³ and R¹⁴ are methyl, also, for example, for -CONR¹³R¹⁴ both R¹³ and R¹⁴ are methyl).

Embodiment No. 57 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

$$R_4$$
 R_5
 R_6
 R_7
 R_8
 R_8
 R_8
 R_8
 R_9
 R_9

wherein all substituents are as defined for formula IA.

5

15

Embodiment No. 58 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

Embodiment No. 59 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

and
$$H_2N-S$$
 OH

5

10

Embodiment No. 60 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

Embodiment No. 61 is directed to compounds of formula IA wherein substituent B is selected from the group consisting of:

Embodiment No. 62 is directed to compounds of formula IA wherein substituent B is:

Embodiment No. 63 is directed to compounds of formula IA wherein substituent B is:

Embodiment No. 64 is directed to compounds of formula IA wherein substituent B is:

Embodiment No. 65 is directed to compounds of formula IA wherein: substituent A is selected from the group consisting of:

substituent A is selected from the group consisting of:

(a)

R⁷ R⁸

R⁸

R⁸

R⁹ R⁹

R

wherein the above rings are unsubstituted or substituted, as described for formula IA: and

5

10

15

wherein in (a) and (b): each R⁷ and R⁸ is independently selected from the group consisting of: H, unsubstituted or substituted alkyl, unsubstituted or substituted aryl, unsubstituted or substituted arylalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted cycloalkylalkyl, -CO₂R¹³, -CONR¹³R¹⁴, fluoroalkyl, alkynyl, alkenyl, and cycloalkenyl, wherein said substituents on said R⁷ and R⁸ substituted groups are selected from the group consisting of: a) cyano, b) -CO₂R¹³,

c) $-C(O)NR^{13}R^{14}$, d) $-SO_2NR^{13}R^{14}$, e) $-NO_2$, f) $-CF_3$, g) $-OR^{13}$, h) $-NR^{13}R^{14}$, i) $-OC(O)R^{13}$, j) $-OC(O)NR^{13}R^{14}$, and k) halogen; and R^{8a} and R^{9} are as defined in formula IA.

Embodiment No. 66 is directed to compounds of formula IA wherein substituent A is selected from the group consisting of:

20

5

10

$$\mathbb{R}^7$$
 \mathbb{R}^8 \mathbb{R}^7 \mathbb{R}^8 and \mathbb{R}^8 \mathbb{R}^8 ,

wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: halogen, alkyl, cycloalkyl, -CF₃, cyano, -OCH₃, and -NO₂; each R⁷ and R⁸ is independently selected from the group consisting of: H, alkyl (e.g., methyl, ethyl, t-butyl, and isopropyl), fluoroalkyl (such as, -CF₃ and -CF₂CH₃), cycloalkyl (e.g., cyclopropyl, and cyclohexyl), and cycloalkylalkyl (e.g., cyclopropylmethyl); and R⁹ is selected from the group consisting of: H, halogen, alkyl, cycloalkyl, -CF₃, cyano, -OCH₃, and -NO₂; and (b)

wherein each R^7 and R^8 is independently selected from the group consisting of: H, alkyl (e.g., methyl, ethyl, t-butyl, and isopropyl), fluoroalkyl (such as, -CF₃ and -CF₂CH₃), cycloalkyl (e.g.,cyclopropyl, and cyclohexyl), and cycloalkylalkyl (e.g., cyclopropylmethyl); wherein R^{8a} is as defined in formula IA, and wherein R^9 is selected from the group consisting of: H, halogen, alkyl, cycloalkyl, -CF₃, cyano, -OCH₃, and -NO₂; each R^7 and R^8 is independently selected from the group consisting of: H, alkyl (e.g., methyl, ethyl, t-butyl, and isopropyl), fluoroalkyl (such as, -CF₃ and -CF₂CH₃), cycloalkyl (e.g.,cyclopropyl, and cyclohexyl), and cycloalkylalkyl (e.g., cyclopropylmethyl).

Embodiment No. 67 is directed to the novel compounds of formula IA wherein substituent A is selected from the group consisting of:

(a) $R^{7}R^{8} \qquad R^{7}R^{8} \qquad$

wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: H, F, Cl, Br, alkyl, cycloalkyl, and –CF₃; R⁷ is selected from the group consisting of: H, fluoroalkyl, alkyl and cycloalkyl; R⁸ is selected form the group consisting of: H, alkyl, -CF₂CH₃ and -CF₃; and R⁹ is selected from the group consisting of: H, F, Cl, Br, alkyl or -CF₃; and

(b)
$$R^7 R^8$$

$$X R^{8a}$$

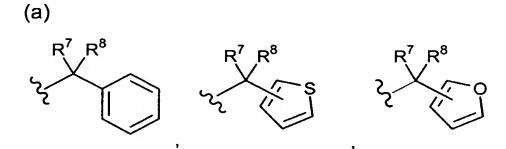
10

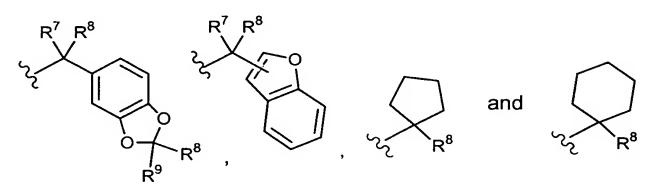
15

wherein R⁷ is selected from the group consisting of: H, fluoroalkyl, alkyl and cycloalkyl; R⁸ is selected form the group consisting of: H, alkyl, -CF₂CH₃ and -CF₃; and R^{8a} is as defined for formula IA.

Embodiment No. 68 is directed to compounds of formula IA wherein substituent

A is selected from the group consisting of:





wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: H, F, Cl, Br, alkyl, cycloalkyl, and –CF₃; R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and

10

15

wherein R^7 is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R^8 is H; and R^{8a} is as defined for formula IA.

Embodiment No. 69 is directed compounds of formula IA wherein substituent A is selected from the group consisting of:

wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: F, Cl, Br, alkyl, cycloalkyl, and –CF₃; R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and

5

10

15

wherein R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and R^{8a} is as defined for formula IA;

Embodiment No. 70 is directed compounds of formula IA wherein substituent A is selected from the group consisting of:

(1) unsubstituted or substituted:

$$\mathbb{R}^7$$
 \mathbb{R}^8 \mathbb{R}^7 \mathbb{R}^8 \mathbb{R}^9 \mathbb

$$\begin{array}{c} (2) \\ \begin{array}{c} R_7 \\ \\ \end{array} \\ \begin{array}{c} R_8 \\ \\ \end{array} \\ R_{8a} \end{array}$$

wherein all substitutents are as defined for formula IA.

5

Embodiment No. 71 is directed to compounds of formula IA wherein substituent A is selected from the group consisting of:

Embodiment No. 72 is directed to compounds of formula IA wherein substituent A is selected from the group consisting of:

Embodiment No. 73 is directed to compounds of formula IA wherein substituent A is selected from the group consisting of:

Embodiment No. 74 is directed to compounds of formula IA wherein substituent

A is selected from the group consisting of:

Embodiment No. 75 is directed to compounds of formula IA wherein substituent A is selected from the group consisting of:

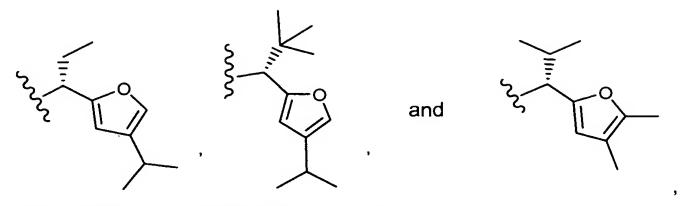
Embodiment No. 76 is directed to compounds of formula IA wherein substituent A is:

Embodiment No. 77 is directed to compounds of formula IA wherein substituent A is:

Embodiment No. 78 is directed to compounds of formula IA wherein substituent A is:

10

Embodiment No. 79 is directed to compounds of formula IA wherein substituent A is selected from the group consisting of:



and substituent B is selected from the group consisting of:

Embodiment No. 80 is directed to compounds of formula IA wherein substituent

A is selected from the group consisting of:

and substituent B is selected from the group consisting of:

1.

10

Embodiment No. 81 is directed to novel compounds of formula IA wherein g is

Embodiment No. 82 is directed to novel compounds as described in any one of Embodiments Nos. 1-80 wherein g is 1.

Embodiment No. 83 is directed to novel compounds is directed to novel compounds of formula IA wherein g is 2.

Embodiment No. 84 is directed to novel compounds as described in any one of Embodiments Nos. 1-80 wherein g is 2.

Embodiment No. 85 is directed to novel compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is as defined in any one of the Embodiment Nos. 65 to 78.

Embodiment No. 86 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is:

and all other substituents are as defined for formula IA.

Embodiment No. 87 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is:

10

5

wherein R⁷ is H, and R⁸ is alkyl (e.g., methyl, ethyl, isopropyl, cyclopropyl and t-butyl), and all other substituents are as defined for formula IA.

Embodiment No. 88 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is:

15

and all other substituents are as defined for formula IA.

Embodiment No. 89 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is:

wherein the furan ring is unsubstituted or substituted as described in the definition of A for formula IA, and all other substituents are as defined for formula IA.

Embodiment No. 90 is directed to compounds of formula IA wherein B is described in any one of the Embodiment Nos. 1 to 64, and A is

5

wherein the furan ring is substituted and all other substituents are as defined for formula IA.

Embodiment No. 91 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64,and A is

10

wherein the furan ring is substituted with at least one (e.g., 1 to 3, or 1 to 2) alkyl group and all other substituents are as defined for formula IA.

Embodiment No. 92 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, A is

15

wherein the furan ring is substituted with one alkyl group and all other substituents are as defined for formula IA.

Embodiment No. 93 is directed to compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is

20

wherein the furan ring is substituted with one C₁ to C₃ alkyl group (e.g., methyl or isopropyl), and all other substituents are as defined for formula IA.

Embodiment No. 94 is directed to novel compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is as defined in any

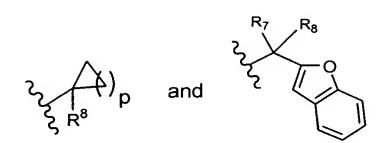
one of the Embodiment Nos. 89 to 93, except that R⁷ and R⁸ are the same or different and each is selected from the group consisting of: H and alkyl.

Embodiment No. 95 is directed to novel compounds of formula IA wherein B is as described in any one of the Embodiment Nos. 1 to 64, and A is as defined in any one of the Embodiment Nos. 89 to 93, except that R⁷ is H, and R⁸ is alkyl (e.g., ethyl or t-butyl).

Embodiment No. 96 is directed to compounds of formula IA wherein:

(1) substituent A in formula IA is selected from the group consisting of:

15



wherein the above rings are unsubstituted or substituted, as described for formula IA: and

wherein in (a) and (b) above: each R^7 and R^8 is independently selected from the group consisting of: H, unsubstituted or substituted alkyl, unsubstituted or substituted aryl, unsubstituted or substituted heteroaryl, unsubstituted or substituted arylalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted cycloalkyl, unsubstituted or substituted cycloalkylalkyl, $-CO_2R^{13}$, $-CONR^{13}R^{14}$, fluoroalkyl, alkynyl, alkenyl, and cycloalkenyl, wherein said substituents on said R^7 and R^8 substituted groups are selected from the group consisting of: a) cyano, b) $-CO_2R^{13}$, c) $-C(O)NR^{13}R^{14}$, d) $-SO_2NR^{13}R^{14}$, e) $-NO_2$, f) $-CF_3$, g) $-OR^{13}$, h) $-NR^{13}R^{14}$, i) $-OC(O)R^{13}$, j) $-OC(O)NR^{13}R^{14}$, and k) halogen; and R^8 are as defined in formula IA; and

(2) substituent B in formula IA is selected from the group consisting of:

wherein R^2 to R^6 and R^{10} to R^{14} are as defined above for the novel compounds of formula IA .

Embodiment No. 97 is directed to compounds of formula IA wherein:

(1) substituent A in formula IA is selected from the group consisting

5 **of**:

(a)
$$R^{7} R^{8}$$

15

10

wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: halogen, alkyl, cycloalkyl, -CF₃, cyano, -OCH₃, and -NO₂; each R⁷ and R⁸ is independently selected

from the group consisting of: H, alkyl (e.g., methyl, ethyl, t-butyl, and isopropyl), fluoroalkyl (such as, -CF₃ and -CF₂CH₃), cycloalkyl (e.g.,cyclopropyl, and cyclohexyl), and cycloalkylalkyl (e.g., cyclopropylmethyl); and R⁹ is selected from the group consisting of: H, halogen, alkyl, cycloalkyl, -CF₃, cyano, -OCH₃, and -NO₂; and

(b)

wherein each R^7 and R^8 is independently selected from the group consisting of: H, alkyl (e.g., methyl, ethyl, t-butyl, and isopropyl), fluoroalkyl (such as, -CF₃ and -CF₂CH₃), cycloalkyl (e.g.,cyclopropyl, and cyclohexyl), and cycloalkylalkyl (e.g., cyclopropylmethyl); wherein R^{8a} is as defined in formula IA, and wherein R^9 is selected from the group consisting of: H, halogen, alkyl, cycloalkyl, -CF₃, cyano, -OCH₃, and -NO₂; each R^7 and R^8 is independently selected from the group consisting of: H, alkyl (e.g., methyl, ethyl, t-butyl, and isopropyl), fluoroalkyl (such as, -CF₃ and -CF₂CH₃), cycloalkyl (e.g.,cyclopropyl, and cyclohexyl), and cycloalkylalkyl (e.g., cyclopropylmethyl); and

(2) substituent B in formula IA is selected from the group consisting of:

20 wherein

5

10

15

R² is selected from the group consisting of: H, OH, -NHC(O)R¹³ or and -NHSO₂R¹³;

R³ is selected from the group consisting of: -SO₂NR¹³R¹⁴, -NO₂, cyano, -C(O)NR¹³R¹⁴, -SO₂R¹³; and -C(O)OR¹³;

R⁴ is selected from the group consisting of: H, -NO₂, cyano, -CH₃, halogen, and -CF₃;

R⁵ is selected from the group consisting of: H, -CF₃, -NO₂, halogen and cyano; R⁶ is selected from the group consisting of: H, alkyl and -CF₃;

each R^{10} and R^{11} is independently selected from the group consisting of: R^{13} , hydrogen, halogen, $-CF_3$, $-NR^{13}R^{14}$, $-NR^{13}C(O)NR^{13}R^{14}$, $-C(O)OR^{13}$, -SH, $-SO_{(t)}NR^{13}R^{14}$, $-SO_2R^{13}$, $-NHC(O)R^{13}$, $-NHSO_2NR^{13}R^{14}$, $-NHSO_2R^{13}$, $-C(O)NR^{13}OR^{14}$, $-OC(O)R^{13}$, $-COR^{13}$, $-OR^{13}$, and cyano;

each R¹³ and R¹⁴ is independently selected from the group consisting of: H, methyl, ethyl and isopropyl; or

R¹³ and R¹⁴ when taken together with the nitrogen they are attached to in the groups -NR¹³R¹⁴, -C(O)NR¹³R¹⁴, -SO₂NR¹³R¹⁴, -OC(O)NR¹³R¹⁴, -CONR¹³R¹⁴, -NR¹³C(O)NR¹³R¹⁴, -SO_tNR¹³R¹⁴, -NHSO₂NR¹³R¹⁴ form an unsubstituted or substituted saturated heterocyclic ring (preferably a 3 to 7 membered ring) optionally having one additional heteroatom selected from the group consisting of: O, S or NR¹⁸; wherein R¹⁸ is selected from the group consisting of: H, alkyl, aryl, heteroaryl, -C(O)R¹⁹, -SO₂R¹⁹ and -C(O)NR¹⁹R²⁰; wherein each R¹⁹ and R²⁰ is independently selected from the group consisting of: alkyl, aryl and heteroaryl; wherein there are 1 to 3 substituents on the substituted cyclized R¹³ and R¹⁴ groups (i.e., the substituents on the ring formed when R¹³ and R¹⁴ are taken together with the nitrogen to which they are bound) and each substituent is independently selected from the group consisting of: alkyl, aryl, hydroxy, hydroxyalkyl, alkoxy, alkoxyalkyl, arylalkyl, fluoroalkyl, cycloalkyl, cycloalkylalkyl, heteroaryl, heteroarylalkyl, amino, -C(O)OR¹⁵, -C(O)NR¹⁵R¹⁶, -SO_tNR¹⁵R¹⁶, -C(O)R¹⁵, -SO₂R¹⁵ provided that R¹⁵ is not H, -NHC(O)NR¹⁵R¹⁶ and halogen; and wherein each R¹⁵ and R¹⁶ is independently selected from the group consisting: of H, alkyl, aryl, arylalkyl, cycloalkyl and heteroaryl.

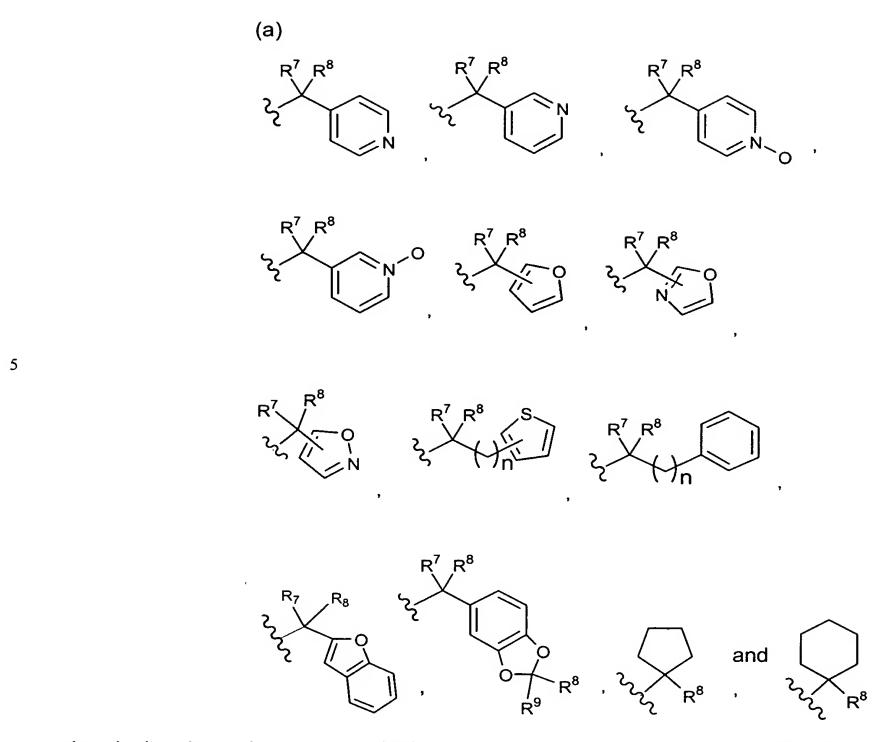
Embodiment No. 98 is directed to compounds of formula IA wherein substituent A in formula IA is even more preferably selected from the group consisting of:

5

10

15

20



wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: H, F, Cl, Br, alkyl, cycloalkyl, and –CF₃; R⁷ is selected from the group consisting of: H, fluoroalkyl, alkyl and cycloalkyl; R⁸ is selected form the group consisting of: H, alkyl, -CF₂CH₃ and -CF₃; and R⁹ is selected from the group consisting of: H, F, Cl, Br, alkyl or -CF₃; and

10

15

wherein R⁷ is selected from the group consisting of: H, fluoroalkyl, alkyl and cycloalkyl; R⁸ is selected form the group consisting of: H, alkyl, -CF₂CH₃ and -CF₃; and R^{8a} is as defined for formula IA.

Embodiment No. 99 is directed to compounds of formula IA wherein:

(1) substituent A in formula IA is selected from the group consisting

5

10

15

of:

wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: H, F, Cl, Br, alkyl, cycloalkyl, and –CF₃; R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and

(b)

R⁷ R⁸

2 P8a

wherein R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and R^{8a} is as defined for formula IA.

(2) substituent B in formula IA is selected from the group consisting of:

$$R^{13}$$
 R^{14}
 R^{14}
 R^{14}
 R^{14}
 R^{14}
 R^{2}
 R^{2}
 R^{2}
 R^{14}
 R^{2}
 R^{3}
 R^{2}
 R^{2}
 R^{3}

20 wherein:

R² is selected from the group consisting of: H, OH, -NHC(O)R¹³ and -NHSO₂R¹³;

 R^3 is selected from the group consisting of: $-C(O)NR^{13}R^{14}$, $-SO_2NR^{13}R^{14}$, $-NO_2$, cyano, $-SO_2R^{13}$; and $-C(O)OR^{13}$;

R⁴ is selected from the group consisting of: H, -NO₂, cyano, alkyl (e.g., -CH₃ and ethyl), -CF₃, and halogen;

R⁵ is selected from the group consisting of: H, -CF₃, -NO₂, halogen and cyano; and

R⁶ is selected from the group consisting of: H, alkyl and -CF₃;

R¹¹ is selected from the group consisting of: H, halogen and alkyl; and each R¹³ and R¹⁴ is independently selected from the group consisting of: H, methyl, ethyl and isopropyl; or

 R^{13} and R^{14} when taken together with the nitrogen they are attached to in the groups -NR¹³R¹⁴, -C(O)NR¹³R¹⁴, -SO₂NR¹³R¹⁴, -OC(O)NR¹³R¹⁴, -CONR¹³R¹⁴, -NR¹³C(O)NR¹³R¹⁴, -SO₁NR¹³R¹⁴, -NHSO₂NR¹³R¹⁴ form an unsubstituted or substituted saturated heterocyclic ring (preferably a 3 to 7 membered ring) optionally having one additional heteroatom selected from O, S or NR¹⁸ wherein R¹⁸ is selected from H, alkyl, aryl, heteroaryl, -C(O)R¹⁹, -SO₂R¹⁹ and -C(O)NR¹⁹R²⁰, wherein each R¹⁹ and R²⁰ is independently selected from alkyl, aryl and heteroaryl, wherein there are 1 to 3 substituents on the substituted cyclized R¹³ and R¹⁴ groups (i.e., on the ring formed when R¹³ and R¹⁴ are taken together with the nitrogen to which they are bound) and each substituent is independently selected from the group consisting of: alkyl, aryl, hydroxy, hydroxyalkyl, alkoxy, alkoxyalkyl, arylalkyl, fluoroalkyl, cycloalkyl, cycloalkylalkyl, heteroaryl, heteroarylalkyl, amino, -C(O)OR¹⁵, -C(O)NR¹⁵R¹⁶, -SO₁NR¹⁵R¹⁶, -C(O)R¹⁵, -SO₂R¹⁵ provided that R¹⁵ is not H, -NHC(O)NR¹⁵R¹⁶ and halogen; and wherein each R¹⁵ and R¹⁶ is independently selected from the group consisting of: H, alkyl, aryl, arylalkyl, cycloalkyl and heteroaryl.

Embodiment No. 100 is directed to compounds of formula IA wherein:

(1) substituent A in formula IA is selected from the group consisting

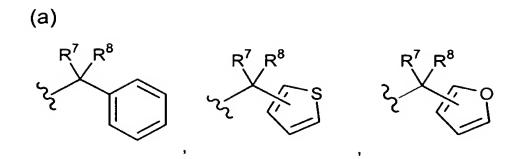
30 of:

5

10

15

20



wherein the above rings are unsubstituted, or the above rings are substituted with 1 to 3 substituents independently selected from the group consisting of: F, Cl, Br, alkyl, cycloalkyl, and –CF₃; R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and

,b) R⁷

Sylvania Real

wherein R⁷ is selected from the group consisting of: H, -CF₃, -CF₂CH₃, methyl, ethyl, isopropyl, cyclopropyl and t-butyl; and R⁸ is H; and R^{8a} is as defined for formula IA;

(2) substituent B in formula IA is selected from the group consisting of:

wherein:

10

15

20

 R^2 is selected from the group consisting of: H, OH, -NHC(O) R^{13} and -NHSO $_2R^{13}$;

 R^3 is selected from the group consisting of: $-C(O)NR^{13}R^{14}$ - $SO_2NR^{13}R^{14}$, - NO_2 , cyano, and - SO_2R^{13} ;

 R^4 is selected from the group consisting of: H, -NO₂, cyano, alkyl (e.g., -CH₃ and ethyl), -CF₃ and halogen;

R⁵ is selected from the group consisting of: H, -CF₃, -NO₂, halogen and cyano; and

R⁶ is selected from the group consisting of: H, alkyl and -CF₃;

R¹¹ is selected from the group consisting of: H, halogen and alkyl; and each R¹³ and R¹⁴ is independently selected from the group consisting of: H and unsubstituted alkyl (e.g., methyl and ethyl).

Embodiment No. 101 is directed to compounds of formula IA wherein:

(1) substituent A in formula IA is selected from the group consisting of:

5

(2) substituent B in formula IA is selected from the group consisting of:

5 wherein:

10

15

20

25

 R^2 is -OH:

R³ is selected from the group consisting of: -SO₂NR¹³R¹⁴ and -CONR¹³R¹⁴:

R⁴ is selected form the group consisting of: H, Br, -CH₃, ethyl and -CF₃;

R⁵ is selected from the group consisting of: H and cyano;

R⁶ is selected from the group consisting of: H, -CH₃ and -CF₃;

R¹¹ is H; and

 R^{13} and R^{14} are independently selected from the group consisting of H and methyl (e.g., for -SO₂NR¹³R¹⁴ both R¹³ and R¹⁴ are H, or both R¹³ and R¹⁴ are methyl, also, for example, for -CONR¹³R¹⁴ both R¹³ and R¹⁴ are methyl).

Embodiment No. 102 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 70 and substituent B is as defined in Embodiment No. 57.

Embodiment No. 103 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 70 and substituent B is as defined in Embodiment No. 58.

Embodiment No. 104 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 70 and substituent B is as defined in Embodiment No. 59.

Embodiment No. 105 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 71 and substituent B is as defined in Embodiment No. 57.

Embodiment No. 106 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 71 and substituent B is as defined in Embodiment No. 58.

Embodiment No. 107 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 71 and substituent B is as defined in Embodiment No. 59.

Embodiment No. 108 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 72 and substituent B is as defined in Embodiment No. 57.

5

10

15

20

25

30

Embodiment No. 109 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 72 and substituent B is as defined in Embodiment No. 58.

Embodiment No. 110 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 72 and substituent B is as defined in Embodiment No. 59.

Embodiment No. 111 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 73 and substituent B is as defined in Embodiment No. 57.

Embodiment No. 112 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 73 and substituent B is as defined in Embodiment No. 58.

Embodiment No. 113 is directed to compounds of formula IA wherein substituent A is as defined in Embodiment No. 73 and substituent B is as defined in Embodiment No. 59.

Embodiment No. 114 is directed to any one of the Embodiment Nos. 1 to 113 wherein the compound of formula IA is a pharmaceutically acceptable salt.

Embodiment No. 115 is directed to any one of the Embodiment Nos. 1 to 113 wherein the compound of formula IA is a sodium salt.

Embodiment No. 116 is directed to any one of the Embodiment Nos. 1 to 113 wherein the compound of formula IA is a calcium salt.

Embodiment No. 117 is directed to a pharmaceutically acceptable salt of any one of the representative compounds of this invention that are described below.

Embodiment No. 118 is directed to a sodium salt of any one of the representative compounds described below.

Embodiment No. 119 is directed to a calcium salt of any one of the representative compounds described below.

Embodiment No. 120 is directed to a pharmaceutical composition comprising at least one (e.g., 1 to 3, usually 1) compound of formula IA as described in any one of Embodiment Nos. 1 to 119 in combination with a pharmaceutically acceptable carrier (or diluent). When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

5

10

15

20

25

30

Embodiment No. 121 is directed to a method of treating any one of the diseases described herein (i.e., the chemokine mediated diseases) comprising administering to a patient in need of such treatment an effective amount (e.g., a therapeutically effective amount) of a compound of formula IA as described in any one of the Embodiment Nos. 1 to 119.

Embodiment No. 122 is directed to a method of treating any one of the diseases described herein (i.e., the chemokine mediated diseases) comprising administering to a patient in need of such treatment an effective amount (e.g., a therapeutically effective amount) of the pharmaceutical composition described in Embodiment No. 120.

Embodiment No. 123 is directed to a method of treating rheumatoid arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 124 is directed to a method of treating rheumatoid arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

Embodiment No. 125 is directed to a method of treating rheumatoid arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually 1) compound from any of Embodiment Nos. 1 to 119 in combination with at least one compound selected from the group consisting of COX-2 inhibitors, COX inhibitors, immunosuppressives (e.g., methotrexate, cyclosporin, leflunimide and sulfasalazine), steroids (e.g., betamethasone, cortisone and dexamethasone), PDE IV inhibitors, anti-TNF- α compounds, MMP inhibitors, glucocorticoids, chemokine inhibitors, CB2-selective inhibitors, and other classes of compounds indicated for the treatment of rheumatoid

arthritis. When more than one compound of Embodiment Nos. 1 to 119 is used, each compound is independently selected from said Embodiment Numbers.

Embodiment No. 126 is directed to a method of treating rheumatoid arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment 120 in combination with at least one compound selected from the group consisting of COX-2 inhibitors, COX inhibitors, immunosuppressives (e.g., methotrexate, cyclosporin, leflunimide and sulfasalazine), steroids (e.g., betamethasone, cortisone and dexamethasone), PDE IV inhibitors, anti-TNF- α compounds, MMP inhibitors, glucocorticoids, chemokine inhibitors, CB2-selective inhibitors, and other classes of compounds indicated for the treatment of rheumatoid arthritis.

5

10

15

20

25

30

Embodiment No. 127 is directed to a method of treating COPD in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 128 is directed to a method of treating COPD in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment 120.

Embodiment No. 129 is directed to a method of treating acute pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 130 is directed to a method of treating acute pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

Embodiment No. 131 is directed to a method of treating acute inflammatory pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of

Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 132 is directed to a method of treating acute inflammatory pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

Embodiment No. 133 is directed to a method of treating chronic inflammatory pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 134 is directed to a method of treating chronic inflammatory pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

Embodiment No. 135 is directed to a method of treating neuropathic pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 136 is directed to a method of treating neuropathic pain in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

Embodiment No. 137 is directed to a method of treating arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 138 is directed to a method of treating arthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

30

25

5

10

15

Embodiment No. 139 is directed to a method of treating osteoarthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of at least one (usually one) compound from any of Embodiment Nos. 1 to 119. When more than one compound is used each compound is independently selected from the group consisting of Embodiment Nos. 1 to 119.

Embodiment No. 140 is directed to a method of treating osteoarthritis in a patient in need of such treatment comprising administering to said patient a therapeutically effective amount of the pharmaceutical composition described in Embodiment No. 120.

5

$$F_{3}C$$

$$F$$

the pharmaceutically acceptable salts thereof, and the pharmaceutically acceptable solvates thereof.

Preferred compounds of this invention are selected from the group consisting

5 of:

the pharmaceutically acceptable salts thereof, and the pharmaceutically acceptable solvates thereof.

More preferred compounds of this invention are selected from the group consisting of:

the pharmaceutically acceptable salts thereof, and the pharmaceutically acceptable solvates thereof.

Most preferred compounds of this invention are selected from the group consisting of:

$$\begin{array}{c} O \\ O \\ S \\ O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\$$

the pharmaceutically acceptable salts, and the pharmaceutically acceptable solvates thereof.

An embodiment of this invention is directed to compounds of formula IA selected from the group consisting of compounds of the formula:

$$\begin{pmatrix} 0 \\ N \\ N \end{pmatrix}$$

$$\begin{pmatrix} N \\ N \\ N \end{pmatrix}$$

$$\begin{pmatrix} 1 \\ N \\ N$$

/

CI
$$H_2N-S$$
 OH H_2N-S OH H

the pharmaceutically acceptable salts thereof, and the pharmaceutically acceptable solvates thereof.

Certain compounds of the invention may exist in different stereoisomeric forms (e.g., enantiomers, diastereoisomers and atropisomers). The invention contemplates all such stereoisomers both in pure form and in admixture, including racemic mixtures. Isomers can be prepared using conventional methods.

Certain compounds will be acidic in nature, e.g. those compounds which possess a carboxyl or phenolic hydroxyl group. These compounds may form pharmaceutically acceptable salts. Examples of such salts may include sodium, potassium, calcium, aluminum, gold and silver salts. Also contemplated are salts formed with pharmaceutically acceptable amines such as ammonia, alkyl amines, hydroxyalkylamines, N-methylglucamine and the like.

5

10

15

20

25

30

Certain basic compounds also form pharmaceutically acceptable salts, e.g., acid addition salts. For example, the pyrido-nitrogen atoms may form salts with strong acid, while compounds having basic substituents such as amino groups also form salts with weaker acids. Examples of suitable acids for salt formation are hydrochloric, sulfuric, phosphoric, acetic, citric, oxalic, malonic, salicylic, malic, fumaric, succinic, ascorbic, maleic, methanesulfonic and other mineral and carboxylic acids well known to those skilled in the art. The salts are prepared by contacting the free base form with a sufficient amount of the desired acid to produce a salt in the conventional manner. The free base forms may be regenerated by treating the salt with a suitable dilute aqueous base solution such as dilute aqueous NaOH, potassium carbonate, ammonia and sodium bicarbonate. The free base forms differ from their respective salt forms somewhat in certain physical properties, such as solubility in polar solvents, but the acid and base salts are otherwise equivalent to their respective free base forms for purposes of the invention.

All such acid and base salts are intended to be pharmaceutically acceptable salts within the scope of the invention and all acid and base salts are considered equivalent to the free forms of the corresponding compounds for purposes of the invention.

Compounds of formula IA can exist in unsolvated and solvated forms, including hydrated forms. In general, the solvated forms, with pharmaceutically acceptable solvents such as water, ethanol and the like, are equivalent to the unsolvated forms for the purposes of this invention.

This invention also includes Prodrugs of the novel compounds of this invention. The term "prodrug," as used herein, represents compounds which are rapidly transformed *in vivo* to the parent compound of the above formula, for example, by hydrolysis in blood. A thorough discussion is provided in T. Higuchi and V. Stella, Pro-drugs as Novel Delivery Systems, Vol. 14 of the A.C.S. Symposium Series, and in

Edward B. Roche, ed., Bioreversible Carriers in Drug Design, American Pharmaceutical Association and Pergamon Press, 1987, both of which are incorporated herein by reference.

5

10

15

20

25

30

This invention also includes the compounds of this invention in isolated and pure form.

For preparing pharmaceutical compositions from the compounds described by this invention, inert, pharmaceutically acceptable carriers can be either solid or liquid. Solid form preparations include powders, tablets, dispersible granules, capsules, cachets and suppositories. The powders and tablets may be comprised of from about 5 to about 95 percent active ingredient. Suitable solid carriers are known in the art, e.g., magnesium carbonate, magnesium stearate, talc, sugar or lactose. Tablets, powders, cachets and capsules can be used as solid dosage forms suitable for oral administration. Examples of pharmaceutically acceptable carriers and methods of manufacture for various compositions may be found in A. Gennaro (ed.), Remington: The Science and Practice of Pharmacy, 20th Edition, (2000), Lippincott Williams & Wilkins, Baltimore, MD..

Liquid form preparations include solutions, suspensions and emulsions. As an example may be mentioned water or water-propylene glycol solutions for parenteral injection or addition of sweeteners and opacifiers for oral solutions, suspensions and emulsions. Liquid form preparations may also include solutions for intranasal administration.

Aerosol preparations suitable for inhalation may include solutions and solids in powder form, which may be in combination with a pharmaceutically acceptable carrier, such as an inert compressed gas, e.g. nitrogen.

Also included are solid form preparations that are intended to be converted, shortly before use, to liquid form preparations for either oral or parenteral administration. Such liquid forms include solutions, suspensions and emulsions.

The compounds of the invention may also be deliverable transdermally. The transdermal composition can take the form of creams, lotions, aerosols and/or emulsions and can be included in a transdermal patch of the matrix or reservoir type as are conventional in the art for this purpose.

Preferably the compound is administered orally.

Preferably, the pharmaceutical preparation is in a unit dosage form. In such form, the preparation is subdivided into suitably sized unit doses containing appropriate quantities of the active component, e.g., an effective amount to achieve the desired purpose.

5

10

15

20

25

30

The quantity of active compound in a unit dose of preparation may be varied or adjusted from about 0.01 mg to about 1000 mg, preferably from about 0.01 mg to about 750 mg, more preferably from about 0.01 mg to about 500 mg, and most preferably from about 0.01 mg to about 250 mg, according to the particular application.

The actual dosage employed may be varied depending upon the requirements of the patient and the severity of the condition being treated. Determination of the proper dosage regimen for a particular situation is within the skill of the art. For convenience, the total dosage may be divided and administered in portions during the day as required.

The amount and frequency of administration of the compounds of the invention and/or the pharmaceutically acceptable salts thereof will be regulated according to the judgment of the attending clinician considering such factors as age, condition and size of the patient as well as severity of the symptoms being treated. A typical recommended daily dosage regimen for oral administration can range from about 0.04 mg/day to about 4000 mg/day, in two to four divided doses.

Classes of compounds that can be used as the chemotherapeutic agent (antineoplastic agent) include: alkylating agents, antimetabolites, natural products and their derivatives, hormones and steroids (including synthetic analogs), and synthetics. Examples of compounds within these classes are given below.

Alkylating agents (including nitrogen mustards, ethylenimine derivatives, alkyl sulfonates, nitrosoureas and triazenes): Uracil mustard, Chlormethine, Cyclophosphamide (Cytoxan®), Ifosfamide, Melphalan, Chlorambucil, Pipobroman, Triethylene-melamine, Triethylenethiophosphoramine, Busulfan, Carmustine, Lomustine, Streptozocin, Dacarbazine, and Temozolomide.

Antimetabolites (including folic acid antagonists, pyrimidine analogs, purine analogs and adenosine deaminase inhibitors): Methotrexate, 5-Fluorouracil, Floxuridine, Cytarabine, 6-Mercaptopurine, 6-Thioguanine, Fludarabine phosphate, Pentostatine, and Gemcitabine.

Natural products and their derivatives (including vinca alkaloids, antitumor antibiotics, enzymes, lymphokines and epipodophyllotoxins): Vinblastine, Vincristine, Vindesine, Bleomycin, Dactinomycin, Daunorubicin, Doxorubicin, Epirubicin, Idarubicin, paclitaxel (paclitaxel is commercially available as Taxol® and is described in more detail below in the subsection entitled "Microtubule Affecting Agents"), Mithramycin, Deoxyco-formycin, Mitomycin-C, L-Asparaginase, Interferons (especially IFN-a), Etoposide, and Teniposide.

5

10

15

20

25

30

Hormones and steroids (including synthetic analogs): 17α-Ethinylestradiol, Diethylstilbestrol, Testosterone, Prednisone, Fluoxymesterone, Dromostanolone propionate, Testolactone, Megestrolacetate, Tamoxifen, Methylprednisolone, Methyltestosterone, Prednisolone, Triamcinolone, Chlorotrianisene, Hydroxyprogesterone, Aminoglutethimide, Estramustine, Medroxyprogesteroneacetate, Leuprolide, Flutamide, Toremifene, Zoladex.

Synthetics (including inorganic complexes such as platinum coordination complexes): Cisplatin, Carboplatin, Hydroxyurea, Amsacrine, Procarbazine, Mitotane, Mitoxantrone, Levamisole, and Hexamethylmelamine.

Methods for the safe and effective administration of most of these chemotherapeutic agents are known to those skilled in the art. In addition, their administration is described in the standard literature. For example, the administration of many of the chemotherapeutic agents is described in the "Physicians' Desk Reference" (PDR), e.g., 2002 edition (Medical Economics Company, Montvale, NJ 07645-1742, USA); the disclosure of which is incorporated herein by reference thereto.

As used herein, a microtubule affecting agent is a compound that interferes with cellular mitosis, *i.e.*, having an anti-mitotic effect, by affecting microtubule formation and/or action. Such agents can be, for instance, microtubule stabilizing agents or agents that disrupt microtubule formation.

Microtubule affecting agents useful in the invention are well known to those of skill in the art and include, but are not limited to allocolchicine (NSC 406042), Halichondrin B (NSC 609395), colchicine (NSC 757), colchicine derivatives (e.g., NSC 33410), dolastatin 10 (NSC 376128), maytansine (NSC 153858), rhizoxin (NSC 332598), paclitaxel (Taxol[®], NSC 125973), Taxol[®] derivatives (e.g., derivatives (e.g., NSC 608832), thiocolchicine (NSC 361792), trityl cysteine (NSC 83265), vinblastine

sulfate (NSC 49842), vincristine sulfate (NSC 67574), epothilone A, epothilone, and discodermolide (see Service, (1996) Science, 274:2009) estramustine, nocodazole, MAP4, and the like. Examples of such agents are also described in the scientific and patent literature, see, e.g., Bulinski (1997) J. Cell Sci. 110:3055-3064; Panda (1997) Proc. Natl. Acad. Sci. USA 94:10560-10564; Muhlradt (1997) Cancer Res. 57:3344-3346; Nicolaou (1997) Nature 387:268-272; Vasquez (1997) Mol. Biol. Cell. 8:973-985; Panda (1996) J. Biol. Chem. 271:29807-29812.

5

10

15

20

25

30

Particularly preferred agents are compounds with paclitaxel-like activity. These include, but are not limited to paclitaxel and paclitaxel derivatives (paclitaxel-like compounds) and analogues. Paclitaxel and its derivatives are available commercially. In addition, methods of making paclitaxel and paclitaxel derivatives and analogues are well known to those of skill in the art (see, e.g., U.S. Patent Nos: 5,569,729; 5,565,478; 5,530,020; 5,527,924; 5,508,447; 5,489,589; 5,488,116; 5,484,809; 5,478,854; 5,478,736; 5,475,120; 5,468,769; 5,461,169; 5,440,057; 5,422,364; 5,411,984; 5,405,972; and 5,296,506).

More specifically, the term "paclitaxel" as used herein refers to the drug commercially available as Taxol® (NSC number: 125973). Taxol® inhibits eukaryotic cell replication by enhancing polymerization of tubulin moieties into stabilized microtubule bundles that are unable to reorganize into the proper structures for mitosis. Of the many available chemotherapeutic drugs, paclitaxel has generated interest because of its efficacy in clinical trials against drug-refractory tumors, including ovarian and mammary gland tumors (Hawkins (1992) *Oncology*, 6: 17-23, Horwitz (1992) *Trends Pharmacol. Sci.* 13: 134-146, Rowinsky (1990) *J. Natl. Canc. Inst.* 82: 1247-1259).

Additional microtubule affecting agents can be assessed using one of many such assays known in the art, e.g., a semiautomated assay which measures the tubulin-polymerizing activity of paclitaxel analogs in combination with a cellular assay to measure the potential of these compounds to block cells in mitosis (see *Lopes* (1997) *Cancer Chemother. Pharmacol.* 41:37-47).

Generally, activity of a test compound is determined by contacting a cell with that compound and determining whether or not the cell cycle is disrupted, in particular, through the inhibition of a mitotic event. Such inhibition may be mediated by disruption of the mitotic apparatus, *e.g.*, disruption of normal spindle formation. Cells

in which mitosis is interrupted may be characterized by altered morphology (e.g., microtubule compaction, increased chromosome number, etc.).

Compounds with possible tubulin polymerization activity can be screened *in vitro*. In a preferred embodiment, the compounds are screened against cultured WR21 cells (derived from line 69-2 wap-ras mice) for inhibition of proliferation and/or for altered cellular morphology, in particular for microtubule compaction. *In vivo* screening of positive-testing compounds can then be performed using nude mice bearing the WR21 tumor cells. Detailed protocols for this screening method are described by Porter (1995) *Lab. Anim. Sci.*, 45(2):145-150.

5

10

15

20

25

30

Other methods of screening compounds for desired activity are well known to those of skill in the art. Typically such assays involve assays for inhibition of microtubule assembly and/or disassembly. Assays for microtubule assembly are described, for example, by Gaskin *et al.* (1974) *J. Molec. Biol.*, 89: 737-758. U.S. Patent No. 5,569,720 also provides *in vitro* and *in vivo* assays for compounds with paclitaxel-like activity.

Methods for the safe and effective administration of the above-mentioned microtubule affecting agents are known to those skilled in the art. In addition, their administration is described in the standard literature. For example, the administration of many of the chemotherapeutic agents is described in the "Physicians' Desk Reference" (PDR), e.g., 1996 edition (Medical Economics Company, Montvale, NJ 07645-1742, USA); the disclosure of which is incorporated herein by reference thereto.

The amount and frequency of administration of the compounds of formula IA and the chemotherapeutic agents and/or radiation therapy will be regulated according to the judgment of the attending clinician (physician) considering such factors as age, condition and size of the patient as well as severity of the disease being treated. A dosage regimen of the compound of formula IA can be oral administration of from 10 mg to 2000 mg/day, preferably 10 to 1000 mg/day, more preferably 50 to 600 mg/day, in two to four (preferably two) divided doses, to block tumor growth. Intermittant therapy (e.g., one week out of three weeks or three out of four weeks) may also be used.

The chemotherapeutic agent and/or radiation therapy can be administered according to therapeutic protocols well known in the art. It will be apparent to those

skilled in the art that the administration of the chemotherapeutic agent and/or radiation therapy can be varied depending on the disease being treated and the known effects of the chemotherapeutic agent and/or radiation therapy on that disease. Also, in accordance with the knowledge of the skilled clinician, the therapeutic protocols (e.g., dosage amounts and times of administration) can be varied in view of the observed effects of the administered therapeutic agents (i.e., antineoplastic agent or radiation) on the patient, and in view of the observed responses of the disease to the administered therapeutic agents.

In the methods of this invention, a compound of formula IA is administered concurrently or sequentially with a chemotherapeutic agent and/or radiation. Thus, it is not necessary that, for example, the chemotherapeutic agent and the compound of formula IA, or the radiation and the compound of formula IA, should be administered simultaneously or essentially simultaneously. The advantage of a simultaneous or essentially simultaneous administration is well within the determination of the skilled clinician.

Also, in general, the compound of formula IA and the chemotherapeutic agent do not have to be administered in the same pharmaceutical composition, and may, because of different physical and chemical characteristics, have to be administered by different routes. For example, the compound of formula IA may be administered orally to generate and maintain good blood levels thereof, while the chemotherapeutic agent may be administered intravenously. The determination of the mode of administration and the advisability of administration, where possible, in the same pharmaceutical composition, is well within the knowledge of the skilled clinician. The initial administration can be made according to established protocols known in the art, and then, based upon the observed effects, the dosage, modes of administration and times of administration can be modified by the skilled clinician.

The particular choice of a compound of formula IA, and chemo-therapeutic agent and/or radiation will depend upon the diagnosis of the attending physicians and their judgement of the condition of the patient and the appropriate treatment protocol.

The compound of formula IA, and chemotherapeutic agent and/or radiation may be administered concurrently (e.g., simultaneously, essentially simultaneously or within the same treatment protocol) or sequentially, depending upon the nature of the proliferative disease, the condition of the patient, and the actual choice of

30

5

10

15

20

chemotherapeutic agent and/or radiation to be administered in conjunction (i.e., within a single treatment protocol) with the compound of formula or IA.

If the compound of formula IA, and the chemotherapeutic agent and/or radiation are not administered simultaneously or essentially simultaneously, then the initial order of administration of the compound of formula IA, and the chemotherapeutic agent and/or radiation, may not be important. Thus, the compound of formula IA may be administered first, followed by the administration of the chemotherapeutic agent and/or radiation; or the chemo-therapeutic agent and/or radiation may be administered first, followed by the administration of the compound of formula IA. This alternate administration may be repeated during a single treatment protocol. The determination of the order of administration, and the number of repetitions of administration of each therapeutic agent during a treatment protocol, is well within the knowledge of the skilled physician after evaluation of the disease being treated and the condition of the patient.

10

15

20

25

30

For example, the chemotherapeutic agent and/or radiation may be administered first, especially if it is a cytotoxic agent, and then the treatment continued with the administration of the compound of formula IA followed, where determined advantageous, by the administration of the chemotherapeutic agent and/or radiation, and so on until the treatment protocol is complete.

Thus, in accordance with experience and knowledge, the practicing physician can modify each protocol for the administration of a component (therapeutic agent--i.e., the compound of formula IA, chemotherapeutic agent or radiation) of the treatment according to the individual patient's needs, as the treatment proceeds.

The attending clinician, in judging whether treatment is effective at the dosage administered, will consider the general well-being of the patient as well as more definite signs such as relief of disease-related symptoms, inhibition of tumor growth, actual shrinkage of the tumor, or inhibition of metastasis. Size of the tumor can be measured by standard methods such as radio-logical studies, e.g., CAT or MRI scan, and successive measurements can be used to judge whether or not growth of the tumor has been retarded or even reversed. Relief of disease-related symptoms such as pain, and improvement in overall condition can also be used to help judge effectiveness of treatment.

BIOLOGICAL EXAMPLES

The compounds of the present invention are useful in the treatment of CXC-chemokine mediated conditions and diseases. This utility is manifested in their ability to inhibit IL-8 and GRO- α chemokine as demonstrated by the following *in vitro* assays.

Receptor Binding Assays:

CXCR1 SPA Assay

For each well of a 96 well plate, a reaction mixture of 10 μ g hCXCR1-CHO overexpressing membranes (Biosignal) and 200 μ g/well WGA-SPA beads (Amersham) in 100 μ l was prepared in CXCR1 assay buffer (25 mM HEPES, pH 7.8, 2 mM CaCl₂, 1mM MgCl₂, 125 mM NaCl, 0.1% BSA) (Sigma). A 0.4 nM stock of ligand, [125I]-IL-8 (NEN) was prepared in the CXCR1 assay buffer. 20X stock solutions of test compounds were prepared in DMSO (Sigma). A 6 X stock solution of IL-8 (R&D) was prepared in CXCR2 assay buffer. The above solutions were added to a 96-well assay plate (PerkinElmer) as follows: 10 μ l test compound or DMSO, 40 μ l CXCR1 assay buffer or IL-8 stock, 100 μ l of reaction mixture, 50 μ l of ligand stock (Final [Ligand] = 0.1 nM). The assay plates were shaken for 5 minutes on plate shaker, then incubated for 8 hours before cpm/well were determined in Microbeta Trilux counter (PerkinElmer). % Inhibition of Total binding-NSB (250 nM IL-8) was determined for IC₅₀ values.

20

25

30

5

10

15

Alternative CXCR1 SPA Assay

Protocol using CXCR1-expressing membranes from Biosignal Packard

For each 50 μ l reaction, a working stock of 0.25 μ g/ μ l hCXCR1-CHO over-expressing membranes with a specific activity of 0.05 pmol/mg (Biosignal Packard) and 25 μ g/ μ l WGA-SPA beads (Perkin Elmer Life Sciences) was prepared in CXCR1 assay buffer (25 mM HEPES, pH 7.8, 0.1 mM CaCl₂, 1mM MgCl₂, 100 mM NaCl) (Sigma). This mixture was incubated on ice for 30 minutes and then centrifuged at 2500 rpm for 5 minutes. The beads and membranes were resuspended in CXCR1 assay buffer to the same concentrations as in the original mixture. A 0.125 nM stock of ligand, [125 I]-IL-8 (Perkin Elmer Life Sciences), was prepared in the CXCR1 assay buffer. Test compounds were first serially diluted by half-logs in DMSO (Sigma) and then diluted 20-fold in CXCR1 assay buffer. The above solutions were added to a

Corning NBS (non-binding surface) 96-well assay plate as follows: 20 μ l test compound or 5% DMSO (final [DMSO] = 2%), 20 μ l of membranes and SPA bead mixture (Final [membrane] = 5 μ g/reaction; Final [SPA bead] = 500 μ g/reaction), 10 μ l of ligand stock (Final [125 l-IL-8] = 0.025 nM). The assay plates were incubated for 4 hours before cpm/well were determined in a Microbeta Trilux counter (Perkin Elmer Life Sciences). IC₅₀ values were quantified using nonlinear regression analysis in GraphPad Prism.

Alternative CXCR1 SPA Assay

5

10

15

20

25

Protocol using CXCR1-expressing membranes from Euroscreen

For each 50 μ l reaction, a working stock of 0.025 μ g/ μ l hCXCR1-CHO over-expressing membranes with a specific activity of 3.47 pmol/mg (Euroscreen) and 5 μ g/ μ l WGA-SPA beads (Perkin Elmer Life Sciences) was prepared in CXCR1 assay buffer (25 mM HEPES, pH 7.8, 2.0 mM CaCl₂, 1mM MgCl₂, 125 mM NaCl) (Sigma). This mixture was incubated on ice for 5 minutes. A 0.125 nM stock of ligand, [125 l]-IL-8 (Perkin Elmer Life Sciences), was prepared in the CXCR1 assay buffer. Test compounds were first serially diluted by half-logs in DMSO (Sigma) and then diluted 13.3-fold in CXCR1 assay buffer. The above solutions were added to a Corning NBS (non-binding surface) 96-well assay plate as follows: 20 μ l test compound or 7.5% DMSO (final [DMSO] = 3%), 20 μ l of membranes and SPA bead mixture (Final [membrane] = 0.5 μ g/reaction; Final [SPA bead] = 100 μ g/reaction), 10 μ l of ligand stock (Final [125 l-IL-8] = 0.025 nM). The assay plates were incubated for 4 hours before cpm/well were determined in a Microbeta Trilux counter (Perkin Elmer Life Sciences). IC₅₀ values were quantified using nonlinear regression analysis in GraphPad Prism.

For the CXCR1 assay, compounds of this invention had an IC₅₀ of <20 μ M. Most preferred compounds (a1) to (a21) had a K_i within the range of 4 nM to 3000 nM. The compound of Example 56 (i.e., (a9)) had a K_i of 4 nM, the compound of Example 201.1 (i.e., (a20)) had a K_i of 123 nM, and the compound of Example 201.9 (i.e., (a21)) had a K_i of 50 nM.

CXCR2 SPA Assay

5

10

15

20

25

30

For each well of a 96 well plate, a reaction mixture of 4 µg hCXCR2-CHO overexpressing membranes (Biosignal) and 200 μg/well WGA-SPA beads (Amersham) in 100 µl was prepared in CXCR2 assay buffer (25 mM HEPES, pH 7.4, 2 mM CaCl₂, 1mM MgCl₂). A 0.4 nM stock of ligand, [125I]-IL-8 (NEN), was prepared in the CXCR2 assay buffer. 20X stock solutions of test compounds were prepared in DMSO (Sigma). A 6 X stock solution of GRO- α (R&D) was prepared in CXCR2 assay buffer. The above solutions were added to a 96-well assay plate (PerkinElmer or Corning) as follows: 10 µl test compound or DMSO, 40 ul CXCR2 assay buffer or GRO- α stock, 100 μ l of reaction mixture, 50 μ l of ligand stock (Final [Ligand] = 0.1 nM). When 40 X stock solutions of test compounds in DMSO were prepared, then the above protocol was used except instead 5 µl test compound or DMSO and 45 µl CXCR2 assay buffer were used. The assay plates were shaken for 5 minutes on a plate shaker, then incubated for 2-8 hours before cpm/well were determined in Microbeta Trilux counter (PerkinElmer). % Inhibition of total binding minus non-specific binding (250 nM Gro-α or 50 μM antagonist) was determined and IC50 values calculated. Compounds of this invention had an IC₅₀ of $<5\mu$ M.

Alternative CXCR2 SPA Assay

Protocol using the CXCR2 50 µl assay

For each 50 μ l reaction, a working stock of 0.031 μ g/ μ l hCXCR2-CHO over-expressing membranes with a specific activity of 0.4 pmol/mg (Biosignal Packard) and 2.5 μ g/ μ l WGA-SPA beads (Perkin Elmer Life Sciences) was prepared in CXCR2 assay buffer (25 mM HEPES, pH 7.4, 2.0 mM CaCl₂, 1 mM MgCl₂) (Sigma). This mixture was incubated on ice for 5 minutes. A 0.50 nM stock of ligand, [125 l]-IL-8 (Perkin Elmer Life Sciences), was prepared in the CXCR2 assay buffer. Test compounds were first serially diluted by half-logs in DMSO (Sigma) and then diluted 13.3-fold in CXCR2 assay buffer. The above solutions were added to a Corning NBS (non-binding surface) 96-well assay plate as follows: 20 μ l test compound or 7.5% DMSO (final [DMSO] = 3%), 20 μ l of membranes and SPA bead mixture (final [membrane] = 0.625 μ g/reaction; final [SPA bead] = 50 μ g/reaction), 10 μ l of ligand stock (final [125 l-IL-8] = 0.10 nM). The assay plates were incubated for 2 hours before

cpm/well were determined in a Microbeta Trilux counter (Perkin Elmer Life Sciences). IC₅₀ values were quantified using nonlinear regression analysis in GraphPad Prism.

Alternative CXCR2 SPA Assay

5

10

15

20

25

30

Protocol using the CXCR2 200 μl assay

For each 200 μ l reaction, a working stock of 0.02 μ g/ μ l hCXCR2-CHO over-expressing membranes with a specific activity of 0.6 pmol/mg (Biosignal Packard) and 2 μ g/ μ l WGA-SPA beads (Perkin Elmer Life Sciences) was prepared in CXCR2 assay buffer (25 mM HEPES, pH 7.4, 2.0 mM CaCl₂, 1 mM MgCl₂) (Sigma). This mixture was incubated on ice for 5 minutes. A 0.40 nM stock of ligand, [125 l]-IL-8 (Perkin Elmer Life Sciences), was prepared in the CXCR2 assay buffer. Test compounds were first serially diluted by half-logs in DMSO (Sigma) and then diluted 20-fold in CXCR2 assay buffer. The above solutions were added to a Corning NBS (non-binding surface) 96-well assay plate as follows: 50 μ l test compound or 10% DMSO (final [DMSO] = 2.5%), 100 μ l of membranes and SPA bead mixture (final [membrane] = 2 μ g/reaction; final [SPA bead] = 200 μ g/reaction), 50 μ l of ligand stock (final [125 l-IL-8] = 0.10 nM). The assay plates were incubated for 2 hours before cpm/well were determined in a Microbeta Trilux counter (Perkin Elmer Life Sciences). IC₅₀ values were quantified using nonlinear regression analysis in GraphPad Prism.

For the CXCR2 assay, most preferred compounds (a1) to (a21) had a K_i within the range of 2 nM to 36 nM. The compound of Example 56 (i.e., (a9)) had a K_i of 7 nM, the compound of Example 201.1 (i.e., (a20)) had a K_i of 3.5 nM, and the compound of Example 201.9 (i.e., (a21)) had a K_i of 2.7 nM.

Calcium Fluorescence Assay (FLIPR)

HEK 293 cells stably transfected with hCXCR2 and $G\alpha\iota/q$ were plated at 10,000 cells per well in a Poly-D-Lysine Black/Clear plate (Becton Dickinson) and incubated 48 hours at 5% CO_2 , 37°C. The cultures were then incubated with 4 mM fluo-4, AM (Molecular Probes) in Dye Loading Buffer (1% FBS, HBSS w. Ca & Mg, 20 mM HEPES (Cellgro), 2.5 mM Probenicid (Sigma) for 1 hour. The cultures were washed with wash buffer (HBSS w Ca, & Mg, 20 mM HEPES, Probenicid (2.5 mM)) three times, then 100 μ l/well wash buffer was added.

During incubation, compounds were prepared as 4X stocks in 0.4% DMSO (Sigma) and wash buffer and added to their respective wells in the first addition plate. IL-8 or GRO- α (R&D Systems) concentrations were prepared 4X in wash buffer + 0.1% BSA and added to their respective wells in second addition plate.

Culture plate and both addition plates were then placed in the FLIPR imaging system to determine change in calcium fluorescence upon addition of compound and then ligand. Briefly, $50~\mu l$ of compound solutions or DMSO solution was added to respective wells and change in calcium fluorescence measured by the FLIPR for 1 minute. After a 3 minute incubation within the instrument, $50~\mu l$ of ligand was then added and the change in calcium fluorescence measured by the FLIPR instrument for 1 minute. The area under each stimulation curve was determined and values used to determine % Stimulation by compound (agonist) and % Inhibition of Total Calcium response to ligand (0.3 nM IL-8 or GRO- α) for IC50 values of the test compounds.

Chemotaxis assays for 293-CXCR2

5

10

15

25

30

A chemotaxis assay is setup using Fluorblok inserts (Falcon) for 293-CXCR2 cells (HEK-293 cells overexpressing human CXCR2). The standard protocol used at present is as follows:

- 1. Inserts are coated with collagenIV (2ug/ml) for 2 hrs at 37°C.
- 2. The collagen is removed and inserts are allowed to air dry overnight.
- 3. Cells are labeled with 10uM calcein AM (Molecular Probes) for 2 hrs. Labeling is done in complete media with 2% FBS.
 - 4. Dilutions of compound are made in minimal media (0.1% BSA) and placed inside the insert which is positioned inside the well of a 24 well plate. Within the well is IL-8 at a concentration of 0.25nM in minimal media. Cells are washed and resuspended in minimal media and placed inside the insert at a concentration of 50,000 cells per insert.
 - 5. Plate is incubated for 2hrs and inserts are removed and placed in a new 24 well. Fluorescence is detected at excitation=485 nM and emission=530 nM.

Cytotoxicity Assays

A cytotoxicity assay for CXCR2 compounds is conducted on 293-CXCR2 cells. Concentrations of compounds are tested for toxicity at high concentrations to

determine if they may be used for further evaluation in binding and cell based assays. The protocol is as follows:

- 1. 293-CXCR2 cells are plated overnight at a concentration of 5000 cells per well in complete media.
- Dilutions of compound are made in minimal media w/0.1% BSA. Complete media is poured off and the dilutions of compound are added. Plates are incubated for 4, 24 and 48hrs. Cells are labeled with 10uM calcein AM for 15 minutes to determine cell viability. Detection method is the same as above.

Soft Agar Assay

10,000 SKMEL-5 cells/well are placed in a mixture of 1.2% agar and complete media with various dilutions of compound. Final concentration of agar is 0.6%. After 21 days viable cell colonies are stained with a solution of MTT (1mg/ml in PBS). Plates are then scanned to determine colony number and size. IC₅₀ is determined by comparing total area vs. compound concentration.

CCR7 membrane preparation.

10

15

20

25

30

Ba/F3-CCR7 membranes were prepared as previously described (Hipkin et al., J.~Biol.~Chem., 272, 1997, 13869-76). Cells were pelleted by centrifugation, incubated in homogenization buffer (10 mM Tris-HCl, 5 mM EDTA, 3 mM EGTA, pH 7.6) and 1 μ M PMSF for 30 min. on ice. The cells were then lysed with a Dounce homogenizer using stirrer type RZR3 polytron homogenizer (Caframo, Wiarton, Ont.) with 12 strokes at 900 RPM. The intact cells and nuclei were removed by centrifugation at 500Xg for 5 min. The cell membranes in the supernatant were then pelleted by centrifugation at 100,000Xg for 30 min. The membranes were then resuspended in glygly buffer (20 mM glycylglycine, 1 mM MgCl₂, 250 mM sucrose, pH 7.2), aliquoted, quick frozen and stored at -80°C.

CCR7 [35S]GTPyS exchange assay.

The exchange of guanosine 5'-[γ - 35 S]triphospate ([35 S]GTP γ S, triethylammonium salt; specific activity = 1250 Ci/mmol; NEN Boston, MA) was measured using a scintillation proximity assay (SPA) as previously described (Cox, et. al., *Mol. Pharmacol.*, 59, 2001, 707-15). For each assay point, 2 µg of membrane was

preincubated for 30 min at room temperature with 200 μ g wheat germ agglutinin-coated SPA beads (WGA-SPA; Amersham, Arlington Heights, IL) in SPA binding buffer (50 mM HEPES, 10 mM MgCl₂, 1 mM EDTA, 100 mM NaCl, 0.1% BSA, pH 7.6). The beads and membranes were transferred to a 96-well Isoplate (Wallac, Gaithersburg, MD) and incubated with 10 μ M guanosine 5'-diphosphate (GDP) in the presence or absence of 2 nM MIP-3 β and/or compounds for 60 min at room temperature. The incubation continued for another 60 min. following the addition of 0.1 nM [35 S]GTP γ S. Membrane-bound [35 S]GTP γ S was measured using a 1450 Microbeta Trilux counter (Wallac, Gaithersburg, MD).

5

10

15

20

25

There were compounds of this invention that had an EC₅₀ of <10 μ M. The compound of Example 2065 had an EC₅₀ of 13nM, the compound of Example 2066 had an EC₅₀ of 16nM, the compound of Example 2105 had an EC₅₀ of 3nM, and the compound of Example 2106 had an EC₅₀ of 12nM.

Compounds of formula IA may be produced by processes known to those skilled in the art, in the following reaction schemes, and in the preparations and examples below.

A general procedure for the preparation of compounds of formula IA is as follows:

Compounds of this invention are prepared by condensing an amine (either A-NH₂ or B-NH₂) with dimethoxythiadiazoledioxide to give the monomethoxythiadiazoledioxide intermediate. Subsequent condensation of this intermediate with the commercially available or prepared amine (either A-NH₂ or B-NH₂) provides the desired chemokine antagonist.

Thiadiazoleoxide compounds are prepared similarly starting with dimethoxythiadiazoleoxide. Sequential condensation with amine A-NH₂ or B-NH₂ as described above provides the desired antagonist.

5

10

15

20

The invention disclosed herein is exemplified by the following preparations and examples which should not be construed to limit the scope of the disclosure.

Alternative mechanistic pathways and analogous structures may be apparent to those skilled in the art.

PREPARATIVE EXAMPLE 1

$$HO_2C$$
 OH OH OH OH OH

3-Nitrosalicylic acid (500 mg, 2.7 mmol), DCC (563 mg) and ethyl acetate (10 mL) were combined and stirred for 10 min. (R) -(-)-2-pyrrolidinemethanol (0.27 mL) was added and the resulting suspension was stirred at room temperature overnight. The solid was filtered and the filtrate washed with 1N NaOH. The aqueous phase was acidified and extracted with EtOAc. The resulting organic phase was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. Purification of the residue by preparative plate chromatography (silica gel, 5% MeOH/CH₂Cl₂ saturated with AcOH) gave the product (338 mg, 46%, MH $^+$ = 267).

PREPARATIVE EXAMPLE 2

Step A

5

10

3-Nitrosalicylic acid (9.2 g), bromotripyrrolidinophosphonium hexafluorophosphate (PyBroP, 23 g) and N,N-diisopropylethylamine (DIEA, 26 mL) in anhydrous CH₂Cl₂ (125 mL) were combined and stirred at 25°C for 30 min. (R) -(+)-3-pyrrolidinol (8.7 g) in CH₂Cl₂ (25 mL) was added over 25 min and the resulting suspension was stirred at room temperature overnight. The mixture was extracted with 1M NaOH (aq) and the organic phase was discarded. The aqueous phase was acidified with 1M HCl (aq), extracted with EtOAc, dried over anhydrous Na₂SO₄, filtered and concentrated *in vacuo* to afford the crude product (7 g) which was used without further purification.

Step B

The crude product from Step A above was stirred with 10% Pd/C (0.7 g) in MeOH (100 mL) under a hydrogen gas atmosphere overnight. The reaction mixture was filtered through celite, the filtrate concentrated *in vacuo*, and the resulting residue purified by column chromatography (silica gel, 10% MeOH/CH₂Cl₂ saturated with NH₄OH) to give the product (2.5 g, 41%, MH+=223).

20

25

15

PREPARATIVE EXAMPLE 2.1

$$H_2N$$
 $NBoc$ NH NH NH

To N-BOC-3-(amino)piperidine (0.5 g) dissolved in CH₂Cl₂ (10 mL) was added benzylisocyanate (3 mmol). After stirring for 2 hrs, amine scavenger resin (1.9 mmol) was added and the mixture was stirred overnight, filtered, the resin back-washed with CH₂Cl₂ and methanol, and the organics concentrated *in vacuo*. Stirring of the crude

material in 4N HCl/dioxane (40 mL) for 2.5 hrs before concentrating *in vacuo* gave the title compound (41%, MH+=369).

PREPARATIVE EXAMPLE 2.2 - 2.6

Following the procedures set forth in Preparative Example 2.1 but using the isocyanate (or chloroformate) indicated in the Table below, the amines were obtained and used without further purification.

5

10

15

Prep Ex.	Amine	Isocyanate	Amine
2.2	H ₂ N NH	NCO	NH H
2.3	H ₂ N NH	NCO	NH H H
2.4	H ₂ N NH	NCO	NH NH NH
2.5	H ₂ N NH	^o CI	O N NH
2.6	H ₂ N NH	NCO	NH NH

PREPARATIVE EXAMPLE 2.7

$$H_2N$$
 $NBoc$
 F
 N
 NH

To N-BOC-3-(amino)piperidine (5 mmol) dissolved in CH₂Cl₂ (30 mL) was added trifluoromethanesulfonic anhydride (5 mmol) and the mixture was stirred overnight. The mixture was concentrated *in vacuo*, diluted with CH₂Cl₂ (10 mL) and treated with trifluoroacetic acid (10 mL). After stirring for 2 hr, the mixture was concentrated *in vacuo* to give the title compound (43%, MH+=233.1).

PREPARATIVE EXAMPLE 2.8

Step A

5

10

15

3-Nitrosalicylic acid (5 mmol) and N-hydroxysuccinimide (5 mmol) were added to a solution of 2% DMF/CH₂Cl₂, followed by DCC (5 mmol). After stirring for 2 hr, the mixture was filtered and concentrated *in vacuo* and the residue used directly in Step B.

Step B

The product from Step A above was suspended in DMF and to this was added morpholino-2-carboxylic acid HCl (5 mmol) in CH₂Cl₂ (10 mL)/DMF (5 mL) and diisopropylethylamine (10 mmol). The mixture was stirred overnight, filtered, basified with 1*N* NaOH (50 mL), washed with CH₂Cl₂, acidified with 5*N* HCl and extracted with EtOAc. The organic phase was dried over Na₂SO₄, filtered and concentrated *in vacuo* to give the desired compound which was used directly in Step C (MH+=296).

20 Step C

Following a similar procedure as in Preparative Example 2 Step B, but using the product from Step B above, the title compound was obtained (23%, MH+=267).

PREPARATIVE EXAMPLE 2.9

Step A

5

10

15

20

2-Piperazinecarboxylic acid and 2-chloro-1,3-pyrimidine were stirred with triethylamine and MeOH. After stirring overnight at reflux, the mixture was filtered and concentrated *in vacuo* to give the desired compound which was used directly in Step B (MH+ = 209).

Step B

Following a similar procedure as Preparative Example 2.8, Step B except using the product from Preparative Example 2.9 Step A above, the desired compound was obtained (41%, MH+ = 374).

Step C

Following a similar procedure as in Preparative Example 2, Step B, but using the product from Step B above, the desired compound was obtained (99%, MH+=344).

PREPARATIVE EXAMPLE 2.10

$$\begin{array}{c|c} & & & \\ &$$

$$\begin{array}{c|c} & & & & \\ & &$$

Following a similar procedure as Preparative Example 2.8, Step A except using 3-nitrobenzoic acid, the desired compound was obtained and used directly in Step B.

Step B

5

10

15

Following a similar procedure as Preparative Example 2.8, Step B except using the products from Preparative Example 2.9, Step A and Preparative Example 2.10, Step A, the desired compound was obtained (86%).

Step C

Following a similar procedure as in Preparative Example 2, Step B, but using the product from Step B above, the desired compound was obtained (67%, MH+=331).

PREPARATIVE EXAMPLE 2.11

N-Benzylpiperidone (2 g, HCl salt, hydrate) was stirred with THF (20 mL), concentrated to dryness, and placed under high vac. The residue was diluted in THF (20 mL), and methyllithium was added (2.5 eq of 1.6N in Et₂O) via syringe. After stirring for 3 hr, the mixture was concentrated *in vacuo*, diluted with water, extracted with CH₂Cl₂, and dried over Na₂SO₄. Filtration and concentrating *in vacuo* gave the desired product (50%, MH+ = 205).

Step B

5

10

Following a similar procedure as in Preparative Example 2, Step B, but using the product from Step A above, the title compound was obtained (95%, MH+=116).

PREPARATIVE EXAMPLE 2.12

15 Step A

20

25

To N-benzyl-N-methylamine (20 mmol) dissolved in acetone (50 mL) was added concentrated HCl (20 mmol), paraformaldehyde (30 mmol) and 2-propanol (2 mL). After stirring at reflux overnight, the mixture was concentrated *in vacuo*, diluted with water, basified to pH 14 and extracted with ether. The organic phase was dried over Na₂SO₄, filtered and concentrated *in vacuo* to give the desired product (98%) which was used directly in Step B.

Step B

The product from Step A above (500 mg) was dissolved in MeOH (20 mL) and to this was added NaBH₄ (50 mg). After stirring for 10 min, the solution was

concentrated *in vacuo* to give the desired compound which was used directly in Step C without purification.

Step C

The product from Step B above was diluted with MeOH (20 mL) and to this was added AcOH (0.1 mL), a catalytic amount of Pd/C (10%) and the mixture stirred under H₂ atmosphere (balloon) overnight. The mixture was filtered, 4N HCI in dioxane (1 mL) was added, and the mixture was concentrated *in vacuo* to give the desired compound that was used directly without purification.

10

15

5

PREPARATIVE EXAMPLE 2.13

Step A

Following a similar procedure as Preparative Example 2, Step A except using methyl glycinate, the desired ester was obtained. The mixture was poured into 200 mL of 1*N* NaOH, then extracted with dichloromethane. The pH was adjusted to 1 and NaCl was added until saturation. After several hours, the resulting precipitate was filtered and washed with cold water to give the desired product (42%).

20 Step B

Following a similar procedure as in Preparative Example 2 Step B, but using the product from Step A above, the title compound was obtained (95%).

PREPARATIVE EXAMPLE 2.14

Step A

Following a similar procedure as in Preparative Example 2.13, Step A except using methyl N-methylglycinate, the desired product was obtained (18%).

Step B

5

10 .

15

20

Following a similar procedure as in Preparative Example 2, Step B, but using the product from Step A above, the title compound was obtained (95%, MH+ = 225).

PREPARATIVE EXAMPLE 2.16

The above n-oxide (2g) was combined with H_2NMe/H_2O (15cm³) and heated to 140°C overnight. Potassium carbonate (1.3g) added and the mixture concentrated *in vacuo*. Extraction with EtOH and concentration of the filtrate *in vacuo* gave 1.56g of crude amine (MH+=125).

PREPARATIVE EXAMPLE 3-10.50

Following the procedures set forth in Preparative Examples 1-2 but using the carboxylic acid, amine, and coupling agent [DCC (Prep. Ex. 1) or PyBrop (Prep. Ex. 2)] listed in the Table below, the indicated amide products were obtained and used without further purification.

Prep	Carboxylic acid	Amine	Product
Ex.			1. Coupling Agent 2. %yield 3. MH ⁺
3	HO ₂ C OH	N—H	NH ₂ OH 1. PyBrop 2. 87, 86
			3. 181
4	HO ₂ C OH	N.H	NH ₂
			1. PyBroP 2. 49 3. 209
5	HO ₂ C OH	NH ₃	H ₂ N NH ₂
			1. PyBroP 2. 95 3. 153
6	HO ₂ C OH	NH ₂	H-N-NH ₂
			1. PyBroP 2. 83 3. 167

7	HO ₂ C OH	ON.H	ON NH ₂
			1. PyBroP 2. 76 3. 223
8	HO ₂ C OH	HO N'H	HO NH ₂ NH ₂
			1. PyBroP 2. 65, 53 3. 209
9	HO ₂ C OH	N.H	NH ₂
			1. PyBroP 2. 59, 69 3. 207
10	HO ₂ C OH	HON_H	HONH ₂
			1. PyBroP 2. 49, 86 3. 237
10.1	HO ₂ C OH	NH ₂	H NH ₂
			 PyBroP 30,88 193

100		T	
10.2	HO ₂ C OH	NH ₂	H NH ₂
			 PyBroP 26,87 195
10.3	HO ₂ C OH	VNH₂	H NH ₂
			 PyBroP 38 209
10.4	HO ₂ C OH	NH ₂	H NH ₂
			 PyBroP 29 209
10.5	HO ₂ C OH	NH ₂	H NH ₂
			1. PyBroP 2. 38 3. 223
10.6	HO ₂ C OH	2.7 O ₂ NH NH	F ₃ C N NH ₂ NH ₂
			 1. PyBroP 2. 32,99 3. 367.9
10.7	HO ₂ C OH	Н	O NH ₂
<u></u>			У ЗОН

			1. PyBroP 2. 35,99
			3. 237
10.8	HO ₂ C OH	NH HO O	1. DCC 2. 30,99 3. 269
10.9	HO ₂ C OH	2.11 NH HO	1. PyBroP 2. 58,95 3. 233.1
10.10	HO ₂ C OH	2.12 OH NH	1. PyBroP 2. 42,95 3. 238.9
10.13	HO ₂ C OH	2.4 NH NH NH NH	1. PyBroP 2. 51,95 3. 307
10.14	HO ₂ C OH	2.2	1. PyBroP 2. 55 3. 347

10.15		2.1	
10.10	HO ₂ C OH	NH NH	NH ₂
	011		1. PyBroP 2. 41 3. 369.1
10.16	HO ₂ C OH	2.3	NH ₂ OH NH ₂
	OH	н н	 PyBroP 56 354.9
10.17	HO ₂ C OH	2.5	ON NH2
			 PyBroP 56 308
10.18	HO ₂ C OH	12.4 OH NH	OH NH ₂
			 1. PyBroP 2. 10,95 3. 252.9
10.19	HO ₂ C OH	HZ HZ	O N OH NH2
			1. PyBroP 2. 42,95 3. 249

40.00	·		
10.20	HO ₂ C OH	HO	OH NH2 OH OH 1. PyBroP
			2. 15,95 3. 264.9
10.21	HO ₂ C OH	HO NH ₂	NH OH NH2
			 PyBroP 64,95 273
10.22	HO ₂ C OH	HO	O NH ₂ OH HO
	· ;		1. PyBroP 2. 45,95 3. 273
10.23	HO ₂ C OH	NH_2	NH OH
			 PyBroP 44,95 281

	,		
10.24	HO ₂ C OH	N N N N N N N N N N N N N N N N N N N	O NH ₂
			 PyBroP 41,95 281.1
10.25	HO ₂ C OH	N H	O NH ₂
			1. PyBroP 2. 48,95 3. 257
10.26	HO ₂ C OH	~~~ NH	1. DCC 2. 15,99
			3. 235
10.28	HO ₂ C OH	HO	NH ₂
			1. PyBroP 2. 52,95 3. 237.1

10.29	HO ₂ C OH	OH HN	HO N OH
			1. PyBroP 2. 31,95 3. 259.1
10.30	HO ₂ C OH	HO	HO OH NH ₂
			 PyBroP 54,95 250.9
10.31	HO ₂ C OH	HO N	HO OH NH ₂
			1. PyBroP 2. 64,95 3. 210.9
10.32	HO ₂ C OH	HO NH ₂	HO NH OH
			 PyBroP 47,95 197

		· · · · · · · · · · · · · · · · · · ·	
10.33	HO ₂ C OH	HO N	1. PyBroP 2. 47,95
10.34		/—NH	3. 273
	HO ₂ C OH	HO	NH ₂ NH ₂
			1. PyBroP 2. 51,95 3. 237.1
10.35	HO ₂ C OH	HN NH ₂	OH ONH ₂
			 PyBroP 60,90 224
10.36	HO ₂ C OH	H NMe ₂	OH ONMe2
			 PyBroP 65,99 252

10.37	HO ₂ C OH	HOMe	OH OME
			1. PyBroP 2. 58,99 3. 239
10.38	HO ₂ C OH	NH	NH ₂ OH O
			1. PyBroP 2. 35,99 3. 221.1
10.39	HO ₂ C OH	NH	NH ₂ OH O
			1. PyBroP 2. 42,99 3. 235.2
10.40	HO ₂ C OH	HNOEt	OH OEt
			1. DCC 2. 32,99 3. 293.1

10.41	HO ₂ C OH	HONH	NH ₂ OH OH
	•		 PyBroP 45,99 223.1
10.42	HO ₂ C OH	HONH	HO N N O O O H
			1. PyBroP 2. 55,81 3. 251.1
10.43	HO ₂ C OH	HONH	HO NH ₂
			1. PyBroP 2. 68,66 3. 224.9
10.44	HO ₂ C OH	HO NH	HO NH ₂
			 PyBroP 68,66 241.1

10.45		12.3	
	HO ₂ C OH	NH	0 NH ₂
			O OH
)
			, , , , , , , , , , , , , , , , , , ,
			1. PyBroP 2. 44,40
			3. 295
10.46	NO ₂		
	HO ₂ C OH	NH	N NH ₂
		HO 0	HOOO
			1. DCC 2. 37,81
			3. 265
10.47		2.6	
	NO ₂	NH	NH ₂
	HO ₂ C´ OH É	H H	N N N O OH W'2
			1. PyBroP
			2. 71,95 3. 293.1
10.48			
10.70		N NH_2	N N N N N N N N
	HO ₂ C OH	N-N	N-N O OH
			1. PyBroP
			2. 35,99 3. 220.9
10.49			
10.45		\longrightarrow NH ₂	H NH
	HO ₂ C OH		OH NH2
			1. DCC
			2. 16,993. 209.0

10.50	HO ₂ C OH	H ₂ N NH	H ₂ N N NH ₂ OH NH ₂
			1. DCC 2. 18,99 3. 264.0

PREPARATIVE EXAMPLE 10.55

Alternative Procedure for Preparative Example 3

Step A

5

To the nitrosalicylic acid (3 g) dissolved dichloromethane (150 mL) at room temperature was added oxalyl chloride (4.3 mL) and DMF (0.01 eq.). After stirring for one day the mixture was concentrated in a vacuum to give a semi solid which was used directly in step B.

10

Step B

To the material from step A diluted in dichloromethane (50 mL) and cooled to 0° C was added dimethyl amine in THF (2N solution, 24.6 mL) and triethylamine (4 eq.). After stirring for 24 hours at room temperature the mixture was concentrated in vacuo, diluted with 1M sodium hydroxide (30 mL) and after a half hour was washed with dichloromethane. The aqueous phase was acidified with 6M HCl (aq), extracted with dichloromethane and the organic phase was washed with water, dried over Na₂SO₄ and concentrated to give the title compound (3.2 g, 93%).

15

Step C

5

10

15

20

A mixture of the product from step B above (6 g), 10% Pd/C (0.6 g), and EtOH (80 mL) was stirred in a parr shaker under hydrogen (40 psi) at room temperature for 2 days. Filtration through celite and concentration *in vacuo* afforded the title product (5.1 g, 99%, MH^+ = 181).

PREPARATIVE EXAMPLE 11

Step A

Following a similar procedure as in Preparative Example 1 except using dimethylamine (2*M* in THF, 33 mL) and 5-methylsalicylic acid (5 g), the desired product was prepared (6.5 g).

Step B

Nitric acid (0.8 mL) in H_2SO_4 was added to a cooled (-20°C) suspension of the product from Step A above (3 g) in H_2SO_4 (25 mL). The mixture was treated with 50% NaOH (aq) dropwise, extracted with CH_2Cl_2 , dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give the product as a crude solid (2.1 g, 44%, MH⁺ = 225).

Step C

The product was prepared in the same manner as described in Step B of Preparative Example 2 (0.7 g, 99%, MH⁺ = 195).

25

PREPARATIVE EXAMPLE 11.1

Step A

The above amine was reacted with the acid using the procedure set forth in Preparative Example 2, Step A to yield the desired amide (54%).

Step B

5

10

15

 $Na_2S_2O_4$ (1.22g) was dissolved in water (4ml) followed by the addition of NH_3/H_2O (300ul). The solution ws then added to the product from Step A (200 mg) in dioxane (4ml) and stirred for 30min. The crude material was purified via flash column chromatography ($CH_2Cl_2/MeOH$, 20:1) to give 100mg of product (56%, MH+=251).

PREPARATIVE EXAMPLE 11.2

Following the procedures set forth in Preparative Example 11.1, Steps A and B, but using N-methylmethoxylamine, the title compound was obtained (86%, MH+=181).

Following the procedure set forth in Preparative Example 1, but using

N-hydroxysuccinimide and 2% DMF in CH₂Cl₂, the desired amide was obtained (33%, MH+=297).

Step B

10

15

Following the procedure set forth in Preparative Example 2, Step B, the amine was prepared (99%, MH+=267).

PREPARATIVE EXAMPLE 11.11 – 11.18

Following the procedures set forth in Preparative Examples 11.11 but using the carboxylic acid, amine, and coupling agent DCC indicated, the indicated amide products were obtained and used without further purification.

Prep Ex.	Carboxylic acid	Amine	Product	1. % Yield 2. MH ⁺
11.11	HO ₂ C OH	HZ,	OH O OH NH2	1. 45,92 2. 310.0
11.12	HO ₂ C OH	CIH.H ₂ N HN	HN O OH NH2	1. 45,95 2. 247.2

11.13	HO ₂ C OH	O OH	O OH OH	1. 85,85 2. 251.1
11.14	HO ₂ C OH	OH NH₂	OH HN OH OH	1. 99,92 2. 211.1
11.15	HO ₂ C OH	O NH	HO NH ₂	1. 48,84 2. 265
11.16	HO ₂ C OH	_Z \ \ \ H_	N N N N N N N N N N	1. 78,91 2. 238.1
11.17	HO ₂ C OH	HO NH	HO N NH2	1. 67,90 2. 265.1
11.18	HO ₂ C OH	HO NH	HO N NH ₂	1. 28,99 2. 267

PREPARATIVE EXAMPLE 12

Following a similar procedure as described in Preparative Example 2 Step A except using dimethylamine in place of R-(+)-3-pyrrolidinol, the desired product was prepared.

Step B

5

10

15

20

25

The product from step A above (8 g) was combined with iodine (9.7 g), silver sulfate (11.9 g), EtOH (200 mL) and water (20 mL) and stirred overnight. Filtration, concentration of the filtrate, re-dissolution in CH_2Cl_2 and washing with 1*M* HCl (aq) gave an organic solution which was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to afford the product (7.3 g, 57%, MH⁺ = 337).

Step C

The product from Step B above (3.1 g) was combined with DMF(50 mL) and Mel (0.6 mL). NaH (60% in mineral oil, 0.4 g) was added portionwise and the mixture was stirred overnight. Concentration *in vacuo* afforded a residue which was diluted with CH_2CI_2 , washed with 1M NaOH (aq), dried over anhydrous $MgSO_4$, filtered and concentrated *in vacuo*. Purification through a silica gel column (EtOAc/Hex, 1:1) gave the desired compound (1.3 g, 41%, MH^+ = 351).

Step D

The product from Step D above (200 mg), Zn(CN)₂ (132 mg), Pd(PPh₃)₄ (130 mg) and DMF (5 mL) were heated at 80°C for 48 hrs, then cooled to room temperature and diluted with EtOAc and 2M NH₄OH. After shaking well, the organic extract was dried over anhydrous MgSO₄, filtered, concentrated *in vacuo* and purified by preparative plate chromatography (Silica, EtOAc/Hex, 1:1) to give the desired compound (62 mg, 44%, MH⁺ = 250).

Step E

BBr₃ (1.3 mL, 1*M* in CH₂Cl₂) was added to a CH₂Cl₂ solution (5 mL) of the product from step D above (160 mg) and stirred for 30 min. The mixture was diluted with water, extracted with CH₂Cl₂, dried over anhydrous MgSO₄, filtered, and concentrated *in vacuo* to give the desired compound (158 mg, MH⁺ = 236).

Step F

5

10

15

20

A mixture of the product from step E above (160 mg), platinum oxide (83%, 19 mg), and EtOH (20 mL) was stirred under hydrogen (25-40 psi) for 1.5 hr. Filtration through celite and concentration *in vacuo* afforded the product (165 mg, $MH^+ = 206$).

PREPARATIVE EXAMPLE 12.1

Step A

Following a similar procedure as in Preparative Example 2, Step A except using 3-(methylaminomethyl)pyridine and 3-nitrosalicylic acid, the desired compound was prepared (41%).

Step B

The compound from Step A above (0.3 g) was diluted with chloroform (15 mL) and stirred with mCPBA (0.4 g) for 2 hr. Purification by column chromatography (silica, 10% MeOH/CH₂Cl₂) gave the pyridyl N-oxide (0.32 g, 100%, MH⁺ = 303.9).

Step C

5

10

20

Following a similar procedure as in Preparative Example 11.1, Step B, but using the product from Step B above, the desired compound was obtained (15%, MH+=274).

PREPARATIVE EXAMPLE 12.2

Step A

3-Nitrosalicylic acid (4 g) in MeOH (100 mL) and concentrated H₂SO₄ (1 mL) were stirred at reflux overnight, concentrated *in vacuo*, diluted with CH₂Cl₂, and dried over Na₂SO₄. Purification by column chromatography (silica, 5% MeOH/CH₂Cl₂) gave the methyl ester (2.8 g, 65%).

15 Step B

Following a similar procedure as in Preparative Example 2, Step B, but using the product from Step A above, the desired compound was obtained (95%, MH+=167.9).

PREPARATIVE EXAMPLE 12.3

To morpholine-2-carboxilic acid (200mg) in EtOH (40mL) at 0°C was added acetyl chloride (3mL) and the mixture was stirred at reflux overnight. Concentration in

vacuo, dilution with CH_2Cl_2 and washing with NaHCO₃ (aq) gave the title compound (99%, MH^+ = 160.1).

PREPARATIVE EXAMPLE 12.4

To N-Boc morpholine-2-carboxylic acid (2g) in THF (5ml) at 0°C was added a solution of borane.THF complex (1N, 10.38ml) and the mixture was stirred for 30min at 0°C, and for 2hr at room temperature. Water (200ml) was added to the reaction and the mixture extracted with CH₂Cl₂, dried with Na₂SO₄, and concentrated *in vacuo* to give 490mg of product (26%). The product was then stirred in 4N HCl/dioxane to give the amine salt.

PREPARATIVE EXAMPLE 13

Step A

15

5

10

Following a similar procedure as in Preparative Example 1 except using dimethylamine (2M in THF, 50 mL) and 4-methylsalicylic acid (15 g), the desired compound was prepared (6.3 g, 35%) .

Step B

5

10

15

20

The product from Step A above (1.5 g) was combined with iodine (2.1 g), NaHCO₃ (1.1 g), EtOH (40 mL) and water (10 mL) and stirred overnight. Filtration, concentration of the filtrate, re-dissolution in CH_2Cl_2 and washing with 1*M* HCI (aq) gave an organic solution which was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. Purification by flash column chromatography (silica gel, 0.5-0.7% MeOH/CH₂Cl₂) gave the product (0.5 q, 20%, MH⁺ = 306).

Step C

Nitric acid (3.8 mL) in AcOH (10 mL) was added to the product from Step B above (0.8 g) and the mixture was stirred for 40 min. The mixture was diluted with water and extracted with CH_2Cl_2 , dried over anhydrous $MgSO_4$, filtered and concentrated *in vacuo* to give the product as an orange solid (0.8 g, 92%, $MH^+ = 351$).

Step D

A mixture of the product from step C above (800 mg), 10% Pd/C (100 mg), and EtOH/MeOH (40 mL) was stirred in a parr shaker under hydrogen (45 psi) for 1.5 hr. Filtration through celite and concentration *in vacuo* afforded the title product after purification by preparative plate chromatography (Silica, 10% MeOH/CH₂Cl₂, saturated with NH₄OH) to give the product (92 mg, 22%, MH⁺ = 195).

PREPARATIVE EXAMPLE 13.1

Following a similar procedure as in Preparative Example 2, Step A except using dimethylamine (2M in THF, 23 ml) and 5-bromosalicylic acid (5g), the desired compound was prepared (4.2g, 75%, MH+=244).

5

10

15

Step B

Nitric acid (10ml) in AcOH (100ml) was added to the product from Step A above (2g) and the mixture was stirred for 20 min. The mixture was diluted with water and extracted with CH₂Cl₂, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give the product as a yellow solid (1.9g, 80%, MH+=289).

Step C

The product from Step B above (1.9g) was partially dissolved in EtOH(50ml). Conc HCl in EtOH (5ml in 40ml), followed by SnCl₂.2H₂O (5.74g) was added and stirred at room temperature overnight. The crude reaction was concentrated *in vacuo*, diluted with CH₂Cl₂ and washed with NaHCO₃, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give the product as a solid (185mg, 9%, MH+=259).

PREPARATIVE EXAMPLE 13.2

Step A

5

10

15

20

Following a similar procedure as in Preparative Example 2, Step A, except using dimethylamine (2M in THF, 29 ml) and 5-chlorosalicylic acid (5g), the desired compound was prepared (4.5g, 78%, MH+=200).

Step B

Nitric acid (10ml) in AcOH (100ml) was added to the product from Step A above (2g) and the mixture was stirred for 20 min. The mixture was diluted with water and extracted with CH₂Cl₂, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give the product as a solid (2.2g, 88%, MH+=245).

Step C

The product from Step B above (2.2g) was partially dissolved in EtOH(50ml). Conc HCl in EtOH (5ml in 40ml), followed by SnCl₂.2H₂O (7.01g) was added and stirred at room temperature overnight. The crude reaction was concentrated *in vacuo*, diluted with CH₂Cl₂ and neutralized with NaOH. The entire emulsion was filtered though celite, the layers were separated and the organic layer was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give a solid (540mg, 22%, MH+=215).

PREPARATIVE EXAMPLE 13.3

Step A

5

10

15

OMe

3-Nitrosalicylic acid (10g), PyBroP (20.52g), and DIEA (28ml) in anhydrous CH₂Cl₂ (200ml) were combined and stirred at room temperature for 10 min. Dimethylamine (2M in THF, 55ml) was added and let the reaction stir over the weekend. The mixture was extracted with 1N NaOH (aq) and the organic phase was discarded. The aqueous phase was acidified with 1N HCl (aq), extracted with CH₂Cl₂, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. The oil was taken up in ether and a solid crashed out, triterated in ether to give 4.45g of a solid (39%, MH+=211).

Step B

The product from Step A (2.99g), K₂CO₃ (9.82g), and iodomethane (8.84ml) were combined in acetone and heated to reflux overnight. The reaction was filtered and concentrated *in vacuo*. The oil was taken up in CH₂Cl₂ and washed with 1N NaOH, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give 3.3g of an oil (99%, MH+=225).

Step C

The crude product from Step B (3.3g) was stirred with 10% Pd/C (350mg) in EtOH (50ml) under a hydrogen gas atmosphere at 20psi overnight. The reaction mixture was filtered through celite and the filtrate was concentrated *in vacuo* to give 2.34 g of a solid (85%, MH+=195).

Step D

The product from Step C (469mg) was dissolved in AcOH (6ml). 1.95M Br₂ in AcOH (1.23ml) was added dropwise to the reaction and the mixture was stirred at room temperature for 1 hour. 50% NaOH was added to the reaction at 0°C and the mixture was extracted with CH₂Cl₂, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. The crude mixture was purified by preparative plate chromatography (Silica, 5% MeOH/ CH₂Cl₂) to give the desired product (298mg, 23%, MH+=273).

15

20

10

5

Step E

BBr₃ (2.14ml, 1M in CH_2Cl_2) was added to a CH_2Cl_2 solution (8ml) of the product from Step D above (290mg) and stirred overnight. A solid formed and was filtered, taken up in MeOH/ CH_2Cl_2 and purified by preparative plate chromatography (Silica, 5% MeOH/ CH_2Cl_2) to give the desired product (137mg, 49%, MH+=259).

PREPARATIVE EXAMPLE 13.4

5

10

15

To the product from Preparative Example 13.3 Step D (200mg) was added phenylboronic acid (98mg), PdCl₂(PPh₃)₂ (51mg), and Na₂CO₃ (155mg) in THF/H₂O (4ml/1ml). The solution was heated at 80°C overnight. EtOAc was added to reaction and washed with 1N NaOH. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. The crude mixture was purified by preparative plate chromatography (5% MeOH/ CH₂Cl₂) to give 128mg of an oil (65%, MH+=271).

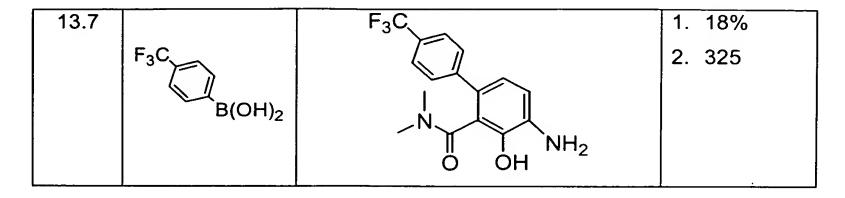
Step B

Following a similar procedure as in Preparative Example 13.3 Step E and using the product from Step A above, the desired compound was prepared (0.1 g, 69%, MH+=257.1).

PREPARATIVE EXAMPLES 13.5-13.7

Following the procedures set forth in Preparative Example 13.4 but using the boronic acid indicated in the Table below, the amine products were obtained.

Prep	Boronic Acid	Product	1.	Yield (%)
Ex.			2.	MH⁺
13.5		FN,	1.	15%
	B(OH) ₂	N NIL	2.	258
	(- //2	O OH NH ₂		
13.6		CF ₃	1.	32%
	CF ₃ B(OH) ₂	NH ₂	2.	325



PREPARATIVE EXAMPLE 13.8

Step A

5

10

15

20

2-Cyanophenol (500mg), sodium azide (819mg), and triethylamine hydrochloride (1.73g) were combined in anhydrous toluene and heated to 99°C overnight. After the reaction cooled down, product was extracted with H₂O. Aqueous layer was acidified with conc. HCl dropwise giving a precipitate, which was filtered to give the product (597mg, 87%, MH+=163).

Step B

Nitric acid (0.034ml) in AcOH (5ml) was added to the product from Step A above (100mg) in AcOH and the mixture was allowed to stir for 1hr. CH₂Cl₂ and H₂O were added to reaction. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give an oil. Trituration in ether gave the product as a solid (12mg, 9%, MH+=208).

Step C

The product from step C (56mg) was stirred with 10% Pd/C (20mg) in EtOH/MeOH (15ml) under a hydrogen gas atmosphere overnight. The reaction mixture was filtered through celite, the filtrate was concentrated *in vacuo* to give 29mg of a solid (62%, MH+=178).

The amine was prepared following the procedure disclosed in WO 01/68570, the disclosure of which is incorporated herein by reference thereto.

PREPARATIVE EXAMPLE 13.10

The amine was prepared following the procedure disclosed in WO 01/68570, the disclosure of which is incorporated herein by reference thereto.

PREPARATIVE EXAMPLE 13.11

15 Step A

Following the procedure described in Preparative Example 88.2, Step A, the ketone was prepared (6.4g, 36%).

To a solution of ketone (1g) and 2-*R*-methylbenzylamine (0.73ml) in anhydrous toluene (20ml) was added 1N TiCl₄ in toluene (3ml) at room temperature for 1.5hrs. The precipitate was filtered and the filtrate was concentrated *in vacuo* and purified via flash column chromatography (Hex/EtOAc, 18/1) to give 800mg of product (71%).

Step C

5

10

15

The imine from above (760mg) and DBU (800ul) were stirred without solvent for 4hr. The crude reaction was concentrated *in vacuo* and purified via flash column chromatography (Hex/EtOAc, 8/1) to give 600mg of product (79%).

Step D

The imine from Step C (560mg) was dissolved in ether (8ml). 3N HCl (5ml) added and let stir at room temperature overnight. The ether layer was separated and concentrated *in vacuo* to give 400mg of the amine hydrochloride product (93%).

PREPARATIVE EXAMPLE 13.12

The title compound was prepared similarly as in Preparative Example 13.11, but using the 2-S-methylbenzylamine instead of 2-R-methylbenzylamine (69%).

PREPARATIVE EXAMPLE 13.13

Step A

At room temperature, CsF (60mg) was added to a mixture of furfuraldehyde (1.3ml) and TMS-CF₃ (2.5g) and stirred at room temperature (24 h) and refluxed for another 12h. 3N HCl (40ml) was added and after 4hr, the mixture was extracted with ether, washed with brine, dried over MgSO₄, and concentrated *in vacuo* to give the product (2.6g, 100%).

Step B

5

10

15

To a solution of alcohol from above (2.6g) in CH₂Cl₂ at room temperature was added Dess-Martin reagent (10g) portionwise and 1 drop of water. After stirring for 3hr at room temperature, 10% Na₂S₂O₃ (60ml) was added and after stirring overnight, the solid was filtered off and the filtrate was extracted with CH₂Cl₂. The organic layer was washed with saturated sodium bicarbonate, dried with MgSO₄, filtered and concentrated *in vacuo*. Ether/hexane (1:2; 30ml) was added to the residue, filtered, and filtrate concentrated *in vacuo* to give the product (2g, 78%).

Step C

Following the procedures described in Preparative Example 13.11, Steps B, C and D, the amine salt was prepared.

20

PREPARATIVE EXAMPLES 13.15-13.17

Following the procedure set forth in Preparative Example 13.13, but using the prepared or commercially available aldehydes, the optically pure amine products in the Table below were obtained.

Prep	Aldehyde	Amine	Product	Yield
Ex.				(%)
13.15	34.12			20
:	0		CF₃ ₹ ∠O	
	H	H ₂ N	CIH.H ₂ N	
	CI		CI	
13.16				31
			ÇF₃ ,O	
	H T	H ₂ N	CIH.H ₂ N	
	Br		Br	
13.17			CF ₃	66
	H T	H ₂ N	CIH.H ₂ N	
			0	
13.17A	34.8			20
13.17A	0 0	ÇF₃	ÇF₃	38
	н	$H_2N \stackrel{\overline{\cdot}}{\longrightarrow} O$	CIH.H ₂ N (V)	
13.17B	, O	ÇF₃	∑F ₃	31
	H	H_2N O	CIHH ₂ N O	
	—	×		;
			,	

$$F_3C$$

CIH.H₂N

 $CIH.H_2$ N

The title compound was prepared from trifluorophenylketone according to the procedures described in Preparative Example 13.11, Steps B, C, and D (68%).

PREPARATIVE EXAMPLE 13.19

Step A

5

10

15

20

Methyl-3-hydroxy-4-bromo-2-thiophenecarboxylate (10.0 g, 42.2 mmol) was dissolved in 250 mL of acetone. Potassium carbonate (30.0 g, 217.4 mmol) was added followed by a solution of iodomethane (14.5 mL, 233.0 mmol). The mixture was heated to reflux and continued for 6 h. After cooled to room temperature, the mixture was filtered, the solid material was rinsed with acetone (\sim 200 mL). The filtrate and rinsing were concentrated under reduced pressure to a solid, further dried on high vacuum, yielding 13.7 g (100%) of methyl-3-methoxy-4-bromo-2-thiophenecarboxylate (MH $^+$ = 251.0).

Step B

Methyl-3-methoxy-4-bromo-2-thiophenecarboxylate (13.7 g), available from step A, was dissolved in 75 mL of THF, and added with a 1.0 M sodium hydroxide aqueous solution (65 mL, 65.0 mmol). The mixture was stirred at room temperature for 24 h. A 1.0 M hydrogen chloride aqueous solution was added dropwise to the mixture until pH was approximately 2. The acidic mixture was extracted with CH₂Cl₂ (100 mL x 2, 50 mL). The combined organic extracts were washed with brine (40 mL), dried with Na₂SO₄, and concentrated under reduced pressure to a solid, 10.0 g

(100%, over two steps) of 3-methoxy-4-bromo-2-thiophenecarboxylic acid (MH^{\dagger} = 237.0).

Step C

5

10

15

20

25

30

To a stirred solution of 3-methoxy-4-bromo-2-thiophenecarboxylic acid (6.5 g, 27.4 mmol) in 140 mL of CH_2Cl_2 , obtained from step B, was added bromotripyrrolidinophosphonium hexafluorophosphate (PyBrop, 12.8 g, 27.5 mmol), a 2.0 M solution of dimethyl amine in THF (34.5mL, 69.0 mmol), and diisopropylethyl amine (12.0 mL, 68.7 mmol). After 3 d, the mixture was diluted with 100 mL of CH_2Cl_2 , and washed with a 1.0 M sodium hydroxide aqueous solution (30 mL x 3) and brine (30 mL). The organic solution was dried with Na_2SO_4 , filtered, and concentrated to an oil. This crude oil product was purified by flash column chromatography, eluting with CH_2Cl_2 -hexanes (1:1, v/v). Removal of solvents afforded a solid, further dried on high vacuum, yielding 6.76 g (93 %) of *N*, *N'*-dimethyl-3-methoxy-4-bromo-2-thiophenecarboxamide (MH $^+$ = 265.0, M+2 = 266.1).

Step D

An oven dried three-neck round bottom flask was equipped with a refluxing condenser, charged sequentially with palladium acetate (95 mg, 0.42 mmol), (R)-BINAP (353 mg, 0.57 mmol), cesium carbonate (9.2 g, 28.33 mmol), and N, N'dimethyl-3-methoxy-4-bromo-2-thiophenecarboxamide (3.74 g, 14.2 mmol, from step C). The solid mixture was flushed with nitrogen. Toluene (95 mL) was added to the solid mixture followed by benzophenone imine (3.6 mL, 21.5 mmol). The mixture was heated to reflux and continued for 10 h. A second batch of palladium acetate (95 mg, 0.42 mmol) and (R)-BINAP (353 mg, 0.57 mmol) in 5 mL of toluene was added. Refluxing was continued for 14 h. The third batch of palladium acetate (30 mg, 0.13 mmol) and (R)-BINAP (88 mg, 0.14 mmol) was added, and reaction continued at 110°C for 24 h. The mixture was cooled to room temperature, diluted with ether (50 mL), filtered through a layer of Celite, rinsing with ether. The filtrate and rinsing were concentrated under reduced pressure to an oil, which was purified twice by flash column chromatography using CH₂Cl₂ and CH₂Cl₂-MeOH (200:1) as eluents. Removal of solvents afforded 4.1 g (79 %) of the amido-thiophene diphenylimine product as a solid (MH $^+$ = 365.1).

Step E

5

10

15

20

25

To a stirred solution of thiophene imine (5.09 g, 13.97 mmol), obtained from step D, in 140 mL of CH₂Cl₂ at -78°C was added dropwise a 1.0 M solution of boron tribromide in CH₂Cl₂. The mixture was stirred for 3 h while the temperature of the cooling bath was increased slowly from -78°C to -15°C. 100 mL of H₂O was added, the mixture was stirred at room temperature for 30 min, then the two layers were separated. The organic layer (as A) was extracted with H₂O (30 mL x 2). The aqueous layer and aqueous extracts were combined, washed with CH₂Cl₂ (30 mL), and adjusted to pH ~ 8 using a saturated NaHCO₃ aqueous solution. The neutralized aqueous solution was extracted with CH₂Cl₂ (100 mL x 3), the extracts were washed with brine, dried with Na₂SO₄, and concentrated under reduced pressure to a light yellow solid, 1.49 g of N, N'-dimethyl-3-hydroxy-4-amino-2-thiophenecarboxamide (first crop). The previous separated organic layer A and organic washing were combined, stirred with 30 mL of a 1.0 M HCl aqueous solution for 1 h. The two layers were separated, the aqueous layer was washed with CH₂Cl₂ (30 mL) and adjusted to pH ~8 using a saturated NaHCO₃ aqueous solution, and the separated organic layer and organic washing were combined as organic layer B. The neutralized aqueous solution was extracted with CH_2Cl_2 (30 mL x 4), the extracts were washed with brine, dried by Na₂SO₄, and concentrated under reduced pressure to give 0.48g of a solid as the second crop of the titled product. Organic layer B from above was washed with brine, and concentrated to an oil, which was separated by preparative TLC (CH₂Cl₂-MeOH = 50:1) to afford 0.45 g of a solid as the third crop of the titled product. The overall yield of the product, N, N'-dimethyl-3-hydroxy-4-amino-2thiophenecarboxamide, is 2.32 g (89%) (MH^{+} = 187.0).

PREPARATIVE EXAMPLE 13.20

Step A

5

10

15

20

25

To the product from Preparative Example 13.19 Step D (1.56g) in CH₂Cl₂ (55ml) was added potassium carbonate (1.8g) followed by dropwise addition of bromine (0.45ml). After 5hr of mixing, water (100ml) was added to the reaction and the layers were separated. The aqueous layer was extracted with CH₂Cl₂, which was then washed with brine, saturated sodium bicarbonate, and brine again. The organic layer was dried with Na₂SO₄, and concentrated *in vacuo*. The residue was purified via flash column chromatography (CH₂Cl₂) to yield 1.6g of product (83%).

Step B

The product from above was reacted in the procedure set forth in Preparative Example 13.19 Step C to give the amine.

PREPARATIVE EXAMPLE 13.21

Step A

To the product from Preparative Example 13.20, Step A (300mg) in THF (7ml) at -78 °C was added a solution of n-BuLi (1.6M in hexanes, 0.54ml). After 1hr, iodomethane (0.42ml) was added dropwise. After 3 hrs of stirring at -78 °C, the reaction was warmed to room temperature overnight. Saturated ammonium chloride and water were added to the reaction and extracted with CH₂Cl₂. The organic layer was washed with saturated sodium bicarbonate and brine, dried over Na₂SO₄, and concentrated *in vacuo*. The crude product was purified by preparative plate chromatography (CH₂Cl₂-MeOH = 70:1 to 50:1) to afford the product (111mg, 43%).

Step B

The product from above was reacted in the procedure set forth in Preparative Example 13.19, Step E to give the amine.

Step A

5

10

15

20

25

To the product from Preparative Example 13.19 (400mg), Step D in CH_2CI_2 -pyridine (14ml) was added N-chlorosuccinimide (220mg). The mixture was stirred for 5hr and then diluted with CH_2CI_2 and washed with water, saturated sodium bicarbonate and brine, and concentrated *in vacuo*. The crude product was purified via preparative plate chromatography (CH_2CI_2 -MeOH = 50:1) to give 180mg of product (64%).

Step B

The product from above (274mg) was reacted in the procedure set forth in Preparative Example 13.19, Step E to give the amine (89mg, 58%).

PREPARATIVE EXAMPLE 13.23

Step A

To a stirred solution of acid (630mg) from Preparative Example 13.19, Step B in CH₂Cl₂ (25ml) was added oxalyl chloride (235ul) followed by a catalytic amount of DMF (10ul). The mixture was stirred for 1hr, then potassium carbonate (1.8g) was added followed by 3-amino-5-methylisoxazole (443mg). The reaction stirred overnight and was quenched with water (25ml). Layers were separated and the organic layer was washed with brine, dried over Na₂SO₄, and concentrated *in vacuo*. The crude product was purified by preparative plate chromatography (CH₂Cl₂) to afford the product (580mg, 78%, MH+=317,319).

The acid from the above (750mg) step was reacted following the procedure set forth in Preparative Example 13.3, Step B to yield 625mg of product (80%, MH+=331).

Step C

5

10

The product from above was reacted following the procedure set forth in Preparative Example 13.19, Step D to yield 365mg of product (53%)

Step D

The product from above was reacted following the procedure set forth in Preparative Example 13.19, Step E to give the amine product (MH+=254).

PREPARATIVE EXAMPLE 13.25

Step A
$$F_3$$
C $Step B$ F_3 C $Step B$ F_3 C F_4

Step A

15

20

To a solution of 2-methylfuran (1g) in ether (30ml) was added n-BuLi (5.32ml) at –78°C. The reaction was warmed to room temperature and then refluxed at 38 °C for 1hr. The reaction was cooled back down to –78°C where the furyl lithium was quenched with trifluorobutyraldehyde and let stir at room temperature overnight. Saturated ammonium chloride added and extracted with ether. Purified via flash column chromatography to yield pure product (2g, 80%)

The azide was prepared using the procedure from Preparative Example 75.75, Step B and the alcohol (1g) from above and carried on crude to Step C below.

5 Step C

The amine was prepared using the procedure from Preparative Example 75.75, Step C to yield 400mg of an oil (53%).

PREPARATIVE EXAMPLE 13.26

$$C_2F_5$$
 Step C H_2N C_2F_5

Step A

10

15

20

Perfluoroiodide (3.6ml) was condensed at –78°C. Ether (125ml) was added followed by the methyllithium.lithiumbromide complex (1.5M in ether, 18.4ml). After 15min, a solution of 5-methylfuraldehyde (2.5ml) in ether was added dropwise. The reaction was warmed to -45°C and let stir for 2hr. Saturated ammonium chloride (30ml) and water (30ml) were added and let stir at room temperature for 1hr. The layers were separated and the aqueous layer was extracted with CH₂Cl₂. The organic layer was washed with brine, dried with Na₂SO₄, filtered and concentrated *in vacuo* to give 5.86g of product (100%).

Step B

The alcohol from above was reacted to form the azide using the procedure set forth in Preparative Example 75.75 Step B.

Step C

5

The azide from above was reacted to form the racemic amine using the procedure set forth in Preparative Example 75.75 Step C.

PREPARATIVE EXAMPLE 13.27

$$C_2F_5$$
 Step D C_2F_5 C_2F_5

Step A

Following the procedure set forth in Preparative Example 13.26, Step A, the alcohol was prepared (100%).

Step B

10

15

20

To a solution of the alcohol (500mg) from step A above in CH₂Cl₂ (20ml) was added N-methyl-morpholine monohydrate (575mg) and a catalytic amount of tetrapropyl ammonium perruthenate (76mg). After 3hr, the mixture was diluted with hexane (10ml) and filtered through a silica pad, rinsing with hexane: CH₂Cl₂ (200ml). The filtrate was concentrated *in vacuo* to give 350mg of product (70.7%)

Step C

The ketone (1.19g) from Step B was dissolved in THF (9.5ml) and cooled to 0 °C. A solution of S-methyl oxazoborolidine (1M in toluene, 1ml) followed by a solution of borane complexed with dimethylsulfide (9.5ml, 2M in THF) was added to the solution. The mixture was stirred at 0 °C for 30min and continued at room temperature for 5hr. The mixture was cooled back down to 0 °C and methanol (15ml)

was added dropwise to the mixture. After 30min, the mixture was concentrated *in vacuo* to give an oily residue.

The residue was dissolved in CH₂Cl₂ and washed with 1N HCl, water, and brine. Dried with Na₂SO₄, filtered and concentrated *in vacuo*. The crude material was purified via flash column chromatography (Hex/ CH₂Cl₂, 1:1) to afford 1.14g of an oil (67%).

Step D

5

10

15

The alcohol (1.14g) from above was reacted to form the azide using the procedure set forth in Preparative Example 75.75 Step B.

Step E

The azide (1.11g) from above was stirred with 10% Pd/C (280mg) in EtOH (40ml) under a hydrogen gas atmosphere overnight. The reaction was filtered through celite, the filtrate was concentrated *in vacuo* to give 700mg of product (70%).

PREPARATIVE EXAMPLE 13.28

Ms represents methanesulfonyl

20 <u>Step A</u>

25

To a stirred solution of 1-(2-thienyl)-1-propanone (3g) in acetic anhydride (6ml) at 0°C was added dropwise a solution of fuming nitric acid in acetic acid (2ml in 10ml). After 30min, the reaction was warmed to room temperature and let stir for 5hrs where a solid precipitated out. Ice was added to the reaction and the solid was filtered. The solid was purified by flash column chromatography (Hex/ CH₂Cl₂, 3:1 and 2:1) to yield 800mg of desired product (20%).

The above nitro-thiophene compound (278mg) was reduced using the procedure set forth in Preparative Example 2, Step B to give 54mg of product (23%).

Step C

5

10

15

20

25

The above amine (395mg), TEA (1ml) and methanesulfonylchloride (0.5ml) were combined in CH₂Cl₂ (35ml) and stirred at room temperature for 1hr. The reaction was quenched with saturated sodium bicarbonate (15ml). The organic layer was washed with brine, dried over Na₂SO₄, filtered and concentrated *in vacuo* to afford product (854mg, 100%).

Step D

To the above product (854mg) in THF (25ml) was added dropwise a solution of tetrabutylammonium fluoride (1M in THF, 2.8ml). The mixture was stirred overnight, then diluted with CH₂Cl₂ (30ml), washed with ammonium chloride and brine, dried over over Na₂SO₄, filtered and concentrated *in vacuo* to afford product (2.36g, >100%).

Step E

The ketone (2.36g) above was reacted via the procedure set forth in Preparative Example 88.2, Step B to yield 547mg of product (86.6%).

Step F

To the product from step E (310mg) in dimethoxyethane (12ml) was added dropwise a solution of LAH (1M in ether, 3.8ml). The mixture was heated to reflux overnight. The reaction was cooled to room temperature, SiO₂ was added as well as water (1ml) dropwise and let stir for 15min. The mixture was filtered and the filtrate was concentratred *in vacuo*. The crude product was purified by preparative plate chromatography (MeOH/ CH₂Cl₂, 15:1) to give the amine product (40mg, 14%).

Step A

5

10

15

To a solution of 3-methoxythiophene (3 g) in dichloromethane (175 mL) at – 78°C was added chlorosulfonic acid (8.5 mL) dropwise. The mixture was stirred for 15 min at –78°C and 1.5 h at room temp. Afterwards, the mixture was poured carefully into crushed ice, and extracted with dichloromethane. The extracts were washed with brine, dried over magnesium sulfate, filtered through a 1-in silica gel pad. The filtrate was concentrated in vacuo to give the desired compound (4.2 g).

Step B

The product from Step A above (4.5 g) was dissolved in dichloromethane (140 mL) and added with triethylamine (8.8 mL) followed by diethyl amine in THF (2*M*, 21 mL). The resulting mixture was stirred at room temperature overnight. The mixture was washed with brine and saturated bicarbonate (aq) and brine again, dried over

sodium sulfate, filtered through a 1-in silica gel pad. The filtrate was concentrated in vacuo to give the desired compound (4.4 g).

Step C

5

10

15

20

25

30

The product from Step B above (4.3 g) was dissolved in dichloromethane (125 mL) and cooled in a -78°C bath. A solution of boron tribromide (1.0 M in dichloromethane, 24.3 mL) was added. The mixture was stirred for 4 h while the temperature was increased slowly from -78°C to 10° C. H_2 O was added, the two layers were separated, and the aqueous layer was extracted with dichloro- methane. The combined organic layer and extracts were washed with brine, dried over magnesium sulfate, filtered, and concentrated in vacuo to give 3.96 g of the desired hydroxy-compound.

Step D

The product from step C above (3.96 g) was dissolved in 125 mL of dichloromethane, and added with potassium carbonate (6.6 g) followed by bromine (2 mL). The mixture was stirred for 5 h at room temperature, quenched with 100 mL of H₂O. The aqueous mixture was addjusted to pH \sim 5 using a 0.5N hydrogen chloride aqueous solution, and extracted with dichloromethane. The extracts were washed with a 10 % Na₂S₂O₃ aqueous solution and brine, dried over sodium sulfate, and filtered through a celite pad. The filtrate was concentrated in vacuo to afford 4.2 g of the desired bromo-compound.

Step E

The product from Step D (4.2 g) was dissolved in 100 mL of acetone and added with potassium carbonate (10 g) followed by iodomethane (9 mL). The mixture was heated to reflux and continued for 3.5 h. After cooled to room temperature, the mixture was filtered through a Celite pad. The filtrate was concentrated in vacuo to a dark brown residue, which was purified by flash column chromatography eluting with dichloromethane-hexanes (1:1, v/v) to give 2.7 g of the desired product.

Step F

The product from step E (2.7 g) was converted to the desired imine compound (3 g), following the similar procedure to that of Preparative Example 13.19 step D.

5 Step G

10

15

20

25

The imine product from step F (3 g) was dissolved in 80 mL of dichloromethane and cooled in a -78° C bath. A solution of boron tribromide (1.0 M in dichloromethane, 9.2 mL) was added dropwise. The mixture was stirred for 4.25 h from -78° C to 5° C. H_2 O (50 mL) was added, and the layers were separated. The aqueous layer was extracted with dichloromethane. The organic layer and extracts were combined, washed with brine, and concentrated to an oily residue. The residue was dissolved in 80 mL of methanol, stirred with sodium acetate (1.5 g) and hydroxyamine hydrochloride (0.95 g) at room temperature for 2 h. The mixture was poured into an aqueous mixture of sodium hydroxide (1.0 M aq, 50 mL) and ether (100 mL). The two layers were separated. The aqueous layer was washed with ether three times. The combined ether washings were re-extracted with H_2 O once. The aqueous layers were combined, washed once with dichloromethane, adjusted to pH \sim 6 using 3.0 M and 0.5 M hydrogen chloride aqueous solutions, and extracted with dichloromethane. The organic extracts were combined, washed with brine, dried over sodium sulfate, and concentrated in vacuo to give 1.2 g of desired amine compound.

PREPARATIVE EXAMPLES 13.30-13.32-A

Following the procedures set forth in Example 13.29, but using commercially available amines, hydroxy-amino-thiophene products in the Table below were obtained.

Prep Ex.	Amine	Product	Yield (%) MH ⁺
13.30	(Bn)₂NH	Bn NH ₂	10% 375.1
13.31	Me(Bn)NH	Bn-NSSS NH2	14% 299.0
13.32	Et(Bn)NH	Bn-NSSSSNH2	22%
13.32A	(Et)₂NH	Et N S NH ₂	25%

Step A

5

10

2-Chlorosulfonyl-3-methoxy-thiophene (4.0 g, 18.8 mmol), the product from Step A of Preparative Example 13.29 was converted to 3-methoxy-2-ethylbenzylsulfonyl-thiophene (5.5 g, 94%, MH⁺ = 312.1) by using ethylbenzyl-amine, following the procedure set forth in Preparative Example 13.29, Step B.

Step B

The product from Step A above (5.5 g, 17.70 mmol) was demethylated following the procedure set forth in Preparative Example 13.29, Step C. The alcohol product was obtained in 4.55 g (87%, MH^+ = 298.0).

Step C

The product from Step B above (4.55 g, 15.30 mmol) was brominated using the procedure set forth in Preparative Example 13.29, Step D. The corresponding bromide was obtained in 4.85 g (84%).

5

Step D

The bromo-alcohol from Step C above (4.84 g, 12.86 mmol) was methylated using the procedure set forth in Preparative Example 13.29, Step E. The product was obtained in 4.82 g (96%).

10

15

20

25

30

Step E

The product from Step D above (4.82 g, 12.36 mmol) was stirred with concentrated sulfuric acid (5 mL) at room temperature ro 3 h. Ice water (30 mL) was added to the mixture followed by CH_2Cl_2 (50 mL). The aqueous mixture was adjusted to pH ~ 6 using a 1.0 M NaOH aqueous solution. The layers were separated. The aqueous layer was extracted with CH_2Cl_2 (50 mL x 3). The combined organic layers were washed with brine, dried over Na_2SO_4 , and concentrated to a dark brown oil, which was purified by flash column chromatography, eluting with CH_2Cl_2 -hexanes (1:1, v/v). Removal of solvents afforded 3.03 g (82%) of the debenzylated product (M^+ = 300.0, M+2 = 302.0).

Step F

The product from Step E (1.34 g, 4.45 mmol) was methylated using the procedure set forth in Preparative Example 13.29, Step E. The desired product was obtained in 1.36 g (97%, M^+ = 314.1, M+2 = 316.0).

Step G

The product from Step F (1.36 g, 4.33 mmol) was converted to imine product (1.06 g, 55%, MH^{+} = 415.1) using the procedure set forth in Preparative Example 13.29, Step F.

Step H

The imine product from Step G (1.06 g, 2.56 mmol) was converted to the desired hydroxy-amino thiophene compound (0.26 g, 43%) using the procedure set forth in Preparative Example 13.29, Step G.

5

PREPARATIVE EXAMPLE 13.34

Step A

10

15

20

2-Chlorosulfonyl-3-methoxy-thiophene (3.8 g, 17.87 mmol), the product from step A of Preparative Example 13. 29, was dissolved in 100 mL of CH_2Cl_2 and 20 mL of pyridine. 3-Amino-5-methyl-isoxazole (3.5 g, 35.68 mmol) was added. The mixture was stirred for 20 h at room temperature, diluted with 100 mL of CH_2Cl_2 , and washed with a 0.5 N HCl aqueous solution (50 mL x 2), H_2O (50 mL), and brine (50 mL). The organic solution was dried with Na_2SO_4 , and conentrated in vacuo to a brown oil. This oil was dissolved in 100 mL of CH_2Cl_2 , washed again with a 0.5 M HCl aqueous solution (30 mL x 3) and brine. After dried over Na_2SO_4 , the organic solution was concentrated in vacuo to a yellow solid, 4.48 g (91%, $MH^+=275.0$) of the desired product.

Step B

The product from Step A above (4.48 g, 16.33 mmol) was dissolved in acetone (100 mL), added with potassium carbonate (5.63 g, 40.80 mmol) and iodomethane (10.1 mL, 163.84 mmol). The mixture was stirred at room temperature for 1.5 h, diluted with 100 mL of hexanes and 50 mL of CH₂Cl₂, and filtered through a 1-in silica

gel pad, rinsing with CH_2Cl_2 . The filtrate was concentrated under reduced pressure to give 4.23 g (90%, MH^{\dagger} = 289.0) of the desired product as a light yellow solid.

Step C

5

10

15

To a stirred suspension of sodium hydride (130 mg, 95%, 5.4 mmol) in 8 mL of N, N'-dimethylforamide at room temperature was added ethanethiol (0.45 mL, 6.0 mmol) dropwise. After 5 min, the mixture became a clear solution, and was added to a stirred solution of the product obtained from Step B above (0.45 g, 1.56 mmol) in 2 mL of N, N'-dimethylforamide in a round bottom flask. The flask was sealed with a ground glass stopper, and the mixture was heated at 90-95°C for 4 h. After cooled to room temperature, the mixture was poured into 20 mL of a 1.0 M NaOH aqueous solution, further rinsed with 20 mL of H₂O. The aqueous mixture was washed with diethyl ether (30 mL x 2), adjusted to PH ~5 using a 0.5 M HCl aqueous solution, and extracted with CH_2Cl_2 (50 mL x4). The combined extracts were washed with brine, dried (Na₂SO₄), and concentrated to a dark yellow solution. This was dissolved in 50 mL of ethyl acetate, washed with H₂O (30 mL x2) and brine (30 mL), dried over Na₂SO₄. Evaporation of solvent gave 0.422 g of the alcohol product (99%, MH^+ = 275.0).

Step D

The alcohol obtained from Step C above (0.467 g, 1.70 mmol) was brominated using the procedure set forth in Preparative Example 13.29, Step D, to afford the corresponding bromide in 0.607 g (100%).

Step E

5

10

15

20

25

The bromide obtained from Step D above (0.607 g, 1.72 mmol) was methylated using the procedure set forth in Preparative Example 13.29, Step E, to give the desired product in 0.408 g (65%, M^+ = 367, M+2 = 369.1).

Step F

The product (0.405 g, 1.103 mmol) from Step E above was converted to the imine compound (0.29 g, 56%) using the procedure set forth in Preparative Example 13.29, Step F.

Step G

The imine product obtained from Step F above (0.29 g, 0.61 mmol) was demethylated using the procedure set forth in Step C above to give the corresponding alcohol as a dark yellow oil, which was dissolved in 5 mL methanol and added with sodium acetate (0.12 g, 1.46 mmol) and hydroxyamine hydrochloride (0.075 g, 1.08 mmol). The resulting mixture was stirred at room temperature for 3 h, and poured into 10 mL of 1.0 M NaOH aqueous solution. 30 mL of H_2O was used as rinsing and combined to the aqueous layer. The aqueous mixture was washed with diethyl ether (40 mL x 3), adjusted to pH ~ 6 using a 1.0 M HCl aqueous solution, and extracted with ethyl acetate (40 mL x 3). The organic extracts were washed with H_2O (20 mL x2), brine (20 mL), dried over Na_2SO_4 , and concentrated in vacuo to give 0.112 g of the desired hydroxy-amino thiophene sulfonamide (64%, MH^+ = 290).

Step A

5

10

To a solution of 2-methyl furan (1.72g) in ether was added BuLi (8.38mL) at -78° C and stirred at room temperature for half an hour. The reaction mixture again cooled to -78° C and quenched with cyclopropyl amide 1 and stirred for two hours at -78° C and slowly warmed to room temperature. The reaction mixture stirred for three hours at room temperature and quenched with the addition of saturated ammonium chloride solution. The mixture was taken to a separatory funnel, washed with water, brine and dried over anhydrous sodium sulfate. Filtration and removal of solvent afforded the crude ketone, which was purified by using column chromatography to afford the ketone 3.0g (87%) as a pale yellow oil.

5

10

To a solution of ketone (1.0g) in THF (5.0mL) at 0°C was added R-methyl oxazoborolidine (1.2Ml, 1M in toluene) dropwise followed by addition of a solution of borane complexed with dimethyl sulfide (1.85mL, 2M in THF). The reaction mixture was stirred for 30minutes at 0°C and than at room temperature for one hour. The reaction mixture was cooled to 0°C and MeOH was added carefully. The mixture was stirred for 20 minutes and was concentrated under reduced pressure. The residue was extracted with ether, washed with water, 1M HCI (10mL), saturated sodium bicarbonate (10.0mL) water and brine. The organic layer was dried over anhydrous sodium sulfate, filtered and removal of solvent afforded the crude alcohol which was purified by silica gel chromatography to afford the pure alcohol 0.91g (91%) as yellow oil.

PREPARATIVE EXAMPLE 13.36

Step A

An equimolar mixture of 2-methylfuran (1.0g) and anhydride (2.6g) was mixed with SnCl₄ (0.05mL) and heated at 100°C for 3 hours. After cooling the reaction mixture, water (10mL) was added, followed by saturated sodium carbonate solution until it becomes alkaline. The reaction mixture was extracted with ether several times and the combined ether layer was washed with water, brine and dried over anhydrous sodium sulfate. Filtration and removal of solvent afforded the crude ketone, which was purified by using silica gel chromatography to afford the ketone 0.9g (43%) as a yellow oil.

15

20

5

10

15

The step B alcohol was obtained following a similar procedure set forth in the preparative example 13.35 Step B.

PREPARATIVE EXAMPLE 13.37

Step A

To a solution of 5-methyl furan-2-aldehyde (1.0g) and 3-bromo-3,3-difluoropropene (2.24g) in DMF (30mL) was added indium powder (1.66g) and lithium iodide (50.0mg). The reaction mixture was stirred over night, diluted with water and extracted with ether. The ether layer was washed with water, brine and purified by silicagel chromatography to afford the pure alcohol 2.8g (92%).

PREPARATIVE EXAMPLES 13.38-13.45

Following a similar procedure set forth in Preparative Examples 13.25 and 13.35, and using the indicated Furan and Electrophile, the following Alcohols in the Table below were prepared.

Prep.	Furan	Electrophile	Alcohol	Yield
13.38		CHO	но	86%
13.39		COOEt	HO	69%

13.40	N OMe	HO	84%
13.41	N OMe	HO	82%
13.42	COOEt	HO	60%
13.43	COOEt	HOO	65%
13.44	F F N OMe	HOFF	82%
13.45	OHCCF ₃	HO CF ₃	89%

PREPARATIVE EXAMPLES 13.50-13.61

Following a similar procedure set forth in Preparative Examples 13.25, and using the indicated Alcohol, the following Amines in the Table below were prepared.

PREP. EX.	ALCOHOL	AMINE	% YIELD
13.50	13.45	H ₂ N O	28%
13.51	13.38	H_2N	58%
13.52	13.36	H_2N	69%
13.53	13.35	H_2N	81%
13.54	13.37	H ₂ N O	82%

13.55	13.39	H ₂ N O	45%
13.56	13.41	H_2N	57%
13.57	13.40	H_2N	58%
13.58	13.44	F F O	54%
13.59	13.42	H_2N O	53%

13.60	13.43	H ₂ N O	50%
13.61	13.37	H ₂ N O	82%

5 Step A

The imine was prepared following the procedure set forth in the preparative example 13.19 from the known bromoester (1.0g) as a yellow solid, Step A to yield 1.1g (79%).

10 <u>Step B</u>

The Step A product (0.6g) was reacted following the procedure set forth in the preparative example 13.19 to give the amine product 0.19g (64%).

Step C

The Step B product (1.0g) was reacted following the procedure set forth in the preparative example 13.19 to give the acid as yellow solid 0.9g (94%)

5 Step D

The Step C product (0.35g) was reacted following the procedure set forth in the preparative example 13.19 to give the amino acid as yellow solid 0.167g (93%).

PREPARATIVE EXAMPLE 13.71

10

Following a similar procedure set forth in Preparative Example 13.33 Step E, but using the product from Preparative Example 13.32, the title compound was obtained (121 mg, 69% yield, MH+ = 223.0).

15

20

PREPARATIVE EXAMPLE 14

Step A

3-Nitro-1,2-phenylenediamine(10 g), sodium nitrite (5.4 g) and acetic acid (20 mL) were heated at 60°C overnight, then concentrated *in vacuo*, diluted with water and extracted with EtOAc. The product precipitated from the organic phase (5.7 g) as a solid and used directly in step B.

5

The product from Step A above (2.8 g) was stirred with 10% Pd/C (0.3 g) in MeOH (75 mL) under a hydrogen gas atmosphere overnight. The reaction mixture was filtered through celite and the filtrate concentrated *in vacuo*, to give the product (2.2 g, MH+=135).

PREPARATIVE EXAMPLE 15

Step A

10

15

20

25

N-methyl-4-bromopyrazole-3-carboxylic acid was prepared according to known methods, see: Yu. A. M.; Andreeva, M. A.; Perevalov, V. P.; Stepanov, V. I.; Dubrovskaya, V. A.; and Seraya, V. I. in *Zh. Obs. Khim,* (Journal of General Chemistry of the USSR) 1982, 52, 2592 (and the references cited therein) the disclosure of whichis incorporated hereinby reference thereto.

Step B

To a solution of N-methyl-4-bromopyrazole-3-carboxylic acid (2.0 g), available from step A, in 65 mL of anhydrous DMF was added bromotripyrrolidinophosphonium hexafluorophosphate (PyBrop, 4.60 g), dimethyl amine (10 mL, 2.0 M in THF) and diisopropylethyl amine (5.2 mL) at 25 °C. The mixture was stirred for 26 h, and concentrated under reduced pressure to an oily residue. This residue was treated with a 1.0 M NaOH aqueous solution, and extracted with ethyl acetate (50 mL x 4). The organic extracts were combined, washed with brine, and dried with anhydrous Na₂SO₄. Removal of solvents yielded an oil, which was purified by preparative thin layer chromatography, eluting with CH₂Cl₂-MeOH (20:1), to give 1.09 g of the amide product (48%, MH $^+$ = 232.0).

Step C

5

10

20

25

To a solution of the amide (0.67 g), obtained from step B, in 8 mL of concentrated sulfuric acid at 0 °C was added potassium nitrate (1.16 g) in small portions. The cooling bath was removed and the mixture was heated at 110 °C for 6 h. After cooling to 25 °C, the mixture was poured into 80 mL of H₂O, and an additional 20 mL of H₂O was used as a rinse. The aqueous mixture was extracted with CH₂Cl₂ (100 mL x 4). The combined extracts were washed with brine (50 mL), sat. NaHCO₃ aqueous solution (50 mL), brine (50 mL), and dried with Na₂SO₄. Evaporation of solvent gave an oil, which solidified on standing. The crude product was purified by flash column chromatography, eluting with CH₂Cl₂-MeOH (1:0, 50:1 and 40:1). Removal of solvents afforded 0.521 g (65%) of the product as a solid (MH⁺ = 277.1)

15 Step D

The product (61 mg) obtained from step C was dissolved in 3 mL of THF. To this solution at -78 °C was added dropwise along the inside wall of the flask a 1.6 M solution of *n*-butyl lithium in hexane. After 45 min, a solution of methyl borate (0.1 mL) in THF (1.0 mL) was added. After 1.5 h, a solution of acetic acid in THF (0.25 mL, 1:10 v/v) was added to the cold mixture. Stirring was continued for 10 min, and a 30 wt % aqueous hydrogen peroxide solution (0.1 mL) was added. An additional portion of hydrogen peroxide aqueous solution (0.05 mL) was added 20 min later. The cooling bath was removed, and the mixture was stirred at 25 °C for 36 h. The mixture was poured into 30 mL of H_2O , and the aqueous mixture was extracted with ethyl acetate (30 mL x 4). The extracts were combined, washed with brine (10 mL), 5% NaHCO₃ aqueous solution (10 mL) and brine (10 mL). The organic layer was dried with Na₂SO₄ and concentrated under reduced pressure to a residue, which was then purified by preparative thin layer chromatography eluting with CH_2CI_2 -MeOH (20:1) to give the hydroxylated product (5 mg, 10%, MH⁺ = 215.3).

30

Step E

By treating the hydroxylated product of Step E with H₂ under the conditions of 10% palladium on carbon in ethanol, one would obtain the desired hydroxyl-amino compound.

5 Step A

Following a similar procedure used in Preparative Example 13, Step C except using the known compound, 4-methyl-pyrimidin-5-ol, the product can be prepared.

Step B

Following a similar oxidation procedure used in Preparative Example 15, Step A except using the compound from Step A above, the product can be prepared.

Step C

Following a similar procedure used in Preparative Example 11, Step A except using the compound from Step B above, the product can be prepared.

Step D

Following a similar procedure used in Preparative Example 12, Step F except using the compound from Step C above, the product can be prepared.

Step A

Following a similar procedure used in Preparative Example 11, Step A except using the known 4-hydroxynicotinic acid, the product can be prepared.

Step B

Following a similar procedure used in Preparative Example 13, Step C except using the compound from Step A above, the product can be prepared.

Step C

Following a similar procedure used in Preparative Example 12, Step F except using the compound from Step C above, the product can be prepared.

15

10

PREPARATIVE EXAMPLE 18

Step A

Following a similar procedure used in Preparative Example 13, Step C except using the compound from Step A above, the product can be prepared.

5 Step B

Stirring the compound from Step A above, a suitable Pt or Pd catalyst and EtOH under hydrogen atmosphere (1-4 atm) the product can be prepared.

PREPARATIVE EXAMPLE 19

10

The amine was prepared following WO 01/68570, the disclosure of whichis incorporated herein by reference thereto.

PREPARATIVE EXAMPLE 19.1

15

The amine was prepared following WO 01/68570, the disclosure of which is incorporated herein by reference thereto.

PREPARTIVE EXAMPLE 20

20

The title compound was prepared according to the procedure set forth in Preparative Example 1, but instead using 4-nitrosalycilic acid (57%, MH+=181).

PREPARATIVE EXAMPLE 21

The above compound is prepared following the procedure set forth in *The Journal of Organic Chemistry*, **1975**, 40(19), 2743-2748, the disclosure of which is incorporated herein by reference thereto.

PREPARTIVE EXAMPLE 22

The compound from Preparative Example 21 (250mg) and the compound from Preparative Example 3 (252mg) were combined in MeOH (15ml) and stirred overnight. The reaction was concentrated under vacuo and used crude (450mg, 99%, MH+=327).

PREPARATIVE EXAMPLE 22.1

15

20

5

3,4-Diethoxy-1,2-5-thiadiazole-1,1-oxide (226mg, 1.4mmol) (prepared according to known methods, see: *J. Am. Chem. Soc.*, **1982**, p. 1375, the disclosure of which is incorporated herein by reference thereto) was added to 3-amino-2-hydroxy-N,N-dimethylbenzamide (252mg, 1.4mmol) in methanol (15mL). The reaction mixture was stirred overnight. The desired product precipitated and was recovered by filtration. Concentration of the mother liquor to half volume afforded a second batch of precipitated product. Combined batches afforded 293mg (65% yield) of product with sufficient purity to be used in subsequent steps. MH⁺ = 346.9.

PREPARATIVE EXAMPLE 22.2

3,4-Diethoxy-1,2-5-thiadiazole-1,1-oxide (226mg, 1.4mmol) (prepared according to known methods, see: *J. Am. Chem. Soc.*, **1982**, p. 1375, the disclosure of which is incorporated herein by reference thereto) was added to R-2-phenylpropylamine (0.195mL, 1.4mmol) in methanol (15mL). The reaction mixture was stirred overnight. Evaporation of solvent afforded an amorphous solid (390mg, 99%) of sufficient purity for use in subsequent steps. $MH^+ = 279.9$.

PREPARATIVE EXAMPLES 22.3-22.7

Following a similar procedure set forth in Preparative Example 22.1 but using the commercially available (or prepared amine) indicated in the Table below, the following thiadiazoleoxide intermediates could be obtained.

Ex.	Amine	Product
22.3	Br NH ₂ O OH	O Me N O Me N N O Me N O Me
22.4	NH ₂	O N O O Me
22.5	NH ₂	N OME

10

22.6	NH ₂ NH ₂ NH ₂	O S O HO H
22.7	N S NH ₂	N S N OME H

PREPARATIVE EXAMPLES 22.8-22.38

Following a similar procedure as that set forth in Preparative Example 22.1 but using the commercially available (or prepared amine) indicated in the Table below, the following thiadiazoleoxide intermediates could be obtained.

Ex.	Amine	Product		
22.8	CI N S NH ₂	CI N N OEt		
22.9	N S NH ₂	N S N OEt		
22.10 CI NH ₂ NH ₂		CI N OEt OEt		

22.16	N N NH2	OH NO OEL
22.17	Br N S O OH	Br N N OEt
22.19	N S NH ₂	O S N O O O O O O O O O O O O O O O O O
22.20	H S-NH ₂	H S N OEt
22.21	N S O OH	O S N O O O O O O O O O O O O O O O O O
22.22	N S NH ₂	S N S N OEt

22.23	P NH ₂	P OH H		
22.24	N OH NH2	OH NOEt		
22.25	$N-N$ $N-N$ NH_2 O OH	O O O O O O O O O O O O O O O O O O O		
22.26	H S NH ₂	O S N O O O O O O O O O O O O O O O O O		
22.27	F ₃ C N N N N O O OH	F ₃ C O S N O O O O O O O O O O O O O O O O O		
22.28	S NH ₂	S O OH H OEI		

	-		
22.29	F ₃ C N N O OH NH ₂	F ₃ C O S N O DEt	
22.30	N S NH ₂	O S N O Et	
22.31	F ₃ C N N O O O O H	F ₃ C N OEt	
22.32	CI N S OH NH ₂	O S N O DEt	
22.34	CI N N S O OH	O=S N OEt OEt	
22.36	N-O NH2	O S N O O O O O O O O O O O O O O O O O	

22.37	N NH2	NO OH HOEL
22.38	Br NH ₂	Br N OEt

PREPARATIVE EXAMPLES 22.39-22.51

Following a similar procedure set forth in Preparative Example 22.1 but using the commercially available (or prepared amine) indicated in the Table below, the following thiadiazoleoxide intermediates could be obtained.

Ex.	Amine	Product		
22.39	F ₃ C—NH ₂	F ₃ C-NOEt		
22.40	N-OOH	ON S.N OEt OH		

22.41	ONO OH	ON-OEt OH
22.42	HO, NH ₂ OH	HO, NOEt OH
22.43	-NN-N-NH2	N.S.N N.OEt OH
22.44	HO_N-NH2	HONO OET
22.45	N-S OH	N-SOOH

22.46		0	
	Br NH ₂ O OH	Br N N OMe	
22.47		0	
	N-S OH	N-S OH	
22.48	N N N NH ₂	O S N O O O O O O O O O O O O O O O O O	
22.49	N N N NH ₂	N N N OEt	
22.50	N N N NH₂	N O S N O O O O O O O O O O O O O O O O	
22.51	ON NON NH2	O O O O O O O O O O O O O O O O O O O	

22.52	NH ₂	O II S N O O O O O O O O O O O O O O O O O O
22.53	CI—NH ₂ H ₂ N—S OH	CI N OEt

5

PREPARATIVE EXAMPLES 23.1-23.9

Following a similar procedure set forth in Preparative Example 22 but using the commercially available (or prepared amine) indicated in the Table below, the following thiadiazoledioxide intermediates were obtained.

Ex.	Amine	Product	1. 2.	Yield MH+
23.1	H ₂ N	MeO N H	1.	99% 281.8

23.2	N OH NH ₂	N OMe	1. 99% 2. 327
23.3	H ₂ N	OH H OO O N N MeO N H	1. 99% 2. 234.0
23.4	CF ₃	O O O N CF3 MeO N O H	1. 99% 2. Not observed
23.5	CF ₃	MeO N CF ₃	1. 99% 2. Not observed
23.6	H ₂ N O	N N N N N N N N N N N N N N N N N N N	 99% Not observed
23.7	H ₂ N O	MeO NH	 99% Not observed
23.8	Br NH ₂ O OH	O S N O O Me	1. 99% 2. 404.9
23.9	O ₂ N NH ₂	O ₂ N N N N OMe	 99% Not observed

PREPARATIVE EXAMPLES 23.30-23.41

Following a similar procedure set forth in Preparative Example 22 but using the commercially available (or prepared amine) indicated in the Table below, the following thiadiazoledioxide intermediates could be obtained.

1	ø	٠	•
į			٠
•	L		2

Ex.	Amine	Product
23.30	H ₂ N	MeO N
23.33	H ₂ N CI	MeO NH
23.34	H ₂ N CI	MeO NH CI
23.35	H ₂ N	N N N N N N N N N N N N N N N N N N N
23.37	CF ₃	MeO N CF3
23.38	CF ₃	MeO N CF3
23.40	N S NH₂ NH₂	N S N OME HO H

PREPARATIVE EXAMPLE 24

Step A

5

10

20

To a solution of *N*-protected amino acid (1.5 g, 6.9 mmol) in CH_2Cl_2 (25 mL) at room temperature was added DIPEA (3.6 mL, 20.7 mmol), and PyBrop (3.4 g, 6.9 mmol) followed by MeNH₂ (6.9 mL, 13.8 mmol, 2.0 M in CH_2Cl_2). The resulting solution was stirred for 18 h at room temperature (until TLC analysis deemed the reaction to be complete). The resulting mixture was washed sequentially with 10% citric acid (3 x 20 mL), sat. aq. NaHCO₃ (3 x 20 mL), and brine (3 x 20 mL). The organic layer was dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography eluting with $CH_2Cl_2/MeOH$ (40:1) to afford 1.0 g (63% yield) of a solid.

15 <u>Step B</u>

To a round bottom charged with the N-protected amide (1.0 g, 4.35 mmol) (from Step A) was added 4N HCl/dioxane (10 mL) and the mixture was stirred at room temperature for 2 h. The mixture was diluted with Et₂O (20 mL) and concentrated under reduced pressure. The crude product was treated with Et₂O (2 x 20 mL) and concentrated under reduced pressure to afford 0.72 g (~100 % yield) of crude product as the HCl salt. This material was taken on without further purification or characterization.

PREPARATIVE EXAMPLES 25-33.1

Following the procedure set forth in Preparative Example 24 but using the commercially available *N*-protected amino acids and amines in the Table below, the amine hydrochloride products were obtained.

	_
И	_
	_
	- 1
٩	_

				-
Prep Ex.	Amino acid	Amine	Product	Yield
LX.				(%)
25	XO NO OH	NH ₃	CIHH ₂ ·N NH ₂	70
26	YOH NHOOH	H ₂ N	CIHH ₂ ·N	71
27	Хо Н О О Н	H ₂ N	CIHH ₂ ·N	66
28	Д Н Он	H ₂ N	CIH.H ₂ N O	65
29	Ход Н Он	H ₂ N	CIH.H ₂ N	90
30	YOU NO OH	H ₂ N	CIHH ₂ ·N H N H	68

31	YOH NOH	H ₂ N	CIHH ₂ ·N	68
32	HZ O=\ O=\ H	H ₂ N	CIHH ₂ ·N	97
33	YOH OH	H ₂ N	CIH.H ₂ N	97
33.1	NH OH	H ₂ N	CIH.H ₂ N	20

PREPARATIVE EXAMPLE 33.2

Step A

BOC-valine (45mg) and PS-carbodiimide (200mg) were suspended in CH₂Cl₂ (4ml). After addition of the CH₂Cl₂-amine solution (0.138N, 1ml), the mixture was shaken overnight. The solution was filtered and the resin was washed with more CH₂Cl₂, and the filtrate was concentrated *in vacuo* to yield the product, which was carried on directly in Step B.

Step B

The crude material from Step A was dissolved in 4N HCl/dioxane (2.5ml) and stirred for 2h. The reaction was concentrated *in vacuo* to yield the desired amine hydrochloride, which was used directly in the next step.

10

PREPARATIVE EXAMPLES 33.3-33.47

If one were to follow the procedure set forth in Example 33.2 but using the commercially available N-protected amino acids in the Table below, one could obtain the amine hydrochloride products in the Table below.

_
~
v
_

Drop	Amino acid	Amino	Desduct
Prep Ex.	Amino acid	Amine	Product
33.3			HCI
	NH OH	H ₂ N	H ₂ N O
33.4		N=N,c	HCI
	DH OH	H ₂ N	H ₂ N N S
33.5			HCI
	O H O H	H ₂ N	H_2N
33.6			HCI
	OH OH	H ₂ N	H_2N
33.7	NH O OH	H ₂ N	HCI O H
33.8	NH OH	H ₂ N	H ₂ N → H HCI

33.9	NH OH	HN	HCI O
33.10	DH OH	H ₂ N	HCI H ₂ N N
33.11	O	H ₂ N	H ₂ N H ₂ N HCI
33.12	XON OH H O	H ₂ N	HCI H ₂ N O
33.13	NO THE OOH	H ₂ N	HCI O
33.14	NH OH	H_2N O	H_2N H_2N H_2N H_3N
33.15	NH OH OH	H ₂ N	H_2N H_2N H_2N
33.16	DE O	H ₂ N	HCI O
33.17	NH OH OH	H ₂ N	H_2N H_2N H_2N H_2N

33.18			
	NH OH	H ₂ N	HCI H ₂ N H
19	OHOO OH	H ₂ N CI	HCI O CI
33.20		H ₂ N	HCI N
33.21	OH OH	H ₂ N CI	HCI O
33.22	XON OH H O	H ₂ N	HCI H ₂ N N
33.23	N H O O O O O O O O O O O O O O O O O O	H ₂ N	HCI H ₂ N O O
33.24	ZII O O O O O	H ₂ N O	HCI O
33.25	ZH O DE O	H ₂ N	HCI O
33.26	XON OH H O	H ₂ N	HCI O

33.27		\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	F H 0
	DH OH	H ₂ N	HCI O
33.28	DH OH	H ₂ N	H ₂ N O
33.29	DE CONTRACTOR OF THE CONTRACTO	H ₂ N	H ₂ N → H HCl
33.30	DH OH OH	H ₂ N NO ₂	HCI NO ₂
33.31	OHOO OH	HN	H ₂ N → N → N → HCI O
33.32	ZH OHO	H ₂ N_,,,,O	HCI O
33.33	ZH OHO	H ₂ N	H ₂ N H N
33.34	ZH O H	H ₂ N F	HCI O F
33.35	NH OH	H ₂ N F	HCI OFF

33.36	,	H-N a	
	NH OH	H ₂ N	HCI H ₂ N H
33.37	NH OH OH	H ₂ N	HCI O
33.38	XON OH OH	H ₂ N	HCI O
33.39	DH O OH	H ₂ N	HCI O H
33.40	OH OH	H ₂ N	HCI O
33.41	XON OH OH	H ₂ N	HCI H ₂ N H
33.42	OH OH	H ₂ N	HCI O H ₂ N O O
33.43	NH OH	H ₂ N	HCI O H

33.44	DE SEE O	HN	HCI HO
33.45		HN	HCI O
33.46	A PART OF THE PART	H ₂ N OH	HCI H ₂ N OH
33.47	DE LES CONTRACTOR DE LA	H_2N	HCI H ₂ N H

Preparative Example 34

To a solution of 3-chlorobenzaldehyde (2.0 g, 14.2 mmol) in THF (5 mL) at 0 °C was added LiN(TMS)₂ (17.0 ml, 1.0 M in THF) dropwise and the resulting solution was stirred for 20 min. EtMgBr (6.0 mL, 3.0 M in Et₂O) was added dropwise and the mixture was refluxed for 24 h. The mixture was cooled to room temperature, poured into saturated aqueous NH₄Cl (50 mL), and then extracted with CH₂Cl₂ (3 x 50 volumes). The organic layers were combined, concentrated under reduced pressure. The crude residue was stirred with 3 M HCl (25 mL) for 30 min and the aqueous layer was extracted with CH₂Cl₂ (3 x 15 mL) and the organic layers were discarded. The aqueous layer was cooled to 0 °C and treated with Solid NaOH pellets until pH = 10 was attained. The aqueous layer was extracted with CH₂Cl₂ (3 x 15 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 25 mL), dried (Na₂SO₄), and concentrated under reduced pressure to afford 1.6 g (66% yield)

10

of the crude amine as an oil (MH⁺ 170). This material was determined to be >90% pure and was used without further purification.

PREPARATIVE EXAMPLE 34.1

The aldehyde (3.5g) and conc. HCl (20ml) were combined and stirred overnight at 40°C. The reaction mixture was poured into cold water and extracted with ether, washed with satd. NaHCO₃ and brine, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give 1.76g of product (55%)

5

10

15

PREPARATIVE EXAMPLE 34.2

Chlorine was bubbled into 100ml of CH₂Cl₂ at 10⁰C. The aldehyde (3.73ml) was charged with 50ml of CHCl₃ and then cooled to 0⁰C. AlCl₃ was added portionwise, followed by the chlorine solution and let stir at room temperature overnight. The reaction was poured into 150ml of ice and 50ml of 3N HCl and stirred for 30min. Organic layer was washed with brine, dried with Na₂SO₄, and concentrated *in vacuo*. The crude product was purified via flash column chromatography (Hex/EtOAc 40/1) to yield 1.5g of pure product.

PREPARATIVE EXAMPLE 34.3

$$F_3C$$

$$Step A$$

$$F_3C$$

$$Step B$$

$$F_3C$$

$$F_3C$$

5 Step A

The ketone (3.25g) was reacted following the procedure set forth in Preparative Example 88.2, Step B to give the oxime (3.5g, 99%).

Step B

10

15

20

The product from step A (1.2g) was stirred with AcOH (3ml) and Pd/C (10%, 300mg) in EtOH (40ml) under a hydrogen atmosphere overnight. The reaction mixture was filtered through celite and the filtrate was concentrated *in vacuo*. The crude material dissolved in ether and washed with 2N NaOH, organic washed with brine, dried with Na₂SO₄, and concentrated *in vacuo* to give product (960mg, 86%).

PREPARATIVE EXAMPLE 34.4

Step A

To a suspension of NaH (1.45g) in DMF (25ml) under a nitrogen atmosphere was added *p*-bromophenol (5g) at 0°C. After stirring for 20min, BrCH₂CH(OEt)₂ (5.3ml) was added and the reaction was heated to reflux overnight. The solution was cooled and poured into ice water (80ml) and extracted with ether. The ether layer was washed with 1N NaOH and brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give 8.4g of crude product (100%).

Step B

5

10

15

To a solution of the product from Step A (8.4g) in benzene (50ml) was added polyphosphoric acid (10g). The mixture was heated at reflux for 4 hrs. The reaction was cooled to 0°C and poured into ice water (80ml) and extracted with ether. The ether layer was washed with saturated sodium bicarbonate and brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give 4.9g of crude product (85%).

Step C

To a solution of the product from Step B (2g) in ether (20ml) at -78°C was added t-BuLi dropwise. After stirring for 20min, DMF (950mg) was added dropwise and the mixture was stirred at -25°C for 3hrs and then warmed to room temperature overnight. Saturated ammonium chloride was added and the solution was extracted with ether. The ether layer was washed with brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give 980mg of crude product (67%).

Step D

To a solution of aldehyde (400g) in ether (10ml) was added LiN(TMS)₂ (1M in THF, 3.3ml) at 0°C dropwise. The solution was stirred at 0°C for 30min and EtMgBr (3M in THF, 1.83ml) was added dropwise. The reaction was refluxed overnight, cooed to 0°C, quenched with saturated ammonium chloride and extracted with ether. The ether was stirred with 3N HCl (20ml), then the aqueous layer was basified with NaOH pellets and extracted with ether. The ether layer was washed with brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give 220mg of product (46%).

25

20

PREPARATIVE EXAMPLE 34.5

Following the procedures set forth in Preparative Example 34.4 Steps A through D, but using *m*-bromophenol (8g), both amines were formed and separated by preparative plate chromatography (63-65%, MH+=175).

PREPARATIVE EXAMPLE 34.6

$$\sim$$

To a solution of 3-methyl-thiophene (5g) in ether (50ml) was added dropwise a solution of n-BuLi (1.6M in hexane, 32ml). The mixture was stirred for 1.5hr at room temperature. DMF (5.1ml) was then added and let stir overnight. The mixture was poured into saturated ammonium chloride and extracted with ether. The ether layer was washed with brine, dried with Na₂SO₄, and concentrated *in vacuo*. The crude product was purified via flash column chromatography (EtOAc/Hex 20:1) to afford 5.27g of an oil (84%).

PREPARATIVE EXAMPLE 34.7

Step A

5

10

15

20

To a solution of 4-bromo-2-furaldehyde (4g) in MeOH (75ml) was added trimethyl- orthoformate (3.8ml). A catalytic amount of *p*-toluene sulfonic acid (195mg) and the mixture was heated to reflux for 3.5hr. The reaction was cooled down and potassium carbonate was added. The mixture was filtered through a silica gel pad. The filtrate was concentrated *in vacuo*, dissolved in CH₂Cl₂ and filtered. The filtrate was again concentrated *in vacuo* to give 4.03g of product (80%).

25 <u>Step B</u>

To a solution of the product from Step A (2.02g) in THF (80ml) at -78°C was added dropwise a solution of n-BuLi (2.5M in hexanes, 4.4ml) and stirred for 1.5hr. A solution of iodomethane (1.7ml) was added and let stir for 2.5hrs at -60°C. The

cooling bath was removed and saturated ammonium chloride was added and let stir for 10min. The layers were separated and the organic layer was washed with brine, dried with Na₂SO₄, and concentrated *in vacuo* to afford 1.34g of crude product.

5 Step C

10

The product from Step B (1.43g) was dissolved in acetone (50ml) and treated with a catalytic amount of p-toluene sulfonic acid (80mg). The mixture was heated to reflux for 2hr. The reaction was cooled down and solid potassium carbonate was added. The mixture was filtered through a silica gel pad and the filtrate was concentrated *in vacuo* to give 1.246g of crude product.

PREPARATIVE EXAMPLE 34.8

15 <u>Step A</u>

20

To a stirred solution of potassium t-butoxide (2.5g) in HMPA (20ml) was added 2-nitropropane (2ml) dropwise. After 5min, a solution of methyl-5-nitro-2-furoate (3.2g) in HMPA (8ml) was added to the mixture and stirred for 16hr. Water was added and the aqueous mixture was extracted with EtOAc. The EtOAc layer was washed with water, dried with MgSO₄, filtered and concentrated *in vacuo*. The crude material was purified by flash column chromatography (Hex/EtOAc, 6:1) to yield 3.6g of product (90%).

Step B

5

10

20

25

30

To a solution of the product from Step A (3.6g) in toluene (16ml) was added tributyltin hydride (5.4ml) followed by AIBN (555mg). The mixture was heated to 85°C for 3.5hr. After cooling, the mixture was separated by flash column chromatography (Hex/EtOAc, 7:1) to afford 2.06g of product (73%).

Step C

To a solution of product from Step B (2.05g) in THF (60ml) at 0°C was added a solution of LAH (1M in ether, 12.8ml). The reaction was stirred at room temperature for 30min. Water and 1M NaOH was added until a precipitate formed, diluted with EtOAc, stirred for 30min and then filtered through a celite pad. The organic filtrate was concentrated *in vacuo* to give 1.56g of product (93%).

15 Step D

To a solution of product from Step C (2.15g) in CH₂Cl₂ (100ml) was added Dess-Martin oxidant (7.26g) in CH₂Cl₂ (45ml) and stirred for 30min. The mixture was diluted with ether (200ml). The organic layer was washed with 1N NaOH, water and brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give oil and solid. The material was extracted with ether and filtered. Some solid crystallized out from the filtrate, filtered again, and the filtrate was concentrated *in vacuo* to give 2.19g of product.

PREPARATIVE EXAMPLE 34.9

Step A

To a solution of carboxylic acid (5g) in CH₂Cl₂ (400ml) at 0⁰C was added N(OCH₃)CH₃.HCl (11.5g), DEC (15.1g), HOBt (5.3g) and NMM (43ml) and stirred for 14hr. The mixture was diluted with CH₂Cl₂ (100ml) and the organic layer was washed with 10% HCl, saturated sodium bicarbonate and brine, dried with Na₂SO₄, and concentrated *in vacuo* to afford 5.74g of crude product (85%).

Step B

To a solution of iodoethane (0.56ml) in ether (5ml) at -78°C was added a solution of t-BuLi (1.7M in pentane, 8.3ml) dropwise. The mixture was warmed to room temperature for 1hr and transferred to a 100ml round bottom charged with the product from Step A (1g) in THF (12ml) at -78°C. The mixture was stirred at -78°C for 1hr and at 0°C for an additional 2hr. 1M HCl was added dropwise followed by CH₂Cl₂. The layers were separated and the organic layer was washed with brine, dried with Na₂SO₄, and concentrated *in vacuo* to afford 620mg of product (76%).

10

15

20

25

5

Step C

To a solution of the product from Step B (620mg) in THF/MeOH (10:1) at 0°C was added NaBH₄ (250mg) in one portion. The mixture was stirred overnight at 0°C, concentrated *in vacuo* and the crude material was dissolved in CH₂Cl₂ and washed with 1N NaOH and brine, dried with Na₂SO₄, and concentrated *in vacuo* to afford 510mg of product.

Step D

The above material was reacted in the procedures set forth in Preparative Example 75.75 Steps B and C to yield 170mg of amine product (28%).

PREPARATIVE EXAMPLE 34.10

The above amine was made analogous to the procedures set forth in Patent WO96/22997 p.56 (the disclosure of which is incorporated herein by reference thereto), but using ethylglycine instead of benzylglycine in the DCC coupling.

PREPARATIVE EXAMPLE 34.11

Step A

To the nitro compound (3.14g) and cyclohexylmethanol (1.14g) in THF (50ml) was added PPH₃ (4.72g) and cooled to 0^oC. Diisopropylazadicarboxylate (3.15ml) was added dropwise and let stir overnight. The reaction was concentrated *in vacuo* and purified via flash column chromatography (Hex/EtOAc, 30:1) to give product (3.3g), which was carried on directly to the next step.

10 <u>Step B</u>

5

To the product from step A (3.3g) in EtOH (50ml) was added 10% Pd/C (1.7g) under a hydrogen atmosphere at 55psi and let stir overnight. The reaction was filtered through celite and concentrated *in vacuo* to give 3.2g of product.

15 PREPARATIVE EXAMPLE 34.12

$$HO \longrightarrow CF_3 \longrightarrow HO \longrightarrow CF_3 \longrightarrow H \longrightarrow CF_3$$

Step A

20

25

A solution of acid (2g) in ether (20ml) was added dropwise to a suspension of LiAlH₄ (350mg) in ether (15ml) at 0°C. The solution was refluxed for 3hr and stirred at room temperature ovenright. 5% KOH was added and reaction was filtered, extracted with ether, dried with MgSO₄, filtered and concentrated *in vacuo* to give the product (1.46g, 79%, MH+=166).

Step B

To a solution of alcohol from above (1.46g) in CH₂Cl₂ at room temperature was added Dess-Martin reagent (5.6g) portionwise and one drop of water and let stir over

the weekend at room temperature. 10% Na₂S₂O₃ was added and stirred for 20min, extracted with CH₂Cl₂, washed with saturated sodium bicarbonate, dried with Na₂SO₄, and concentrated *in vacuo* to afford 1.1g of product (76%).

PREPARATIVE EXAMPLE 34.13

The above compound was prepared according to the procedure set forth in EP 0 555 153 A1 (the disclosure of which is incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 34.14

The aldehyde (500mg) from above was reacted following the procedure set forth in the Preparative Example 13.4, Step A to yield 372mg of product (76%).

PREPARATIVE EXAMPLE 34.15-34.16

Following the procedures set forth in Preparative Example 34.8 but using the nitroalkanes indicated in the table below, the aldehydes were prepared.

PREP.	NITROALKANE	ALDEHYDE	YIELD
Ex.			(%)
34.15	√-NO ₂	H	17
34.16	—NO₂	н	21

10

15

PREPARATIVE EXAMPLE 34.17

Step A

5

10

To a stirred suspension of 5-bromo-2-furoic acid (15.0 g, 78.54 mmol) in 225 mL of CH₂Cl₂ at room temperature was added oxalyl chloride followed by a catalytic amount of *N,N'*-dimethylforamide. After 1 h, ethanol (20 mL) was added followed by triethylamine (22 mL). Reaction was continued for 15 h. The mixture was concentrated under reduced pressure to a residue, which was extracted with excess volume of hexanes, and hexanes-CH₂Cl₂ (3:1, v/v). The extracts were filtered, the filtrated was concentrated to a yellow oil, dried on high vacuum, yielding 17.2 g (93%) of the desired ester.

Step B

The ester product obtained from Step A above (17.2 g, 73.18 mmol) was converted to 2-ethyl-4-tertbutyl-5-bromo-furoate (7.9 g, 37%) using the literature procedure: *J. Am. Chem.Soc.*, **1939**, *61*, 473-478 (the disclosure of which is incorporated herein by reference thereto).

20 <u>Step C</u>

The ester product obtained from Step B above (7.9 g, 27.13 mol) was reduced to the alcohol (6.32 g) using the procedure set forth in Preparative Example 34.8, Step C.

Step D

The product obtained from Step C above (6.32 g) was dissolved in 140 mL of THF and cooled in a –78°C bath. A 2.5 M solution of n-butyllithium in hexanes (22 mL, 55.0 mmol) was added dropwise along the side wall of the flask. After 15 min, H₂O (~70 mL) was added. Cooling bath was removed, the mixture was stirred for an additional 1h. Brine (50 mL) and CH₂Cl₂ (300 mL) were added, the two layers were separated, the aqueous layer was extracted with CH₂Cl₂ (100 mL), and the combined organic layers ere dried by Na₂SO₄. Evaporation of solvents afforded 5.33 g (crude) of the debrominated product as a reddish brown oil.

10

5

Step E

The alcohol product obtained from Step D above (5.33g) was oxidized to the corresponding aldehyde (3.06 g, 74% over three steps) using the procedure set forth in Preparative Example 34.8, Step D.

15

20

25

PREPARATIVE EXAMPLE 34.18

Step A

To a stirred solution of cyclopropyl bromide (4.0 mL, 50 mmol) in 120 mL of ether at –78°C was added dropwise a 1.7M solution of t-butyllithium in pentane (44.5 mL, 75.7 mmol). After 10 min, cooling bath was removed, stirring was continued for 1.5 h. The mixture was cooled again in a –78°C bath, and 3-furaldehyde (3.5 mL, 41.9 mmol) was added. Reaction was continued for 1 h, and quenched with a saturated NH4Cl aqueous solution. The aqueous mixture was extracted with CH₂Cl₂ (100 mL x 3). The organic extracts were washed with brine, dried by Na₂SO₄, filtered, and concentrated in vacuo to give 5.3 g (91%) of the alcohol product as a yellow oil.

Step B

5

10

15

20

Chloro trimethylsilane (27.2 mL, 214.2 mmol) was added dropwise to a vigorously stirred suspension of sodium iodide (32 g, 213.5 mmol) in 100 mL of acetonitrile. After 5 min, a solution of the alcohol obtained from Step A above (4.93 g, 35.68 mmol) in 100 mL of acetonitrile was added dropwise. Stirring was continued for 5 min. H_2O (100 mL) was added, the layers were separated, and the aqueous layer was extracted with ether (100 mL x 2). The organic layers were combined, washed with a 10 % $Na_2S_2O_3$ aqueous solution and brine, and dried over Na_2SO_4 . Evaporation of solvents gave a dark brown oil, which was filtered through a 5-in silica gel column, eluting with CH_2Cl_2 -hexanes (1:3.5, v/v). Removal of solvents afforded 4.22 g (47%) of the iodo product as a light yellow oil.

Step C

The iodo-product obtained from Step B above (2.2 g, 8.8 mmol) was dissolved in 60 mL of ether, and stirred in a –78°C bath. A 1.7 M solution of t-butyllithium in pentane (10.4 mL, 17.7 mmol) was added dropwise. After 20 min, cooling bath was removed. Reaction was continued for 2.5 h, and quenched with H₂O (20 mL). The aqueous mixture was stirred overnight and separated. The aqueous layer was extracted with ether (30 mL). The combined organic layers were washed with brine, dried by Na₂SO₄, and filtered through a Celite pad. Removal of solvent gave 1.10 g (100%) of 3-butylfuran as a reddish-yellow oil.

Step D

3-Butylfuran (1.1 g, 8.8 mmol), obtained from Step C above, was dissolved in 60 mL of ether, and stirred in a –78°C bath. A 1.7 M solution of t-butyllithium in pentane (6.0 mL, 10.2 mmol) was added dropwise along the side wall of the flask. The mixture was stirred for 3 h from –78°C to 0°C, and continued for 1 h at room temperature. A solution of *N*,*N*'-dimethylforamide (1.1 mL, 14.23 mmol) was added. Reaction was continued overnight, and quenched with a saturated NH₄Cl aqueous solution. The two layers were separated, the aqueous layer was extracted with CH₂Cl₂ (30 mL x 2). The combined organic layers were washed with brine, dried with Na₂SO₄, and concentrated to an oil, which was purified by preparative TLC (CH₂Cl₂-

hexanes = 1 :1.5, v/v) to give 0.48 g (36%) of the aldehyde (contaminated by some 3-butyl-2-furaldehyde).

PREPARATIVE EXAMPLE 34.19

Step A

5

10

15

3-Ethylfuran was prepared from 3-hydroxymethylfuran according to literature procedure: *J. Org. Chem.*, **1983**, *48*, 1106-1107 (the disclosure of which is incorporated herein by reference thereto).

Step B

3-Ethylfuran obtained from Step A above was converted to 4-ethyl-2-furaldehyde using the procedure set forth in Preparative Example 34.32, Step D.

PREPARATIVE EXAMPLES 35-51.20

Following the procedure set forth in Preparative Example 34 but using the commercially available aldehydes and Grignard reagents listed in the Table below, the amine products below were obtained.

Prep Ex.	Aldehyde	Grignard Reagent	Amine	1.Yield 2. MH ⁺
35	F	EtMgBr	H ₂ N F	1. 65% 2. 154
36	H	EtMgBr	H ₂ N	1. 75% 2. 180

37	H CI	EtMgBr	H ₂ N CI	1. 78% 2. 170
38	CF ₃	EtMgBr	H ₂ N CF ₃	1. 34% 2. 204
39	н	EtMgBr	H ₂ N	1. 68% 2. 150
40	O OCF ₃	EtMgBr	H ₂ N OCF ₃	1. 40% 2. 220
41	H F	EtMgBr	H ₂ N F	1. 73% 2. 154
42	H OCF3	EtMgBr	H ₂ N OCF ₃	1. 52% 2. 220
43	H C	EtMgBr	H ₂ N O	1. 55% 2. 180
44	HCF3	EtMgBr	H ₂ N CF ₃	1. 20% 2. 204

	P	EtMgBr		
	H		H ₂ N	1. 80%
45	осн ₃		OCH ₃	2. 166
	0	EtMgBr		
46	н	Euriger		1. 35%
40	OCF ₃		H ₂ N	2. 220
			OCF ₃	2. 220
		: Daller Da		
4.7		<i>i</i> -PrMgBr		4 000/
47	H'		H ₂ N	1. 20%
				2. 150
48	OMe	EtMgBr		
			H ₂ N OMe	1. 77%
				2.
				$[M-NH_2]^{\dagger} =$
				149
	0	EtMgBr		
49	H		H ₂ N F	1. 77%
	<u></u>			2. 172
	F		F	
	0			
50	H	EtMgBr	H ₂ N	1. 78%
			1121	2. [M-
			T	NH ₂] ⁺ =
				147
	Q		1/_	1. 10%
51	H	EtLi		2. 116
	/ \		H ₂ N	
<u> </u>			<u> </u>	<u> </u>

51.2	H	EtMgBr	H ₂ N O	1. 37% 2. 161
51.3	H OFF	EtMgBr	H ₂ N OF	1. 63% 2. 216
51.4	H O	EtMgBr	H ₂ N	1. 71% 2. 228
51.5	H	EtMgBr	H ₂ N F	1. 89% 2. 168
51.6	H	EtMgBr	H ₂ N O	1. 20% 2. 228
51.8	H CF ₃	EtMgBr	H_2N CF_3	1. 36% 2. 222
51.10	H	MgBr	H ₂ N O	1. 95% 2. 152.1
51.11	Н	EtMgBr	H ₂ N O OH	1. 61% 2. 138.1 MH ⁺ -H ₂ O

51.12	H N-	EtMgBr	H ₂ N O N-	1. 70% 2. 184.1
51.18	O H	EtMgBr	H ₂ N	1. 42% 2. 147 [M-NH ₂] ⁺
51.19		EtMgBr	H ₂ N CI	1. 67% 2. 204
51.20	H CI	EtMgBr	H ₂ N CI	1. 33% 2. 188

PREPARATIVE EXAMPLES 51.25 - 51.31

Following the procedure set forth in Example 34 but using the commercially available aldehydes and Grignard reagents listed in the Table below, the amine products were obtained.

5

Prep Ex.	Aldehyde	Grignard Reagent	Amine	Yield (%)
51.25	H	EtMgBr	H ₂ N	20
51.26	H C	MgBr	H ₂ N O	77

51.27	(34.2) O O CI	EtMgBr	H ₂ N CI	51
51.28	(78.1) H Q N	BrMg	H ₂ N O N	56
51.29	(78.1) O H	MgBr	H ₂ N ON	54
51.30	(34.12) O H O F F	EtMgBr	H ₂ N F	80
51.31	H	————MgBr	H ₂ N	10

$$F_3C$$
 S
 S
 S
 F_3C
 S
 S

Step A

5

10

15

20

A mixture of 2-(trifluoroacetyl)thiophene (2 mL, 15.6 mmol), hydroxylamine hydrochloride (2.2 g, 2 eq), diisopropylethylamine (5.5 mL, 2 eq) and MeOH (50 mL) was stirred at reflux for 48-72 hrs, then concentrated *in vacuo*. The residue was diluted with EtOAc, washed with 10% KH₂PO₄ and dried over Na₂SO₄ (anhydrous). Filtration and concentration afforded the desired oxime (2.9 g, 96%) which was used directly in Step B without further purification.

Step B

To a mixture of the product from Step A above in TFA (20 mL) was added Zn powder (3 g, 3 eq) portionwise over 30 min and stirred at room temperature overnight. The solid was filtered and the mixture reduced *in vacuo*. Aqueous NaOH (2 *M*) was added and the mixture was extracted several times with CH₂Cl₂. The organic phase was dried over anhydrous Na₂SO₄, filtered and concentrated to afford the desired product (1.4 g, 50%).

PREPARATIVE EXAMPLES 53-61

Following the procedure set forth in Preparative Example 52 but using the commercially available ketones listed in the Table below, the following amines were obtained.

Prep Example	Ketone	Amine	1.Yield (%) 2. MH ⁺
53	S	H ₂ N S	1. 11 2. 128
54	S	H_2N	1. 33 2. 142
55	S	H_2N	1. 49 2. 156
56	S	H_2N	1. 5 2. 154
57		H_2N	1. 47 2. 174
58	S	H_2N	1. 71 2. 190
59	SN	H_2N S N	1. 78 2. 191

60	S	H_2N S	1. 80 2. 190
61	S	H_2N	1. 9 2. 156

$$H_2N$$
 S
 H_2N
 OH
 S
 H_2N
 S

To a cooled (0-5°C) suspension of L- α -(2-thienyl)glycine (0.5 g) and LiBH₄ (2M in THF, 3.8 mL) in anhydrous THF (10 mL) was slowly added a THF (5 mL) solution of iodine (0.8 g). After stirring at room temperature for 15 min, the mixture was stirred at relux overnight. After cooling to room temperature, MeOH was added dropwise until gas evolution ceased and after 30 min, the mixture was evaporated. The oily residue was stirred in 20 mL KOH for 4 hrs, diluted with brine and extracted with EtOAc.

5

10

15

20

The organic phase was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to afford a crude mixture. Purification by flash column chromatography (50% EtOAc/ CH₂Cl₂, silica) afforded the product (0.3 g, 63%, MH⁺ = 144).

PREPARATIVE EXAMPLE 63

$$NC S \longrightarrow H_2N S$$

CeCl₃-7H₂O was dried at 140-150°C for 22 hr. To this solid was added THF (80 mL, anhydrous) and after stirring for 2 hr, the suspension was cooled to –78°C and to it was added methyl lithium over 30 min. After stirring for an additional 30 min 2-thiophenecarbonitrile dissolved in anhydrous THF (4.5 mL) was added and the

resulting mixture stirred for an additional 4.5 hr at –78°C. Concentrated aqueous NH₃ (25 mL) was added and the mixture was warmed to room temperature and filtered through celite. The filtrate was extracted with dichloromethane, dried over anhydrous Na₂SO₄, filtered and concentrated *in vacuo* to afford a crude mixture. Purification by flash column chromatography (5% MeOH, CH₂Cl₂, silica) afforded the desired product (1.2 g, 62%).

PREPARATIVE EXAMPLE 64

10 Step A

15

20

5

To a solution of (D)-valinol (4.16 g, 40.3 mmol) in CH₂Cl₂ (60 mL) at 0 °C was added MgSO₄ (20 g) followed by dropwise addition of 3-fluorobenzaldehyde (5.0 g, 40.3 mmol). The heterogenous solution was stirred at 0°C for 2h and was allowed to warm to room temperature and stir overnight (14h). The mixture was filtered and the drying agent was washed with CH₂Cl₂ (2 x 10 mL). The filtrate was concentrated under reduced pressure to afford 8.4 g (100%) of an oil which was taken onto the next step without further purification.

Step B

To a solution of the imine (8.4 g, 40.2 mmol) from Step A in CH₂Cl₂ (60 mL) at room temperature was added Et₃N (6.2 mL, 44.5 mmol) followed by dropwise addition of TMSCl (5.7 mL, 44.5 mmol). The mixture was stirred for 6h at room temperature whereupon the ppt that had formed was filtered off and washed with CH₂Cl₂ (2 x 10 mL). The combined filtrate was concentrated under reduced pressure and was taken

up in Et₂O/hexane (1:1/150 mL). The precipitate was filtered off and the filtrate was concentrated under reduced pressure to afford 10.1 g (89%) of the protected imine as an oil. This material was taken onto the next step without further purification.

Step C

5

10

15

20

25

30

To a solution of Etl (4.0 g, 25.6 mmol) in Et₂O (40 mL) at -78 °C was added t-BuLi (30.1 mL, 51.2 mmol, 1.7 M in pentane) and the mixture was stirred for 10 min. The mixture was warmed to room temperature, stirred for 1 h, and was recooled to -40 °C. A solution of the imine (6.0 g, 21.4 mmol) from Step B in Et₂O (30 mL) was added dropwise via addition funnel to afford a bright orange mixture. The reaction mixture was stirred for 1.5 h at -40 °C then 3M HCl (50 mL) was added and the mixture was allowed to warm to room temperature. Water (50 mL) was added and the layers were separated. The aqueous layer was extracted with Et₂O (2 x 30 mL) and the organic layers were combined and discarded. The aqueous layer was cooled to 0 °C and carefully treated with solid NaOH pellets until pH = 12 was attained. The aqueous layer was extracted with Et₂O (3 x 30 mL) and the combined layers were washed with brine (1 x 30 mL). The organic layer was dried (Na₂SO₄), filtered, and concentrated under reduced pressure to afford 4.8 g (94% yield) of the amine as an oil. This material was taken on crude to the next step without further purification.

Step D

To a solution of amine (4.5 g, 18.8 mmol) from Step C in MeOH (80 mL) at room temperature was added MeNH₂ (25 mL, 40% in water) followed by addition of a solution of H₅IO₆ (14.0 g, 61.4 mmol) in H₂O (25 mL). The heterogenous mixture was stirred for 1.5 h (until the reaction was complete by TLC) and the precipitate was filtered off. The resulting filtrate was diluted with water (50 mL) and the mixture was extracted with Et₂O (4 x 60 mL). The combined organic layers were concentrated to a volume of ~30 mL whereupon 3M HCl (75 mL) was added. The mixture was stirred overnight (12h at room temperature) after which the mixture was concentrated to remove the volatiles. The aqueous layer was extracted with Et₂O (3 x 40 mL) and the organic layers were discarded. The aqueous layer was cooled to 0 °C and was carefully treated with solid NaOH pellets until pH ~12 was reached. The aqueous layer was extracted with Et₂O (3 x 60 mL) and the combined organic layers were dried (MgSO₄). The organic layer was concentrated under reduced pressure to afford

2.8 g (97% yield) of the desired amine as an oil [MH⁺ 154]. This compound was proven to be >85% pure by ¹H NMR and was used crude in the subsequent coupling step.

5

PREPARATIVE EXAMPLES 65-75.10J

Following the procedure set forth in Preparative Example 64 but using the prepared or commercially available aldehydes, amino alcohols, and organolithium reagents in the Table below, the optically pure amine products in the Table below were obtained.

10

Prep Ex.	Aldehyde	Amino Alcohol	Organo lithium	Product	1.Yield (%) 2. MH ⁺
65	H	H ₂ N OH	EtLi	H ₂ N F	1. 62 2. 154
66	н	H ₂ N OH	EtLi	H ₂ N F	1. 70 2. 154
67	H	H ₂ N OH	<u></u> Li	H ₂ N F	1. 54 2. 166
68	H	H ₂ N OH	 Li	H ₂ N F	1. 67 2. 166
69	H	H ₂ N OH	EtLi	H ₂ N F	1. 67 2. 154

70	H	H ₂ N OH	EtLi	H ₂ N S	1. 42 2. 142
71	H	H ₂ N OH	EtLi	H ₂ N S	1. 36 2. 142
72	H	H ₂ N OH	Li	H ₂ N	1. 62 2. 148
73	H	H ₂ N OH	t-BuLi	H ₂ N S	1. 27 2. 256
74	H	H ₂ N OH	t-BuLi	H ₂ N	1. 15 2. 164
75	H	H ₂ N OH	FF	H ₂ N	1. 7 2. 204
75.1	H	H ₂ N OH	EtLi	H ₂ N O	1. 65 2. 123 [M-NH ₂] ⁺
75.2	H	H ₂ N OH	EtLi	H ₂ N O	1. 62 2. 123 [M-NH ₂] ⁺
75.3	H S	H ₂ N OH	EtLi	H ₂ N S	1. 93 2. 139 [M-NH ₂] ⁺

75.4	H	H ₂ N OH	tBuLi	H ₂ N S	1. 50 2. 167 [M-NH ₂] ⁺
75.5	(34.6) O H	H ₂ N OH	tBuLi	H ₂ N S	1. 48 2. 167 [M-NH ₂] ⁺
75.6	(34.6) O H	H ₂ N OH	EtLi	H ₂ N S	1. 97 2. 139 [M- NH ₂] [†]
75.7	(34.6) O H S	H ₂ N OH	iPrLi	H_2N	1. 87 2. 153 [M- NH ₂] [†]
75.8	(34.6) O H	H ₂ N OH	Li	H ₂ N S	1. 94 2. 151 [M-NH ₂] [†]
75.9	(34.8) O H	H ₂ N OH	EtLi	H ₂ N O	1. 75 2. 151 [M-NH ₂] [†]
75.10	(34.8) O H	H ₂ N OH	tBuLi	H ₂ N O	1. 30 2. 179 [M-NH ₂] [†]

75.10A	(34.7) O H	H ₂ N OH	Li	H ₂ N O	1. 61 2. 135 [M-NH ₂] ⁺
75.10B	(34.19) O H	H ₂ N OH	EtLi	H ₂ N	1. 24 2. 154
75.10C	(34.18) O H	H ₂ N OH	EtLi	H ₂ N O	1. 32 2. 165 [M-NH ₂] [†]
75.10D	(34.8) O H	H ₂ N OH	MeLi	H ₂ N O	1. 47 2. 137 [M-NH ₂] ⁺
75.10E	(34.8) O H	H ₂ N OH	iPrLi	H ₂ N O	1. 30 2. 165 [M- NH ₂] ⁺
75.10F	(34.8) H	H ₂ N OH	Li	H ₂ N O	1. 67 2. 163.0 [M-NH ₂] ⁺
75.10G	(34.17) O H	H ₂ N OH	EtLi	H ₂ N O	1. 24 2. 165 [M-NH ₂] ⁺

75.10H	(34.15) O H	H ₂ N OH	EtLi	H ₂ N O	1. 70 2. 194
75.10J	(34.16)	H ₂ N OH	EtLi	H ₂ N	1. 54 2. 208

PREPARATIVE EXAMPLES 75.11-75.59

Following the procedure set forth in Preparative Example 64 but using the prepared or commercially available aldehydes, amino alcohols, and organolithium reagents in the Table below and carrying the amine on crude, the optically pure amine products in the Table below were obtained.

5

Prep Ex.	Aldehyde	Amino Alcohol	Organo lithium	Product	Yield (%)
75.11	H	H ₂ N OH	Li	H ₂ N O	52
75.12	H	H ₂ N OH	Li	H_2N	50
75.13	H O	H ₂ N OH	iPrLi	H ₂ N O	57

75.14	H	H ₂ N OH	iPrLi	H ₂ N O	54
75.15	H	H ₂ N OH	iPrLi	H ₂ N S	58
75.16	H S	H ₂ N OH	Li	H ₂ N S	61
75.17	H	H ₂ N OH	EtLi	H ₂ N	72
75.18	H S	H ₂ N OH	Li	H ₂ N S	68
75.19	H	H ₂ N OH	iPrLi	H ₂ N S	77
75.20	H S	H ₂ N OH	t-BuLi	H ₂ N S	15
75.21	H	H ₂ N OH	MeLi	H ₂ N S	50
75.22	H C	H ₂ N OH	EtLi	H ₂ N	23

75.24	H	H ₂ N OH	EtLi	H ₂ N	20
75.27	H	H ₂ N OH	EtLi	H_2N	65
75.28	H	H ₂ N OH	iPrLi	H ₂ N	61
75.29	H	H ₂ N OH	EtLi	H ₂ N F	90
75.30	H_2N	H ₂ N OH	iPrLi	H ₂ N O	62
75.31	H	H ₂ N OH	iPrLi	H_2N	43
75.32	H	H ₂ N OH	Li	H_2N	50
75.33	H	H ₂ N OH	Li	H ₂ N F	50
75.34	P F	H ₂ N OH	tBuLi	H ₂ N F	51
75.35	н	H ₂ N OH	MeLi	H ₂ N O	51

75.36	H	H ₂ N OH	tBuLi	H_2N S N	57
75.37	H	H ₂ N OH	tBuLi	H ₂ N O	60
75.38	H	H ₂ N OH	EtLi	H ₂ N O	73
75.39	H O	H ₂ N OH	MeLi	H ₂ N O	48
75.41	H	H ₂ N OH	Li	H ₂ N O	52
75.42	H	H ₂ N OH	EtLi	H ₂ N S	40
75.43	H	H ₂ N OH	tBuLi	H_2N	20
75.44	H	H ₂ N OH	t-BuLi	H ₂ N O	79
75.45	H	H ₂ N OH	iPrLi	H ₂ N O	55

75.46	(75.57) O H O N	H ₂ N OH	tBuLi	H ₂ N O N	39
75.47	(75.57) O H	H ₂ N OH	iPrLi	H ₂ N ON	55
75.48	(75.57) O H Q N	H ₂ N OH	Li	H ₂ N QN	34
75.49	(34.7) O H	H ₂ N OH	EtLi	H ₂ N	61
75.50	(34.7) O H	H ₂ N OH	tBuLi	H_2N	25
75.51	(34.2) O H O CI	H ₂ N OH	iPrLi	H ₂ N O	33
75.52	(34.2) O H O CI	H ₂ N OH	tBuLi	H ₂ N O	30

75.53	(34.2) O H O CI	H ₂ N OH	EtLi	H ₂ N CI	39
75.54	(34.2) O H O CI	H ₂ N OH	Li	H_2N O CI	38
75.55	T O	H ₂ N OH	EtLi	H ₂ N O	64
75.56	H	H ₂ N OH	EtLi	H ₂ N O	46
75.57	(75.57) O H O N	H ₂ N OH	EtLi	H ₂ N O N	62
75.58	O S N	H ₂ N OH	iPrLi	H ₂ N S	24
75.59	(34.1) H O CI	H ₂ N OH	EtLi	H ₂ N CI	70
75.60	H	H ₂ N OH	t-BuLi	H ₂ N	60

75.61	H	H ₂ N OH	iPrLi	H ₂ N O	60
75.62	H	H ₂ N OH	t-BuLi	H ₂ N O	57
75.63	H	H ₂ N OH	EtLi	H ₂ N O	94
75.64	H	H ₂ N OH	t-BuLi	H_2N O	46
75.65	H CI	H ₂ N OH	t-BuLi	H ₂ N CI	60
75.66	H S	H ₂ N OH	t-BuLi	H ₂ N S	15
75.67	H	H ₂ N OH	t-BuLi	H_2N	60

Step A

To a solution of aldehyde (2.5g) in ether (50ml) at 0°C was added EtMgBr (4.56ml) dropwise. The heterogenous mixture was stirred for 2hr at 0°C and then poured into a beaker of saturated ammonium chloride (25ml), ice and CH₂Cl₂ (30ml). After the biphasic mixture stirred for 10min, the organic layer was separated, washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo* to afford the product (2.41g, 95%)

Step B

5

10

15

20

To a solution of alcohol from Step A above (1g) in toluene at room temperature was added DPPA. The mixture was cooled to 0°C and DBU was added and let stir for 12hr at room temperature. The layers were separated and the organic layer was washed with water, 1N HCl and dried over Na₂SO₄, filtered, and concentrated *in vacuo*. Purified by preparative plate chromatography (hexane/EtOAc 20/1) to give the product (840mg, 75%).

Step C

To a solution of azide (730mg) from Step B above in THF (7ml) was added PPh₃ (1g). The heterogenous solution was stirred for 12hr, whereupon water (1.5ml) was added. The mixture was refluxed overnight, cooled to room temperature and concentrated *in vacuo*. Ether and 1N HCl were added to the residue. The aqueous layer was cooled to 0°C, basified with NaOH pellets and extracted with ether. The ether layer was dried over MgSO₄, filtered, and concentrated *in vacuo* to afford the product (405mg, 62%).

25

30

Step D

To a solution of azide in THF at -10°C was added LiAIH₄ portionwise. The heterogenous solution was stirred at room temperature for 1hr and then refluxed for 4hr. The solution was cooled to 0°C and water, 2M NaOH and ether were added to the reaction. The mixture was filtered through a celite pad. The filtrate was treated with 3N HCI. The aqueous layer was cooled to 0°C, basified with NaOH pellots and extracted with ether. The ether layer was dried over MgSO₄, filtered, and concentrated *in vacuo* to afford the product.

PREPARATIVE EXAMPLE 75.76-75.90

Following a similar procedure set forth in Preparative Example 75.75, and using the reduction procedure indicated, the following amines were obtained.

Prep	Aldehyde	Reducing	Product	
Ex.		Step		% Yield
75.76	н	D	H_2N	43
75.77	H	С	H_2N	36
75.78	H S CI	D	H ₂ N S CI	32
75.79	H	С	H_2N	42
75.80	H_2N	D	H_2N	56
75.81	H	D	H_2N	35
75.82	H Br	С	H_2N O Br	13
75.83	H CI	С	H ₂ N CI	42

75.84	H	С		39
	F F		H ₂ N O	
			F-/F	
75.85	H CI	С	H_2N CI	26
75.86	H	С	H_2N F F	25
75.87	T Z=\ \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	С	H_2N S N	14
75.88	(34.14) O H	С	H_2N	49
75.89	(34.13) O F H O H	С	H ₂ N F H	34
75.90	H Br	C	H ₂ N Br	44
75.92	H	С	H ₂ N	74
75.93	H	С	H ₂ N	81

Preparative Example 75.200

$$H_2N$$
 O
 O

If one were to follow a similar procedure as that in Preparative Example 64, but using the aldehyde from Preparative Example 1004A and cyclopentyllithium instead of ethyllithium, the title aldehyde could be prepared.

5

Preparative Example 75.201

$$H_2N$$
 $\overline{\overline{z}}$
 O

If one were to follow a similar procedure as in Preparative Example 75.200, but using 5-methylfuranaldehyde instead of the aldehyde from Preparative Example 1004A, the title aldehyde could be prepared.

PREPARATIVE EXAMPLE 76

$$H_2N$$

The desired compound was prepared according to methods previously described in *J. Med. Chem.* 1996, *39*, 3319-3323 (the disclosure of which is incorporated herein by reference thereto).

Step A

5

10

15

20

To a solution of amine from Preparative Example 75.90 (2.22g) in CH₂Cl₂ (50ml) at 0°C was added TEA (3.03ml) followed by BOC₂O (2.85g). The heterogenous mixture was allowed to stir at room temperature overnight. 10% Citric acid was added to the reaction and the layers were separated. The organic layer was washed with saturated sodium bicarbonate, brine and dried with Na₂SO₄, filtered, and concentrated *in vacuo*. The crude material was purified by flash column chromatography (Hex/EtOAc 10:1) to afford 2.7g of an oil (81%).

Step B

Following the procedure from Preparative Example 13.4, Step A, but using the product from Step A above (450mg) and 3-thiophene boronic acid (284mg), the product was prepared (325mg, 71%).

Step C

To the product from Step B (325g) was added 4M HCl in dioxane (1.31ml) and let stir for 1hr. The reaction was concentrated *in vacuo* and taken up in CH₂Cl₂ and concentrated *in vacuo* again. This procedure was repeated 5 times to afford a semisolid (89%).

PREPARATIVE EXAMPLE 76.2-76.3

Following the procedures set forth in Preparative Example 76.1, but using the commercially available boronic acids, the indicated amines were prepared.

Prep	Boronic Acid	Product	Yield (%)
Ex.			
76.2	N B(OH) ₂	CIH.H ₂ N	70
76.3	(HO) ₂ B N	CIH.H ₂ N	35

PREPARATIVE EXAMPLE 76.10

Step A

5

10

15

The product from Preparative Example 75.75, Step A (2.5g) was reacted via the Preparative Example 13.11, Step B to give the ketone (1.93g, 78%).

Step B

To a solution of ketone from Step A above (500mg) in THF (5ml) at 0°C was added S-2-methyl-CBS-oxazaborolidine (0.98ml) dropwise followed by BH_{3.}Me₂S (1.48ml). The mixture was stirred at 0°C for 2hr and was allowed to warm to room temperature and stir overnight. The mixture was cooled to 0°C and treated with

MeOH (10ml). After stirring for 20min, the reaction was concentrated *in vacuo*. The residue was dissolved in CH₂Cl₂ and washed with 1M HCl, saturated sodium bicarbonate, water and brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The crude material was purified by preparative plate chromatography (Hex/EtOAc 4:1) to afford 650mg of an oil (89%).

Step C

The chiral alcohol from Step B above was reacted via the Preparative Example 75.75 Step B to give the azide.

10

5

Step D

The azide from Step C above was reacted via the Preparative Example 75.75 Step C to give the amine product.

15

PREPARATIVE EXAMPLE 76.11

$$H_2N$$
 B

The desired compound was prepared as in Preparative Example 76.10, but using the R-2-methyloxazaborolidine in step B.

20

PREPARATIVE EXAMPLE 77

The desired compound was prepared according to methods previously described in *J. Med. Chem.* 1996, *39*, 3319-3323 (the disclosure of which is incorporated herein by reference thereto).

25

PREPARATIVE EXAMPLE 78

The desired compound was prepared according to methods previously described in *Chem. Pharm. Bull.* 1991, *39*, 181-183 (the disclosure of which is incorporated herein by reference thereto).

5

PREPARATIVE EXAMPLE 78.1

$$H_2N$$

The desired compound was prepared according to methods previously described in J. Organometallic Chem. 1998, 567, 31-37 (the disclosure of which is incorporated herein by reference thereto).

10

15

20

PREPARATIVE EXAMPLE 79

The desired compound was prepared according to methods previously described in *Chem. Pharm. Bull.* 1991, *39*, 181-183 (the disclosure of which is incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 80

The desired compound was prepared according to methods previously described in (a) *Synthesis* **1987**, 998-1001, (b) *Synthesis* **1996**, 641-646, and (c) *J. Med. Chem.* **1991**, *34*, 2176-2186 (the disclosures of each reference being incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 81

25

The desired compound was prepared according to methods previously described in (a) Synthesis 1987, 998-1001, (b) Synthesis 1996, 641-646 and

(c) *J. Med. Chem.* **1991**, *34*, 2176-2186 (the disclosures of each reference being incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 82

The desired compound was prepared according to methods previously described in *J. Med. Chem.* **1988**, *31*, 2176-2186 (the disclosure of which is incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 83

To a solution of carboxylic acid (1.5 g, 7.89 mmol) in H₂O/acetone (1:10/12 mL total) at 0°C was added Et₃N (1.43 mL, 10.3 mmol) followed by addition of ethyl chloroformate (0.83 mL, 8.68 mmol). The resulting mixture was stirred for 30 min after which a solution of NaN₃ (0.77g, 11.8 mmol) in H₂O (2 mL) was added dropwise. The resultant heterogenous mixture was stirred for 1 h at 0°C, then cold water (5 mL) and Et₂O (10 mL) were added. The layers were separated and the aqueous layer was extracted with Et₂O (2 x 10 mL). The organic layers were combined, toluene (20 mL) was added, and the organic layers were dried (MgSO₄) and concentrated under reduced pressure to a volume of 20 mL. t-BuOH (5 mL) was added and the mixture was refluxed for 12h. The mixture was concentrated under reduced pressure and the crude residue was taken up in 3M HCl (30 mL) and was heated at reflux for 12h. The mixture was cooled to room temperature and extracted with Et₂O (3 x 15 mL). The aqueous layer was cooled to 0 °C and solid NaOH pellets were added until pH ~12 was reached. The aqueous layer was extracted with Et₂O (3 x 30 mL) and the combined organic layers were dried (MgSO₄) and concentrated under reduced pressure to afford 0.78 g (61% yield) of an oil [MH⁺ 162]. This material was used without further purification.

5

10

15

20

25

$$H_2N$$

The corresponding cyclopropyl analog was prepared according to the procedure outlined in Preparative Example 83.

5

PREPARATIVE EXAMPLE 85

$$HO$$
 H_2N

The corresponding cyclohexyl analog was prepared according to the procedure outlined in Preparative Example 83.

10

15

PREPARATIVE EXAMPLE 86

The desired compound was prepared according to methods previously described in *J. Org. Chem.* **1978**, *43*, 892-898 (the disclosure of which is incorporated herein by reference thereto).

Step A

5

10

15

25

2-Methylthiophene (3g) was dissolved in THF and cooled to -40°C. N-butyllithium (2.5M in hexane, 12.24ml) added dropwise and let stir at -40°C for 30min. CuBr.(CH₃)₂S (6.29g) added and let warm to -25°C where the trifluoroaceticanhydride (4.32ml) was added. The reaction was stirred at -15°C over the weekend. The reaction was quenched with saturated ammonium chloride and extracted with EtOAc. The organic layer washed with brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give 4.59g of an oil (78%).

Step B

The product from Step A (4.58g), hydroxylamine hydrochloride (3g), sodium acetate (4.4g), EtOH (75ml) and H_2O (7.5ml) were combined and heated to 75^0C overnight. The reaction was concentrated *in vacuo*, taken up 1N HCl, extracted with ether, dried with MgSO₄, filtered and concentrated *in vacuo* to give 4.58g of the product (93%, MH+=210).

20 Step C

The product from Step B above (4.5g) was dissolved in TFA (40ml) and cooled to 0°C. Zn powder (4.2g) was added portionwise and let reaction warm to room temperature and stir overnight. The reaction was concentrated *in vacuo*, taken up in 1N NaOH, extracted with ether, dried with MgSO₄, filtered and concentrated *in vacuo* to give 3.43g of the product (80%).

To a solution of KH (0.45 g, 11.3 mmol) in THF (15 mL) at room temperature was added amine hydrochloride (0.85 g, 5.1 mmol) portionwise to afford a heterogenous reaction mixture. The mixture was allowed to stand overnight (12h) and Mel (0.32 mL, 5.1 mmol) was added dropwise. The mixture was stirred for 6h after which the mixture was carefully poured into cold brine (125 mL). The mixture was extracted with Et₂O (3 x 25 mL) and the organic layers were combined. The organic layer was dried (Na₂SO₄), filtered, and concentrated under reduced pressure to afford the crude product as an oil. This material was carried on crude to the coupling step without further purification or characterization.

5

10

15

20

25

PREPARATIVE EXAMPLE 89.1

To a solution of KH (1.1g) in THF (20ml) at room temperature was added (R)-2-amino-1-butanol 48ml) dropwise to afford a heterogenous mixture. The mixture was allowed to stand overnight (18hr) and then MeI (1.59ml) was added dropwise. The mixture was stirred for 4hr after which brine was added. Extracted with ether, dried with K₂CO₃, filtered and concentrated *in vacuo* to afford 1.75g of an oil.

PREPARATIVE EXAMPLE 89.2

$$H_2N$$
 OH
 H_2N
 $O-$

To a solution of KH (1.1g) in THF (20ml) at room temperature was added (S)-2-amino-1-butanol 48ml) dropwise to afford a heterogenous mixture. The mixture was allowed to stand overnight (18hr) and then Mel (1.59ml) was added dropwise. The mixture was stirred for 4hr after which brine was added. Extracted with ether, dried with K₂CO₃, filtered and concentrated *in vacuo* to afford 1.75g of an oil.

The corresponding *cis* analog was prepared in an analogous fashion utilizing the procedure described in Preparative Example 89. This material was also used without further purification.

PREPARATIVE EXAMPLE 91

The desired compound was prepared according to methods previously described in *J. Org. Chem.* **1987**, *52*, 4437-4444 (the disclosure of which is incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 92

15

5

The desired compound was prepared according to methods previously described in *Bull. Chem. Soc. Jpn.* **1962**, *35*, 11-16 (the disclosure of which is incorporated herein by reference thereto).

20

PREPARATIVE EXAMPLE 93

The desired amine was prepared from the corresponding ketone according to standard methods previously described in (a) *Synthesis* **1987**, 998-1001, (b)

Synthesis 1996, 641-646 and (c) J. Med. Chem. 1991, 34, 2176-2186 (the disclosures of each being incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 94

5

10

The desired amine was prepared from the corresponding ketone according to standard methods previously described in(a) *Synthesis* **1987**, 998-1001, (b) *Synthesis* **1996**, 641-646 and (c) *J. Med. Chem.* **1991**, *34*, 2176-2186 (the disclosures of each being incorporated herein by reference thereto).

PREPARATIVE EXAMPLE 95

15 Step A

20

25

Lithium hexamethyldisilylazide (34 mL, 1*M* in THF) was added dropwise to a –78°C THF (20 mL) solution of isobutyronitrile (2.8 mL). After 40 min, cyclopropylmethylbromide (5 g) was added and the mixture warmed to and stirred at 25°C overnight. After cooling to 0°C, 1*M* HCl (aq) was added and the mixture was extracted with diethyl ether, dried over anhydrous Na₂SO₄, filtered and concentrated *in vacuo* at 0°C to give the desired product (4.5 g).

Step B

Methyl Lithium (17 mL, 1.4 *M* in Et₂O) was added to the product from Step A above (1.5 g) in Et₂O (anhydrous) at 0°C. The mixture was stirred at 0-25°C

overnight, then diluted with 3M HCl (aq), extracted with CH₂Cl₂, dried over anhydrous Na₂SO₄, filtered, concentrated *in vacuo* at 0°C and used directly in Step C.

Step C

5

10

15

20

25

The product from Step B above was added to a slurry of NaBH₄ (1.4 g) in isopropanol (50 mL) at 0°C, then the mixture was stirred at reflux for 8 hr and at room temperature for 48 hrs. Water was added and the mixture was stirred for 30 min, then extracted with diethyl ether, dried over anhydrous Na₂SO₄, filtered and concentrated *in vacuo*. The residue was diluted with CH₂Cl₂ and extracted with 3M HCl. The organic phase was discarded and the aqueous phase was basified with NaOH (aq) and extracted with CH₂Cl₂. Drying over anhydrous Na₂SO₄, filtering, and concentration *in vacuo* gave the desired compound (0.5 g).

PREPARATIVE EXAMPLE 96 Step A Step B Step B

Step A

2-Thiophenecarbonyl chloride (2.0mL, 18.7mmol) was dissolved in 100mL dichloromethane. After addition of diisopropylethylamine (4.1mL, 23.4mmol) and Boc-piperazine (3.66g, 19.7mmol), the mixture was stirred for 4h at room temperature. The resulting mixture was put into water (500mL) and acidified with 3N HCl to pH~1. Extraction with dichloromethane (2x100mL) and drying over sodium sulfate resulted in sufficiently pure product that was used in the next step without any further purification. ¹H NMR (300MHz, d₆-DMSO) 1.60 (s, 9H), 3.29 (dd, 4H), 3.69 (dd, 4H), 7.23 (dd, 1H), 7.49 (d, 1H), 7.79 (d, 1H).

Step B

The crude material from Step A was dissolved in trifluoroacetic acid/dichloromethane (75mL, 4/1). After stirring for 2h, the reaction mixture was put

into 1N sodium hydroxide (400mL). Extraction with dichloromethane (2x100mL) and drying over sodium sulfate resulted in sufficiently pure product that was used in Step C without any further purification. ¹H NMR (300MHz, d₆-DMSO) 2.81 (dd, 4H), 3.63 (dd, 4H), 7.21 (dd, 1H), 7.46 (d, 1H), 7.82 (d, 1H).

5

10

15

Step C

The crude material (3.50g, 17.8mmol) from Step B was dissolved in dichloromethane (100mL). After addition of diisopropylethylamine (18.7mL, 107mmol), 3-nitrosalicylic acid (3.3g, 18.0mmol), and PyBrOP (10.4g, 22.3mmol), the resulting mixture was stirred over night at room temperature before being put into 1N sodium hydroxide (200mL). Extraction with dichloromethane (2x200mL) removed all PyBrOP by-products. The aqueous phase was acidified with 3N HCl and subsequently extracted with dichloromethane (3x 100mL). The combined organic phases of the acidic extraction were dried over sodium sulfate, concentrated, and finally purified by column chromatography (dichloromethane/methanol = 10/1) to yield the desired product (2.31g, 34 % over 3 steps). ¹H NMR (300MHz, d₆-DMSO) 3.30-3.90 (m, 8H), 7.10-8.20 (m, double signals due to E/Z-isomers, 6H), 10.82 (s, 1H).

Step D

20

25

The nitro-compound (2.3g, 6.4mmol) from Step C was dissolved in methanol (50mL) and stirred with 10% Pd/C under a hydrogen gas atmosphere over night. The reaction mixture was filtered through Celite and washed thoroughly with methanol. Finally, the filtrate was concentrated *in vacuo* and purified by column chromatography (dichloromethane/methanol = 10/1) to yield the desired product (1.78g, 84%). ¹H NMR (300MHz, d₆-DMSO) 3.30-3.90 (m, 8H), 7.22 (m, 2H), 7.55 (d, 1H), 7.71 (d, 1H), 7.88 (d, 1H), 8.15 (d, 1H), 10.85 (bs, 1H).

Step A

5

10

Picolinic acid (3.0g, 24.3mmol) was suspended in SOCl₂ (15mL). After addition of dimethylformamide (5 drops), the reaction mixture was stirred for 4 hours. Evaporation of the solvent yielded the corresponding acid chloride as HCl-salt. Without any further purification, the solid was suspended in 120mL dichloromethane. After addition of diisopropylethylamine (12.7mL, 73mmol) and Boc-piparazine (4.8g, 25.5mmol), the reaction was stirred over night at room temperature. The resulting mixture was put into water (500mL) and extracted with dichloromethane (2x100mL). Drying over sodium sulfate resulted in sufficiently pure product that was used in Step B without any further purification. ¹H NMR (300MHz, d₆-DMSO) 1.63 (s, 9H), 3.21 (dd, 4H), 3.61 (dd, 4H), 7.57 (dd, 1H), 7.63 (d, 1H), 7.98 (dd, 1H), 8.70 (d, 1H).

15 <u>Step B</u>

The crude material from Step A was dissolved in trifluoroacetic acid/dichloromethane (75mL, 4/1). After stirring for 2days, the reaction mixture was put into 1N sodium hydroxide (400mL). Extraction with dichloromethane (2x100mL) and drying over sodium sulfate resulted in sufficiently pure product that was used in Step C without any further purification. ¹H NMR (300MHz, d₆-DMSO) 2.77 (dd, 2H), 2.83 (dd, 1H), 3.38 (dd, 2H), 3.64 (dd, 1H), 7.58 (dd, 1H), 7.62 (d, 1H), 8.00 (dd, 1H), 8.67 (d, 1H).

Step C

25

20

The crude material (1.35g, 7.06mmol) from Step B was dissolved in dichloromethane (50mL). After addition of diisopropylethylamine (3.7mL, 21.2mmol), 3-nitrosalicylic acid (1.36g, 7.41mmol), and PyBrOP (3.62g, 7.77mmol), the resulting mixture was stirred over night at room temperature before being put into 1N sodium hydroxide (300mL). Extraction with dichloromethane (2x100mL) removed any

PyBrOP products. The aqueous phase was acidified with 3N HCI. Adjustment of the pH with saturated sodium carbonate solution to almost neutral crushed the desired compound out of solution. The aqueous phase was subsequently extracted with dichloromethane (3x 100mL). The combined organic layers of the neutral extraction were dried over sodium sulfate, concentrated, and finally purified by column chromatography (dichloromethane/methanol = 20/1) to yield the desired product (1.35g, 16% over 3 steps). ¹H NMR (300MHz, d₆-DMSO) 3.30-3.95 (m, 8H), 7.22 (m, 1H), 7.61 (m, 1H), 7.73 (d, 2H), 8.03 (m, 1H), 8.17 (m, 1H), 8.69 (m, 1H), 10.82 (s, 1H).

10

5

Step D

The nitro-compound (1.35g, 3.79mmol) from Step C was dissolved in methanol (60mL) and stirred with 10% Pd/C under a hydrogen gas atmosphere over night. The reaction mixture was filtered through Celite and washed thoroughly with methanol.

Finally, the filtrate was concentrated *in vacuo* and purified by column chromatography (dichloromethane/methanol = 20/1) to yield the desired product (1.10g, 89 %). ¹H NMR (300MHz, d₆-DMSO) 3.50-3.85 (m, 8H), 6.47 (dd 1H), 6.74 (m, 2H), 7.59 (dd, 1H), 7.71 (d, 1H), 8.04 (dd, 1H), 8.68 (d, 1H).

20

25

PREPARATIVE EXAMPLE 98

Step A

1-Methyl-2-pyrrolecarboxylic acid (2.5g, 20.0mmol) was dissolved in dichloromethane (50mL). After addition of PyBrOP (16.3g, 35.0mmol), diisopropylethylamine (14.0mL, 73.0mmol) and Boc-piparazine (5.5g, 30.0mmol), the reaction was stirred over night at room temperature before being put into 1N sodium hydroxide (200mL). Extraction with dichloromethane (2x100mL) removed all PyBrOP by-products. The aqueous phase was acidified with 3N HCI. Adjustment of the pH

with saturated sodium carbonate solution to almost neutral precipitated the desired compound. The aqueous phase was subsequently extracted with dichloromethane (3x 100mL). The combined organic phases of the neutral extraction were dried over sodium sulfate. Removal of the solvent resulted in sufficiently pure product that was used in Step B without any further purification. ¹H NMR (300MHz, d₆-DMSO) 1.59 (s, 9H) 3.21 (dd, 4H), 3.61 (dd, 4H), 3.74 (s, 3H), 6.11 (dd, 1H), 6.33 (d, 1H), 7.01 (d, 1H).

Step B

10

15

20

25

5

The crude material from Step A was dissolved in trifluoroacetic acid/dichloromethane (75mL, 4/1). After stirring for 3h, the reaction mixture was put into 1N sodium hydroxide (400mL). Extraction with dichloromethane (3x100mL) and drying over sodium sulfate resulted in sufficiently pure product that was used in Step C without any further purification. ¹H NMR (300MHz, d₆-DMSO) 2.79 (dd, 4H), 3.62 (dd, 4H), 3.76 (s, 3H), 6.11 (dd, 1H), 6.37 (d, 1H), 6.96 (d, 1H).

Step C

The crude material (3.15g, 16.3mmol) from Step B was dissolved in dichloromethane (100mL). After addition of diisopropylethylamine (8.5mL, 49.0mmol), 3-nitrosalicylic acid (3.13g, 17.1mmol), and PyBrOP (9.11g, 19.6mmol), the resulting mixture was stirred over night at room temperature before being put into 1N sodium hydroxide (400mL). Extraction with dichloromethane (2x100mL) removed all PyBrOP products. The aqueous phase was then carefully acidified with 3N HCl until the color of the solution changes from orange to yellow and the desired compound crashed out of solution. The aqueous phase was subsequently extracted with dichloromethane (3x 100mL). The combined organic layers of the acidic extraction were dried over sodium sulfate and concentrated *in vacuo* to yield the desired product. ¹H NMR (300MHz, d₆-DMSO) 3.35-3.85 (m, 8H), 3.79 (s, 3H), 6.13 (dd, 1H), 6.45 (d, 1H), 7.01 (s, 1H), 7.22 (dd, 1H), 7.70 (d, 1H), 8.16 (d, 1H), 10.83 (s, 2H).

30

Step D

The crude nitro-compound from Step C was suspended in methanol (60mL) and stirred with 10% Pd/C under a hydrogen gas atmosphere over night. The reaction mixture was filtered through Celite and washed thoroughly with methanol. The filtrate

was concentrated *in vacuo* and purified by column chromatography (dichloromethane/methanol = 10/1) to yield the desired product (2.61g, 40 % for 4 steps). ¹H NMR (300MHz, d₆-DMSO) 3.45-4.80 (m, 8H), 3.79 (s, 3H), 6.17 (dd, 1H), 6.45 (m, 2H), 6.78 (m, 2H), 7.01 (d, 1H).

5

10

15

20

PREPARATIVE EXAMPLE 99

Step A

2-Bromopyridine N-oxide hydrochloride (1.13g, 5.37mmol) and Boc-piperazine (1.50g, 8.06mmol) were heated to 80° C in pyridine (10mL) over night. The reaction mixture was put into water (300mL) and then extracted with dichloromethane (2x100mL). The combined organic phases were dried over sodium sulfate, concentrated, and finally purified by column chromatography (dichloromethane/methanol = 10/1) to yield the desired product (500mg, 33 %).

¹H NMR (300MHz, d-CDCl₃) 1.60 (s, 9H), 3.46 (dd, 4H), 3.78 (dd, 4H), 6.99 (m, 2H), 7.37 (dd, 1H), 8.33 (d, 1H).

Step B

The purified product (500mg, 1.79mmol) was stirred for 30 min with 4N HCl/dioxane (15mL). Evaporation of the solvent yielded the crude amine (465mg) as multiple HCl-salt which was used in Step C without any further purification.

¹H NMR (300MHz, d₆-DMSO) 3.38 (m, 4H), 4.81 (m, 4H), 7.34 (dd, 1H), 7.55 (d, 1H), 7.86 (dd, 1H), 8.55 (d, 1H).

25 <u>Step C</u>

The crude material (370mg, 1.48mmol) from Step B was suspended in dichloromethane (20mL). After addition of diisopropylethylamine (2.6mL, 14.8mmol),

3-nitrosalicylic acid (406mg, 2.22mmol), and PyBrOP (1.21g, 2.59mmol), the mixture was stirred over night at room temperature before being put into 1N sodium hydroxide (50mL). Extraction with dichloromethane (2x50mL) removed all PyBrOP products. The aqueous phase was then carefully acidified (pH ~ 4-5) with 3N HCl and extracted with dichloromethane (3x 50mL). The combined organic layers of the acidic extraction were dried over sodium sulfate, concentrated *in vacuo* and purified by column chromatography (dichloromethane/methanol = 10/1) to yield the desired product (330mg, 65%).

LCMS calculated: 344.1, found: (M+1)⁺ 345.1

10

15

5

Step D

Sodium hydrosulfite (1.05g) was dissolved in water (3.0mL) to yield a 1.5N solution. Addition of dioxane (3.0mL) was followed by injection of conc. ammonium hydroxide (0.60mL, yields a 1.0N concentration). After addition of the nitrocompound (100mg, 0.29mmol), the reaction mixture was stirred for 0.5h. Subsequently, the solvent was removed and the residue suspended in dichloromethane/methanol (10/1). Filtration through Celite removed most of the salts. Final purification by column chromatography (dichloromethane/methanol = 5/1) yielded the desired product (68mg, 75%).

20

LCMS calculated: 314.14, found: (M+1)⁺ 315.1

PREPARATIVE EXAMPLE 100

4-Bromopyridine hydrochloride (3.0g, 15.4mmol) was dissolved in water (15mL). After addition of N-benzylpiperazine (14.8mL, 85.0mmol) and 500mg copper sulfate, the reaction mixture was heated overnight to 140° C. The resulting product was extracted with ether (5x75mL), dried over sodium sulfate and concentrated. Final purification by column chromatography (dichloromethane/methanol/NH₄OH = 10/1/0.1) yielded the desired product (2.16g, 55%). ¹H NMR (300MHz, d-CDCl₃) 2.68 (dd, 4H), 3.45 (dd, 4H), 6.76 (d, 2H), 7.40 (m, 5H), 8.38 (d, 2H).

10 <u>Step B</u>

5

15

20

25

The benzylamine (2.16g, 8.54mmol) from Step A, ammonium formate (2.71g, 43.0mmol) and Pd(C) (10%, 1.0g) was suspended in methanol (50mL) and refluxed for 3h. The palladium was filtered off and the filtrate was concentrated. The sufficiently pure product was used in Step C without any further purification. ¹H NMR (300MHz, d-CDCl₃) 2.48 (bs, 1H), 3.13 (dd, 4H), 3.41 (dd, 4H), 7.78 (d, 2H), 8.39 (d, 2H).

Step C

The crude material (1.15g, 7.06mmol) from Step B was dissolved in dichloromethane (50mL). After addition of diisopropylethylamine (4.7mL, 42.4mmol), 3-nitrosalicylic acid (1.94g, 10.6mmol), and PyBrOP (5.78g, 12.3mmol), the resulting mixture was stirred over night at room temperature before being put into 1N sodium hydroxide (300mL). Extraction with dichloromethane (2x100mL) removed all PyBrOP products. The aqueous phase was carefully acidified to pH ~ 5-6 with 3N HCl and extracted with dichloromethane (3x 100mL). The combined organic layers of the neutral extraction were dried over sodium sulfate, concentrated, and finally purified by column chromatography (dichloromethane/methanol/NH₄OH = 10/1/0.1) to yield the desired product (850mg, 37% for 2 steps).

30 Step D

The nitro-compound (850mg, 2.59mmol) from Step C was dissolved in methanol (40mL) and stirred with 10% Pd/C under a hydrogen gas atmosphere over night. The reaction mixture was filtered through Celite and washed thoroughly with methanol. Finally, the filtrate was concentrated *in vacuo* and purified by column

chromatography (dichloromethane/methanol/

 $NH_4OH = 10/1/0.1$) to yield the desired product (650g, 84 %). ¹H NMR (300MHz, d₆-DMSO) 3.40-3.75 (bm, 8H), 6.49 (dd, 1H), 6.76 (m, 2H), 6.93 (d, 2H), 8.28 (d, 2H).

PREPARATIVE EXAMPLE 101

Step 1

5

10

15

25

N,N'-Dibenzyl-ethane-1,2-diamine (20mL, 0.0813mol), triethylamine (22.66mL, 0.1626mol) and benzene (100mL) were combined in a round bottom flask. A solution of 2,3-dibromo-propionic acid ethyl ester (11.82mL, 0.0813mol) in benzene (50mL) was added dropwise. The solution was refluxed over night and monitored by TLC (20% ethyl acetate/hexane). The reaction was cooled to room temperature, then filtered and washed with benzene. The filtrate was concentrated then purified by column chromatography (15% ethyl acetate/hexane). The product was isolated as an oil (25.42g, 0.0752mol, 92%). MS: calculated: 338.20, found: 339.2

¹H NMR (300 MHz, CDCl₃) 1.23 (t, 3H), 2.48 (m, 3H), 2.62 (m, 1H), 2.73 (m, 1H), 3.07 (m, 1H), 3.30 (m, 1H), 3.42 (d, 1H), 3.56 (m, 2H), 3.91 (d, 1H), 4.17 (m, 2H), 7.27 (m, 10H).

20 Step 2

In a Parr shaker vessel, the ester (25.43g, 0.075mol) and methanol (125mL) were combined. The vessel was purged with argon and palladium catalyst (5% on carbon, 2.5g) was added. The system was shaken under an atmosphere of hydrogen overnight. TLC (20% ethyl acetate/hexane) indicated that reaction was complete. The reaction mixture was filtered through a pad of Celite and washed with methanol. The filtrate was concentrated and the product isolated as a solid (11.7g, 0.074mol, 98%).

MS: calculated: 158.11, found: 159.2^{-1} H NMR (300 MHz, CDCl₃) 1.27 (t, 3H), 2.70 (m, 4H), 2.96 (m, 1H), 3.13 (dd, 1H), 3.43 (dd, 1H), 4.18 (m, 2H).

PREPARATIVE EXAMPLE 102

Piperazine-2-carboxylic acid ethyl ester (3.11g, 0.0197mol),

diisopropylethylamine (5.15mL, 0.0296mol) and methylene chloride (200mL) were combined in a round bottom flask. While stirring at room temperature, a solution of N,N-dimethylcarbamoyl chloride (1.81mL, 0.0197mol) in methylene chloride (20mL) was added dropwise. The reaction was stirred for one hour. After this time the reaction was concentrated and carried on to the next step without further purification. (99% yield).

MS: calculated: 229.14, found:230.1

¹H NMR (300 MHz, CDCl₃) 1.30 (t, 3H), 2.85 (s, 6H), 3.10 (m, 3H), 3.31 (m, 2H), 3.60 (m, 2H), 4.21 (q, 2H).

PREPARATIVE EXAMPLE 103-104

Following the procedure described for Preparative Example 102, the Products listed in the table below were prepared using the commercially available chloride shown and piperazine-2-carboxylic acid ethyl ester from Preparative Example 101.

Example	Chloride	Product	1.Yield (%) 2. (M+1) ⁺
103	O>s/ / °,0	O OEt	1. 99 2. 237.1
104	CO	O OEt	1. 62 2. 253.1

15

PREPARATIVE EXAMPLE 105

Step 1

5

10

15

20

25

30

3-Nitrosalicylic acid (3.61g, 0.0197g), DCC (2.03g, 0.0099mol) and ethyl acetate (130mL) were combined in a round bottom flask and stirred for 15min. 4-Dimethylcarbamoyl-piperazine-2-carboxylic acid ethyl ester (4.51g, 0.0197g) was added, and the reaction was stirred for 72 hours. The reaction mixture was concentrated then dissolved in dichloromethane. The organic phase was washed once with 0.1N sodium hydroxide. The aqueous phase was back extracted once with dichloromethane. The aqueous phase was acidified and wash three times with ethyl acetate. The aqueous phase was concentrated and purified by column chromatography (5% methanol/DCM).

MS: calculated: 394.15, found:395.0

¹H NMR (300 MHz, CDCl₃) 1.32 (t, 3H), 2.86 (m, 7H), 3.15 (m, 1H), 3.51 (m, 4H), 4.24 (m, 3H), 7.15 (m, 1H), 7.66 (m, 1H), 8.20 (m, 1H), 10.86 (bs, 1H).

Step 2

4-Dimethylcarbamoyl-1-(2-hydroxy-3-nitro-benzoyl)-piperazine-2-carboxylic acid ethyl ester (0.80g, 0.002mol) and methanol (50mL) were combined in a round bottom flask. The system was purged with argon. To the solution was added 5% palladium on carbon (~100mg). The flask was purged with hydrogen and stirred overnight. The reaction was filtered through a pad of celite and washed with methanol. The material was concentrated then purified by column chromatography (6% methanol/DCM). Isolated product (0.74g, 0.002mol, 100%).

MS: calculated: 364.17, found:365.1

¹H NMR (300 MHz, CDCl₃) 1.27 (t, 3H), 2.85 (m, 8H), 3.18 (1H), 3.45 (m, 3H), 4.19 (m, 3H), 3.90 (m, 3H)

Step 3

1-(3-Amino-2-hydroxy-benzoyl)-4-dimethylcarbamoyl-piperazine-2-carboxylic acid ethyl ester (0.74g, 0.002mol) was suspended in a solution of dioxane (10mL)

and water (10mL). Lithium hydroxide (0.26g, 0.0061mol) was added and the mixture stirred for two hours. The solution was acidified to pH=6 with 3N HCl then extracted with butanol. The extracts were combined, dried over sodium sulfate and concentrated.

5

MS: calculated: 336.14, found:337.1

¹H NMR (300 MHz, CD₃OD) 2.86 (m, 7H), 3.23 (m, 3H), 3.54 (m, 3H), 6.92 (m, 2H), 7.23 (m, 1H).

PREPARATIVE EXAMPLE 106-107

10

15

Following the procedure described for Example 105, the Products listed in the table below were prepared using the amine from the Preparative Example indicated and 3-nitrosalacylic acid.

Example	Aniline	Product	1.Yield (%)
			2. (M+1) ⁺
			3. Note
400	100		
106	103		1. 91
		O CO ₂ H	2. Not
		HONO	observed
		H ₂ N N S	3. Rainey
			nickel used in
			Step 2
107	104		1. 24
		O CO ₂ H	2. 360.0
		HON	3. For Step
		H_2N	1 used
			PyBrop/
			DIEA in DCM

PREPARATIVE EXAMPLE 108

$$HO \longrightarrow OH$$
 $HN \longrightarrow OH$ OOH OOH

5

3-Nitrosalicylic acid (1.0g, 5.5mmol) was dissolved in ethyl acetate (20mL). 1,3-Dicyclohexylcarbodiimide (0.568g, 2.8mmol) was added and the mixture was stirred for approximately 10 minutes and cooled to 0°C. During this time a precipitate formed. Azetidine (0.39mL, 5.8mmol) was added and the reaction was stirred overnight and allowed to warm to room temperature. After this time the reaction was cooled to 0°C and filtered. The collected solid was washed with chilled ethyl acetate. The filtrate was concentrated and purified by column chromatography (80% EtOAc/Hex) to give the product (476mg, 39.0%).

 1 H NMR (300 MHz, CDCl₃) δ2.40(m, 2H), 4.38(m, 4H), 6.97(m, 1H), 7.62(d, 1H), 10 8.12(d, 1H), 12.88(m, 1H) ppm.

Step B

$$N = 0$$
 $N = 0$
 $N =$

The nitro compound (0.48q, 2.1mmol) from Preparative Example 32 Step A 15 was dissolved in methanol (25ml) and stirred with 10% Pd/C under a hydrogen gas atmosphere overnight. The reaction mixture was filtered through celite, the filtrate concentrated in vacuo to give the product (344mg, 90%). ¹H NMR (300 MHz, CDCl₃)

 δ 2.52(m, 2H), 4.57(bs, 4H), 6.75(m, 1H), 6.90(m, 2H), 12.71(bs, 1H) ppm.

PREPARATIVE EXAMPLE109

In essentially the same manner as described in Preparative Example 108 above, the morpholino-amine product was obtained.

20

Piperazine (4.9g, 0.057mol) was dissolved in dichloromethane (100mL). N,N'-Dimethylcarbamoyl chloride (1.0mL, 0.011mol) was added dropwise to the solution at room temperature. The reaction was stirred for one hour. After this time 1N potassium hydroxide (200mL) was added. The layers were separated and the aqueous layer was extracted three times with dichloromethane. The organic fractions were combined and dried over sodium sulfate. Filtration and concentration provided the product, without further purification, as an oil (1.16g, 13%).

¹H NMR (CDCl₃, 300 MHz) 1.95 (s, 1H), 2.83 (s, 6H), 2.86 (m, 4H), 3.20 (m, 4H).

MS: calculated: 157.12, found: 158.1.

5

10

15

20

25

30

PREPARATIVE EXAMPLE 111

Piperazine (4.9g, 0.057mol) was dissolved in 1N HCl (100mL). A solution of phenylsulfonylchloride (1.45mL, 0.011mol) in acetonitrile (25mL) was added dropwise to the solution at room temperature. The reaction was stirred for 30 minutes. After this time the reaction was extracted two times with ethyl acetate. The solution was then made basic with 1N potassium hydroxide and extracted three times with dichloromethane. The dichloromethane fractions were combined and dried over magnesium sulfate. Filtration and concentration provided the product, without further purification, as a solid (1.22g, 9.4%).

¹H NMR (CDCl₃, 300 MHz) 2.94 (m, 8H), 7.56 (m, 3H), 7.76 (m, 2H). MS: calculated: 226.08, found: 227.1.

PREPARATIVE EXAMPLE 112

Piperazine (4.9g, 0.057mol) was dissolved in dichloromethane (100mL). Methanesulfonyl chloride (0.85mL, 0.011mol) was added dropwise to the solution at room temperature. The reaction was stirred for 30 minutes. After this time 1N potassium hydroxide (200mL) was added. The layers were separated and the

aqueous layer was extracted three times with dichloromethane. The organic fractions were combined and dried over sodium sulfate. Filtration and concentration provided the product, without further purification, as a solid (1.07g, 11%).

¹H NMR (CDCl₃, 300 MHz) 1.75 (s, 1H), 2.78 (s, 3H), 2.97 (m, 4H), 3.20 (m, 4H).

MS: calculated: 164.06, found: 165.1.

PREPARATIVE EXAMPLE 113

10 Step A

15

20

25

Boc-Piperazine (3.0g, 0.0161mol) was dissolved in dichloromethane (100mL). Propylisocyanate (1.51mL, 0.0161mol) was added to the solution at room temperature. The reaction was stirred for over night. After this time the reaction was diluted with 1N potassium hydroxide (200mL) and extracted six times with dichloromethane. The organic fractions were combined and dried over magnesium sulfate. Filtration and concentration provided the product as a solid.

Step B

The product of Step A above, was dissolved in a 30% trifluoroacetic acid/dichloromethane solution and stirred overnight. After this time a 1N potassium hydroxide solution (200 mL) was added to the reaction. The aqueous layer was extracted a total of six times with dichloromethane. The organic fractions were combined and dried over sodium sulfate. Filtration and concentration provided the product (1.37g, 50%).

¹H NMR (CDCI₃, 300 MHz) 0.92 (t, 3H), 1.52 (m, 2H), 2.89 (m, 4H), 3.01 (s, 1H), 3.18 (m, 2H), 3.37 (m, 4H), 4.61 (bs, 1H).

MS: calculated: 171.14, found: 172.0.

PREPARATIVE EXAMPLE 114

Piperazine (4.9g, 0.0569mol) was dissolved in 1N HCI (70mL). A solution of phenylchloroformate (1.43mL, 0.0114mol) in acetonitrile (25mL) was added dropwise to the solution at room temperature. The reaction was stirred for 30 minutes. After this time the reaction was extracted two times with ethyl acetate. The solution was then made basic with 1N potassium hydroxide and extracted three times with dichloromethane. The dichloromethane fractions were combined and dried over magnesium sulfate. Filtration and concentration provided the product, without further purification, as a solid (2.12g, 18%).

¹H NMR (CDCl₃, 300 MHz) 1.78 (s, 1H), 2.91 (m, 4H), 3.59 (m, 4H), 7.11 (2H), 7.19 (m, 1H), 7.36 (m, 2H).

MS: calculated: 206.24, found: 207.1.

PREPARATIVE EXAMPLE 115-117

Following the procedure described for Example 112, the Products listed in the table below were prepared using the commercially available chloroformate shown and piperazine.

Example	Chloroformate	Product	1.Yield (%)
			2. (M+1) ⁺
115	=0	0=	
	CI O	N O	1. 54
	;	HN	2. 144.9
116	0	0	
	CI	N O	1. 17
		HN	2. 173.0
117			
	CIOO	N O	1. 69
		HN	2. 173.0

PREPARATIVE EXAMPLE 118

5

Boc-Piperazine (3.01g, 0.0161mol) was dissolved in dichloromethane (100mL) along with diisopropylethylamine (5.61mL, 0.0322mol). Benzoylchloride (1.87mL, 0.0161mol) was added dropwise to the solution at room temperature. The reaction was stirred for several hours. After this time the reaction was concentrated and the product was purified by column chromatography (10% MeOH/DCM). Boc-Protected product was isolated as a solid (5.21g).

¹H NMR (CDCl₃, 300 MHz) 1.47 (s, 9H), 3.45 (m, 8H), 7.41 (m, 5H). MS: calculated: 290.16, found: 290.8.

10

5

Step B

The product from Step A above, was dissolved in a 50% trifluoroacetic acid/dichloromethane solution and stirred overnight. After this time the reaction was diluted with 1N potassium hydroxide (200mL) and the organic layer was separated. The aqueous phase was then extracted six times with dichloromethane. The organic fractions were combined and dried over magnesium sulfate. Filtration and concentration provided product (2.93g).

¹H NMR (CDCl₃, 300 MHz) 1.92 (s, 1H), 2.87 (m, 4H), 3.52 (m, 4H), 7.39 (s, 5H).

20

15

MS: calculated: 190.11, found: 191.1.

PREPARATIVE EXAMPLE 119

Step A

25

30

Boc-Piperazine (3.0g, 0.0161mol) was dissolved in dichloromethane (100mL) along with diisopropylethylamine (3.1mL, 0.0177mol). N,N'-dimethylsulfamoyl chloride (1.73mL, 0.0161mol) was added dropwise to the solution at room temperature. The reaction was stirred for several hours. After this time the reaction was diluted with water (100mL). The layers were separated and the aqueous layer was extracted six times with dichloromethane. The organic fractions were combined and dried over magnesium sulfate. Filtration and concentration provided the product, without further purification, as a solid (4.53g).

¹H NMR (CDCl₃, 300 MHz) 1.47 (s, 9H), 2.84 (s, 6H), 3.21 (m, 4H), 3.48 (m, 4H).

MS: calculated: 293.14, found: 194.1 (M-Boc)[†].

5 Step B

The product from Step A above, was dissolved in a 30% trifluoroacetic acid/dichloromethane solution and stirred overnight. After this time the reaction was diluted with water and 1N potassium hydroxide was used to make the aqueous layer slightly basic. The aqueous layer was extracted a total of seven times with dichloromethane. The organic fractions were combined and dried over sodium sulfate. Filtration and concentration provided the product (2.96g).

¹H NMR (CDCl₃, 300 MHz) 2.03 (s, 1H), 2.83 (s, 6H), 2.92 (m, 4H), 3.23 (m, 4H).

MS: calculated: 193.09, found: 194.1.

15

10

PREPARATIVE EXAMPLE 120

Step A

In essentially the same manner as that described in Preparative Example 105, Step 1, using 3-nitrobenzoic acid instead of 3-nitrosalicylic acid, the methyl ester product was prepared.

Step B

$$O_2N$$
 O_2N
 O_2N
 O_2N
 O_3N

25

30

The methyl ester (1.79g, 6.1mmol) from Step A above, was dissolved in dioxane/water (20mL/15mL) at room temperature. Lithium hydroxide (0.258g, 6.2mmol) was added to the solution. After a few hours more lithium hydroxide was added (0.128g, 3.0mmol) and the reaction was stirred for another hour. After this time the reaction was concentrated and then taken up in water. The solution was

extracted two times with ether. The aqueous phase was then acidified and extracted three times with ethyl acetate. The organic fractions were then dried over sodium sulfate, filtered and concentrated. Product was isolated by column chromatography (95% EtOAc/Hex, 0.05% HOAc) to give the product (1.66 g, 98%).

¹H NMR (300 MHz, CDCl₃) 1.49(m, 2H), 1.68(m, 1H), 1.82(m, 2H), 2.44(m, 1H) 3.32(m, 1H), 3.58(m, 1H), 5.57(m, 1H), 7.65(m, 1H), 7.80(m, 1H), 8.32(m, 2H), 10.04(bs, 1Hppm).

Step C

$$O_2N$$
 O_2N
 O_2N

10

15

20

5

The nitro compound was dissolved in an excess of methanol (20mL) and covered by a blanket of argon. 5% Palladium on carbon was added (catalytic) and a hydrogen balloon was attached to the flask. The atmosphere of the system was purged under vacuum and replaced with hydrogen. This step was repeated for a total of three times. The reaction was then stirred under hydrogen overnight. After this time the balloon was removed and the solution was filtered through celite followed by several rinses with methanol. The filtrate was concentrated and dried on the vacuum line to provide the desired aniline product (1.33 g, 90%).

¹H NMR (300 MHz, CDCl₃) 1.40(m, 2H), 1.50(m, 1H), 1.68(m, 2H), 2.33(m, 1H) 3.18(m, 1H), 3.62(m, 1H), 5.39(m, 1H), 6.12(bs, 2H), 6.75(m, 2H), 7.12(m, 1H)ppm.

Mass Spectra, calculated: 248, found: 249.1 (M+1)⁺

PREPARATIVE EXAMPLES 121-123

25

Following the procedure described in Preparative Example 120, but using the commercially available amine and benzoic acid indicated, the intermediate products in the table below were obtained.

Ex.	Carboxylic	Amine	Product	1.Yield
	Acid			(%)
				2. (M+1) ⁺
				3. Note
121	NO ₂	NH-HCI		1. 21
	ОН	IN 1-1 IOI	NH ₂	2. 251.0
	\ 0	0 0-	O OH	
	но		ООН	
122	NO ₂	NH-HCI		1. 21
	ОН	101	N-NH ₂	2. 265.0
	\	0 0-	OH	3.
	HO		0 0-	Skipped
				step B
123	NO ₂	NH-HCI		1. 15
	ОН	1417-1101	N-NH ₂	2. 264.0
	\ 0	0 N− H	OH	3.
	HO		0 N− H	Skipped
			11	step B

PREPARATIVE EXAMPLE 124

$$HO_2C$$
 OH NO_2 + $N-H$ Step A OH OH OH

5 Step A

10

3-Nitrosalicylic acid (500 mg, 2.7 mmol), 1,3-dicyclohexylcarbodiimide (DCC) (563 mg) and ethyl acetate (10 mL) were combined and stirred for 10 min. (*R*)-(-)-2-pyrrolidinemethanol (0.27 mL) was added and the resulting suspension was stirred at room temperature overnight. The solid was filtered off and the filtrate was either concentrated down and directly purified or washed with 1N NaOH. The aqueous phase was acidified and extracted with EtOAc. The resulting organic phase was dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. Purification of the

residue by preparative plate chromatography (silica gel, 5% MeOH/CH₂Cl₂ saturated with AcOH) gave the desired compound (338 mg, 46%, MH^{+} = 267).

Step B

5

10

The product from Step A above was stirred with 10% Pd/C under a hydrogen gas atmosphere overnight. The reaction mixture was filtered through celite, the filtrate concentrated *in vacuo*, and the resulting residue purified by column chromatography (silica gel, 4% MeOH/CH₂Cl₂ saturated with NH₄OH) to give the product (129mg, 43%, MH+=237).

PREPARATIVE EXAMPLES 125-145

Following the procedure described for Preparative Example 124, but using the commercially available amine or the amine from the Preparative Example indicated and 3-nitrosalicylic acid, the products in the table below were obtained.

Ex.	Amine Comm. Avail./ From Prep.Ex.	Product	1.Yield(%) 2. (M+1) ⁺
125	NH	N NH2	1. 37 2. 298.1
126	O NH	O OH NH ₂	1. 31 2. 310.1
127	O N NH	ON NH2	1. 68 2. 294.1
128	CINNH	CI NH ₂ OH	1. 54 2. 365.9

129	O N NH	ON NH2	1. 45 2. 316.1
130	110	ON NH2	1. 59 2. 293.1
131	111	S-N NH ₂	1. 32 2. 362.0
132	114	O N NH ₂	1. 36 2. 342.0
133	112	S-N NH2	1. 65 2. 300.0
134	O NH	HNN NH2	1. 48 2. 321.1
135	N N NH	N N N O OH	1. 50 2. 300.1
136	N NH	N N O OH	1. 56 2. 299.2
137	115	ON NH ₂ OH	1. 79 2. 280.1

138	116	ON NH2	1. 64 2. 307.1
139	N NH	NNN OH	1. 73 2. 304.2
140	O N NH	ON NH2	1. 34 2. 264.0
141	117	ON NH2	1. 40 2. 307.1
142	113	HN NH2	1. 91 2. 307.1
143	118	O N N O OH	1. 9.0 2. 326.0
144	119	N-S-N NH2	1. 42 2. 329.0
145	NNH	NH ₂	1. 6.5 2. 236.1

PREPARATIVE EXAMPLE 146

5

10

15

20

To a solution of tosylaziridine (*J. Am. Chem. Soc.* **1998**, *120*, 6844-6845, the disclosure of which is incorporated herein by refernce thereto) (0.5 g, 2.1 mmol) and $Cu(acac)_2$ (55 mg, 0.21 mmol) in THF (5 mL) at 0 °C was added PhMgBr (3.5 ml, 3.0 M in THF) diluted with THF (8 mL) dropwise over 20 min. The resulting solution was allowed to gradually warm to rt and was stirred for 12h. Sat. aq. NH_4Cl (5 mL), was added and the mixture was extracted with Et_2O (3 x 15 mL). The organic layers were combined, washed with brine (1 x 10 mL), dried (MgSO₄) and concentrated under reduced pressure. The crude residue was purified by preparative TLC eluting with hexane/EtOAc (4:1) to afford 0.57 g (86% yield) of a solid. The purified tosylamine was taken on directly to the next step.

Step B

To a solution of tosylamine (0.55 g, 1.75 mmol) in NH $_3$ (20 mL) at -78 °C was added sodium (0.40 g, 17.4 mmol). The resulting solution was stirred at -78 °C for 2 h whereupon the mixture was treated with solid NH $_4$ Cl and allowed to warm to rt. Once the NH $_3$ had boiled off, the mixture was partitioned between water (10 mL) and CH $_2$ Cl $_2$ (10 mL). The layers were separated and the aqueous layer was extracted with CH $_2$ Cl $_2$ (2 x10 mL). The organic layers were combined,), dried (NaSO $_4$), and concentrated under reduced pressure to a volume of \sim 20 mL. 4N HCl in dioxane (5 mL) was added and the mixture was stirred for 5 min. The mixture was concentrated under reduced pressure and the resultant crude residue was recrystallized from EtOH/Et $_2$ O to afford 0.30 g (87% yield) of a solid.

25

PREPARATIVE EXAMPLES 147-156.10

Following the procedure set forth in Preparative Example 146 but using the requisite tosylaziridines and Grignard reagents listed in the Table below, the following racemic amine hydrochloride products were obtained.

Prep Ex.	Tosyl aziridine	Grignard Reagent	Amine hydrochloride	Yield (%)
147	NTs	MeMgBr	NH ₂ ·HCI	19
148	NTs	EtMgBr	√″ _{NH₂} ·HCI	56
149	NTs	<i>n</i> -PrMgBr	NH ₂ ·HCl	70
150	NTs	<i>i</i> -PrMgCl	NH ₂ ·HCl	41
151	NTs	BnMgCl	''NH ₂ ·HCI	61
152	NTs	MeMgBr	NH ₂ ·HCl	61
153	NTs	EtMgBr	NH ₂ ·HCl	66
154	NTs	<i>n</i> -PrMgBr	''',NH ₂ ·HCI	80

155	NTs	<i>i</i> -PrMgBr	·'NH2·HCI	27
156	NTs	BnMgCl	·',NH2·HCI	79
156.1	NTs	MgBr	H ₂ N	52
156.2	NTs	MgBr	H ₂ N	49
156.3	TsN	BrMg	H ₂ N	61
156.4	TsN	BrMg	H ₂ N	57
156.5	TsN	MgBr	H ₂ N	64
156.6	TsN	MgBr	H ₂ N	64
156.7	TsN	MgBr	H ₂ N	45

156.8	TsN	BrMg	H ₂ N	23
156.9	TsN	MgBr	H ₂ N	40
156.10	TsN	BrMg	H ₂ N	15

PREPARATIVE EXAMPLE 156.11

Step A

5

10

To a solution of the amine (118mg) from Preparative Example 148 in CH₂Cl₂ (10ml) was added triethylamine (120ul), R-Mandelic Acid (164mg), DCC (213mg) and DMAP (8.8mg)and let stir for 40hr. The mixture was diluted with CH₂Cl₂ and washed with saturated ammonium chloride, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The crude material was purified by preparative plate chromatography (Hex/EtOAc 4:1) to afford both isomers (A, 86mg, 45%) (B, 90mg, 48%).

Step B

To isomer B (90mg) from above in dioxane (5ml) was added 6M H₂SO₄ (5ml). The reaction was heated to 80°C over the weekend. 2M NaOH added to basify the reaction and extracted with ether. Ether layer washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The residue was stirred in 4N HCl in dioxane for 30min, concentrated *in vacuo* and recrystallized in EtOH/ether to afford 55mg of product (98%).

10 Step C

5

15

20

25

Isomer A (86mg) was reacted following the procedure set forth in Step B above to give the amine salt.

PREPARATIVE EXAMPLE 156.12

The above nitro compound was reduced following the Preparative Example 2, Step B.

PREPARATIVE EXAMPLE 156.13

To a solution of 1,2-phenylenediame (1.5g) in CH₂Cl₂ (30ml) at 0⁰C was added TEA (2.91ml), followed by dropwise addition of MeSO₂Cl (1.07ml). The mixture was allowed to warm to room temperature and stir overnight. 1M HCl added and the layers were separated. The aqueous layer was adjusted to pH=11 with solid NaOH, extracted with CH₂Cl₂. The basified aqueous layer was then neutralized using 3N HCl and extracted with CH₂Cl₂, dried with Na₂SO₄, filtered, and concentrated *in vacuo* to give 1.8g of product (71%).

PREPARATIVE EXAMPLE 156.14

The above compound was prepared using the procedure set forth in Preparative Example 156.13, but using PhSO₂CI.

5

PREPARATIVE EXAMPLE 156.15

The nitro compound was reduced following a similar procedure as in Preparative Example 2, Step B.

10

PREPARATIVE EXAMPLE 156.16

Step A

15

The known acid (410mg) above (*J.Med.Chem.* **1996**, 34,4654, the disclosure of which is incorporated herein by reference thereto.) was reacted following the procedure set forth in Preparative Example 2, Step A to yield 380mg of an oil (80%).

Step B

20

The amide (200mg) from above was reacted following the procedure set forth in Preparative Example 2, Step B to yield 170mg of an oil (100%).

PREPARATIVE EXAMPLE 156.17

To a solution of ketone (500mg) in EtOH/water (3:1, 4ml) at room temperature was added hydroxylamine hydrochloride (214mg) followed by NaOH to afford a heterogenous mixture. The reaction was not complete so another equivalent of hydroxylamine hydrochloride was added and refluxed overnight. The reaction was cooled to 0°C and treated with 3N HCl and extracted with CH₂Cl₂, washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo* to give 500mg of product (92%).

10 <u>Step B</u>

5

15

20

25

To a solution of oxime (300mg) in THF (5ml) at 0°C was added LiAlH₄ (266mg) portionwise. The heterogenous solution was stirred at room temperature for 14hr and then refluxed for 8hr. The solution was cooled to 0°C and water, 2M NaOH, water and ether were added to the reaction. The mixture was filtered through a celite pad. The filtrate was treated with 3N HCl. The aqueous layer was cooled to 0°C, basified with NaOH pellets and extracted with ether. The ether layer was dried over MgSO₄, filtered, and concentrated *in vacuo* to afford the product (143mg, 69%).

PREPARATIVE EXAMPLE 156.18

$$H_3CO$$
 CO_2H $Step A$ H_3CO Me OMe $Step B$

Step A

Methoxyacetic acid (14 mL) in CH₂Cl₂ (120 mL) and cooled in an ice-water bath was treated with DMF (0.9 mL) and oxalyl chloride (21 mL). After stirring at RT overnight, the mixture was concentrated *in vacuo* and redissolved in CH₂Cl₂ (120 mL). N-methyl-N-methoxylamine (20 g) was added and the mixture stirred at RT

overnight. Filtration and concentration in vacuo afforded the desired amide (21 g, 89%).

Step B

5

10

15

20

To a solution of the above amide (260mg) in THF (5ml) at -78 °C was added a solution of 2-thienyllithium (1M in THF, 2.15ml). The solution was stirred for 2hr at -78 °C and warmed to -20 °C for an additional 2hr. The reaction was quenched with saturated ammonium chloride and extracted with CH₂Cl₂, washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo* to give 250mg of product (82%).

Step C

The ketone from above (250mg) was reacted via the procedure set forth in Preparative Example 156.17 Steps A and B to yield 176 mg of the amine (79%).

PREPARATIVE EXAMPLE 156.19

Step A

To a solution of 3-chlorothiophene (1.16ml) in ether (20ml) at $-10\,^{\circ}$ C was added n-BuLi (2.5M in hexane, 5ml). After solution was stirred at $-10\,^{\circ}$ C for 20min, propionaldehyde (0.82ml) in ether (20ml) was added dropwise and let warm to room temperature slowly. The reaction was quenched with saturated ammonium chloride and extracted with CH₂Cl₂, washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo* to give 1.37g of product (62%).

25 <u>Step B</u>

The alcohol from Step A above was reacted via the procedures set forth in Preparative Example 75.75, Steps B and C to give the amine.

PREPARATIVE EXAMPLE 156.20

Step A

5

10

15

To a solution of magnesium metal (360mg) in THF (15ml) at 0 °C was added 2-bromothiophene (1.45ml) in THF (10ml) dropwise over 20min. The solution was warmed to room temperature for 3hr, recooled to 0 °C whereupon a solution of cyclopropylacetonitrile (1g) in ether (30ml) was added dropwise via a syringe and let warm to room temperature and stir overnight. 3M HCl was added and washed with CH₂Cl₂. The aqueous layer was basified with NaOH pellets and extracted with ether, dried with Na₂SO₄, filtered, and concentrated *in vacuo* to give 625mg of product (68%).

Step B

The ketone was reacted via the procedure set forth in Preparative Example 156.17 Step A to give the oxime.

Step C

The oxime from above was reacted via the procedure set forth in Preparative Example 156.17 Step B to give the amine.

20

25

PREPARATIVE EXAMPLE 156.21

Step A

To a solution of $CH_3ONHCH_3.HCI$ (780mg) and acid chloride (1g) in CH_2Cl_2 at 0 °C was added dry pyridine (1.35ml) to afford a heterogenous mixture. The solution was warmed to room temperature and stirred overnight. 1M HCl was added to the

reaction and the organic layer was separated, washed with brine, dried with Na₂SO₄, filtered, and concentrated *in vacuo* to give 1g of product (85%).

Step B

5

10

15

To a solution of Etl (614ul) in ether (5ml) at –78°C was added t-BuLi (1.7M in pentane, 9ml) dropwise. The mixture was warmed to room temperature for 1hr, cooled to –78°C where the amide (1g) from Step A in THF (4ml) was added and allowed to warm to 0 °C for 2hr. 1M HCl was added to the reaction and extracted with CH₂Cl₂, washed with brine, dried with Na₂SO₄, filtered, and concentrated *in vacuo* to give 500mg of product (63%).

Step C

To a solution of ketone (800mg) in THF/water (10:1, 20ml) at 0 °C was added sodium borohydride (363mg) portionwise. The solution was stirred for 2hr at 0 °C. The mixture was concentrated *in vacuo*, the residue was dissolved in CH₂Cl₂, washed with 1N NaOH and brine, dried with Na₂SO₄, filtered, and concentrated *in vacuo* to give 560mg of product (69%).

Step D

The alcohol from above was reacted via the procedures set forth in Preparative Example 75.75, Steps B and C to give the amine (176mg, 59%).

PREPARATIVE EXAMPLE 156.22

Cyclopropylacetonitrile (12 mmol) in Et₂O (50 mL) at 0°C was treated with PhMgBr (14 mmol) and the mixture was stirred for 2 hrs at 0°C, then at RT overnight. Hydrochloric acid (3 *M*) was added, and after stirring for an additional 12 hrs, the mixture was extracted with CH₂Cl₂, washed with brine, dried over Na₂SO₄, filtered and concentrated *in vacuo* to give the desired ketone (1.34 g, 70%).

Step B

5

15

Following the procedures set forth in Preparative Example 156.20 Steps B and C, the amine was prepared.

PREPARATIVE EXAMPLE 156.23

The above amine was prepared using the procedures set forth in WO 98/11064 (the disclosure of which is incorporated herein by refernce thereto.

PREPARATIVE EXAMPLE 157

Step A

By taking the known carboxylic acid (*J. Med. Chem.* **1996**, *39*, 4654-4666, the disclosure of wnich is incorporated herein by reference thereto) and subjecting it to the conditions outlined in Preparative Example 112, the product can be prepared.

20

Step B

Following a similar procedure used in Preparative Example 2, Step A, except using dimethylamine and the compound from Step A above, the product can be prepared.

5

Step C

Following a similar procedure used in Preparative Example 2, Step B, except using the compound from Step B above, the product can be prepared.

10

PREPARATIVE EXAMPLE 158

Following a similar procedure used in Preparative Example 157, Steps A-C, except using trifluoromethylsulfonylchloride in Step A above, the product can be prepared.

15

PREPARATIVE EXAMPLE 500.1

Step A

By using the nitro-amide from Preparative Example 13.3, Step A, the amidine structure can be prepared following a similar procedure to that in *Tetrahedron Lett.*, **2000**, 41 (11), 1677-1680 (the disclosure of which is incorporated herein by refernce thereto).

Step B

25

20

By using the product from Step A and the procedure set forth in Preparative Example 2, Step B, one could obtain the desired amine-amidine.

ALTERNATE PREPARATIVE EXAMPLE 500.2 Step A NO2 Step C NH2

Step A

By treating the nitro-amide from Preparative Example 13.3, Step B with POCl₃ and subsequently MeNH₂, according to procedures known in the art, one would obtain the desired compound.

Step B

By treating the product from Step A according to the procedure set forth in Preparative Example 13.3, Step E, one could obtain the desired compound.

ÓН

Step C

By using the product from Step B and the procedure set forth in Preparative Example 2 Step B, one would obtain the desired compound.

15

5

By following a similar procedure as that described in *Zh. Obshch. Khim.*, 27, **1957**, 754, 757 (the disclosure of which is incorporated herein by reference thereto), but instead using 2,4-dichlorophenol and dimethylphosphinic chloride, one would obtain the desired compound.

Step B

5

10

15

20

25

By following a similar procedure as that described in *J. Organomet. Chem.*; 317, **1986**, 11-22 (the disclosure of which is incorporated herein by reference thereto), one would obtain the desired compound.

Step C

By following a similar procedure as that described in *J. Amer. Chem. Soc.*, 77, **1955**, 6221 (the disclosure of which is incorporated herein by reference thereto), one would obtain the desired compound.

Step D

By following a similar procedure as that described in *J. Med. Chem.*, 27, **1984**, 654-659 (the disclosure of which is incorporated herein by reference thereto), one would obtain the desired compound.

ALTERNATE PREPARATIVE EXAMPLE 500.4

Step A

By following a similar procedure as that described in *Phosphorous, Sulfur Silicon Relat. Elem.*; EN; 61, 12, **1991**, 119-129 (the disclosure of which is

incorporated herein by reference thereto), but instead using 4-chlorophenol, one would obtain the desired compound.

Step B

By using a similar procedure as that in *Phosphorous, Sulfur Silicon Relat. Elem.*; EN; 61, 12, **1991**, 119-129 (the disclosure of which is incorporated herein by reference thereto), but instead using MeMgBr, the desired compound could be prepared.

10 <u>Step C</u>

By following a similar procedure as that described in *J. Amer. Chem. Soc.*, 77, **1955**, 6221 (the disclosure of which is incorporated herein by reference thereto), one would obtain the desired compound.

15 <u>Step D</u>

By following a similar procedure as that described in *J.Med. Chem.*, 27, **1984**, 654-659 (the disclosure of which is incorporated herein by reference thereto), one would obtain the desired compound.

20

PREPARATIVE EXAMPLE 500.5

$$H \longrightarrow NH_2$$

By following a similar procedure as that set forth in *J. Org. Chem.* **1998**, 63, 2824-2828 (the disclosure of which is incorporated herein by reference thereto), but using CH₃CCMgBr, one could obtain the desired compound.

5

10

15

By following the procedure set forth in Preparative Example 13.1, Step B using 3-methoxythiophene, one could obtain the desired product.

Step B

By using the product from step A and following the procedure set forth in Preparative Example 13.19, Step E, the desired compound could be obtained.

Step C

By using the product from Step B and following the procedure set forth in Preparative Example 13.29, Step D, one could obtain the desired compound.

Step D

By using the product from Step C and following the procedure set forth in Preparative Example 13.3, Step B, the desired compound could be obtained.

Step E

By treating the product from Step D with n-BuLi at -78°C in THF and quenching the resulting anion with CO₂ according to standard literature procedure, one could obtain the desired compound following aqueous acid work up.

Step F

By using the product from Step E and the procedure set forth in Prepartive Example 13.19, Step C, one could obtain the desired compound.

10

5

Step G

By using the product from step F and following the procedure set forth in Preparative Example 13.19, Step E, the desired compound could be obtained.

15 <u>Step H</u>

By using the product from Step G and following the procedure set forth in Preparative Example 2, Step B, the desired compound could be obtained.

PREPARATIVE EXAMPLE 500.7

Step A

5

10

15

If one were to use a similar procedure to that used in Preparative Example 13.3 Step B, except using the hydroxy acid from *Bioorg. Med. Chem. Lett.* 6(9), 1996, 1043 (the disclosure of which is incorporated herein by reference thereto), one would obtain the desired methoxy compound.

Step B

If one were to use a similar procedure to that used in Preparative Example 13.19 Step B, except using the product from Step A above, one would obtain the desired compound.

Step C

If one were to use a similar procedure to that used in *Synth. Commun.* 1980, 10, p. 107 (the disclosure of which is incorporated herein by reference thereto),

except using the product from Step B above and t-butanol, one would obtain the desired compound.

Step D

5

If one were to use a similar procedure to that used in *Synthesis*, 1986, 1031 (the disclosure of which is incorporated herein by reference thereto), except using the product from Step C above, one would obtain the desired sulfonamide compound.

Step E

If one were to use a similar procedure to that used in Preparative Example 13.19 Step E, except using the product from Step D above, one would obtain the desired compound.

PREPARATIVE EXAMPLE 500.8

Step A

If one were to treat the product from Step C of Example 1125 with BuLi (2.2 eq.) in THF followed by quenching of the reaction mixture with N,N,-dimethylsulfamoyl chloride (1.1 eq.) then one would obtain

Step B

If one were to use the product of Step A above and follow Step E of Preparative Example 500.7, then one would obtain the title compound.

15

PREPARATIVE EXAMPLE 500.9

Step A

5

10

15

To a solution of 3-methoxythiophene (3 g) in dichloromethane (175 mL) at -78° C was added chlorosulfonic acid (8.5 mL) dropwise. The mixture was stirred for 15 min at -78° C and 1.5 h at room temp. Afterwards, the mixture was poured carefully into crushed ice, and extracted with dichloromethane. The extracts were washed with brine, dried over magnesium sulfate, filtered through a 1-in silica gel pad. The filtrate was concentrated in vacuo to give the desired compound (4.2 g).

Step B

The product from Step A above (4.5 g) was dissolved in dichloromethane (140 mL) and added with triethylamine (8.8 mL) followed by diethyl amine in THF (2M, 21 mL). The resulting mixture was stirred at room temperature overnight. The mixture was washed with brine and saturated bicarbonate (aq) and brine again, dried over

sodium sulfate, filtered through a 1-in silica gel pad. The filtrate was concentrated in vacuo to give the desired compound (4.4 g).

Step C

The product from Step B above (4.3 g) was dissolved in dichloromethane (125 mL) and cooled in a -78°C bath. A solution of boron tribromide (1.0 M in dichloromethane, 24.3 mL) was added. The mixture was stirred for 4 h while the temperature was increased slowly from -78°C to 10°C. H₂O was added, the two layers were separated, and the aqueous layer was extracted with dichloro- methane.

The combined organic layer and extracts were wahed with brine, dried over magnesium sulfate, filtered, and concentrated in vacuo to give 3.96 g of the desired hydroxy-compound.

Step D

10

The product from step C above (3.96 g) was dissolved in 125 mL of dichloromethane, and added with potassium carbonate (6.6 g) followed by bromine (2 mL). The mixture was stirred for 5 h at room temperature, quenched with 100 mL of H₂O. The aqueous mixture was addjusted to pH ~ 5 using a 0.5N hydrogen chloride aqueous solution, and extracted with dichloromethane. The extracts were washed with brine, dried over sodium sulfate, and filtered through a celite pad. The filtrate was concentrated in vacuo to afford 4.2 g of the desired bromo-compound.

Step E

25

30

The product from Step D (4.2 g) was dissolved in 100 mL of acetone and added with potassium carbonate (10 g) followed by iodomethane (9 mL). The mixture was heated to reflux and continued for 3.5 h. After cooled to room temperature, the mixture was filtered through a Celite pad. The filtrate was concentrated in vacuo to a dark brown residue, which was purified by flash column chromatography eluting with dichloromethane-hexanes (1:1, v/v) to give 2.7 g of the desired product.

Step F

The product from step E (2.7 g) was converted to the desired imine compound (3 g), following the similar procedure to that of Preparative Example 13.19 step D.

Step G

5

10

15

20

25

The imine product from step F (3 g) was dissolved in 80 mL of dichloromethane and cooled in a -78° C bath. A solution of boron tribromide (1.0 M in dichloromethane, 9.2 mL) was added dropwise. The mixture was stirred for 4.25 h from -78°C to 5°C. H₂O (50 mL) was added, and the layers were separated. The aqueous layer was extracted with dichloromethane. The organic layer and extracts were combined, washed with brine, and concentrated to an oily residue. The residue was dissolved in 80 mL of methanol, stirred with sodium acetate (1.5 g) and hydroxyamine hydrochloride (0.95 g) at room temperature for 2 h. The mixture was poured into an aqueous mixture of sodium hydroxide (1.0 M aq, 50 mL) and ether (100 mL). The two layers were separated. The aqueous layer was washed with ether three times. The combined ether washings were re-extracted with H₂O once. The aqueous layers were combined, washed once with dichloromethane, adjusted to pH ~ 6 using

3.0 M and 0.5 M hydrogen chloride aqueous solutions, and extracted with dichloromethane. The organic extracts were combined, washed with brine, dried over sodium sulfate, and concentrated in vacuo to give 1.2 g of desired amine compound.

PREPARATIVE EXAMPLE 600

Step A

Following the procedure set forth in Preparative Example 13.19 Step D, the imine was prepared from the known bromoester (1.0g) to yield 1.1g (79%) as a yellow solid.

Step B

The product of Step A (0.6g) was reacted following the procedure set forth in Preparative Example 13.19 Step E to give the amine product 0.19g (64%).

5 Step C

The product of Step B (1.0g) was reacted following the procedure set forth in Preparative Example 13.19 Step B to give the acid as yellow solid 0.9g (94%).

Step D

10

15

20

25

30

The product of Step C (0.35g) was reacted following the procedure set forth in Preparative Example 13.19 Step E to give the amino acid as yellow solid 0.167g (93%).

PREPARATIVE EXAMPLE 601

Step A

To a solution of 2-methyl furan (1.72g) in ether was added BuLi (8.38mL) at -78° C and stirred at room temperature for half an hour. The reaction mixture again cooled to -78° C and quenched with cyclopropyl amide 1 and stirred for two hours at -78° C and slowly warmed to room temperature. The reaction mixture stirred for three hours at room temperature and quenched with the addition of saturated ammonium chloride solution. The mixture was taken to a separatory funnel, washed with water, brine and dried over anhydrous sodium sulfate. Filtration and removal of solvent afforded the crude ketone, which was purified by using column chromatography to afford the ketone 3.0g (87%) as a pale yellow oil.

Step B

To a solution of ketone (1.0g) from Step A above in THF (5.0mL) at 0°C was added *R*-methyl oxazoborolidine (1.2Ml, 1M in toluene) dropwise followed by addition of a solution of borane complexed with dimethyl sulfide (1.85mL, 2M in THF). The reaction mixture was stirred for 30minutes at 0°C and than at room temperature for one hour. The reaction mixture was cooled to 0°C and MeOH was added carefully.

The mixture was stirred for 20 minutes and was concentrated under reduced pressure. The residue was extracted with ether, washed with water, 1M HCI (10mL), saturated sodium bicarbonate (10.0mL) water and brine. The organic layer was dried over anhydrous sodium sulfate, filtered and removal of solvent afforded the crude alcohol which was purified by silica gel chromatography to afford the pure alcohol 0.91g (91%) as yellow oil.

PREPARATIVE EXAMPLE 601.A

$$\begin{array}{c|c} & & \\ & &$$

10 <u>Step A</u>

5

If one were to follow the procedure set forth in Preparative Example 601, but using the cyclopentylamide instead of the cyclopropylamide (prepared according to standard procedures), then one would obtain the desired alcohol.

15 <u>Step B</u>

If one were to follow the procedure set forth in Preparative Example 13.25, but instead using the alcohol from Step A above, then one would obtain the title amine.

PREPARATIVE EXAMPLE 601.B

Step A

If one were to follow the procedure set forth in Preparative Example 601.A, but using 4-isopropylfuran instead of 5-methylfuran, then one would obtain the desired alcohol.

Step B

5

10

15

20

25

If one were to follow the procedure set forth in Preparative Example 13.25, but instead using the alcohol from Step A above, then one would obtain the title amine.

PREPARATIVE EXAMPLE 602

Step A

An equimolar mixture of 2-methylfuran (1.0g) and anhydride (2.6g) was mixed with SnCl₄ (0.05mL) and heated at 100⁰C for 3 hours. After cooling the reaction mixture, water (10mL) was added, followed by saturated sodium carbonate solution until it becomes alkaline. The reaction mixture was extracted with ether several times and the combined ether layer was washed with water, brine and dried over anhydrous sodium sulfate. Filtration and removal of solvent afforded the crude ketone, which was purified by using silica gel chromatography to afford the ketone 0.9g (43%) as a yellow oil.

Step B

The title alcohol was obtained following a similar procedure set forth in the Preparative Example 601.

PREPARATIVE EXAMPLE 603

To a solution of 5-methyl furan-2-aldehyde (1.0g) and 3-bromo-3,3-difluoropropene (2.24g) in DMF (30mL) was added indium powder (1.66g) and lithium iodide (50.0mg). The reaction mixture was stirred over night, diluted with water and

extracted with ether. The ether layer was washed with water, brine and purified by silica gel chromatography to afford the pure alcohol 2.8g (92%).

PREPARATIVE EXAMPLES 603A AND 603D

If one were to follow the procedure of Preparative Example 64, using the aldehydes, amino alcohols and organolithiums in the Table below, then the optically pure amine Products in the Table below would be obtained.

Prep. Ex.	Aldehyde	Amino Alcohol	Organolithium	Product
603A	(see Preparative Example 1004B)	H ₂ N OH	iPrLi	H ₂ N O
603B		H ₂ N OH	tBuLi	H ₂ N O
603C	H Z	H ₂ N OH	tBuLi	H ₂ N O

603D	H	H ₂ N OH	tBuLi	H ₂ N O
	(see Preparative Example 1004B)			

PREPARATIVE EXAMPLES 604-611

Following a similar procedure set forth in Preparative Examples 13.25 or 601 the following Alcohols were prepared.

Prep Ex	Furan	Electrophile	Alcohol	Yield
604		СНО	но	86%
605		COOEt	HOO	69%
606		O OMe	но	84%

607	O OMe	но	82%
608	COOEt	HO	60%
609	COOEt	но	65%
610	F F N OMe	HO F	82%
611	OHC_CF ₃	HO CF ₃	89%

PREPARATIVE EXAMPLES 620-631

Following a similar procedure to that set forth in Preparative Example 13.25 the following Amines were prepared from the corresponding Alcohols.

	T		
Prep Ex	ALCOHOL	AMINE	% YIELD
620	HO CF ₃	H ₂ N O	28
621	но	H_2N	58
622	но	H ₂ N O	69
623	но	H ₂ N O	81
624	F O	H ₂ N O	82
625	но	H ₂ N O	45

626	но	H_2N	57
627	но	H_2N	58
628	HO O	F H ₂ N O	54
629	H OH	H ₂ N O	53
630	HOO	H ₂ N O	50
631	HO O	F F O	82%

Step A

5

10

15

20

Oxalyl chloride (3 mL, 34.27 mmol) was added dropwise to a mixture of 2-methoxy-6-(trifluoromethyl)benzoic acid (1.5 g, 6.81 mmol) (prepared according to known method, see: EP0897904B1), *N,N*-dimethylformamide (0.3 mL), and dichloromethane (40 mL) with stirring at rt. The reaction mixture was stirred overnight. Evaporation of solvent and excess oxalyl chloride and drying under vacuum afforded 2-methoxy-6-(trifluoromethyl)benzoyl chloride as a solid, which was used without purification.

Step B

A solution of 2-methoxy-6-(trifluoromethyl)benzoyl chloride (ca. 6.81 mmol) from Step A above in dichloromethane (20 mL) was added dropwise to a mixture of 4-(dimethylamino)pyridine (42 mg, 0.34 mmol), triethylamine (2.8 mL, 20.09 mmol), and 2 M dimethylamine solution in tetrahydrofuran (7 mL, 14 mmol), and dichloromethane (30 mL) with stirring at rt. The reaction mixture was stirred overnight. A mixture of dichloromethane and water was added. The organic phase was separated, washed with 1N HCl solution, water, and saturated sodium bicarbonate solution and concentrated. The residue was purified by column chromatography (ethyl acetate:hexanes, 3:1 v/v) to give the product as a white solid (1.24 g, 74% over two steps).

Step C

A mixture of the amide from Step B above (1.8 g, 7.28 mmol), carbon tetrachloride (25 mL), and iron powder (305 mg, 5.46 mmol) was cooled to 0 °C. Bromine (0.94 mL, 18.34 mmol) was added dropwise with stirring. After addition, the mixture was stirred at rt for 1 h and at 50 °C for 3 h. The mixture was cooled to rt, diluted with dichloromethane, and slowly poured to a cold 10% NaHSO₃ solution. After stirring at rt for 0.5 h, the organic layer was separated and concentrated to give the product as a white solid (2.26 g, 95%).

10

15

5

Step D

Concentrated sulfuric acid (10 mL) was added dropwise to a flask charged with the bromide from Step C above (600 mg, 1.84 mmol) at 0 °C with stirring. A mixture of nitric acid (0.2 mL, 4.76 mmol) and concentrated sulfuric acid (0.3 mL) was then added dropwise. After addition, the mixture was stirred at rt for 3 h. The mixture was added to ice-water, neutralized with 15% NaOH solution to pH 7, and extracted with dichloromethane. The organic layer was concentrated to give the product as a white solid (621 mg, 91%). mp 92 °C, *m*/e 371 (MH⁺).

20 Step E

A solution of the compound from Step D above (1.2 g, 3.23 mmol) in dichloromethane (50 mL) was cooled to –75 °C. 1 M BBr₃ solution in dichloromethane (7.5 mL, 7.5 mmol) was added dropwise with stirring. The mixture was stirred at –75 °C for 2 h. The mixture was added to ice-water. After stirring at rt for 0.5 h, the mixture was extracted with dichloromethane. The organic was concentrated and the residue was purified by column chromatography (dichloromethane-methanol, 9:1 v/v) to give the product as a yellow solid (1.05 g, 91%). *m/e* 357 (MH⁺).

Step F

30

25

A mixture of the compound from Step E above (1.08 g, 3.02 mmol), methanol (30 mL), and 10% Pd-C (250 mg) was subjected to hydrogenation at 50 psi at rt for 6 h. The mixture was filtered through a layer of Celite. The filtrate was concentrated to give the title compound as a pale yellow solid (930 mg, 96%). mp 132 °C, *m/e* 249.

Step A Step B HO Ph Step D Step C

Step A

5

10

15

20

Ph

To a cooled (-70°C) etherial (45 mL dry) solution of 3-bromothiophene (3.8 mL) was added BuLi (30 mL of 1.6M in hexane) dropwise, and the mixture was stirred at -70°C for 20 min. Acetophenone (4.6 mL) in ether (6 mL) was added dropwise with stirring at -70°C. After 3 hrs, the mixture was warmed to RT and sat. NH₄Cl (aq) was added and the mixture was extracted with ether. The organic phase was dried (Na₂SO₄) and concentrated in vacuo to give the title compound which was used in Step B without further purification.

Ph

Step B

The crude product from Step A above was stirred with oxalic acid (0.375 g) at 70°C under reduced pressure for 3 hr, then cooled to RT and extracted with ether. The organic phase was dried (Na₂SO₄) and concentrated in vacuo to give the product as a pale yellow liquid (5.7 g, 78% for Steps A-B).

Step C

To the product from Step B above (4.2 g) diluted with dichloromethane (30 mL) and containing triethylsilane (6 mL) was added TFA (3 mL) in dichloromethane (7.5 mL). After stirring at RT for 10 min, the mixture was concentrated in vacuo to give the product as a colorless liquid (4.61 g, 80%).

Step D

5

10

15

20

25

To an etherial (3.5 mL dry) solution of the thiophene product (1.5 g) from Step C above was added BuLi (3.2 mL of 2.5M), and the mixture was heated at reflux for 15 min, cooled to RT, and DMF (0.8 mL) in ether (3.5 mL) was added dropwise. After stirring for 30 min, sat. NH₄Cl (aq) was added and the mixture was extracted with ether. The organic phase was dried (Na₂SO₄) and concentrated in vacuo to give the title compound (1.71 g, 98%).

Preparative Example 1003

Step A

The aldehyde (0.50 g) was combined with ethylene glycol (1 mL), benzene (40 mL) and pTSA monohydrate (30 mg) and stirred at reflux for 20 hr. Cool to room temperature, add EtOAc and sat. NaHCO₃ (aq) solution, separate the organic phase, concentrate in vacuo, and purify by silica gel chromatography (EtOAc-Hex, 1:4) to give a colorless liquid (60 mg)

Step B

The product from Step A above (0.607 g) was stirred at 45°C overnight with 1N NaOH (aq), then cooled to room temperature, acidified with 3N HCl and extracted with EtOAc. Washing with brine and concentration in vacuo gave a solid (5.0 g).

Step C

Following a similar procedure as that used in Preparative Example 1, except using the product from Step B above and dimethylamine in THF (2M), the product was obtained (1.21g crude).

Step D

The product from Step C above was dissolved in THF and stirred with 0.3N HCl (aq) and stirred at RT for 4 hr. Concentration in vacuo gave a pale yellow oil (1.1 g, 67%).

5

Preparative Example 1004

Step A

To a cooled (-78°C) solution of methoxybenzofuran-2-carboxylic acid (1 g) was added DIBAL (30 mL, 1M in THF). After stirring for 20 min, the mixture was warmed to RT and stirred for 4 hr, then poured into sat. NH₄Cl (aq) (35 mL). After stirring at RT for 20 min, 6M HCl (aq) was added and the mixture was extracted with EtOAc, the organic phase dried and then concentrated in vacuo. Purification by silica gel chromatography (EtOAc-hexane, 3:7) afforded the alcohol as a solid (0.4 g, 97%).

15

20

10

Step B

A mixture of the product from Step A above (0.9 g), EtOAc (50 mL) and MnO2 (5.2 g) was stirred at RT for 22 h, then filtered and concentrated in vacuo. The solid was redissolved in EtOAc (50 mL), MnO2 (5.2 g) was added and the mixture was stirred for 4 additional hrs. Filtration, concentration and silica gel purification (EtOAc-Hexane, 1:3) gave the title compound as a solid (0.60 g, 67%).

PREPARATIVE EXAMPLE 1004A

Step A

To a stirred solution of potassium t-butoxide (2.5g) in HMPA (20ml) was added 2-nitropropane (2ml) dropwise. After 5min, a solution of methyl-5-nitro-2-furoate (3.2g) in HMPA (8ml) was added to the mixture and stirred for 16hr. Water was added and the aqueous mixture was extracted with EtOAc. The EtOAc layer was washed with water, dried with MgSO₄, filtered and concentrated *in vacuo*. The crude material was purified by flash column chromatography (Hex/EtOAc, 6:1) to yield 3.6g of product (90%).

10 Step B

To a solution of product from Step A (3.6g) in toluene (16ml) was added tributyltin hydride (5.4ml) followed by AIBN (555mg). The mixture was heated to 85°C for 3.5hr. After cooling, the mixture was separated by flash column chromatography (Hex/EtOAc, 7:1) to afford 2.06g of product (73%).

15

20

25

30

5

Step C

To a solution of product from Step B (2.05g) in THF (60ml) at 0°C was added a solution of LAH (1M in ether, 12.8ml). The reaction was stirred at room temperature for 30min. Water and 1M NaOH was added until a precipitate formed, diluted with EtOAc, stirred for 30min and then filtered through a celite pad. The organic filtrate was concentrated *in vacuo* to give 1.56g of product (93%).

Step D

To a solution of product from Step C (2.15g) in CH₂Cl₂ (100ml) was added Dess-Martin oxidant (7.26g) in CH₂Cl₂ (45ml) and stirred for 30min. The mixture was diluted with ether (200ml). The organic layer was washed with 1N NaOH, water and brine, dried with MgSO₄, filtered and concentrated *in vacuo* to give oil and solid. The material was extracted with ether and filtered. Some solid crystallized out from the filtrate, filtered again, and the filtrate was concentrated *in vacuo* to give 2.19g of product.

PREPARATIVE EXAMPLE 1004B

Step A

To a suspension of 5-bromo-2-furoic acid (15g) in CH₂Cl₂ (275ml) at room temperature was added oxalyl chloride (6.9ml) followed by a catalytic amount of N,N'-dimethylformamide ((0.3ml). The mixture was stirred for 1hr, whereupon, EtOH (20ml) and TEA (22ml) were added and then let stir overnight. The mixture was concentrated *in vacuo* and extracted with hexanes and hexanes/ CH₂Cl₂. The extracts were concentrated *in vacuo* to give an oil (17.2g, 93%).

10

5

Step B

The product from Step A (17.2g), aluminum trichloride (19.52g) and carbon disulfide (150ml) were combined in a flask. A solution of n-octadecyl bromide (24.4g) in carbondisulfide (50ml) was added dropwise over 45min. The reaction was stirred for 2.5hr, whereupon, 300ml of crushed ice and water were added. The layers were separated and the organic layer was washed with saturated sodium bicarbonate, water, and brine. The organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The crude material was purified by flash column chromatography (hexanes/ CH₂Cl₂, 3:1) to yield 7.91g of product (37%).

20

25

30

15

Step C

To the product from step B (7.9g) in THF (140ml) at –10°C was added a solution of LAH (1M in THF, 28.5ml). The solution was stirred for 2.5hrs at 15°C. Water and 1M NaOH were added carefully to the mixture, followed by EtOAc and let stir for 1.5hr. The reaction was filtered through a silica pad and the filtrate was concentrated *in vacuo* to yield 6.48g of crude product (100%).

Step D

The product from Step C (6.32g) was dissolved in THF (140ml) and cooled to -78°C. A solution of t-BuLi (2.5M in hexanes, 22ml) was added dropwise and let stir

for 15min. An excess of water (70ml) was then added and let the reaction stir another hour. CH₂Cl₂ (300ml) and brine (50ml) were added and the layers were separated. The organic layer was dried with Na₂SO₄ and concentrated *in vacuo* to give 5.33g of crude product.

5

10

Step E

To a solution of the product from Step D (5.33g) in CH₂Cl₂ (100ml) was added a solution of Dess-Martin periodinane in CH₂Cl₂ (15wt%, 12.6g). The mixture was stirred for 1.5hr and then diluted with ether (400ml) and washed with 1N NaOH, water and brine. The organic layer was dried with Na₂SO₄ and filtered through a magnesium sulfate/silica pad. The filtrate was concentrated *in vacuo* and purified via flash column chromatography (hex/EtOAc, 50:1, 25:1) to yield 3.06g of an oil (74%).

Preparative Example 1005

Following a similar procedure as that described in Preparative Example 1004, except using 5-chlorobenzofuran-2-carboxylic acid (1.5 g), the title compound was obtained (solid, 0.31 g, 24%).

20

25

15

Preparative Example 1006

Step A

The sulfonyl chloride from Preparative Example 13.29 Step A (1.5 g) was stirred with AlCl3 and benzene for 15 min at 20 $^{\circ}$ C. Treatment with NaOH, extraction with Et₂O, concentration in vacuo, and purification by column chromatography (silica, hexane-EtOAc, 5:2) gave the phenylsulfone (1.5g, 84%, MH⁺ = 255).

Step B

Following similar procedures as those used in Preparative Example 13.29 Steps C-G, except using the sulfone from Step A above, the title compound was prepared (0.04 g, 27%, MH^{+} = 256).

5

10

15

Preparative Example 1030

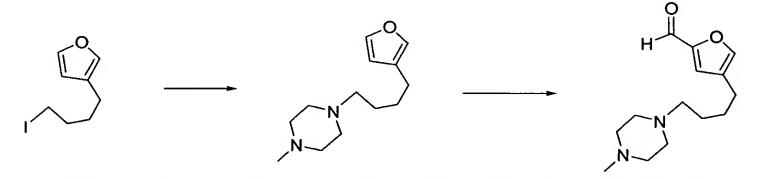
Step A

The product of Preparative Example 34.18 Step B (2 g, 8 mmol) was stirred with morpholine (0.9 mL, 10.29 mmol) and K_2CO_3 (2.2 g, 15.9 mmol) in 50 mL of acetone at RT to obtain the morpholinobutylfuran derivative (1.22 g, 73%).

Step B

Following a similar procedure as that in Preparative Example 34.18 Step D, but using the product (1.2 g) from Step A above, the title aldehyde was prepared (0.9 g, 66%, 1:0.7 regioisomeric mixture).

Preparative Example 1030-A



20

Following a similar procedure as that in Preparative Example 1030 Steps A-B, but using N-methylpiperazine instead of morpholine, the title aldehyde could be prepared.

Preparative Example 1030-B

Following a similar procedure as in Preparative Example 1030 Steps A-B, but using N,N-dimethylamine instead of morpholine, the title aldehyde could be prepared.

Preparative Example 1031

5

10

15

20

A solution of 5-bromobenzofuran (950 mg, 4.82 mmol) in anhydrous ether (12 mL) was cooled to -78 °C. 1.7 M *tert*-BuLi solution in pentane (6 ml, 10.2 mmol) was added dropwise under argon. After addition, the mixture was stirred at -78 °C for 20 min, followed by addition of a mixture of DMF (0.8 mL) and ether (1 mL). The mixture was allowed to warm to rt and stirred for 0.5 h. Ethyl acetate was added. The mixture was poured to saturated ammonium chloride solution. The organic layer was separated and concentrated. The residue was purified by column chromatography (ethyl acetate-hexanes, 1:5 v/v) to give the title compound as a pale yellow solid (490 mg, 70%).

PREPARATIVE EXAMPLES 1040-1054

Following the procedure set forth in Preparative Example 64 but using the commercially available (or prepared) aldehyde, aminoalcohols, and organolithium reagents in the Table below, the optically pure amine products in the Table below were obtained.

Prep . Ex.	Aldehyde	Amino Alcohol	Organo- lithium	Product	1.Yield (%) 2. (M+1) ⁺
1040	H	H ₂ N OH	EtLi	H ₂ N O	1. 24% 2. 267
1041	H	H ₂ N OH	EtLi	H ₂ N O	1. 94% 2. 176 (m/e)
1042	O H Ph	H ₂ N OH	EtLi	H ₂ N S	1. 67% 2. 229 (M-16)
1043	H	H ₂ N OH	i-PrLi	H ₂ N O	1. 60% 2. 151 [M-16]
1044	H CON(Me) ₂	H ₂ N OH	EtLi	H ₂ N CON(Me) ₂	1. 74% 2. 194 (M-16)

1045	O H	H ₂ N OH	EtLi	H_2N	1. 33% 2. 165 [M-NH2] ⁺
1046	o H	H ₂ N OH	EtLi	H ₂ N	1. 31 2. 179 [M-NH2] ⁺
1047	H	H ₂ N OH	t-BuLi	H_2N O CI	1. 31% 2. 188
1048	H O	H ₂ N OH	t-BuLi	H ₂ N O	1. 10% 2. 154
1049	H	H ₂ N OH	EtLi	H ₂ N O	1. 73% 2. 137 [M-NH2] [†]
1051	H OFF	H ₂ N OH	t-BuLi	H ₂ N OF	1. 17%

1054			t-BuLi		1. 79%
	O I			\downarrow	2. 151
	H //- Q	H ₂ N OH		÷	(M-16)
				H ₂ N O	

PREPARATIVE EXAMPLES 1100-1126

Following the procedure set forth in Preparative Example 34 but using the commercially available aldehydes and Grignard/Organolithium reagents listed in the Table below, the amine products were obtained.

Prep. Ex.	Aldehyde	Òrgano- metallic Reagent	Product	1.Yield (%) 2. (M+1) ⁺
1100	H NMe ₂	t-BuLi	H ₂ N NMe ₂	1. 83% 2. 190 (M- 16)
1101	H	t-BuLi	H ₂ N	1. 46% 2. 204
1102	OMe	t-BuLi	OMe H ₂ N	1. 48% 2. 194
1103	H OMe	t-BuLi	H ₂ N OMe	1. 51% 2. 194

1104	H	t-BuLi	H ₂ N CI	1. 12% 2. 238
1105	O H OMe	t-BuLi	H ₂ N OMe	1. 39% 2. 234
1106	OMe	t-BuLi	H ₂ N OMe	1. 44% 2. 194 (m/e)
1107	H	t-BuLi	H ₂ N N	1. 57% 2. 150 (M- 16)
1108	O H OMe OMe	t-BuLi	H ₂ N OMe OMe	1. 31% 2. 224

1109	H	t-BuLi	H_2N	1. 11% 2. 224
1110	H	t-BuLi	H ₂ N	1. 57% 2. 224
1111	H	t-BuLi	H_2N	1. 21% 2. 224
1112	H	c-Pentyl-Li	H ₂ N	1. 58% 2. 190
1113	OCF ₃	t-BuLi	H_2N OCF ₃	1. 20% 2. 248

1114	O H CF ₃	t-BuLi	H ₂ N CF ₃	1. 24% 2. 232
1115	H	EtLi	H_2N O O	1. 32% 2. 177 (M- NH2)
1116	H	t-BuLi	H ₂ N O	1. 26% 2. 205 (M- NH2)
1117	H 0 - 2 - 2 -	t-BuLi	H ₂ N	1. 50% 2. 190 (M- NH2)
1118	H—F F	t-BuLi	H_2N F	1. 29% 2. 200

1119	H—CI CI	t-BuLi	H ₂ N———CI	1. 28% 2. 232
1120	H—————————————————————————————————————	t-BuLi	H_2N	1. 76% 2. 224
1121	H	t-BuLi	H ₂ N	1. 40% 2. 206
1122	н	t-BuLi	H_2N	1. 38% 2. 236
1123	H	t-BuLi	H ₂ N	1. 70% 2. 192

1124	H	t-BuLi	H ₂ N	1. 81% 2. 204
1125	H Br	t-BuLi	H ₂ N O Br	33%
1126	H Br	t-BuLi	H_2N O Br	50%

PREPARATIVE EXAMPLES 1200A-1203A

Following the procedure set forth in Preparative Example 13.29 but using the commercially available amines, the hydroxyaminothiophene products listed in the Table below were obtained.

Prep. Ex.	Amine	Product	1.Yield (%) 2. (M+1) [†]
1200A	H N N	O S NH ₂ OH NH ₂ NH ₂	 3% 342

1201A	O_N.H	N S NH ₂	1. 41% 2. 265
1202A	N, H	N S NH ₂	1. 17% 2. 237
1203A	, H	N S NH ₂	1. 1%

PREPARATIVE EXAMPLE 1300

The title compound from Preparative Example 13.32 (0.35 g) was treated with concentrated sulfuric acid (3 mL) for 6 hrs, then poured on ice, and the pH adjusted to 4 with NaOH. Extraction with EtOAc, and drying of the organic phase over Na₂SO₄ gave the title compound (159 mg, 64%, MH⁺ = 223).

PREPARATIVE EXAMPLE 1301

Step A

5

10

Following the procedure set forth in Preparative Example 605 but using the commercially available fluoroisopropylester, the alcohol product was obtained (1.2 g, 84%, M-OH = 155).

Step B

5

Following the procedure set forth in Preparative Example 625 but using the alcohol from Step A above, the amine product was obtained (350 mg, 35%, M-NH2 = 155).

PREPARATIVE EXAMPLE 1302

Step A

Following a similar procedure as that used in Preparative Example 13.29 Step B, except using the commercially available arylsulfonylchloride (0.15 g) and diethylamine (2.2 eq), the dimethylsulfonamide was obtained (0.12 g, 71%, MH⁺ = 323).

15 Step B

Following a similar procedure as that used in Preparative Example 13.29 Step C, except using the product from Step A above (0.12 g), the phenol was obtained (0.112 g, 98%).

Step C

Following a similar procedure as that used in Preparative Example 10.55 Step C, except using the product from Step B above (0.112 g), the title compound was obtained (0.1 g, 99%, MH^{+} = 245).

5

10

PREPARATIVE EXAMPLE 1303

Following a similar procedure as that used in Preparative Example 1302 Steps A-C, except using piperidine in Step A (0.078 g) instead of diethylamine, the title compound was obtained (0.070 g, 35%, MH^{+} = 257).

PREPARATIVE EXAMPLE 1304

Following a similar procedure as that used in Preparative Example 1302 Steps A-C, except using dimethylamine (2M in THF) in Step A instead of diethylamine, the title compound was obtained (1.92g, 72%, MH⁺ = 217).

PREPARATIVE EXAMPLE 1305

Step A

Following a similar procedure as that used in Preparative Example 1302 Step A, except using the phenethylamine indicated (4.99 g), the product was obtained (5.96 g, 86%, MH^{+} = 210).

Step B

The compound from Step A above (5.0 g) was added to 30 g of PPA at 150°C and the resulting mixture stirred for 20 min, before being poured on ice and extracted with dichloromethane. The organic phase was dried over MgSO4, concentrated in vacuo and purified by silica gel chromatography (EtOAc:MeOH, 95:5) to give the product (0.5 g, 9%).

Step C

Following a similar procedure as that used in Preparative Example 13.3 Step D, except using the compound from Step B above (0.14 g), the product was obtained (0.18 g, 87%, MH^{+} = 256).

5

Step D

Following a similar procedure as that used in Preparative Example 11 Step B, except using the compound from Step C above (0.18 g), the product was obtained (0.17 g).

10

Step E

Following a similar procedure as that used in Preparative Example 13.3 Step B, except using the compound from Step D above (0.17 g), the product was obtained (0.17 g, 95%, MH^{+} = 315).

15

Step F

Following a similar procedure as that used in Preparative Example 13.29 Step C, except using the product from Step E above (0.17 g), the nitrophenol was obtained (0.165 g, 99%, MH^+ = 303).

20

Step G

Following a similar procedure as that used in Preparative Example 10.55 Step C, except using the product from Step F above (0.165 g), the title compound was obtained (0.128 g, 86%, MH^{+} = 193).

25

Step A

5

10

15

Following a similar procedure as that used in Preparative Example 11 Step B, except using the lactam (0.179 g), the title compound was obtained (0.25 g, 25%).

Step B

Following a similar procedure as that used in Preparative Example 13.29 Step C, except using the product from Step A above (0.055 g), the phenol was obtained (0.045 g, 99%).

Step C

Following a similar procedure as that used in Preparative Example 10.55 Step C, except using the product from Step B above (0.045 g), the title compound was obtained (0.022 g, 57%, MH^{+} = 179).

PREPARATIVE EXAMPLE 1307

Following a similar procedure as that used in Preparative Example 2, except using 3(*R*)-hydroxypyrrolidine HCl (1.36 g), the title compound was obtained (2.25 g, 89%).

$$\bigcap_{N} \bigcap_{OH} NH_2$$

Following a similar procedure as that used in Preparative Example 2, except using morpholine, the title compound was obtained (3.79 g).

5

10

15

PREPARATIVE EXAMPLE 1309

Step A

Following a similar procedure as that used in Preparative Example 13.29 Step B, except using the commercially available nitrophenylsulfonylchloride and diethylamine (2.2 eq), the dimethylsulfonamide was obtained (90%, MH⁺ = 231).

Step B

Following a similar procedure as that used in Preparative Example 10.55 Step C, except using the product from Step B above, the title compound was obtained (45%, MH⁺ = 201).

PREPARATIVE EXAMPLE 1310

Step A

Following a similar procedure as that used in Preparative Example 13.29 Step B, except using the commercially available nitrobenzoylchloride and the commercially available amine indicated, the benzamide was obtained (13%, $MH^{+} = 253$).

5

Step B

Following a similar procedure as that used in Preparative Example 10.55 Step C, except using the product from Step A above, the title compound was obtained (94%, MH⁺ = 223).

10

15

PREPARATIVE EXAMPLE 1311

Step A

To a benzene (20 mL) solution of methoxythiophenesulfonylchloride (1.5 g) was added AlCl₃ (2.0 g) at RT. After 15 min, the mixture was added to 0.1N HCl (aq) with stirring, then extracted with Et₂O. Washing the organic phase with bring, drying over MgSO₄, concentration in vacuo and purification by silica gel chromatography (Hexane:EtOAc, 5:2) gave the title compound (1.5 g, 84%).

20 <u>Step B</u>

Following a similar procedure as that used in Preparative Example 13.29 Steps C-G, except using the product from Step A above, the title compound was obtained (3%, MH⁺ = 380).

25

Step A

5

10

Following a similar procedure as that used in Preparative Example 1311 Step A, except using the commercially available sulfonylchloride, the diphenylsulfone was obtained (880 mg, 80%).

Step B

Following a similar procedure as that used in Preparative Example 11 Step B, except using the product from Step A above, the title compound was obtained (0.90 g, 97%).

Step C

Following a similar procedure as that used in Preparative Example 10.55 Step

C, except using the product from Step B above (0.16 g), the title compound was obtained (0.106 g, 95%).

Step A

5

10

15

Following a similar procedure as that used in Preparative Example 1311 Step A, except using the commercially available phenol (2 g), the nitroacid was obtained (~ 13 mmol).

Step B

Oxallyl chloride (3.5 mL) and two drops of DMF was added to the product from Step A above (~ 13 mmol) dissolved in dichloromethane (100 mL). After stirring at RT overnight, the mixture was concentrated in vacuo, diluted with dichloromethane (50 mL), cooled to 0°C. Dimethylamine in THF (20 mL of 2N) and TEA (8 mL) were added. After 3 hr of stirring, the mixture was concentrated in vacuo, aq NaOH (1M) was added, and the mixture was extracted with dichloromethane. The pH of the aq layer was adjusted to pH = 2 using 6N HCl (aq), and extracted with dichloromethane. The combiuned organic extracts were washed with brine, dried, concentrated in vacuo, and the product purified by silica gel chromatography (700 mL dichloromethane/20 mL MeOH/ 1 mL AcOH) to give the title compound (800 mg, 27% for two steps).

Step C

Following a similar procedure as that used in Preparative Example 10.55 Step C, except using the product from Step B above (780 mg), the title compound was obtained (0.46 g, 68%).

5

PREPARATIVE EXAMPLE 1314

Step A

15

20

10

Methyl-4-bromo-3-hydroxy-2-thiophenecarboxylate (20g, 84.36 mmol) was dissolved in 400 mL of acetone. Potassium carbonate (58g, 420.3 mmol) was added followed by iodomethane (45 mL, 424 mmol). The resulting mixture was heated at reflux for 4.5 h. After cooling, the mixture was filtered through a thin Celite pad, rinsing with methylene chloride. The filtrate was concentrated in vacuo to give 22.5 g of methyl-4-bromo-3-methoxy-2-thiophenecarboxylate (crude, 100%, MH⁺ = 251.0) as a dark green solid.

Step B

The product from Step A above (22.5g, 84.36 mmol) was dissolved in 60 mL of tetrahydrofuran and added with 125 mL of a 1.0 M NaOH aqueous solution. The mixture was stirred at room temperature for 4 d, then washed with ether (60 mL x 2),

acidified to pH ~ 2 using a 1.0 M HCl aqueous solution. Solids were precipitated out after acidification, and collected by filtration. The solid was dissolved in methylene chloride-ethyl acetate (~4:1, v/v). The organic solution was washed with H_2O and brine, dried with Na_2SO_4 , and concentrated in vacuo to a light yellow solid, further dried on hight vacuum, yielding 17.95 g of 4-bromo-3-methoxy-2-thiophene carboxylic acid (90%, MH^+ =237.0).

Step C

5

10

15

The carboxylic acid (3.26 g, 13.75 mmol) available from Step B above was treated with 30 mL of concentrated sulfuric acid. The mixture was sealed in a one-neck round bottom flask, and heated at 65°C for 4.5 h. After cooled to room temperature, the mixture was poured into 200 mL of crushed ice, and extracted with methylene chloride (100 mL x 3). The organic extracts were combined, washed successively with H₂O (50 mL x 2), sat. NaHCO₃ (50 mL x 3), and brine (50 mL). The organic solution was dried with Na₂SO₄, and concentrated in vacuo to a dark brown oil, which was purified by flash column chromatography (biotage, SiO₂ column) using hexanes-methylene chloride (3:1, v/v) as eluents. Removal of solvents afforded 1.83 g of 3-bromo-4-methoxy thiophene (69%) as a light yellow oil.

20 Step D

To a stirred solution of 3-bromo-4-methoxythiophene (550 mg, 2.85 mmol), prepared in Step C above, in 30 mL of methylene chloride at -78°C was added dropwise along the inside wall of the flask chlorosulfonic acid (0.48 mL, 7.21 mmol). The mixture was stirred at -78°C for 10 min, continued at room temperature for 1 h, and filtered through a 1-in silica gel pad, rinsing with methylene chloride. The filtrate was concentrated in vacuuo to give 270 mg of 4-bromo-3-methoxy-2-thiophene sulfonylchloride (33%) as a light yellow oil.

Step E

30

25

To a stirred solution of thiophene sulfonylchloride (270 mg, 0.926 mmol) prepared in Step D above in 15 mL of methylene chloride at room temperature was added triethylamine followed by *N*-methyl-tertbutylamine (0.25 mL, 2.094 mmol). After 20 h, the mixture was diluted with 50 mL of methylene chloride, and washed with H₂O and brine. The organic solution was dried over Na₂SO₄, filtered, and

concentrated to an oily residue, which was purified by preparative TLC (methylene chloride as eluent) to afford 73 mg of the titled bromo-sulfonamide (23%) as a near colorless oil.

5 Step F

10

15

20

25

30

A one-neck round bottom flask was charged with bromo-sulfonamide (73 mg, 0.2133 mmol, from Step E above), palladium acetate (5 mg, 0.0223 mmol), binap (0.03212 mmol), cesium carbonate (139 mg, 0.4266 mmol), and benzophenonimine (0.06 mL, 0.358 mmol). The mixture was evacuated via house vacuum, and refilled with nitrogen. A 3 mL of anhydrous toluene was added. The mixture was evacuated again, refilled with nitrogen, and heated at reflux for 2.5 d. After cooled to room temperature, methylene chloride (50 mL) was added, the mixture was filtered through a Celite pad, rinsing with methylene chloride. The filtrated was concentrated in vacuo to give 205 mg (crude, MH⁺ = 443.1) of the desired imine product as a dark brown oil, used in next step without purification.

Step G

The imine from Step F above (205 mg, crude, 0.2133 mmol) was dissolved in 5 mL of methanol, and added with sodium acetate (81 mg, 0.9873 mmol) followed by hydroxylamine hydrochloride (68 mg, 0.98 mmol). The mixture was stirred at room temperature for 6.5 h, quenched with the addition of 10 mL of a 1.0 M NaOH aqueous solution. The aqueous mixture was extracted with methylene chloride (30 mL x 3). The extracts were combined, washed with brine, dried by Na_2SO_4 , and concentrated in vacuo to a dark yellow oil, which was purified by preparative TLC (methylene chloride – methanol = 100:1, v/v) to give 34 mg (57% over two steps, MH^+ = 279.0) of methoxy-thiophenesulfonamide amine as a light yellow oil, solidified on standing.

Step H

To a stirred suspension of sodium hydride (60%, 45 mg, 1.13 mmol) in 3 mL of anhydrous *N*,*N*'-dimethylformamide (DMF) was added dropwise ethanethiol (0.1 mL, 1.34 mmol). After 10 min, the mixture tured into a clear solution, and 1 mL of this solution was taken up in a syringe and added dropwise to a stirred solution of methoxy-thiophenesulfonamide amine in 1 mL of DMF. The mixture was heated up to 95°C, and continued for 3.5 h. After cooling, the mixture was poured into 20 mL of

a 1.0 M NaOH aqueous solution. The aqueous mixture was washed with methylene chloride (30 mL x 3). The organic washings were combined, re-extracted with a 1.0 M NaOH aqueous solution (15 mL) and H_2O (15 mL). The aqueous layer and aqueous extracts were combined, adjusted to pH~6 using a 1.0 M HCl aqueous solution,and extracted with methylene chloride (75 mL x 3). The organic extracts were washed with brine, dried (Na₂SO₄), and concentrated in vacuo to a dark yellow oil. This oil was dissolved in ethyl acetate (50 mL), washed with H_2O (10 mL x 2) and brine (10 mL). The organic solution was dried (Na₂SO₄), and concentrated in vacuo to afford 36 mg (100%, MH⁺ = 265.0) of hydroxyl-thiophene sulfonamide amine as a yellow oil.

10

5

PREPARATIVE EXAMPLE 1315

Step A

15

Following the procedures described in Preparative Example 1314 Step E, 4-bromo-3-methoxy-2-thiophene-sulfonyl chloride (190 mg, 0.65 mmol, available from Step D, Preparative Example 1314) was converted to the titled tert-butyl sulfonamide (56 mg, 26%, MH⁺ = 328.1) upon treatment of triethylamine (0.28 mL, 2.0 mmol) and tert-butylamine (0.15 mL, 1.43 mmol) in 10 mL of methylene chloride.

20

Step B

tert-Butyl sulfonamide (98 mg, 0.3 mmol) available from Step A above was converted to the imine product (296 mg, crude, MH⁺ = 429.1) by using the procedure described in Step F of Preparative Example 1314.

Step C

The imine product (296 mg, crude, \sim 0.3 mmol) was transformed to the desired thiophene-amine (23 mg, 30% over two steps, MH $^+$ = 265.0) by using the procedure described in Step G of Preparative Example 1314.

5

Step D

If one were to apply the procedure set forth in Step H of Preparative Example 1314, but using the thiophene amine available from Step C above, one would obtain the titled hydroxyl thiophene sulfonamide amine.

10

PREPARATIVE EXPAMPLE 1316

Step A

Following the procedures set forth in Preparative example 13.29 Step B through F, but using diethylamine, 3-methoxy-2-thiophenesulfonyl chloride (available from Step A, Preparative example 13.29) was converted to titled diethylsulfonamido thiophene imine (MH^+ = 429.1)

Step B

20

25

15

Thiophene-imine (1.5 g, 3.5 mmol), available from Step A above, was dissolved in 30 mL of CH₂Cl₂, and added with potassium carbonate (1.2 g, 8.70 mmol) followed by drop wise addition of bromine (0.32 mL, 6.25 mmol). After stirred for 2 d, H₂O was added. The two layers were separated. The aqueous layer was extracted with CH₂Cl₂ (50 mL x 2). The organic layers were combined, washed with a 10% Na₂S₂O₃ aqueous solution (40 mL x 2) and brine (40 mL), dried over Na₂SO₄, and concentrated in vacuo to a dark brown oil. This oil was separated by preparative

TLC (CH₂Cl₂ as eluent), to give 0.96 g (54%) of the desired bromo-imine as a bright yellow oil ($M^+ = 507$, M + 2 = 509)

Step C

5

10

20

Bromo-imine (0.95g, 1.87 mmol), available from Step B above, was dissolved in 15 mL of anhydrous THF, cooled in a -78°C bath, and treated with a 2.5 M solution of n-butyl lithium in hexanes (1.2 mL, 3.0 mmol) drop wise along the side wall of the flask. After 30 min, lodomethane (0.35 mL, 5.62 mmol) was added drop wise. Reaction was continued for 5 h, during which time the cooling bath was allowed to warm slowly to 0°C. The mixture was quenched by H_2O (25 mL), and extracted with CH_2CI_2 (50 mL x 2). The organic extracts were washed with brine, dried with Na_2SO_4 , and concentrated in vacuo to give 0.93 g (crude, > 100%) of the desired methylated imine as a dark yellow oil (MH^+ = 443.1)

15 Step D

The crude methyl-imine (0.93 g), prepared in step C above, was converted to the methyl-hydroxyl-amine (0.21 g, 41%, MH^{+} = 265.0) by using the procedures described in Step G of Preparative Example 13.29.

PREPARATIVE EXAMPLES 1203-1234

Following a similar procedure set forth in Preparative Example 22 but using the commercially available or prepared amine from the Preparative Example indicated in the Table below, the following thiadiazoledioxide intermediates were obtained.

Prep Ex	Amine/Aniline	Product	1.Yield (%)
1203	F_3C OH OH NMe_2	F ₃ C N N OMe	2.(M+1) ⁺ 1. 99% 2. 395.0

1205	OH OH	ON S N N OME N OH	1. 33% 2. 382.1
1206	N N N NH ₂	N N N O O O O O O O O O O O O O O O O O	1. 71% 2. 445.1
1207	$O \longrightarrow NH_2$ $O \longrightarrow NH_2$ $O \longrightarrow NH_2$	ON NOME OH	1. 85% 2. 369.1
1209	N NH2	N N OME	1. 69% 2. 363.0
1210	NH ₂	N O O O N O O N O O N O O N O O O N O O O N O O O N O O O N O O O N O O O N O O O N O O O O N O O O O N O	1. 92% 2. 405.0 (M+Na)
1211	N S NH ₂ NH ₂	N S N OME	1. 45% 2. 403.0

1212	Br N S NH ₂	Br N N N OMe	1. 85% 2. 442.8
1216	N S O OH NH2	ON ON OME	1. 50% 2. 389.0
1217	NH ₂ O OH	ON ON OME	1. 67% 2. 339.0
1218	OHOH	ON ONE OH OH	1. 87% 2. 370.9
1219	O NH ₂ OH N	ON OME OME OH	1. 70% 2. 404.0
1220	OHOH	ON ON OME OH OH	1. 83% 2. 419.0

1221	OHOH	O O N S N O Me O H O H	1. 99% 2. 383.1
1222	OH OH OH	ON ON OME OH	1. 99% 2. 405.0
1223	NH ₂ OH N N N N N N N N N N N N N N N N N N	ON ON OME OH OH	1. 69% 2. 473.0
1224	N N N N N N N N N N N N N N N N N N N	O S N III OME	1. 99% 2. 461.0
1225	H_2N	MeO N H	1. 99% 2. 314

1226	H ₂ N O	MeO N N H	1. 99% 2. 354
1227	H ₂ N O	MeO N H	1. 92% 2. 300
1228	H ₂ N	MeO N N N N N N N N N N N N N N N N N N N	1. 99% 2. 342
1229	H_2N	MeO N N H	1. 99% 2. 300
1230	H ₂ N	MeO N N H	1. 99% 2. 286

1231	H ₂ N O	MeO N N H	1. 99% 2. 300
1232	Br NH ₂	Br N N OMe N OMe	used crude
1233	NH ₂	ON SIN OME HOME	1. 92% 2. 396
1234	$N \longrightarrow NH_2$ OH	ON NOME OF THE O	used crude

EXAMPLE 1

The thiadiazole intermediate from Preparative Example 22 (65mg, 0.2mmol), isopropylamine (17ul, 0.2mmol) and DIEA (100ul) were combined in MeOH (2ml) and stirred overnight at room temperature. The reaction was purified by liquid chromatography to yield product (22mg, 31%, MH+=31%).

5

10

EXAMPLES 2-71

Following a similar procedure set forth in Example 1 but using the commercially available (or prepared amine) and the thiadiazoledioxide intermediate from the Preparative Example(s) indicated in the Table below and stirring the reaction mixtures at room temperature up to reflux, the following thiadiazoledioxide products were obtained.

Ex.	Prep Ex	Product	1. Yield
	of Thiadiazole		2. MH+
	Intermediate and		
	Amine		
2	22	O, O N, S, N	1. 43
		N,S,N	2. 379.9
	and		
		N H H	
	H ₂ N	O OH H	
3	22	Q,_O ,_S,_	1. 54
		N, S, N	2. 393.9
	and		
		OH H H	
	H ₂ N		
4	22	O, S,	1. 18
1		N,S,N	2. 387.9
	and		
		OH H H	
	H ₂ N		

5	22	Q, ,,O	1. 55
	and	O O O N O O O O O O O O O O O O O O O O	2. 430
	H ₂ N		
6	22 and	ON ON NO N	1. 58 2. 433.8
	H ₂ N O		
7	22 and	O S N N N N N N N N N N N N N N N N N N	1. 14 2. 367.9
	H ₂ N	О ОН ''	
8	22 and		1. 34 2. 381.9
	H ₂ N	O OH H N	
9	22 and		1. 43 2. 429.8
	H ₂ N	о он н н	

10	22	0,0	1. 47
	and		2. 415.8
		OH H H	
	H ₂ N		
11	22	O O O	1. 32% 2. 381.9
	and	N N N N N N N N N N N N N N N N N N N	2. 001.0
		Ö ÖH ^H	
12	⁷ NH ₂ 22	0,50	1. 27
	and	N N N N Ph	2. 514.9
	H_2N H_2N Ph O		
13	22	O S O	1. 24 2. 441.8
	and	O OH H H	2. 411.0
	NH ₂	0 011	
14	22	0,0	1. 42
	and	N N	2. 427.9
	H ₂ N	N OH H H	

15	and H ₂ N	ON NO N	1. 33 2. 461.7
16	22 and H₂N ← O CI	ON SIN OH H H CI	1. 12 2. 453.7
17	and H ₂ N	ON SIN SIN SIN SIN SIN SIN SIN SIN SIN SI	1. 29 2. 461.4
18	and H ₂ N		1. 44 2. 495.7

19	22	O O	1. 40
	and	N N CF ₃	2. 473.8
	CF ₃	OH H H	
20	22	O O N N	1. 34
	and	CF ₃	2. 469.9
	CF ₃	N OH H	
	H ₂ N	Ö	
21	22	O O N S N	1. 31
	and	CF ₃	2. 513.8
	©F ₃	N OH H H	
	H ₂ N O		
22	22		1. 30
	and	N N	2. 496.0 (M+Na) ⁺
		OH OH OH	
	H ₂ N		

23	and H ₂ N	ON NOH H	1. 23 2. 489.9
24	and H ₂ N	ON NO N	1. 55 2. 473.8
25	and H ₂ N	ON N H H OH OH	1. 4 2. 447.8

26	and H ₂ N S	ON NON NON NON NON NON NON NON NON NON	1. 8 2. 501.8
27	and H ₂ N	O S N N N N N N N N N N N N N N N N N N	1. 18 2. 482.0 (M+Na) ⁺
28	and H ₂ N		1. 42 2. 467.9 (M+Na) ⁺
29	and H₂N		1. 24% 2. 435.8

30	22 and		1. 51 2. 463.8
	and	NOH H H	
	H ₂ N S		
31	22	O O N	1. 52 2. 475.8
	and /	N OH H H	
	H ₂ N O	Ö	
32	22	0,0	1. 32
	and	N N N N N N N N N N N N N N N N N N N	2. 477.8
		NO OH H H	
	H ₂ N		
33	22	0 0 N	1. 58 2. 449.7
	and 	N N S	
	H ₂ N	N OH	

34	22 and	O S N N N N N N N N N N N N N N N N N N	1. 50 2. 483.9 (M+Na) ⁺
	H ₂ N	, O	
35	22	O O N S N	1. 33 2. 447.8
	and	H H	
	Min.	OH H H	
	H ₂ N		
36	22	O O N S N	1. 53 2. 449.8
	and	N N	2. 7.0.0
	H ₂ N S	N OH H	
37	22	O O N N	1. 15 2. 459.8
	and	N N CF ₃	
	H ₂ N O	N OH H	
38	23.2	0 N S N	1. 34 2. 430.0
	and	N H H	2. 400.0
	H ₂ N	ОН .	

39	23.1	O_2N N N N	1. 13 2. 403.8
	and	N N Ph	2. 400.0
	O ₂ N NH ₂	OH	
40	23.1	NC NSN	1. 28 2. 383.8
	and	N N Ph OH H H	2. 303.0
	NC NH ₂		
41	23.1	O, O N S N	1. 42 2. 358.8
	and	N N Ph	2. 000.0
	NH ₂		
42	23.1	O S N N	1. 19 2. 435.8
	and	N N Ph	2. 100.0
	O S NH ₂		
43	23.1	HO N S N	1. 34 2. 402.8
	and	N N Ph	2. 702.0
	HO NH ₂		

44	23.1	O O	1. 23
	and	N N	2. 383.8
		NC N N Ph OH H H	
	NC NH ₂		
45	23.1	O, , O	1. 30
	and	N N Ph	2. 376.8
	F NH ₂		
46	23.1	0,0	1. 35
	and	N N Ph	2. 415.8
	O NH ₂	ни он н	
47	23.1	0,0	1. 25
	and	N N Ph NH ₂ OH H H	2. 401.8
	ONH ₂ OH		
48	23.1	O, O	1. 13
	and	O O S N N Ph	2. 471.7
	O S NH ₂		

49	23.1	N N S N	1. 10 2. 373.9
	and	N N Ph OH H H	
	N NH ₂ OH		
50	23.5	0, 0 N N	1. 15 2. 467.7
	and	NC OH H H	2. 407.7
	NC OH		
51	23.4	O N N	1. 21 2. 413.8
	and	NC NH N CF ₃	
	NC OH	ОН	
52	23.3	O O N S N	1. 16 2. 336.9
	and	N N	2. 000.0
	NC OH	NC´ \ H H	
53	23.6	O O N S N	1. 29
	and	NC NH N H	2. 774.6 (dimer)
	NC OH	OH	

54	23.6 and NH ₂ OH NO ₂	O ₂ N N N N O N H N O O O O N N N N N O O O O	1. 20 2. 814.4 (dimer)
55	23.8 and H ₂ N	Br N N O O O O O O O O O O O O O O O O O	1. 1 2. 539.9
56	and H ₂ N	Br N N N N N N N N N N N N N N N N N N N	1. 9 2. 567.9
57	23.1 and NH₂ OH	O S N N H H	1. 11 2. 443.9

58	23.1	O O N	1. 14 2. 509.8
	and	Br—NNN—N—	2. 000.0
	P P P P P P P P P P	O OH H H	
59	23.1	O O N S N	1. 15 2. 471.7
	and	CI—N N H H	2. 4/1./
	CI—NH ₂	H ₂ N-S OH	
	H ₂ N-S OH		
60	23.1		1. 41 2. 485.0
	and	H H	
	NH ₂	OH OH	
61	23.1	0 0 N S N	1. 33 2. 471.9
	and	N N N	2. 471.9
	O N O	OH OH	
62	23.1	O O N	1. 21 2. 514.0
	and	HO ₂ C OH H H	Z. 014.U
	HO_2C N OH OH	Ö	

63	23.1	0,0	1. 11
	and	N N	2. 513.9
	and	OH H H	
	NH ₂	HO ₂ Cliff	
	HO ₂ Ciii OH		
64	23.7	O O N S N	1. 34
	and	N N	2. 415.7
	NH ₂	NC OH	
	NC OH		
65	23.7	0 0 N N	1. 31
	and	O S N N N	2. 503.7
	13.29	N HO H H	
66	23.8	O S N	1. 22
	and	Br N N	2. 511.7
		он н н	
	H ₂ N O	-	

67	and H ₂ N	ON N H H S	 33 539.8
68	and H ₂ N		1. 24 2. 469.9
69	23.1 and NH ₂ OH	ON ON NO N	1. 12 2. 470.0
70	23.1 and N OH OH	ON NO N	1. 20 2. 456.0

71	23.9 and	O ₂ N N N N OH H	1. 8 2. 361.9
72	H ₂ N 22 and 75.60	OH N N N N N N N N N N N N N N N N N N N	1. 48 2. 502.1

EXAMPLES 100-188

If one were to follow a procedure similar to that set forth in Example 2, but using the commercially available (or prepared amine) and the thiadiazoledioxide intermediate from the Preparative Example(s) indicated in the Table below and stirring the reaction mixtures at room temperature up to reflux, the thiadiazoledioxide products in the Table below could be obtained.

Ex.	Prep Ex of Thiadiazole dioxide Intermediate	Amine	Product
100		H ₂ N	ON N H H
101	22	H ₂ N N	ON N N N N N H H
103	22	H ₂ N	O O O O O O O O O O O O O O O O O O O

5

104	22		Q O N S N
		H ₂ N	O OH H H
105	22	H ₂ N	
106	22	H ₂ N S	Q O N S N N H O OH H
107	22	H ₂ N S	
108	22	H ₂ N S	
109	22	H ₂ N S	
110	22	H_2N	
111	22	H_2N	ON N N H
112	22	H ₂ N	ON ON NO N

113	22		O, O N S, N
		N O	
		H D	O OH H H
114	22	H_2N	ON NO N
115	22	H_2N	ON ON NO N
116	22	H_2N	ON ON NO N
117	22	H ₂ N CI	
118	22	H_2N	
119	22	H ₂ N Br	O N N H O Br
120	22	H_2N O Br	ON N N N N N H Br

121	22	H ₂ N O	ON NH NH
122	22	H ₂ N O	ON NO N
123	22	H ₂ N O	
124	22	H ₂ N O	
125	22	H ₂ N O	
126	22	H ₂ N O	ON ON NO N
127	22	H ₂ N O	O O N N N N N N N N N N N N N N N N N N

	1	Ť ·	
128	22	H ₂ N CI	N N N N OCI
129	22	H_2N O Br	N S N H Br
130	22	H ₂ N	ON NH NH
131	22	H_2N	
132	22	H ₂ N	O Z H H N N N N N N N N N N N N N N N N N
134	23.32	NNN H	N N H H N N N N N N N N N N N N N N N N
135	23.32	13.19	O S N N N N N N N N N N N N N N N N N N

136	23.6	13.19	O S N N N N N N N N N N N N N N N N N N
137	23.31	13.19	ON N H H
138	23.6	13.29	
139	23.39	H ₂ N CI	O Z Z H
140	23.39	H_2N O CI	
141	23.39	H ₂ N	O O O N N N N N N N N N N N N N N N N N
142	23.39 ⁻	H ₂ N	O O O O O O O O O O O O O O O O O O O
143	23.39	H ₂ N	O O O N S N N O O O O O O O O O O O O O

	<u> </u>		
144	23.39	H ₂ N O	NS N N N N N N N N N N N N N N N N N N
145	23.39	H ₂ N O	N S N N N N N N N N N N N N N N N N N N
146	23.39	H ₂ N Br	N S N N N N N N N N N N N N N N N N N N
147	23.39	H_2N O Br	NS N N N N N N N N N N N N N N N N N N
148	23.6	500.1 or 500.2	ON SIN NO
149	23.32	500.1 or 500.2	
150	23.6	N OH OH	ON SIN H
151	23.38	NH ₂	O, O N, N CF ₃ N N CF ₃ N N N N N N N N N N N N N N N N N N N
152	23.37	3	ON ON CF3

153	23.32	13.32-A	S N N N N N N N N N N N N N N N N N N N
154	23.33	13.32-A	O N N H O CI
155	23.31	13.32-A	S HO HO H
156	23.34	13.32-A	S HO HO CI
157	23.40	H ₂ N—O	S N N N N N N N N N N N N N N N N N N N
158	23.40	H ₂ N	S HO H H
159	23.40	H ₂ N	

160	23.41	H ₂ N	O S N
·			N-S OH H H
161	23.41	H ₂ N CI	N-S OH H H CI
162	23.41	H ₂ N	ON NON NON NON NON NON NON NON NON NON
163	23.41	H ₂ N	
164	23.42	H ₂ N	
165	23.42	H ₂ N O	CI Z H H OH OH
166	23.42	H ₂ N—OCI	CI—N—S OH CI

167	23.40	H ₂ N	N N N N N N N N N N N N N N N N N N N
168	23.41	H ₂ N	
169	23.42	H ₂ N O	
170	22	H ₂ N O	
171	23.8	H ₂ N CI	Br NO CI
172	23.8	H ₂ N O	Br OH OH OH
173	23.8	H ₂ N O	Br N N OH OH

174	23.8	H ₂ N	O S N
		CI	Br N N O CI
175	23.8	H ₂ N	Br O O N N N N N N N N N N N N N N N N N
176	23.8	H ₂ N	Br OH OH
177	23.32	13.32-A	H N S H H
178	23.32	NH ₂ OH O Br Z Z -	Br OH H
179	23.30	15	HO Z H

180	23.32	15	N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-
181	23.33	15	O S N N N H H O CI
182	23.30	500.3 or 500.4	O S N H H OH OH
183	23.32	500.3 or 500.4	O S N N O O O O O O O O O O O O O O O O
184	23.33	500.3 or 500.4	ON S N O CI
185	23.30	500.1 or 500.2	ON N H H

186	23.32	500.1 or 500.2	O S N N N N N N N N N N N N N N N N N N
187	23.33	500.1 or 500.2	O S N O CI
188	23.32	13.71	O S N N N N N N N N N N N N N N N N N N

EXAMPLE 200

The thiadiazoleoxide from Preparative Example 22.1 (55mg, 0.17mmol) was added to R-2-phenylpropylamine (0.024mL, 0.17mmol) in methanol (2ml) with diiospropylethylamine (100 μ L). The reaction mixture was microwaved at 100W for 4hr then purified by preparative HPLC. Concentration of desired fractions gave the pure product (12.4mg, 18%). MH⁺ = 413.9.

EXAMPLE 201

The thiadiazoleoxide from Preparative Example 22.2 (56mg, 0.2mmol) was added to 2-amino-6-cyanophenol (27mg, 0.2mmol) in methanol (2ml) with diiospropylethylamine (100 μ L). The reaction mixture was microwaved at 100W for 24hr then purified by preparative HPLC. Concentration of desired fractions gave the pure product (11mg, 15%). MH⁺ = 367.9.

EXAMPLE 201A

To a solution of the Thiadiazole mono-oxide intermediate from Preparative Example 22.1 (100mg, 0.3086mmol) and the Furyl amine from Preparative Example 75.1 (43mg, 0.3086mmol) in methanol (2mL) was added sodium trifluoro acetate (84mg, 0.6172mmol), followed by drop wise addition of diisopropylethylamine (80mg, 0.6172mmol). The reaction mixture was stirred over night at room temperature, solvents were removed under reduced pressure and the product was purified by preparative thin layer chromatography using Dichloromethane-Methanol (20:1) to afford the compound as a white solid. (Yield: 98mg, 74%, m.p = 140°C)

EXAMPLES 201.1 – 201.9

Following a similar procedure to that set forth in Example 201, but using the amine and the thiadiazoleoxide intermediate from the Preparative Example indicated in the Table below, the thiadiazoleoxide products in the Table below were prepared.

5

10

15

20

F.,	D	D	Draduck	4 1/2 1 1
Ex.	Prep Ex	Prep Ex	Product	1. Yield
	(Thiadiazole	(Amine)		2. mp
	oxide			(°C)
	Intermediate)			3. MH ⁺
201.1	22.1	75.9		1. 9%
			0 S	2. 110
			N S N	3. 446
			N N	
			н н <i>)</i> / Q	
			OH OH	
004.0	00.4	75.40		1 201
201.2	22.1	75.49	0	1.6%
	ļ		O 11 S	2. 107
			N, N	3. 418
			N N	
			N OH H H	
			/ (
			1	
201.3	23.9	64.1		1. 32%
			0 II .S.	2. 156
			N, N	3. 440
			\	
			N H H	
			OH	

201.4	22.1	64.4	OH S N N N H H	1.50% 2. 171 3. 380
201.5	22.1	64.3	OIS N H H	1. 49% 2. 171 3. 380
201.6	22.1	75.92	OH OH OH	1.25% 2. 102 3. 482
201.7	23.3	75.1		1. 9% 2. 145 3. 460

201.8	22.1	623		1.31% 2. 60 3. 444
201.9	22.1	75.61	0=s 2 2 H OH OH	1.37% 2. 172 3. 446

EXAMPLES 202-245

If one were to follow a procedure similar to that set forth in Example 201A, but using the commercially available (or prepared amine) and the thiadiazoleoxide intermediate from the Preparative Example(s) indicated in the Table below and stirring the reaction mixtures at room temperature up to reflux, the thiadiazoleoxide products in the Table below could be obtained.

Ex.	Prep Ex of Thiadiazole oxide	Amine	Product
202	Intermediate 22.1	H ₂ N	O=S P H H OH OH OH OH OH OH OH OH OH OH OH OH

5

203	22.1	H ₂ N—O	0=s N N H O
204	22.1	H ₂ N O	0=0 2 2 H 0H 0H
205	22.1	H ₂ N O	0=s 2 1 H
206	22.1	H ₂ N—CI	0=s 2
207	22.1	H ₂ N O	O S N N O O O O O
208	22.3	H ₂ N	Br N N H H

209	22.3	H ₂ N O	Br N N OH OH
210	22.3	H ₂ N O	0H H Z Z H Z Z H OH OH OH OH OH OH
211	22.3	H ₂ N O	0=s 2 H OH OH
212	22.3	H ₂ N CI	O=S N N H OH OH
213	22.3	H ₂ N CI	Br N H H OCI
214	22.4	H ₂ N	

215	22.4	H ₂ N—O	
216	22.4	H ₂ N O	
217	22.4	H ₂ N	0=s 2 1 0=s 2 1 0 5 0 0 6 0
218	22.4	H ₂ N CI	
219	22.4	H ₂ N O CI	
220	22.5	H ₂ N	
221	22.5	H ₂ N—O	

			
222	22.5	H ₂ N—O	O S O O H H
223	22.5	H ₂ N O	
224	22.5	H ₂ N CI	0=s + O O O O O O O O O O O O O O O O O O
225	22.5	H ₂ N CI	
226	22.6	H ₂ N	
227	22.6	H ₂ N—O	
228	22.6	H ₂ N—O	

229	22.6	H ₂ N → O	
230	22.6	H ₂ N CI	O=S, N H H O CI
231	22.6	H ₂ N — CI	
232	22.7	H ₂ N	0=0 2 2 H 0=0 0 H 2 2 H
233	22.7	H ₂ N O	
234	22.7	H ₂ N O	
235	22.7	H ₂ N O	
236	22.7	H ₂ N—CI	

237	22.7	H ₂ N O	
238	22.1	H ₂ N	OH S N N N N N N N N N N N N N N N N N N
239	22.1	H ₂ N Br	O=S N OH Br
240	22.3	H ₂ N	0=S Z H H OH OH
241	22.3	H ₂ N Br	Br N N N N N N N N N N N N N N N N N N N
241.1	22.4	H ₂ N	

241.2	22.4		0
		H ₂ N Br	CI N N N N N N N N N N N N N N N N N N N
241.3	22.5		0= <i>s</i>
		H ₂ N	N S OH H H
241.4	22.5		0 = \$
		H ₂ N Br	N S OH H H Br
242	22.6		O = S N
		H ₂ N	N S N N N N N N N N N N N N N N N N N N
243	22.6		. O .S.
		H ₂ N Br	S N N N N N N N N N N N N N N N N N N N
244	22.7		O S
		H ₂ N	S N N N N N N N N N N N N N N N N N N N
245	22.7		0 s
-1 -1		H ₂ N Br	N S N N N N N N N N N N N N N N N N N N

EXAMPLES 246-373

If one were to follow a procedure similar to that set forth in Example 201A, but using the amines from the Preparative Examples indicated in the Table below (or the

commercially available amines indicated in the Table below), and the thiadiazoleoxide intermediates from the Preparative Examples indicated in the Table below and stirring the reaction mixtures at room temperature up to reflux, the thiadiazoleoxide products in the Table below could be obtained.

		_	_
1	ľ	_	
8			١
1	L		e

Ex.	Prep. Ex. (Thiadiazole oxide Intermediate)	Prep. Ex. (Amine)	Product
246	22.25	75.44	
248	22.23	74	
249	22.23	75.44	O=S Z H OH

251	22.24	74	
252	22.24	75.44	
254	22.26	75.44	O S N N N N N N N N N N N N N N N N N N
255	22.1	H ₂ N (commercially available)	O= N
256	22.1	H ₂ N (commercially available)	

257	22.1	75.66	
258	22.1	75.30	
259	22.1	75.50	
260	22.1	603D	
262	22.1	75.10	O H O H O H O H O H O H O H O H O H O H

263	22.1	75.10E	
264	22.29	75.44	F ₃ C
265	22.39	603C	F ₃ C
266	22.30	75.44	
267	22.6	75.1	

268	22.6	75.9	
269	22.39	75.44	F ₃ C O S N N H
270	22.6	75.61	
271	23.9	75.44	
272	22.27	75.44	F ₃ C N N N N N N N N N N N N N N N N N N N

273	22.28	75.10	
274	22.28	75.61	
275	22.31	75.44	F ₃ C N N N N N N H N H
276	22.32	75.61	CI NS N = ON S N H
277	22.32	75.10	

278	22.32	75.44	
2.0	22.02	,	CI N S N T N T N T N T N T N T N T N T N T
279	22.32	75.9	CI N S N H
280	22.32	75.60	
281	22.32	75.30	
282	22.32	75.52	
283	22.47	75.60	

284	22.34	75.52	
285	22.48	75.44	
286	22.1	75.67	O = S N N N N N N N N N N N N N N N N N N
287	22.36	75.44	
289	22.41	75.44	OHO HIS Z
291	22.32	75.201	

292	22.32	75.200	
293	22.47	75.201	
294	22.47	75.200	
295	22.4	75.201	
296	22.4	75.200	

297	22.45	75.201	
298	22.45	75.200	
299	22.46	75.201	Br N N H
300	22.46	75.200	Br N N N N N N N N N N N N N N N N N N N
301	22.39	75.201	F ₃ C OH N S N = OH N H

302	22.39	75.200	F ₃ C N S N S N S N S N S N S N S N S N S N
303	22.39	601.A	F ₃ C N N N N N N N N N N N N N N N N N N N
304	22.39	601.B	F ₃ C N N N N N N N N N N N N N N N N N N N
305	22.1	75.93	
306	22.1	64.11	

307	22.48	75.1	
308	22.37	75.9	
309	22.50	51.26	
310	22.51	75.9	
311	22.16	75.1	
312	22.51	51.26	

313	22.38	75.1	Br N N N N N N N N N N N N N N N N N N N
314	22.7	75.61	H-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z-Z
315	22.7	75.10	O S N S N S N S N S N S N S N S N S N S
316	22.7	75.10E	
317	22.7	603A	

318	22.7	603B	
319	22.7	603C	
320	22.19	75.10	N S N N N N N N N N N N N N N N N N N N
321	22.19	75.10E	
322	22.19	603A	

323	22.19	603B	
324	22.19	603C	
325	22.20	75.10	H-N-S-N-H-N-H-N-H-N-H-N-H-N-H-N-H-N-H-N-
326	22.20	75.10E	H-N S N H
327	22.20	603A	H S N N N N N N N N N N N N N N N N N N

328	22.20	603B	
329	22.20	603C	H S N N N N N N N N N N N N N N N N N N
330	22.21	75.10	
331	22.21	75.10E	
332	22.21	75.10E	

333	22.21	603A	
334	22.21	603B	
335	22.21	603C	
336	22.19	75.61	
337	22.20	75.61	H-N-SOOH N-H

338	22.21	75.61	
339	22.22	75.61	H-Z
340	22.22	75.10	
341	22.22	75.10E	
342	22.22	603A	

343	22.22	603B	
344	22.22	603C	
345	22.1	74	
346	22.1	75.45	

347	22.1	75.1	O = S N N H H
348	22.1	75.44	
349	22.39	75.1	F ₃ C N N N N N N O O O O O O O O O O O O O
350	22.47	64.2	OH SN N N H H

351	22.47	75.1	
352	22.1	51.26	
353	22.41	(commercially available)	0=0 H Z Z H H Z Z H
354	22.1	13.18	OII S N F F F N N H H H

355	22.1	75.60	
356	22.1	(commercially available)	OH N N O
357	22.1	75.29	OIIS N N H H N N F
358	22.1	72	OIS NOH H

359	22.1	75.34	O I S N N N H H O H
360	22.1	75.62	
361	22.39	75.61	F ₃ C N N N O OH
362	22.39	H ₂ N (commercially available)	F ₃ C N N N N N N OH OH

363	22.42	(commercially available)	HO,,, OH H H
364	22.43	(commercially available)	OH OH OH
365	22.1	75.63	
366	22.44	(commercially available)	HO NOH H

367	22.1	(commercially available)	
368	22.1	75.27	
369	22.1	64.6* H ₂ N (commercially available)	
370	22.46	64.1* H ₂ N (commercially available)	OIS N N N N OH OH

371	22.45	75.1	
372	22.47	75.61	
373	22.1	75.19	0=0 2 2 1 2 2 H 0H 0H
374	22.39	75.19	F ₃ C N N N N N N N N N N N N N N N N N N N

375	22.46	75.9	Br N N N N OH H H
376	22.52	75.9	
377	22.52	75.44	
378	22.46	75.10E	OIS Z H H OH OH

379	22.52	75.10E	
380	22.5	75.9	
381	22.4	75.9	
382	22.39	75.10E	F ₃ C

383	22.46	75.61	Br N N N N OH
384	22.52	75.61	
385	22.5	75.61	
386	22.4	75.61	

387	22.7	75.10E	
388	22.7	75.9	
389	22.22	75.9	
390	22.22	75.44	

391	22.53	75.9	CI N
392	22.53	75.10E	CI N N N N N N N N N N N N N N N N N N N
393	22.53	75.44	CI N N N N N N N N N N N N N N N N N N N
394	22.53	75.61	CI N N N H H H

EXAMPLES 2001-2113

Following a similar procedure as that set forth in Example 1 but using the commercially available or prepared amine from the Preparative Example indicated in the Table below, the following thiadiazoledioxide products were obtained.

			
Ex	Amine & PrepEx of	Product	1.Yield
	intermediate		(%)
			2. (M+1) ⁺
2001		0,6,0	1. 33%
	H ₂ N S	O O N N /	2. 539.8
		N N S	
	Ph	OH H	
	&	NMe ₂ Ph	
	22		
2002	V	0,0	1. 72%
		N N	2. 461.0
	H ₂ N N	$N \longrightarrow N$	
		H H	
	&	O≕ OH NMe ₂	
	22	-	
2003		0,0	1. 45%
		N,S,N	2. 497.9
	H ₂ N	N N N N N N N N N N N N N N N N N N N	
	0	он н н Гором	
	&		
2004	22	O O	1 270/
2004		N S N	1. 37% 2. 447.9
	H ₂ N \		2. 447.3
	1	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	
	&	O=\ OH \	
	22	NMe ₂	
1			I

5

2005	H ₂ N O & 22	ON SN N N N N N N N N N N N N N N N N N N	1. 5% 2. 497.9
2006	H ₂ N O & 1203	F_3C N	1. 59% 2. 529.9
2007	H ₂ N O & 22	ON N N O N N O O O O O O O O O O O O O	1. 64% 2. 461.7
2008	H ₂ N MeO & 22	ON N N N NH N H MeO NMe ₂	1. 55% 2. 487.9
2009	H ₂ N OMe & 22	ONS N N N OH OH NMe ₂	1. 14% 2. 488.0
2010	H ₂ N & 22	ON N N N O N N N N N N N N N N N N N N	1. 24% 2. 469.9

2011	H ₂ N OMe & 22	ON S N OME OME OME	1. 14% 2. 488.9
2012	H ₂ N O OMe & 22	$O \longrightarrow O$ $N \longrightarrow N$ $O \longrightarrow O$ $O \longrightarrow $	1. 10% 2. 527.8
2013	H₂N O & 1203	F_3C N	1. 14% 2. 529.8
2014	H ₂ N O CI & 22	ON N N O O O O O O O O O O O O O O O O	1. 32% 2. 531.8
2015	H ₂ N O & 22	ON N N O O O O O O O O O O O O O O O O	1. 6% 2. 498

2016	H ₂ N CONMe ₂ & 22	ON NO N	1. 6% 2. 504.9
2017	H₂N & 22	ON S N N N N N N N N N N N N N N N N N N	1. 8% 2. 458.2
2018	H ₂ N NMe ₂ & 22	O N	1. 92% 2. 501.0
2019	HN & 1224		1. 75% 2. 596.2
2020	H ₂ N 0 & 1205	ON N HN ON N N N N N N N N N N N N N N N	1. 33% 2. 517.2

2021	H ₂ N & 1206		1. 71% 2. 580.2
2022	H ₂ N 0 & 1207	ON NHN OH	1. 85% 2. 504.2
2023	22 & H ₂ N OMe	ON N H OME OME	1. 92% 2. 518.2
2024	22 H ₂ N O		1. 69% 2. 418.1
2025	22 & OMe OMe	ON ON OME OME	1. 86% 2. 518.2
2026	1206 & HN O		1. 75% 2. 552.2

2027	22 & H ₂ N MeO OMe	O O O N S N H MeO OMe	1. 83% 2. 518.2
2028	1207 & H ₂ N	ON ON HN HN ON	1. 79% 2. 516.1
2029	1224 & H ₂ N	ON OH HIN OH	1. 75% 2. 568.2
2031	H ₂ N 0 & 1207		1. 92% 2. 476.1
2032	H ₂ N 0 8 1209		1. 39% 2. 497.7
2033	HN 0 0 8 1209		1. 46% 2. 509.8

2034	& 1210 H ₂ N	O O O N N N N N N N N N N N N N N N N N	1. 89% 2. 489.9
2035	H ₂ N 0 & 1209		1. 13% 2. 469.6
2036	H ₂ N 0 & 1211	N S N N N N N N N N N N N N N N N N N N	1. 36% 2. 509.5
2037	H ₂ N 0 & 1212	Br N N N N N N N N N N N N N N N N N N N	1. 48% 2. 577.7
2038	NH ₂ NH ₂ & 23.1	ON N H H	1. 12% 2. 470.0
2039	H ₂ N 0 & 23.8	O Z Z H O Z	1. 1% 2. 539.8
2040	O OH & 23.1	ON ON N H	1. 19% 2. 456.0

2041	H ₂ N O & 1216	ON SOOH H	1. 73% 2. 523.7
2042	H ₂ N & 1217	ON NH H	1. 23% 2. 442.0
2043	H ₂ N → O & 1216	O O O O O O O O O O O O O O O O O O O	1. 38% 2. 495.6
2044	4 1209	OS N N N N N N N N N N N N N N N N N N N	1. 25% 2. 497.6
2045	HO NH ₂ NH ₂ NH ₂ NH ₂ NH ₂ NH ₂ NH ₂ NH ₂	HO N N H H	1. 4% 2. 466.0
2046	O NH ₂ OH O & 23.1		1. 33% 2. 471.9

2047	H ₂ N 0 & 23.8	Br N HN O	1. 9% 2. 567.9
2049	H ₂ N → O & 1207	ON NON HN ON OH	1. 28% 2. 526.0 (M+Na)
2050	H ₂ N & 22		1. 17% 2. 410.0
2051	H ₂ N → O & 1218	ON N N H OOH OOH	1. 43% 2. 505.8
2052	H ₂ N → O & 1219	O'S' HN OH OH	1. 45% 2. 539.0

2053	H ₂ N → O & 23.8	Br N N N O N O N O N O N O N O N O N O N	1. 22% 2. 511.6
2054	H ₂ N O & 1220	ON HIN OOH OOH	1. 55% 2. 531.9
2055	H ₂ N → O & 1221	ON HN OH OH	1. 46% 2. 517.9
2056	H ₂ N → O & 1222	O S N O O O O O O O O O O O O O O O O O	1. 38% 2. 517.9
2057	H ₂ N → 0 & 23.8	Br N N H O O O O O O O O O O O O O O O O O	1. 1% 2. 539.8

2058	H ₂ N ~ & 22	ON N N H H	1. 23% 2. 396.0
2059	0 NH ₂ OH NH ₂ & 1205		1. 41% 2. 485.0
2060	H ₂ N → 0 & 1223		1. 55% 2. 608.0
2062	OH NH ₂ OH & 23.1	ON N HN HN OH HN	1. 11% 2. 513.9
2063	H ₂ N O & 22	O N O N N N N N N N N N N N N N N N N N	1. 5% 2. 461.7

2064	O NH ₂ OH OH & 23.1	ON HN HN OH	1. 21% 2. 514.0
2065	N S NH ₂ NH ₂ OH & 1225	ON SIN NO OH H H	1. 41% 2. 532
2066	N S NH ₂ NH ₂ HO & 1225		1. 48% 2. 546
2067	N S NH ₂ HO & 1226	ON SH3C CH3 NN N CH3 ON N N ON	1. 23% 2. 586
2068	NH2 NH2 NH2 NH2 NH2 NH2	N S N N N N N N N N N N N N N N N N N N	1. 58% 2. 572

2069	S-71	0,0	1. 43%
	N S OH 8 1227	HO H H	2. 518
2070	N S NH ₂ NH ₂ OH & 1228	N S N N N N N N N N N N N N N N N N N N	1. 28% 2. 560
2071	0 N S NH₂ N O OH & 1229	ON SIN NO NO ON NO	1. 69% 2. 518
2072	N S NH ₂ NH ₂ & 23.7		1. 27% 2. 532
2073	N S NH ₂ NH ₂ & 1230	OOO OH H H H	1. 23% 2. 504
2074	N S NH ₂ NH ₂ & 1231	O O O O O O O O O O O O O O O O O O O	1. 44% 2. 518

2075	N S NH ₂	OO	1. 48% 2. 546
2076	& 1225 H ₂ N 0 & 22	ON NON NON NON NON NON NON NON NON NON	1. 48% 2. 502
2077	H ₂ N — 0	ON N N N OH H H	1. 24% 2. 476
2078	H ₂ N — O		1. 22% 2. 448
2079	H ₂ N — CI & 22	ON NON NON NON NON NON NON NON NON NON	1. 28% 2. 482

2080	<u> </u>	O, O	1. 38%
	H_2N		2. 528
		N N	
	T Br	N-OH H H	
	& 22	Y Br	
2081	""	O, O	1. 27%
	H ₂ N	N N	2. 488
	\	, /	
	& 22		
2082			1. 29%
	H ₂ N	N N	2. 486
	>_o	H H	
		N-Q OH " "	
	& 22 \		
2083		NSN \	1. 36% 2. 462
	H_2N		2. 102
	F	HH	
	& 22	OH ————————————————————————————————————	
2084		O, O	1. 40%
	H_2N	N N	2. 498
		, N N	
		N OH H H	
	& 22		

2085	H ₂ N — 6 F F 8 22	ON NON NON NON NON NON NON NON NON NON	1. 67% 2. 510
2086	H ₂ N -	N S N N N N N N N N N N N N N N N N N N	1. 71% 2. 542
2087	H ₂ N F F & 22	N N N F F F	1. 58% 2. 526
2088	H_2N F $8 22$	ON N N H H	1. 70% 2. 476
2089	H ₂ N — O O — & 22		1. 42% 2. 548

2090		0,,,0	1. 47%
	H ₂ N-	N N V	2. 518
		H H	
	Q	OH ()	
		o o	
	& 22		
2091	<u> </u>	O, O	1. 37%
	$H_2N-\langle$	N N	2. 494
	F	$N \longrightarrow N$	
	>= /	N OH H H	
	F & 22	OH F	
2092		O, O	4 2004
2092	<u> </u>	N.S.N.	1. 39% 2. 526
	H ₂ N		
	>_cı	N N	
	CI	N- OH H H (
	& 22	CI	
2093	<u> </u>	O, O	1. 34%
	H ₂ N-	N N	2. 486
		$N \longrightarrow N$	
	> -/	N H H	
	& 22	OH S	
2094		0,_0	1. 48%
	H ₂ N		2. 500
		N H H	
	0-1	N- OH " "	
	& 22		

2095		0,,,0	1. 38%
		N.S.N	2. 530
	H ₂ N—		
		N H H	
		N- OH H H	
	\ <u>\</u>		
	\	0	<u> </u>
	& 22	· ·	
2096	V	0,,0	1. 21%
	$H_2N-\langle$	N. S. N	2. 514
		H H	
		N OH H H	
	/ \		
	& 22		
2097		0, 0	1. 13%
	H ₂ N	Br N'SN	2. 649
	\	Н Н	
	0	SOOH HH	
	& 1232		
2000		0 0	
2098	\\ \mu_{m_i} \ \	NIS.NI	1. 49%
	H_2N	N N V	2. 517
		N N	
	& 1233	H H PO	
	G 1255	SO OH	

2099	H ₂ N - 0 & 1233	ON SIN NO	1. 22% 2. 567
2100	H ₂ N — O & 1233	ON NON NON NON NON NON NON NON NON NON	1. 40% 2. 571
2101	H ₂ N O & 23.8	0 2 2 H 0 2 H 0 2 H 0 Br 2	1. 15% 2. 582
2102	H ₂ N - 0 & 1209		1. 15% 2. 538

2103	H ₂ N O & 23.8	O, O N, S, N Br-N, N OH H H	1. 9% 2. 568
2104	H ₂ N O F F & 22		1. 29% 2. 538
2105	H ₂ N — O & 1234		1. 15% 2. 476
2106	H ₂ N — 0 & 22	ON NON NON NON NON NON NON NON NON NON	1. 11% 2. 476
2107	H ₂ N 0 & 22	N N F N N N H H H	1. 26% 2. 466

2108	F _F H ₂ N	ON ON F F N N N ON F ON THE POSITION OF THE P	1. 25% 2. 470
2109	& 22 H ₂ N	OH ON NO	1. 34% 2. 448
2110	& 22 H ₂ N	NOH H H	1. 24% 2. 480
	& 22		
2111	H ₂ N — O N — 8 22		1. 17% 2. 516
2112	H ₂ N - & 22	0, 0 N N N N N H H	1. 22% 2. 498

While the present invention has been described in conjunction with specific embodiments set forth above, many alternatives, modifications and variations thereof will be apparent to those of ordinary skill in the art. All such alternatives, modifications and variations are intended to fall within the spirit and scope of the present invention.

5